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Thermal conductivities of solid and molten silicates: implications for dynamos in Mercury-like proto-planets

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Abstract

Remanent magnetization and active magnetic fields have been detected for several telluric planetary bodies in the solar system (Earth, Mercury, Moon, Mars) suggesting the presence of core dynamos active at the early stages of the planet formation and variable lifetimes. Among the factors controlling the possibility of core dynamos generation, the dynamics of the surrounding silicate mantle and its associated thermal properties are crucial. The mantle governs the heat evacuation from the core and as a consequence the likeliness of an early thermally driven dynamo. In the case of planets with a thick mantle (associated with supercritical Rayleigh numbers), the core heat is efficiently removed by mantle convection and early thermally-driven dynamos are likely. At the opposite, planets with a thin mantle (associated with subcritical Rayleigh numbers) might evacuate their inner heat by diffusion only, making early thermally-driven dynamos difficult. Within the Solar System, Mercury is a potential example of such a regime. Its small mantle thickness over the planet radius ratio might be inherent to its small orbital semi-axis and hence, might be ubiquitous among the terrestrial objects formed close to their star.

To constrain the likeliness of a thermally driven dynamo on “Mercury-like” planets (i.e. with large R_c/R), we present new thermal diffusivity measurements of various solid, glassy and molten samples. We applied the Angstrom method on cylindrical samples during multi-anvil apparatus experiments at pressures of 2 GPa and temperatures up to 1700 K. Thermal diffusivities and conductivities were estimated for solid and partially molten peridotites, with various melt fractions, and for basaltic and rhyolitic glasses and melts. Our study demonstrates that melts have similar thermal properties despite a broad range of composition investigated. The melts reveal much lower thermal conductivities than the solids with almost an order of magnitude of decrease: 1.70 (± 0.19) to 2.29 (± 0.26) W/m/K against 0.18 (± 0.01) to 0.41 (± 0.03) W/m/K for peridotites at high temperatures and various melts respectively. Partially molten samples lie in between and several predictive laws are proposed as a function of the melt fraction and solid/melt texture.

Using our results into forward calculations of heat fluxes for dynamo generation for Mercury-like planets, we quantify the effect of mantle melting on the occurrence of thermally driven dynamos. The presence of a mushy mantle and partial melting could significantly reduce the ability of the mantle to evacuate the heat from the core and can prevent, shut or affect the presence of a planetary magnetic field. The buoyancy and fate of molten material in such bodies can thus influence the magnetic history of the planet. Future observations of Mercury-like planets accreted near their star and the detections of their magnetic signatures could provide constraints on their inner state and partial melting histories.

Keywords: Thermal diffusivity; thermal conductivity; melts; geodynamo; Mercury;

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Author contributions:

GM, AM, DA and DF set up the MAA apparatus and heating system for Angström measurements. DF and GM conceived the experiments. DF performed, processed and analyzed the experiments. FS significantly contributed to Raman micro-analyses. NC helped for sample selections, analyses and preparations prior experimental runs. JM, DA, GM and DF realized the modelling and discussion section. DF, JM and DA wrote the manuscript. All of the authors have agreed and contributed to the manuscript.

Introduction

The presence of internally generated magnetic fields is a variable feature among telluric bodies in the inner solar system. While few are currently active, such as for Mercury and Earth, several are now extinct, as observed for the Moon, Venus and Mars. On terrestrial planets currently exhibiting a dynamo, the generated magnetic fields characteristics are very different. Earth's magnetic field is very intense 25-65 μT , originating from the core and present for at least 3.5 Ga (Tarduno et al. 2010). Mercury's field strength is much weaker, representing around a 1% of Earth's one (Kabin et al. 2008; Anderson et al., 2011) and its shape is also unique among the different detection in the solar system (Tian et al., 2015). According to the remnant magnetization measured in the crust, it was proposed that such weak magnetic activity occurred during the last 3.9 Ga (Johnson et al. 2015). The source of such a weak-and-prolonged dynamo is still largely debated (Manthilake et al. 2019). In the meantime, there are several evidences for an intrinsic dynamo during the early stages of both Mars and Moon (Acuña et al., 1999; Hood et al., 2010). Their dynamo seems to cease around 4.1-3.9 Ga ago for Mars (Johnson and Phillips, 2005; Lillis et al., 2008, Lillis et al., 2013) and exhibits a somewhat complicated history for the Moon with a strong dynamo between 4.25 and 3.5 Ga, followed by a weak persistence up to 2.5-2 Ga ago (Tikoo et al. 2014, Lawrence et al. 2008, Garrick-Bethell et al., 2009, Mighani et al. 2020). For Venus, the early presence of a dynamo remains yet unconstrained but may be detectable in future explorations (Nimmo et al. 2002; O'Rourke et al. 2019). More broadly, evidences of paleomagnetic anomalies indicate that the angrite parent bodies, originating from inner regions of the solar system, were subject to an early internally generated dynamo (Weiss et al. 2008). All these elements suggest that transient dynamos might be a somewhat common feature in telluric bodies (Monteux et al. 2011).

The presence of early dynamos is highly conditioned by the internal structure of the planet and its capacity to release the heat accumulated during the accretion processes (accretion, metal/silicate differentiation, core and mantle crystallization) as well as short-lived radiogenic heating. As heat conduction is an inefficient heat transport process in silicates at high temperatures (Hofmeister and Branlund, 2015), the onset of a mantle global convection is a crucial step in planet's thermal history. Convection starts when the Rayleigh number (Ra) of the terrestrial mantle is larger than the critical Rayleigh number (Ra_c). The higher is Ra, the stronger the convection, and the more efficient the heat transport. In contrast, heat is only transported by conduction for a terrestrial mantle with a $Ra < Ra_c$. As Ra scales with mantle thickness (h_{mantle}) as h_{mantle}^3 , convection should take place easily in planets with a thick silicate mantle, even after the solidification of the early magma ocean stage. Consequences are efficient evacuation of the inner heat and the possible occurrence of a dynamo. At the opposite, bodies with a thinner mantle lead to smaller Ra values, making convection unlikely and dynamos more difficult to generate.

Mercury is the most interesting planet for our study. Indeed, its mantle is thin 420 ± 30 km (Hauck et al. 2013) and its core occupies almost 55% of planet's volume and 65% of the planet mass (Strom and Spague, 2003; Charlier and Namur, 2019). Hence, the mantle of Mercury is controversially at the limit between conductive and convective regimes (Breuer et al. 2007). Different scenarios could

101 explain the small $h_{\text{mantle}}/\text{planet radius (R)}$ ratio on Mercury (Charlier and Namur, 2019): (1) primordial
102 nebular processes, yielding to the enrichment of metal over silicate materials in the inner solar system
103 (Ebel and Grossman, 2000; Wurm et al. 2013; Weidenschilling 1978), (2) highly energetic accretional
104 collisions inducing a major loss of the silicate fraction (Benz et al. 1988, Asphaug and Reufer, 2014),
105 and (3) post-accretion scenarios, with major vaporization of the volatile and silicate elements from
106 the planet during magma ocean stage (Fegley and Cameron 1987; Boujibar et al., 2015). If scenarios
107 (1) and (3) are dominant, then "Mercury-like" planets with small h_{mantle}/R ratio would be ubiquitous
108 within all planetary systems. Moreover, such a small ratio would be prevailing during the whole
109 accretionary processes. Mercury-like bodies could adopt a wide range of possible compositions
110 depending on their history. For example, bodies accreted from reduced enstatite and/or carbonaceous
111 bencubbinite chondrites (Malavergne et al. 2010) could present a mantle composition similar to
112 terrestrial Iherzolite, but with a sulfur content potentially as high as 11 wt.% (Namur et al. 2016).
113 Accordingly, the diversity of the silicate samples found on Earth in the forms of rocks, melts and
114 glasses is a good proxy to decipher the properties of a range of mantle-relevant silicate compositions
115 on Mercury-like bodies. Depending on the planet size and the h_{mantle}/R ratio, the internal pressure
116 ranges from a few MPa to several GPa. We note that extensive mantle melting likely occurred at
117 different stages of the history of such Mercury-like bodies. Major energy incomes are expected from
118 the vicinity to the young Sun (2500 to 3500 K according Charlier and Namur, 2019), internal energy
119 release (chemical and gravitational differentiation, core crystallization), and presence of short-period
120 radioactive elements (Al^{26} , K^{40} etc.). The very high temperatures likely induce extensive melting of
121 the thin mantle up to the core-mantle boundary (CMB).

122 As mentioned above, dynamo generation could be difficult for Mercury-like planets, due to the
123 subcritical value of Ra possibly disabling mantle convection. In such case, heat transfer by conduction
124 would dominate the planet history and the thermal conductivity of the silicate mantle is a key
125 parameter governing the early core heat flow. Silicates thermal conduction properties are now well
126 characterized at ambient conditions (Hofmeister and Branlund, 2015). Among them, the most
127 common geological minerals were characterized in the forms of single crystal and polycrystalline
128 aggregates: olivine (Osako et al. 2004; Xu et al. 2004; Pertermann and Hofmeister 2006; Gibert et al.
129 2005), periclase (Hofmeister and Branlund, 2015), feldspar (Pertermann et al. 2008; Hofmeister et al.
130 2009; Branlund and Hofmeister, 2012) and pyroxenes (Hofmeister 2012, Hofmeister and Pertermann
131 2008) as well as peridotite rocks (Gibert et al. 2005; Beck 1978), which have been extensively studied
132 due to their important geological implications. The thermal diffusivities of minerals are linked to the
133 characteristics of their lattice structure and modes of phonon generation and propagation as a function
134 of temperature. For silicates, lattice thermal diffusivities usually decrease with increasing temperature
135 following a $1/T$ dependence. At the opposite, the diffusivities increase while increasing pressure,
136 however, the temperature dependence is much greater than that of pressure over the considered ranges
137 for small planets. The resulting implication is that mantle rocks and minerals are poor thermal
138 conductors in planetary interiors. Recent measurements of thermal diffusivities of glasses and melts
139 at ambient pressure suggested that the non-crystalline silicates are even more insulating than the
140 minerals (Hofmeister et al. 2009, 2014, Romine et al. 2012).

141 Measurements of thermal diffusivities at relevant conditions of planetary mantles encompass
142 important difficulties, which were overcome by the use of different techniques (see Hofmeister and
143 Brandlund 2015 for a critical review). Measurements were reported over a wide P and T range for
144 large volume samples of olivine, periclase, bridgmanite using Angström or Pulse method in solid
145 pressure apparatus (Osako et al. 2004; Xu et al. 2004; Manthilake et al. 2011a and 2011b, Zhang et
146 al. 2019). Up to now, accurate measurements of geologically relevant silicate glass, partially molten
147 systems and melts at planetary interior conditions remain scarce if not absent. The available results
148 report a nearly flat evolution of thermal diffusivities with the temperature above 1000 K for silicates,
149 glass and melts (Hofmeister et al. 2014; Hofmeister and Banlund, 2015), suggesting that the

150 measurements above the melting temperature could be safely extrapolated to planetary P-T
151 conditions.

152 In this study, we aim at better constraining the thermal properties of Mercury-like protoplanets
153 where convection is unlikely and heat is mostly removed by diffusion. We perform HP-HT *in situ*
154 thermal diffusivities measurements of solid, partially molten and fully molten silicate for various
155 compositions in Multi-anvil apparatus and using Angström method. Then, we constrain the likeliness
156 of a thermally driven dynamo during the early stages of the evolution of a Mercury-like planet. We
157 consider a wide range of planet sizes ($1 \text{ km} < R < R_{\text{Mercury}}$) and thermal states (with solid and partially
158 molten mantles) on the dynamo likeliness.

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161 **Experimental and analytical methods**

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163 High-pressure assemblies:

164 High-pressure and high-temperature experiments were performed using a 1500-ton Kawai type
165 Multi-anvil apparatus. All experiments were conducted at 2 GPa, based on a previous press-load vs
166 sample pressure calibration (Boujibar et al. 2014), providing an uncertainty of ~ 0.1 GPa at pressures
167 < 5 GPa. We used octahedral pressure media with full length edges composed of MgO doped with
168 Cr_2O_3 (5 wt. %) in a 25/17 multi-anvil configuration (octahedron edge length / anvil truncation edge
169 length) (Figure 1). Our assembly was designed to accommodate the specific requirements for
170 measurements of thermal conductivity of relatively large samples (4 mm long for 4-3.5 mm diameter).
171 All ceramic parts of the cell assembly, including the pressure media, were fired at 1373 K prior
172 assembling in order to remove the absorbed moisture. Oxygen fugacity of the sample was not
173 controlled during the experiments but is expected to be quite reducing due to the presence of the
174 graphite furnace. The use of a steeped graphite furnace helped reducing thermal gradients. Thermal
175 loss from the sample zone was further reduced by the use of a thick zirconia (ZrO_2) sleeve around the
176 furnace. Thermal gradients in our assembly were computed using the software developed by
177 Hernlund et al. 2006. The models show temperature gradients within the sample volume limited to
178 ~ 7.5 K/mm vertically and even less radially/horizontally (Figure S1). On the other hand, uncertainties
179 of our thermocouple reading (i.e. where the conductivity measurement is performed, see below) are
180 less than 5 K on absolute and 0.1 K in relative temperatures.

181 Previous experimental studies described the difficulties to perform a good measurement of
182 thermal diffusivity for a molten sample, due to a potential sample deformation (Hofmeister et al.
183 2009, 2014; Romine et al. 2012). However, deformation remained minor in our experiments, as
184 evidenced by the good reproducibility of the measurements during the repeated cycles of heating and
185 cooling (Figures 1 and 2) as well as the shape of the recovered samples (Figures S6, S7 and S8).

186 Two tungsten-rhenium (W_{95}Re_5 - $\text{W}_{74}\text{Re}_{26}$) thermocouples of 75 μm of diameter were used to
187 measure temperature oscillations at the center (in 300 and 600 μm drilled holes in solids and glasses,
188 respectively) and at the edge of the cylindrical sample. Special care was given to ensure that the
189 junction points of the two thermocouples were located in the same sample plane perpendicular to the
190 cylinder axis. Measurements of the thermal conductivity of melts and glasses have always been
191 particularly challenging due to the risk of thermocouple short circuit, because melts, even dry, are
192 good electrical conductors (Tyburczy and Waff, 1983; Ni et al. 2011). To prevent this effect,
193 thermocouples were inserted in alumina tubes of 0.6 mm diameter and 170 μm wall thickness. While
194 these tubes show good resistance to cold compression and almost no reaction with the samples, even
195 at very high temperature, some leakages for low viscosity basaltic melts have been identified. Their
196 occurrence was taken in account for thermal diffusivity estimations (see Supplementary Text S1 and
197 Figures S2 and S3).

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200 Angström method for thermal diffusivity measurements

201 We aim at determining thermal diffusivities of geological samples such as peridotites, mafic and
202 felsic glasses and their melts at high pressure and high temperature using a double contact method:
203 the Angström method. Experimental configurations, assemblies and data treatment are similar to
204 several previous studies (Fujisawa et al. 1968; Kanamori et al. 1969; Katsura 1993; Xu et al. 2004;
205 Manthilake et al. 2011a and 2011b). Briefly, a temperature wave is generated radially in a cylindrical
206 sample by the surrounding heater sleeve of the multi-anvil assembly (Figure 1). Oscillations are
207 generated with controlled frequency and period by modulating the power supply. Periodic
208 temperature signals are recorded by two thermocouples fixed in the center and on the edge of the
209 sample cylinder. At frequencies higher than 1.5 Hz, the signal can get noisier due to a limited time
210 resolution of the recording system.

211 The recorded signals, for each thermocouple channel, are fitted by a nonlinear least square solver
212 (lsqcurvefit on Matlab© using Levenberg-Marquardt algorithm):

$$213 \quad \Delta T = A_0 + A_1 t + A_2 \sin\left(2\pi A_3 t + \frac{A_4 \pi}{180}\right) \quad (1)$$

214 Following this method, we obtained the amplitude of the temperature variation (A_2), frequency
215 (A_3), and phase (A_4) of the recorded wave as a function of time (t), for the two thermocouples. Errors
216 are quantified based on the residue on the non-linear curve fitting (the *nlparci* function in Matlab©).
217 To infer thermal diffusivity or conductivity from these parameters, the equation of conductive heat
218 transport has to be solved. Here we consider the sample as an infinite uniform cylinder and assume
219 that the heat flow is negligible in the vertical direction, thanks to a relatively long cylindrical heater.
220 The following equation, expressed in cylindrical coordinates by Carslaw and Jaeger (1959), has to
221 be inverted in order to retrieve the diffusivity of the sample:

$$222 \quad \frac{dT}{dt} = D \left(\frac{d^2 T}{dr^2} + \frac{1}{r} \frac{dT}{dr} \right) \quad (2)$$

223 where r is the radial distance from the axis, T the temperature, t the time, and D the thermal diffusivity.
224 The boundary condition of our setup is:

$$225 \quad \frac{dT}{dr} = 0, \text{ at } r = 0 \quad (3)$$

226 If we consider harmonic excitation at a distance $r = R$ from the axis in normal or complex form:

$$227 \quad T_R = B_0 + B_1 \cos wt \leftrightarrow T_R = b_0 + b_1 \operatorname{Re}(\exp iwt) \quad (4)$$

228 where B_0 , B_1 , b_0 and b_1 are constants and Re is the real part of the exponential. The solution of the
229 radial flow equation with the boundary conditions developed above can be expressed as:

$$230 \quad T_R = b_0 + b_1 \left[J_0(\sqrt{-i} * x) * \frac{\exp(iwt)}{J_0(\sqrt{-i} * l)} \right] \quad (5)$$

231 where:

$$232 \quad l = (w/\kappa)^{1/2} R \text{ and } x = (w/\kappa)^{1/2} r \quad (6)$$

233 for $0 \leq r \leq R$, where w is the angular frequency and J_0 is the Bessel function of the first kind (integer
234 order $n = 0$). At $r = 0$ and $r = R$ we have:

$$235 \quad T_0 = b_0 + b_1 \theta \cos(wt - \varphi) \quad (7)$$

$$236 \quad T_R = b_0 + b_1 \cos wt \quad (8)$$

237 where:

$$238 \quad \theta = \frac{1}{\sqrt{\operatorname{bei}(u)^2 + \operatorname{ber}(u)^2}} \quad (9)$$

$$239 \quad \varphi = \tan^{-1} \left(\frac{\operatorname{bei}(u)}{\operatorname{ber}(u)} \right) \quad (10)$$

240 where θ is the amplitude ratio and φ the phase shift between the two harmonic temperature
241 measurements and bei and ber are imaginary and real parts of the Bessel function of the first kind,
242 respectively. The solution to the eq. (4) can be written using the dimensionless argument u , from

243 which thermal diffusivity (D) can be directly estimated knowing angular frequency (ω) and sample
244 radius (d).

$$245 \quad u = d(\omega/D)^{1/2} \quad (11)$$

246 Thanks to the Eqs.9, 10, and 11, the diffusivity was then retrieved via forward Monte Carlo
247 simulation and neighborhood algorithm (Sambridge, 2002). In this step, different values of
248 diffusivities are generated and theoretical phase shifts and amplitude ratios are calculated. These
249 values are compared to the values measured in our experiments. When the differences between the
250 computed solution and the experimental determination tend to 0 (minimization step), the correct
251 diffusivity is then obtained if the sample radius (d) and angular frequency (ω) are known. We note
252 that mathematical solutions appear every 360° for the phase shift. Such erroneous solutions are
253 checked manually and discarded.

254 For most of our experiments, we observe a significant variation of the refined raw-diffusivity
255 value as a function of the heat-source frequency. This effect was already reported in the literature.
256 Different equations were proposed to refine a real value of diffusivity, corresponding to the infinite
257 frequency asymptote, based on non-linear equations. While Manthilake et al. (2011b) used:

$$258 \quad D = D_\infty + A_0 \exp(A_1 * f) \quad (12)$$

259 Xu et al. (2004) used:

$$260 \quad D = D_\infty + A_0 \exp\left(\frac{-f}{f_0}\right) \quad (13)$$

261 where D is diffusivity, f source frequency, f_0 asymptote frequency and A_0 and A_1 constants. In these
262 equations, D_∞ , A_n and f_0 are inverted parameters. For a better fit of our experimental data and to
263 minimize the uncertainties on the parameters, we adopt Eq. 13. On the other hand, for experiments
264 presenting no systematic dependence of the raw-diffusivity with frequency, we consider as real value
265 the average value between the raw-diffusivity values measured at all frequencies (Xu et al. 2004;
266 Manthilake et al. 2011b).

267 Experimental uncertainties

269 Experimental measurements of thermal diffusivities and their further transformation into
270 thermal conductivities generate uncertainties originating from the estimations of pressure,
271 temperature, sample dimensions and the data fitting itself. Experimental uncertainties on pressure and
272 temperature are presented above (~ 0.1 GPa and ~ 5 K, respectively). Sample lengths prior to the
273 sample loading (d_0) and after the melting experiments were determined with a high precision digital
274 gauge (accuracy of $\sim 1 \mu\text{m}$) and using the Scanning Electron Microscope (FEG-SEM), respectively
275 (see Table 1). Then, the sample radius during the experiment at high pressure and temperature was
276 calculated using:

$$277 \quad d(P, T) = d_0 * \left(1 - \alpha(T - 298) + \frac{P}{K}\right)^{-1/3} \quad (14)$$

278 where α is the thermal expansion and K the bulk modulus of the sample. For these calculations only
279 samples radii before experiments were considered, as post mortem measurements are affected by
280 decompression cracks (highlighted by the larger values measured after the experiments in Table 1).
281 Moreover, the distance between the two thermocouples could not be measured precisely for a few
282 samples. The values of all experimental parameters are provided in Supplementary Materials (Text
283 S4 and Figure S4). Altogether, final uncertainty on the sample length is between 5 and 10 μm .

284 There are other uncertainties associated with the procedure of data fitting for the determination
285 of raw thermal diffusivities. Experimental phase shifts and amplitude ratios (Eqs. 9 and 10) are
286 determined with a precision generally better than 1% and majored by 3% in the worst cases. In the
287 course of the Monte Carlo simulation, the differences between experimental and theoretical
288 diffusivities are recorded and used a posteriori to refine uncertainties within 1σ errors.

289 Then, real thermal diffusivities are refined from the raw-diffusivities using either (i) the
290 asymptotic non-linear fit for experiments presenting a dependence in frequency (Eq. 13, which yields

291 important uncertainties on the refined parameters) or by averaging (see Methods). The error on
292 average (σ_{AVG}) is:

293
$$\sigma_{AVG} = \sqrt{\left(\frac{1}{n}\right)^2 * \sum_{1:n} \sigma_D^2} \quad (15)$$

294 where n is the number of diffusivity measurements performed at different frequencies and σ_D is the
295 error on each diffusivity. Raw diffusivities recovered from Monte Carlo processing have errors of
296 ~1%, similar to those of phase shift and amplitude ratio. If the fitting step is realized with Eq. 13,
297 errors of raw data are used as weights in the inversion. The standard deviations are usually between
298 1 and 10% of the real asymptotic diffusivity. If the averaging method is selected, the standard
299 deviation, estimated via Eq. 15, is usually about 1% of the final value.

300 A final source of uncertainties come from other technical issues and apparatus reproducibility,
301 which are inherent to such challenging experiments. We considered that final error must be majored
302 by 5 % of the value. The relative uncertainties become even higher after conversion into thermal
303 conductivities due to uncertainties and simplifications on the sample density and heat capacity at high
304 pressure and temperature (see Supplementary Text S4).

305

306 Experimental procedure:

307 We performed a suite of three heating and cooling cycles to provide an important number of data
308 and maximize the quality of recovered data. A first cycle was run up to 500 °C for moisture removal
309 by 50°C steps (Figure. 2). The second cycle was run up to glass transition temperature (T_g) by 100°C
310 steps. T_g varies according to rock composition and its water content (Giordano et al., 2005). For our
311 dry samples, the temperature of 800°C happened to be above T_g for all our compositions. The
312 temperature was then decreased down to 150°C by 100°C steps. The quality of the measurement is
313 usually better during the cooling cycle once (1) the sample has thermally equilibrated with the
314 assembly, (2) moisture has been removed, and (3) a better contact was achieved between the sample
315 and the thermocouples due to local flow in the solid-state. The redistribution of matter under high
316 temperature cancels most of the potential artifacts associated with the presence of pores/voids
317 between the sample and the thermocouples (Hofmeister et al. 2009). Hence, data recorded during the
318 last cycles were considered for final values.

319 In the last cycle, we heated the sample up to its melting temperature by 100°C steps or 50°C near
320 the melting point. The sample was maintained above its melting point for less than 1 hour, to avoid
321 sample leakage and chemical reaction with the surrounding parts. The sample was then quenched to
322 600-550°C. This step produces a glassy sample from which measurements at low temperature are
323 performed.

324 Temperature oscillations of the two thermocouples were measured at each temperature step for
325 both heating and cooling cycles, at 12 different oscillating frequencies between 0.1 and 1.5 Hz. The
326 measurements were performed after at least 2 min of thermal equilibration to reach a stable regime
327 (the smaller is the frequency, the longer is the required equilibration time). Then, the recording
328 duration was 1 to 2 minutes or at least 10 oscillating periods. The measurement at the 12 frequencies
329 took between 20 to 30 minutes. Thus, the duration of each experiment was more than 12 hours.

330

331 Chemical and textural analyses:

332 Recovered samples were cut parallel and perpendicular to the cylindrical furnace and each
333 section was polished with great care. We could observe the position of the two thermocouples
334 junctions, measure the sample radius and perform the textural and chemical analyses (see
335 Supplementary Figures S6, S7, and S8).

336 Micro-textures were observed with a Scanning Electron Microscope (SEM) JEOL Jeol JSM-
337 5910 LV using an accelerating voltage of 15 kV and a working distance of 11.4 mm. The 2D phase
338 proportions of our partially molten peridotite samples were obtained from the analyses of qualitative

339 chemical maps obtained by energy-dispersive X-ray spectroscopy (EDX) in the SEM. The images
340 were binarized and phases were individually separated allowing textural analyses with the FOAMS
341 software (Shea et al., 2010). A more detailed description is given in Freitas et al. (2019). On the other
342 hand, quantitative chemical analyses were performed on both our starting materials and experiment
343 products using the electron probe micro analyzer (EPMA). Chemical and textural analyses of starting
344 materials are reported in the Supplementary Text S2, Figures S6 to S13 and Tables S1, S2, S3 and
345 S4. Analyses of recovered runs are detailed in Supplementary Text S3, supplementary Figures S6 to
346 S7 and Tables S1, S2, S3 and S4.

347 Water contents were estimated using the ICP-AES for the peridotite starting materials and via
348 Raman spectroscopy for the recovered samples. Raman spectra were collected with a Renishaw InVia
349 confocal Raman micro spectrometer, equipped with a 532 nm diode laser and a Leica DM 2500M
350 optical microscope. Measurements were carried out using a 2400 grooves/mm grating, a 100×
351 microscope objective, a slit aperture set to either 20 μm or 65 μm and a laser power of 8 mW for
352 glasses and 16 or 75 mW for olivine. The resulting lateral and axial resolutions were of ~1 and 3 μm,
353 respectively, and the spectral resolution was better than 1 cm⁻¹. Daily calibration of the spectrometer
354 was performed based on the 520.5 cm⁻¹ peak of Si. Spectra were recorded from ~100 to 1300 cm⁻¹
355 (alumino-silicate network domain) and from ~3000 to 3800 cm⁻¹ (water domain), with variable
356 acquisition times ranging between 5 and 120 s for silicate bands and 120 and 240 s for water domain
357 depending on the water content (Figures S11 to S17). For water quantification in olivine and glass,
358 we followed the procedures reported by Bolfan-Casanova et al. (2014) and Schiavi et al. (2018). We
359 used both (1) the external calibration procedure, which is based on a set of hydrous olivine standards
360 from (Bolfan-Casanova et al. 2014) and different types of silicate glasses ranging from basaltic to
361 rhyolitic compositions (Schiavi et al., 2018; Médard and Grove, 2008), and (2) the internal calibration
362 procedure, based on the correlation between the water concentration in olivine or glass and the relative
363 areas of the water and silicate Raman bands (OH/Si integrated intensity ratio). The discrepancy
364 between the two methods is small. Water contents in the standard materials were previously
365 determined using the FTIR technique.

366

367 **Results**

368

369 We performed a total of 11 thermal diffusivity experiments on various chemical compositions,
370 using the Angström method (details in Table 1). In this section, we first describe phase shifts and
371 amplitude ratios between the two thermocouples and their conversion into thermal diffusivities and
372 conductivities. We detail the post-mortem chemical and textural analyses in the Supplementary Text
373 S3, Tables S1 to S4 and Figures S6 to S13.

374

375 Phase shift, amplitude ratio and the refined raw-thermal diffusivities

376 Signals recorded during the experiments are sinusoidal oscillations with varying frequencies. The
377 thermocouple located at the sample center (labeled TC1 in Figure 3) presents a phase delay and a
378 smaller amplitude compared to the thermocouple located at the sample edge (labeled TC2 in Figure
379 3). Typical raw signals and their fits are presented in Figure 3, while the refined (absolute) phase
380 shifts and amplitude ratio are reported in Figure 4. Both the magnitude of phase shift and amplitude
381 ratio change significantly with the type of sample and the experimental conditions, including the
382 source frequency. At a constant frequency, the phase shift increases with increasing temperature. At
383 a constant temperature, the phase shift increases with increasing the excitation frequency. The
384 amplitude ratio is decreasing with increasing frequency and temperature.

385 Globally, the refined thermal diffusivities present a comparable evolution of temperature at all
386 signal frequencies (see an example in Figure 5). When the frequency dependence is larger than the
387 experimental uncertainty (Figure 6), we use Eq. 13 to refine the true asymptotic value of the thermal
388 conductivity. Alternatively, when the temperature dependence is below the experimental uncertainty

389 or when no clear frequency trends is visible, we average the different raw-diffusivity values (see
390 Methods). In the wide majority of the cases, diffusivity values inferred from phase shifts appear to be
391 more robust and with a lesser degree of uncertainty, compared to values inferred from difference of
392 amplitude, in agreement with previous studies (Kanamori et al. 1969 and Xu et al. 2004). Hence,
393 despite similar values obtained with the two methods, values refined from phase shift were preferred.
394

395 *Results for peridotite*

396 For peridotite, our two successful experiments present a smooth evolution with temperature,
397 yielding diffusivities values decreasing from $1.7(\pm 0.1)e^{-6}$ to $7.5(\pm 0.4)e^{-7}$ m²/s for sample M804 and
398 $1.5(\pm 0.8)e^{-6}$ to $5.5e^{-7}$ m²/s for sample M847 (Figure 6a), with an uncertainty of about 5 to 10% for
399 each sample. We attribute the relative discrepancy to a difference of sample mineralogy between the
400 two samples, due to the coarse grain size of the core drilled peridotite (see Supplementary Texts S2
401 and S3 and Figures S8 and S9). Olivine has a higher intrinsic thermal diffusivity than the other phases
402 present in peridotite such as pyroxenes and spinel (Hofmeister and Branlund, 2015). M804 may
403 contain more olivine, inducing higher diffusivities than M847. In this respect, the results for two
404 samples are thus compatible with each other and the differences are representative of the variability
405 of thermal diffusivity that can be expected among the compositional variability of peridotites (from
406 lherzolites to dunites).

407 Thermal diffusivities of our peridotite samples scale as a function of $\sim 1/T$, as expected from the
408 standard equation provided by Hofmeister and Branlund, 2015:

$$409 D_{lat} = a * T^{-b} + c * T \quad (16)$$

410 where T is the temperature (K), and a , b and c are adjustable parameters (Table 2). The quality of the
411 fit is excellent up to a temperature of ~ 1300 K corresponding to the onset of peridotite melting. We
412 therefore exclude the data points above 1300 K to model the thermal properties of solid peridotite.
413

414 *Results for basalts*

415 For our basaltic samples, the measured diffusivities plot over a broad range of values, from
416 $1.0(\pm 0.1)e^{-6}$ to $3.0(\pm 0.2)e^{-7}$ m²/s (Figure 6b), with an uncertainty of $\sim 5\%$ for each experiment. The
417 discrepancy is particularly important at low temperatures. Also, a same sample yields thermal
418 diffusivity values significantly different along the different cycles of the experimental procedure
419 (between C2, H3 and the final quench, see Figure S5). Samples with the highest diffusivities at low
420 temperatures present a rapid decrease of diffusivity with increasing temperature. On the other hand,
421 samples with the lowest diffusivities show very small temperature dependence. It yields to a
422 convergence of all diffusivity measurements at ~ 1000 K. Based on a study of rhyolitic glasses,
423 Romine et al. (2012) reported a moderate temperature dependence, similar to our samples presenting
424 a low diffusivity, and an increase of 0.0192 mm²/s of the glass diffusivity per percent of crystallinity.
425 For our starting materials with less than 5 vol.% of crystals (see Supplementary Text S2 and S3,
426 Figure S7), the effect of microlites could account for ~ 0.1 mm²/s of variation in our diffusivity values,
427 which correspond to less than $\sim 10\%$ of observed differences. However, the presence of 35 vol.% of
428 crystals in the recovered sample M807 would explain not only its high thermal diffusivity at low
429 temperature, but also its strong temperature dependence that is typical of crystals (see Beck et al.,
430 1978; Romine et al. 2012; Hofmeister and Branlund 2015 and our peridotite trends in Figure 6a). The
431 recrystallization of M807 is not surprising, since the second cycle of annealing was performed above
432 its T_g (Figures 2 and S5). The crystallinity of other samples depends on the applied cycles of
433 annealing at a temperature eventually close to their T_g . Nonetheless, diffusivities of all samples
434 converge at increasing temperatures, because the conductivity of crystals is not much greater than
435 that of the glass at high temperature, especially if microlites are low diffusivity silicates such as
436 pyroxenes, plagioclase or spinels (see Hofmeister and Branlund 2015 for mineral diffusivity
437 compilations). Over the 5 basaltic samples investigated in this study, the diffusivity trends indicate
438 either significant recrystallization of M807 (DR07-MORB) and M662 (EPR-MORB), or negligible

439 crystallization of M844 and M846 (DR11-MORB) and M836 (synthetic haplobasalt). The sample
440 crystallization is likely to evolve during the thermal diffusivity measurements in step H3 (up to
441 melting point) of the experiments performed at a temperature significantly above the T_g. For this
442 reason, the crystallinity determined on the recovered samples is only a qualitative measurement of
443 the sample properties at high temperatures.
444

445 *Results for rhyolites*

446 Measured diffusivities of rhyolite samples also plot over a broad range of values from $1.4(\pm 0.8)e^{-6}$
447 to $4.0(\pm 0.2)e^{-7}$ m²/s, with an uncertainty of about 5% for each experiment (Figure 6c). The
448 discrepancy appears similar than for the basalt samples, as the sample presenting higher diffusivities
449 also show a major temperature dependence at low temperature. The presence of less than 2 vol.% of
450 crystals in the starting material could account for a diffusivity increase of 0.038 mm²/s at maximum
451 (Romine et al., 2012). Still, the diffusivity trends suggest a major recrystallization at high-temperature
452 for M808 (Güney Dag) and M843 (ATHO), some crystallization for M843 (ATHO), and negligible
453 crystallization for M850 (Güney Dag).
454

455 *Properties of melts and partially molten samples*

456 In addition to the evolutions described above, a strong decrease of the thermal diffusivity is
457 observed for most of our samples at the highest temperatures (Figure 6). The decrease occurs at
458 temperatures around 1300 K for peridotites, 1200 K for basalts and >1050 K for rhyolites. Such
459 temperatures are in agreement with the melting or glass transition temperatures, depending if the
460 sample is a peridotite or a glass, recrystallized or not. Similar changes were already reported at
461 temperatures above the glass transition (Hofmeister 2009, 2014, Romine et al. 2012).
462

463 The amplitude of the decrease is 45-50% in peridotites, which recovered samples present a degree
464 of partial melting (F) up to 23%, 35-70% for basalts and <30% for rhyolites samples. The more
465 pronounced decrease in molten peridotites is probably due to a more contrasted change of the local
466 structure at the melting point, between the minerals and the melt (see discussion). On the other hand,
467 the minor change of diffusivity for rhyolites at T_g could be related to their high SiO₂-content, which
468 preserves a polymerized structure in the melt above the glass transition.

469 Thermal conductivities

470 Thermal conductivities (κ) can then be computed from thermal diffusivities following:

$$471 \quad \kappa(P, T) = D(P, T) * \rho(P, T) * C_p(P, T) \quad (17)$$

472 where $\rho(P, T)$ and $C_p(P, T)$ are the sample density and heat capacity, respectively. For our calculations,
473 we considered ρ and C_p values in standard conditions when the P and/or T dependences were not
474 available in the literature (see Supplementary Text S4 and Figure S4).
475

476 Conductivities calculated for peridotites evolve from 4.32 (± 0.48) to 1.70 (± 0.19) W/m/K with
477 increasing the temperature up to the melting point. Values for basalts range from 1.7 ($\pm 8.0e^{-2}$) to 0.5
478 ($\pm 4.0e^{-2}$) W/m/K, whereas those for rhyolite lie between 1.1 ($\pm 9.0e^{-2}$) and 0.3 ($\pm 3.0e^{-2}$) W/m/K for
479 low to high temperatures, respectively. At the melting temperature, our partially molten peridotites
480 display values of 1.19 (± 0.16) to 0.93 (± 0.10) W/m/K, whereas melts present relatively lower values
481 of 0.34 (± 0.2) to 0.18 (± 0.1) W/m/K for basaltic and 0.41 (± 0.3) to 0.31 (± 0.3) for rhyolitic
482 compositions (See Figure 7 and Table 3). We note that the important difference in conductivity
483 between basalts and rhyolites (Figure 7) is predominantly due to differences between their heat
484 capacities and densities, while their thermal diffusivities are found similar (Figure 6).
485

485 **Interpretation of results**

486 General Temperature dependence

488 For all compositions investigated in this study, thermal diffusivities decrease with increasing
489 temperature until reaching a plateau at temperatures between 700 and 1000 K. Based on experiments
490 performed at room pressure, it was observed that the plateau occurs at about the Debye temperature
491 of the mantle minerals (Hofmeister et al. 2009, 2014). For this reason, it was proposed that thermal
492 diffusivities vary largely with temperature until the complete activation of the vibration modes
493 (phonons in minerals). The temperature range observed in our study for the occurrence of a plateau
494 is fully compatible with this interpretation.

495 For basaltic glasses, a comparable but more moderate decrease of thermal diffusivity was
496 reported up to a saturation temperature corresponding well to the glass T_g (Hofmeister et al. 2009,
497 2014, Romine et al. 2012). In our experiments, the decrease is of ~30% to 60% over the investigated
498 temperature range, depending on the experiment. Such amplitude is compatible with the ~40%
499 decrease observed during the heating of pyroxene glasses (Hofmeister et al. 2009).

500 For rhyolite samples, the thermal conductivity increases slightly with increasing the temperature
501 (Figure 7). This is due to the heat capacity that increases more with temperature than the density
502 increases and diffusivity decreases (Eq. 17). The increase is, however, smaller than reported in
503 Romine et al. (2012), due to the use of a different C_p (Neuville et al. 1993) (see Supplementary Text
504 S4) and a stronger temperature dependence of thermal diffusivities observed in our experiments
505 because of different crystallizations states.

506

507 Effect of radiative conduction

508 Romine et al. (2012) reported an increase in thermal diffusivities of the melts at very high
509 temperatures at ambient pressure. They attributed this feature to an increased role of the radiative
510 component. Such a component can dominate the thermal diffusivity for a sample transparent to the
511 infrared and visible photons at high temperatures. For thin samples, this effect can become
512 problematic if the mean free path of photons is longer than the sample length (ballistic photons, see
513 Hofmeister and Branlund, 2015). No significant increase in thermal diffusivity and conductivity is
514 observed in our high-pressure experiments, except maybe for the rhyolitic samples (Figures 6 and 7).
515 The difference with the previous work is most probably related to the opacity of our basalts and
516 peridotites samples, hence limiting the radiative transfers.

517

518 Effect of glass/melt composition

519 Overall, our conductivity values are compatible with the values available in the literature (see
520 Figure 8). Differences in absolute values are nonetheless present. Rhyolite melts (0.31-0.41 W/m/K)
521 are found slightly more conductive than basaltic ones (0.18-0.34 W/m/K). Within the same family of
522 glass, thermal conductivity varies by 0.10 to 0.15 W/m/K for rhyolitic and basaltic melts, respectively.
523 This is slightly larger (of at least 10%) than the experimental uncertainty, an effect possibly due to
524 larger uncertainties on the dimensions of the molten sample. No clear trend can be retrieved from the
525 comparison between our different basaltic or rhyolitic compositions. Among the major elements, iron
526 could be of major importance, due to its critical impact on glass/melt density. Indeed, the thermal
527 diffusivity of glasses was reported to decrease with increasing density (Hofmeister 2014). The
528 comparison between our rhyolites and basalts is coherent with such a trend. However, our haplobasalt
529 (M836) presents diffusivity values comparable with natural basalts (M662, M807, M844), as well as
530 Fe-bearing (M843 and M848) and Fe-free (M808 and M850) rhyolites (Figure 6), despite a variation
531 of the Fe-content from 0 to 10 wt.% in these different samples. Other elements could also impact the
532 melt thermal diffusivity, in particular Si and Al, which favor polymerization of the liquid (and alkali
533 elements for the opposite effect) (Ni et al. 2015). Still, within the experimental uncertainty, we
534 observe no clear trend related to these elements, despite a variation of the SiO₂ content by more than
535 20%. Additionally, water, with a total content smaller than 1.10 wt.%, should have a negligible effect
536 on thermal conductivity (Romine et al. 2012, Ni et al. 2015). It could, however, impact the melt
537 density at a low degree of partial melting (Hofmeister 2014). As the water contents estimated in the

538 recovered samples are similar to the ones obtained in starting materials (see Supplementary Text S2
 539 and S3), water should not induce any strong diffusivity variation in our data-set. We, therefore,
 540 conclude that chemical effects are secondary compared to the structural ones.

542 Mixing models for thermal conductivities of partially molten peridotite

543 Our results show that peridotites, and to a lower extent the glasses, present higher thermal
 544 diffusivities compared to the melts. This result is not surprising: thermal conductivity is strongly
 545 dependent on the local structure and its vibrational properties. The disorder in the atomic structure
 546 increases from solid, glasses to molten states (Hofmeister et al. 2014). The composition of the melt
 547 appears to be of secondary importance. The final set of thermal diffusivities and conductivities values
 548 selected for the applications are given in Tables 3, S5 and Figures 9 and S14.

549 To propose a predictive law for thermal conductivity of partially molten rocks several mixing
 550 equations are now tested. Such equation generally describes the effect of a small amount of
 551 conductive phase into an insulating matrix. For thermal conduction, the problem is reversed because
 552 the melt is less conductive than the solid. In this section, we explore the different predictive models
 553 of thermal conductivity of binary mixtures:

554 1) Linear mixing model consider parallel thermal resistor:

$$555 \kappa_{bulk} = \kappa_s * (1 - F) + F\kappa_m \quad (18)$$

556 Where s corresponds to the solid, m the melt and F the volume fraction of the melt.

557 2) tube / Ashbie model consider 1/3 of tubes of melt aligned in the heat flow direction (Grant and
 558 West, 1965, Schmeling 1986):

$$559 \kappa_{bulk} = \frac{1}{3} * \kappa_s * (1 - F) + F\kappa_m \quad (19)$$

560 3) cube model (Waff, 1974) representing cubes of solids into a melt matrix

$$561 \kappa_{bulk} = [1 - F^{2/3}] * \kappa_s \quad (20)$$

562 4) Archie's law, an empirical relation developed for electrical conductivity (Watanabe and Kurita,
 563 1983)

$$564 \kappa_{bulk} = C * (1 - F)^n * \kappa_s \quad (21)$$

565 Where C and n are constants.

566 5) Thermal resistors in series:

$$567 \kappa_{bulk} = \kappa_s * \frac{\kappa_m / \kappa_s}{\kappa_m / \kappa_s + F(1 - \kappa_m / \kappa_s)} \quad (22)$$

568 6) Hashin Shtrikman lower bound (HS⁻), representing insulating melt spheres into a conductive solid
 569 matrix:

$$570 \kappa_{bulk} = \kappa_s * \frac{F}{1 / (\kappa_m / \kappa_s) + \frac{1 - F}{3\kappa_s}} \quad (23)$$

571 7) Maxwell-Eucken relation:

$$572 \kappa_{bulk} = \kappa_s * \frac{\kappa_m + 2\kappa_s + 2F(\kappa_m - \kappa_s)}{\kappa_m + 2\kappa_s - F(\kappa_m - \kappa_s)} \quad (24)$$

573 8) Landauer relation, based on resistors in series:

$$575 \kappa_{bulk} = \frac{1}{4} * \left[\kappa_m(3F - 1) + \kappa_s(2 - 3F) + \left\{ (\kappa_m(3F - 1) + \kappa_s(2 - 3F))^2 + 8\kappa_s\kappa_m \right\}^{\frac{1}{2}} \right] \quad (25)$$

576 9) Russel-Rayleigh relation:

$$578 \kappa_{bulk} = \frac{\kappa_s [\kappa_s + F^{2/3}(\kappa_m - \kappa_s)]}{\kappa_s + (\kappa_m - \kappa_s)(\kappa_m^{\frac{2}{3}} - F)} \quad (26)$$

579

580 These equations provide a different evolution of the thermal conductivity with the fraction of melt
581 (F). In Figure S14, we present the results when either fixing κ_{solid} and κ_{melt} to the average value of our
582 measurements, or adjusting their values to minimize the misfit between the mixing models and our
583 results at varying F values (Table S5). Among the variety of fits obtained, it appears that the effect of
584 partial melting is underestimated in most of the cases. Only the thermal resistors in series is capable
585 to reproduce adequately the strong curvature observed experimentally at low F values, as well as the
586 end-member values of solid and melt conductivities. For this reason, we use series model (eq. 22) for
587 further discussions.

588 We note a lack of data points at high melt fractions, preventing to decipher more precisely the
589 mixing trend. At very high temperatures, the experimental measurements become difficult on natural
590 peridotite melting, in particular due to melt escape and chemical reactions with the experimental cell.
591 Complementary data could be acquired working with analog system such as basalt/olivine mixture
592 and could represent a further research direction.

593 Thermal conductivity of partially molten peridotite: influence of texture

594 The 3D solid/melt arrangement is known to influence significantly the geophysical properties
595 of partially molten systems (von Bargen and Waff, 1986; Laporte et al., 1997; Laporte and Provost,
596 2000, Minarik and Watson, 1995; Yoshino et al., 2005 Maumus et al. 2005; Ten Grotenhuis et al.
597 2005, Freitas et al. 2019). Their distribution is classically described using dihedral angle value, which
598 translates the ability of a liquid to wet the grain boundaries as a consequence of interfacial energies.
599 The dihedral angle decreases when increasing pressure, temperature, water content and decreasing
600 silica/alumina content of the melt (Yoshino et al. 2007, Mibe et al. 1998, 1999, Laporte et al., 1997,
601 Watson et al 1991). Several studies with similar basaltic or peridotite melts (dry) have shown that
602 basalt-like melts at mantle conditions have dihedral angles significantly lower than the
603 interconnection threshold of 60° , with values between $30\text{-}40^\circ$ at 2 GPa (Laporte et al., 1997, Yoshino
604 et al. 2005, 2007). Our partially molten samples display a coherent texture and dihedral angles with
605 these observations. Dihedral angles of 23.3° and 19.0° were measured from our samples containing
606 6.4% and 23.3% of melt, respectively (Figure S15), in good agreement with previous data given that
607 these mafic melts are moderately hydrous (Table S3). For each melt fraction, a thin layer of melt
608 surround most of the grains, in particular olivines, which in 3D will result into the insulation of the
609 solid gains from their surroundings (Figure S9). This is very well visible on our low melt fraction
610 sample (F=6.4%) where the layers of melt are few microns thick (M804). Even if melt is more
611 abundant near clinopyroxenes and spinel sites in M847 (F=23.3%), the melt pockets are
612 interconnected with similar thin melt layers ($<10\ \mu\text{m}$) (Figure S9). As a result, thermal conductivity
613 is expected to drop brutally in the first degrees of melting. Still, some grain boundaries should remain
614 un-wetted until the melt fraction rise significantly. For this reason, thermal conductivity should only
615 stabilize at melt fraction corresponding to solid grains completely isolated from each other. This trend
616 is visible in our data and parallel model (Figures 9 and S14) with a strong decrease in the first 10%
617 of melting highlighted by M804, the change of slope seems to occur around 15% and values decrease
618 more slowly in the 15-50% range as seen in M847, to stabilize and display near-melt values above
619 50%. The complete isolation of solids should occur at “packing” threshold, which is a function of the
620 solid shapes and size distribution and is expected to occur between 40 to 60% of melting. Thus, the
621 first degrees of melting are very crucial in the case of a wetting liquid and affecting importantly the
622 thermal properties.

623 Thermal conductivity of peridotite: effect of the grain size

624 The modelling of thermal conductivity of peridotites and low F molten peridotites should also
625 take in account the effect of grain boundaries thermal resistances as grain size may vary in the
626 different geological contexts (from $100\ \mu\text{m}$ to $>1\ \text{cm}$), as seen in natural meteoritic examples, (Barrat
627 et al. 1999, Busek, 1977, Keil, 2010, Floran et al. 1978). Indeed, grain boundary scattering could be
628
629

630 important when the mean free paths of phonons approach the grain size. This effect, which only
 631 concerns solids and low fractions of melt, can be quantified with the following equation (Smith et al.
 632 2003, Smith et al. 2013):

$$633 \quad \frac{1}{\kappa_{poly}} = \frac{1}{\kappa_{single}} + nR_{boundary} \quad (27)$$

634 Where κ_{poly} is the thermal conductivity of the polycrystalline sample, κ_{single} the thermal
 635 conductivity of a reference single crystal (average from olivine data of Hofmeister et al. 2016, Table
 636 3), n represents the surface of grain boundaries along the heat flow direction per unit length, and
 637 $R_{boundary}$ the thermal resistance of grain boundary plane. The n value should be almost constant with
 638 temperature (Smith et al. 2013) and is estimated between $4e^{-4}$ and $7e^{-3}$ m for our two peridotite
 639 samples via analyses of SEM images (grain size ranging from 25 to 140 μm , see Table S4 for textural
 640 parameters). We calculate $R_{boundary}$ values between $1.9e^{-6}$ and $5.9e^{-6}$ $\text{W}/\text{m}^2/\text{K}$ for M804 and $9.3e^{-6}$ and
 641 $3.2e^{-5}$ $\text{W}/\text{m}^2/\text{K}$ for M847. These values are compatible or slightly higher than hydrous polycrystalline
 642 olivine samples (Zhang et al. 2019).

643 As a result, the thermal conductivities quantified in our experiments are underestimations of
 644 natural ones as the grain size is $<100 \mu\text{m}$, in experiments compared to grain sizes of $100 \mu\text{m}$ up to >1
 645 cm typical of mantle peridotites and reduced meteorites (which could be relic of bodies interiors,
 646 from cumulates (Floran et al. 1978), enstatite chondrite/achondrite (Keil, 2010), diogenite (Barrat et
 647 al. 1999) to pallasite (Busek, 1977)). Melts and high F partially molten systems are not affected by
 648 such effect, thus the observed decrease of thermal conductivity at the melting temperature is probably
 649 smaller in our experiments.

651 **Implications for geodynamos on Mercury-like proto-planets**

652 Suitable conditions for a dynamo

654 For a thermally driven dynamo to operate in a terrestrial planet, four conditions were found to be
 655 necessary (e.g., Montoux et al., 2011): the core heat flow must be at least adiabatic (1), the thermal
 656 convection within the core has to supply enough power to compensate the losses due to ohmic
 657 dissipation (2), the Reynolds magnetic number must be supercritical (complex turbulent convection)
 658 (3) and the mantle heat flow has to overcome the core heat flow needed to induce a dynamo (4). These
 659 conditions can be expressed in terms of heat flow balance (See Figure 10 for a schematic
 660 representation) and are detailed here:

661
 662 *(1) The metallic core has to convect, meaning that the heat flow out of the core needs to overcome
 663 the adiabatic heat flow (Stevenson et al., 1983).*

664 For this, the core thermal conductivity (κ_{core}) is a dominant parameter. A large κ_{core} value
 665 increases the heat flux along the core adiabat and reduces the lifetime of a thermally driven dynamo
 666 (Breuer et al., 2015). Several laboratory measurements suggested that the thermal conductivity of
 667 polycrystalline iron at Mercury's core conditions is 113–125 $\text{W}/\text{m}/\text{K}$ (see Deng et al., 2013 and
 668 references therein). However, such values for Mercury are recently challenged with several recent
 669 studies proposing a much lower conductivity. In a first one, the conductivity of pure Fe and Fe-Si
 670 alloys is reported at 30-40 $\text{W}/\text{m}/\text{K}$ and 35-40 $\text{W}/\text{m}/\text{K}$, respectively (Sibert et al. 2019). Then, it is
 671 proposed that the thermal conductivity of Fe-S at the P-T conditions of Mercury's core is as low as
 672 $\sim 4 \text{ W}/\text{m}/\text{K}$, thus 1-2 orders of magnitude lower than that of pure iron (Pommier et al. 2019,
 673 Manthilake et al. 2019).

674 This first condition can be expressed as:

$$675 \quad Q_{CMB} > Q_{Ad} = \frac{\kappa_c \alpha_c g_c T_{CMB}}{C_{p,c}} 4\pi R_c^2 \quad (28)$$

677 To estimate this flux, we assume that T_{CMB} is the melting temperature of pure iron at P_{CMB} . This
 678 assumption gives a conservative value of the core heat flow in comparison with considering T_{ICB}
 679 since the core liquidus is steeper than the core adiabat. We estimate the relation between the melting
 680 temperature of pure iron and the pressure using the following expression obtained by fitting the
 681 experimental results from Anzellini et al. 2013 with a Simon and Glatzel equation:

$$682 \quad T_{m,Fe} = 1800 \left(\frac{P_{CMB}}{27.9} + 1 \right)^{1/2.08} \quad (29)$$

683 Such a melting temperature typically lies between the solidus and liquidus of a chondritic mantle
 684 for the same pressure conditions (Monteux et al. 2020 and references therein). We also assume that
 685 k_c , α_c and $C_{p,c}$ are constant (see values in Table 4) and P_{CMB} is calculated as follows (Monteux and
 686 Arkani-Hamed, 2014):

$$687 \quad P_{CMB} = P(r = R_c) = \frac{2}{3} \pi G \rho_c^2 (R_c^2 - r^2) + \frac{2}{3} \pi G \rho_{Si}^2 (R^2 - R_c^2) + \frac{4}{3} \pi G \rho_{Si} R_c^3 (\rho_c - \rho_{Si}) \left(\frac{1}{R_c} - \frac{1}{R} \right) \quad (30)$$

688
 689 (2) *The energy supplied by thermal convection to the geodynamo has to compensate for the loss due*
 690 *to ohmic decay (Buffett, 2002).*

691 This imposes a condition on the core heat flow at the CMB. In fact, core heat flow will need to
 692 overcome a critical value. Assuming that dynamo is generated only by thermal convection in the core,
 693 we can write:

$$694 \quad Q_{CMB} > Q_{Ad} + \frac{4v_c \bar{B}^2 C_{p,c}}{0.8 \mu_c \alpha_c G \rho_c R_c} \quad (31)$$

695 This heat flux is estimated by considering that the characteristic magnetic length scale equals the
 696 radius of the core. The parameter \bar{B} is the average strength of the magnetic field inside the core and
 697 was estimated using a scaling from Christensen and Aubert (2006):

$$698 \quad \bar{B} = 0.9 \mu_c^{\frac{1}{2}} \rho_c^{\frac{1}{6}} \left(\frac{g_c Q_B (R_c - r_i)}{4\pi R_c r_i} \right)^{\frac{1}{3}} \quad (32)$$

699 With $Q_B = \alpha_c Q_{CMB} / C_{p,c}$ the buoyancy flux and r_i the radius of the inner core. In the scaling from
 700 Christensen and Aubert (2006), the inner core size cannot be set to 0. On Mercury the size of the
 701 inner core is currently not well constrained even if recent constraints via geodetic analysis (Genova
 702 et al. 2018) suggest its presence and a possible important size (r_i/R_c between 0.3 and 0.7). As we focus
 703 here on the effect of thermal cooling on dynamo generation (i.e. we do not consider the effect of
 704 compositional convection related to inner core growth), we consider a small inner core with $r_i/R_c =$
 705 0.01. The scaling law used to calculate the average strength of the magnetic field inside the core (Eq.
 706 32) is valid for the Earth but overestimates \bar{B} in the case of thin shell dynamos such as the one
 707 operating within Mercury (Christensen and Aubert (2006)). Mariner 10 spacecraft measurements
 708 showed that Mercury's magnetic field was 100 times weaker than the Earth's one. To account for this
 709 discrepancy, we consider that the average strength of the magnetic field is 1% \bar{B} obtained from Eq. 32
 710 when solving Eq. 31. We also consider that v_c , μ_c and ρ_c are constants (see values in Tab. 4). We
 711 note that most of the power needed to overcome the criterion related to Eq. 31 can be supplied by
 712 thermal core convection (i.e. criterion related to Eq. 28) especially for large metallic cores.

713
 714 (3) *The magnetic Reynolds number (Re_m) must be supercritical in order to have convective motions,*
 715 *inducing a complex structure needed to carry the magnetic field lines (U. R. Christensen and Aubert,*
 716 *2006).*

717 Reynolds magnetic number is calculated using Christensen and Aubert, 2006 formulation's and
 718 assuming that the characteristic magnetic length scale is the radius of the core:

$$719 \quad Re_m = \left(\frac{Q_{CMB} G \alpha_c}{3 C_{p,c}} \right)^{1/3} \frac{R_c}{v_c} > 10 - 100 \quad (33)$$

720 The value of the critical magnetic Reynolds number ($Re_{m,c}$) is usually constrained for models
 721 implying a large inner core. For Mercury-like planets, this value is less documented. Here we make
 722 the conservative assumption that this value ranges between 10 and 100.

723
 724 (4) *The mantle heat flow has to overcome the core heat flow needed to generate a dynamo.*

725 The mantle's Rayleigh number (Ra) conditions the efficiency of heat evacuation from the mantle:

726
$$Ra = \frac{\alpha_{Si} \rho_{Si} g \Delta T \delta_{Si}^3}{\eta_{Si} \kappa_{Si}} \quad (34)$$

727 With α_{Si} the mantle thermal expansion coefficient, ρ_{Si} the mantle density, ΔT the temperature
 728 difference between the core and the surface of the planet, δ_{Si} the mantle thickness, η_{Si} the mantle
 729 viscosity, κ_{Si} the mantle heat diffusivity ($\kappa_{Si} = k_{Si}/(\rho_{Si} C_{p,Si})$), and $C_{p,Si}$ the mantle heat capacity. If
 730 Ra is lower than a critical value ($Ra_C \approx 1000$), the heat is evacuated from the mantle by conduction
 731 and the mantle heat flow Q_{Si} is:

732
$$Q_{Si} = \frac{k_{Si} \Delta T}{\delta_{Si}} 4\pi R^2 \quad (35)$$

733 Alternatively, if the $Ra > Ra_C$, the heat is evacuated from the mantle by convection and Q_{Si} scales
 734 with $Ra^{1/3}$ (Solomatov, 2007):

735
$$Q_{Si} = 0.089 \frac{k_{Si} \Delta T}{\delta_{Si}} 4\pi R^2 Ra^{1/3} \quad (36)$$

736 In any case, for a thermally-driven dynamo to operate, Q_{Si} has to be larger than the core heat flow
 737 Q_{Core} . For these calculations, we consider here that the surface gravity is dominated by the gravity at
 738 the CMB ($g = g_c$). On a planet with a core over planet radius ratio (R_c/R) of ~ 0.5 , the mantle is thick
 739 enough so that $Ra > Ra_C$ and the heat is efficiently evacuated by convection. On such bodies, the
 740 dynamo is generated easily and should appear early in the planet history. If the heat evacuation is
 741 very efficient, the planet might cool down rapidly, which can potentially lead to a short-lived dynamo.
 742 On a Mercury-like body with a large R_c/R ratio, the mantle is thin compared to the core and its
 743 Rayleigh number, critical criterion for thermal convection, hardly overcomes Ra_C . Hence, conduction
 744 should be the main heat evacuation process within the mantle of Mercury-like planets, which can
 745 limit the occurrence of an early thermally-driven dynamo.

746
 747 Simulation results

748 We now compare the core heat flow needed to generate a dynamo within a Mercury-like body
 749 (with e.g. $R_c/R_{planets}$) (Figures 11 and 12). To do this, we use a maximum possible range of solid
 750 conductivities between $\kappa_{Si} = 4.32$ and 1.70 W/m/K. We consider a maximum planet radius $R=2440$
 751 km with a thin mantle shell to stay in temperature and pressure conditions equivalent to the
 752 experimental conditions detailed above ($P \leq 2$ GPa and $T \leq 1700$ K). We also report the critical size of
 753 the protoplanet ($R = 215$ km) above which Re_m is larger than 100 (Eq. 33). The Figure 11 shows that
 754 a solid conductive mantle is able to evacuate enough heat to induce a thermally driven dynamo if the
 755 planet radius is larger than ~ 215 km, regardless of the value of thermal conductivity in the outer core,
 756 at least up to more than 40 W/m/K. In contrast, the change of core conductivities from e.g 40 W/m/K
 757 (Sibert et al., 2019) to 4 W/m/K (Manthilake et al. 2019) has a major effect on the maximum size of
 758 a planet that can operate a thermally driven dynamo. For a core conductivity $\kappa_{core} = 40$ W/m/K, the
 759 maximum planet radius evolves from 1220 to 1880 km, for $\kappa_{Si} = 1.70$ and 4.32 W/m/K, respectively.
 760 When $\kappa_{core} = 4$ W/m/K, the maximum planet radius is more than 2500 km.

761 We now perform the same type of calculation for a mushy mantle using the silicate-melt
 762 conductivity refined in this study (Tables 3 and 4). We consider here neither a complete magma ocean
 763 that would evacuate the core heat very efficiently by turbulent convection (Monteux et al., 2016) nor
 764 a mushy mantle with a large melt fraction (larger than $40-60\%$) which would imply a strong decrease
 765 of the bulk viscosity of the mantle (Picard et al., 2013). We consider here a purely conductive mantle
 766 where the melt fraction is smaller than 20% . Hence, if this melt fraction is concentrated at the core

767 mantle boundary, the molten layer is thin enough to avoid its convection, while if the liquid is equally
768 distributed within the mantle, we can reasonably assume that the bulk viscosity is weakly affected by
769 the liquid phase and close to the viscosity of subsolidus silicate material. Still, the presence of melt
770 may affect mantle properties. An important parameter is the fate of melts, which is primarily
771 controlled by the solid-melt density contrast. The molten material can either be evacuated at the
772 surface during early volcanism or cumulate at the lowermost mantle. For peridotites, negative melt
773 buoyancy is unlikely to happen at pressures lower than 7 GPa (Sakamaki et al. 2006, Matsukage et
774 al. 2005, Freitas et al. 2017). However, the buoyancy of melts generated from the partial melting of
775 a body with composition largely different from peridotite remains uncertain. Molten reservoirs may
776 be trapped at shallow depths at the end of mushy mantle cooling (Monteux et al., 2020) or during the
777 last stages of fractional crystallization within small bodies (e.g. Frossard et al., 2019). On a Moon-
778 like body, a melt layer may be trapped below an anorthositic crust while on a Mercury-like body,
779 anorthite is denser than the melt and then may not float allowing the melt to form a late shallow
780 magma ocean. We consider here that the melt has neutral buoyancy and is equally distributed within
781 the planetary mantle, with the consequence that the mantle viscosity is primarily controlled by the
782 solid fraction of the mantle.

783 Due to the lower thermal conductivity of the melt compared to the solid mantle, the mushy mantle
784 should limit the heat flow that can be extracted from the core. In our calculation, we consider for
785 simplicity that the entire mantle has the same conductivity as the melt. As a result, the range of
786 planetary radii for which a dynamo is plausible is much narrower than for a fully solid mantle (Figure
787 12). For a core conductivity of 40 W/m/K, a thermally driven dynamo can only operate for planetary
788 sizes ranging between 215 and 350-570 km, depending on the value of κ_{Si} and for $\kappa_{core} = 40$ W/m/K.
789 The maximum planetary radius becomes 1000-1500 km for a core conductivity of 4 W/m/K.
790 Therefore, the melting of a Mercury-like planetary mantle could limit the establishment of a magnetic
791 field on the planet. This effect could last over different timescales, depending on the size of the planet
792 and of the core, the initial core temperature, the fate of the melt material, etc.

793 The differences obtained for fully solid (Figure 11) and molten (Figure 12) models show that the
794 presence of melt within a “Mercury-like” planet mantle could result into better thermal insulation of
795 the core, which may prevent the occurrence of a thermally driven dynamo. Of course, planets
796 presenting mantle with supercritical Rayleigh number would not be affected by this effect, because
797 mantle convection could extract enough heat from the core at the CMB. As a consequence, the solid
798 mantle viscosity plays a key role because (1) it controls the planet size above which mantle convection
799 is likely, through the Ra value, and (2) it governs the efficiency of mantle heat evacuation once the
800 critical Rayleigh number is overcome (see Eq. 36). Decreasing the mantle viscosity from 10^{22} to 10^{18}
801 Pa.s decreases the critical planet size for mantle convection by one order of magnitude (from 1600-
802 1900 km to 160-200 km for a solid mantle and from 850-1050 km to 80-110 km for a mushy mantle,
803 see Figures 11 and 12). The figures also show that the lower is mantle viscosity, the easier the planet
804 can meet the criteria for a thermally driven dynamo when the mantle of the planet is convecting. We
805 note here that considering the influence of melt on the bulk mantle viscosity should lead to a decrease
806 of the critical Rayleigh number and as a consequence of the critical planetary size where heat is only
807 evacuated by thermal conduction (i.e. green triangles would move to the left). However, this effect
808 should be negligible for the small melt fraction considered here.

809 The impact of mantle partial melting crucially depends on the melt quantity and its location within
810 the mantle. Whether it is distributed in a mushy mantle or accumulated as a pond at a given mantle
811 depth affects both, locally and globally, the mantle viscosity and the thermal diffusivity. As a
812 consequence, the dynamic regime can evolve from conduction to convection, changing the thermal
813 outputs at the CMB and making the heat flux enough, or insufficient, to power a thermally-driven
814 dynamo. In this framework, transient phenomena such as mantle overturns (Elkis-Tanton et al. 2003,
815 Ballmer et al. 2017) implying major melt migration will result in a brutal change in the planet
816 magnetic signature. Similarly, the chemical stratification resulting in the production of a crust

817 (anorthositic, basaltic etc.) may help insulating the planet interior in its early history suggesting
818 weaker but longer sustained dynamos.

819

820 **Conclusions**

821

822 In this study, we constrain the thermal diffusivities of silicate melts with a wide range of
823 composition at high pressure and temperature conditions using the multi-anvil apparatus and the
824 Angström method. We observed that melting induces a significant decrease in both thermal
825 diffusivities and conductivities. We measure melt thermal diffusivities ranging from 0.18 (± 0.02)
826 W/m/k to 0.41 (± 0.04) W/m/K. The composition of the melt does not have a significant impact on
827 diffusivities. We then use our results to address the thermal properties of the molten silicate reservoirs,
828 and the likeliness of a thermally driven dynamo within a Mercury-like planets presenting a large
829 metallic core and a relatively thin silicate mantle (large R_c/R_{planet}). The mantle of such bodies is not
830 expected to be convecting easily ($Ra < Ra_c$) leading to an inefficient heat evacuation and difficulty
831 generating a dynamo compared to planets with a thicker mantle (where easily $Ra > Ra_c$). Our results
832 illustrate that the presence of a mushy mantle above the CMB in such a Mercury-like planet can
833 significantly reduce the ability of the mantle to evacuate the heat from the core and limit the likeliness
834 of a thermally-driven dynamo. As a consequence, the fate and the lifetime of such a mushy reservoir
835 can have a profound impact on the thermal history of Mercury-like planets. Future observations of
836 Mercury-like planets accreted in orbits close to their star and the eventual detection of their intrinsic
837 magnetic field would in return constrain the state of the mantle surrounding their metallic core.

838

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846

847 **References**

- 848 1. Acuna, M. H., Connerney, J. E. P., Lin, R. P., Mitchell, D., Carlson, C. W., McFadden, J., et al. (1999).
849 Global distribution of crustal magnetization discovered by the Mars Global Surveyor MAG/ER
850 experiment. *Science*, 284(5415), 790–793.
- 851 2. Anderson, B. J., Johnson, C. L., Korth, H., Purucker, M. E., Winslow, R. M., Slavin, J. A., et al. (2011).
852 The global magnetic field of Mercury from MESSENGER orbital observations. *Science*, 333(6051),
853 1859–1862. <https://doi.org/10.1126/science.1211001>
- 854 3. Andraut, D., Bolfan-Casanova, N., Nigro, G.L., Bouhifd, M.A., Garbarino, G., Mezouar, M., (2011).
855 Solidus and liquidus profiles of chondritic mantle: implication for melting of the Earth across its history.
856 *Earth Planet. Sci. Lett.*304, 251–259. <http://dx.doi.org/10.1016/j.epsl.2011.02.006>.
- 857 4. Anzellini, S., Dewaele, A., Mezouar, M., Loubeyre, P., Morard, G., (2013). Melting of iron at Earth’s
858 inner core boundary based on fast X-ray diffraction. *Science*340, 464–466.
- 859 5. Asphaug, E., & Reufer, A. (2014). Mercury and other iron-rich planetary bodies as relics of inefficient
860 accretion. *Nature Geoscience*, 7(8), 564–568. <https://doi.org/10.1038/ngeo2189>
- 861 6. Bajgain, S., Ghosh, D. B., & Karki, B. B. (2015). Structure and density of basaltic melts at mantle
862 conditions from first-principles simulations. *Nature Communications*, 6, 1–7.
863 <https://doi.org/10.1038/ncomms9578>
- 864 7. Ballmer, M. D., D. L. Lourenco, K. Hirose, R. Caracas, and R. Nomura (2017), Reconciling magma-
865 ocean crystallization models with the present-day structure of the Earth's mantle, *Geochem., Geophys.,*
866 *Geosyst.*, 18(7), 2785-2806, doi:10.1002/2017gc006917.
- 867 8. Bambier, A., Berger, E., Mergoil, J., Valadas, B., Veyret, Y., Weisbord, A., (1985). Carte géologique
868 de la France à 1/50 000 (n 840), Burzet: Des sources de l’Ardèche aux sources de la Loire. Carte et
869 notice explicative.
- 870 9. Barrat, J.A, Gillet, PH, Lesourd, M., Blicher-Toft, J., & G.R Poupeau. (1999). The Tatahouine
871 diogenite: Mineralogical and chemical effects of sixty-three years of terrestrial residence. *Meteoritics*
872 *& Planetary science*, 34, 91-97.
- 873 10. Beck, A. E., Darbha, D. M., & Schloessin, H. H. (1978). Lattice conductivities of single-crystal and
874 polycrystalline materials at mantle pressures and temperatures. *Physics of the Earth and Planetary*
875 *Interiors*, 17(1), 35–53. [https://doi.org/10.1016/0031-9201\(78\)90008-0](https://doi.org/10.1016/0031-9201(78)90008-0)
- 876 11. Benz, W., Slattery, W. L., & Cameron, A. G. W. (1988). Collisional stripping of Mercury’s mantle.
877 *Icarus*, 74(3), 516–528. [https://doi.org/10.1016/0019-1035\(88\)90118-2](https://doi.org/10.1016/0019-1035(88)90118-2)
- 878 12. Berger, E., (1973). Le volcanisme récent de l’Ardèche. Thèse, Orsay, 402 p.
- 879 13. Berger, E., Forette, M. C., (1975). Réactions de basaltes alcalins des Causses, du Vivarais et du Velay
880 (France) avec les xénocristaux et les minéraux des enclaves homoéogènes. *Bulletin de Minéralogie*,
881 98(6), 366-373.
- 882 14. Berger, E., (1981). Enclaves ultramafiques, mégacristaux et leurs basaltes-hôtes en contexte océanique
883 (Pacifique Sud) et continental (Massif Central Français). Thèse d’État, Université de Paris-Sud,
884 Orsay,470 p.
- 885 15. Bigazzi, G., Yegingil, Z., Ercan, T., Oddone, M., Özdoğan, M., (1993). Fission track dating obsidians
886 in Central and Northern Anatolia. *Bulletin of Volcanology*, 55(8), 588-595.

- 887 16. Bolfan-Casanova, N., Montagnac, G., Reynard, B., (2014). Measurement of water contents in olivine
888 using Raman spectroscopy. *Am. Mineral.* 99:149–156. doi: 10.2138/am.2014.4444.
- 889 17. Bougault, H., & Treuil, M. (1980). Mid-atlantic ridge: Zero-age geochemical variations between azores
890 and 22°N. *Nature*, 286(5770), 209–212. <https://doi.org/10.1038/286209a0>
- 891 18. Boujibar, A., Andrault, D., Ali, M., Bolfan-casanova, N., Devidal, J., & Trcera, N. (2014). Metal –
892 silicate partitioning of sulphur , new experimental and thermodynamic constraints on planetary
893 accretion. *Earth and Planetary Science Letters*, 391, 42–54. <https://doi.org/10.1016/j.epsl.2014.01.021>
- 894 19. Boujibar, A., Andrault, D., Bolfan-Casanova, N., Bouhifd, M. A., & Monteux, J. (2015).
895 Cosmochemical fractionation by collisional erosion during the Earth's accretion. *Nature*
896 *Communications*, 6, 1–7. <https://doi.org/10.1038/ncomms9295>
- 897 20. Breuer, D., Hauck, S. A., Buske, M., Pauer, M., Spohn, T., (2007). Interior Evolution of Mercury. *Space*
898 *Sci. Rev.* 132, 229–260.
- 899 21. Breuer, D., Rueckriemen, T., & Spohn, T. (2015). Iron snow, crystal floats, and inner-core growth:
900 modes of core solidification and implications for dynamos in terrestrial planets and moons. *Progress in*
901 *Earth and Planetary Science*, 2(1), 39. <https://doi.org/10.1186/s40645-015-0069-y>
- 902 22. Buffett, B.A., 2002. Estimates of heat flow in the deep mantle based on the power requirements for the
903 geodynamo. *Geophys. Res. Lett.* 29 (12), 1–4.
- 904 23. Busek, P. R., (1997). Pallasite meteorites-mineralogy, petrology and geochemistry. *Geochimica et*
905 *Cosmochimica Acta*, 41, 711-740.
- 906 24. Cao, H., Aurnou, J. M., Wicht, J., Dietrich, W., Soderlund, K. M., & Russell, C. T. (2014). A dynamo
907 explanation for Mercury's anomalous magnetic field. *Geophysical Research Letters*, 41, 4127–4134.
908 <https://doi.org/10.1002/2014GL060196>.
- 909 25. Charlier, B., and Namur, O., (2019). The Origin and Differentiation of Planet Mercury, *Elements*, 15
910 (1): 9–14. <https://doi.org/10.2138/gselements.15.1.9>.
- 911 26. Chen, B., Li, J., & Hauck, S. A. (2008). Non-ideal liquidus curve in the Fe-S system and Mercury's
912 snowing core. *Geophysical Research Letters*, 35, L07201. <https://doi.org/10.1029/2008GL033311>.
- 913 27. Christensen, U. R., & Aubert, J. (2006). Scaling properties of convection-driven dynamos in rotating
914 spherical shells and application to planetary magnetic fields. *Geophysical Journal International*, 166(1),
915 97–114. <https://doi.org/10.1111/j.1365-246X.2006.03009.x>
- 916 28. Christensen, U. R., & Wicht, J. (2008). Models of magnetic field generation in partly stable planetary
917 cores: Applications to Mercury and Saturn. *Icarus*, 196(1), 16–34.
918 <https://doi.org/10.1016/j.icarus.2008.02.013>.
- 919 29. Christensen, U. R. (2006). A deep dynamo generating Mercury's magnetic field. *Nature*, 444(7122),
920 1056–1058. <https://doi.org/10.1038/nature05342>.
- 921 30. Deng, L., Seagle, C., Fei, Y., & Shahar, A. (2013). High pressure and temperature electrical resistivity
922 of iron and implications for planetary cores. *Geophysical Research Letters*, 40, 33–37.
923 <https://doi.org/10.1029/2012GL054347>.
- 924 31. Druitt, T. H. (1995). Settling behaviour of concentrated dispersions and some volcanological
925 applications. *Journal of Volcanology and Geothermal Research*, 65(1–2), 27–39.
926 [https://doi.org/10.1016/0377-0273\(94\)00090-4](https://doi.org/10.1016/0377-0273(94)00090-4)
- 927 32. Dumberry, M., & Rivoldini, A. (2015). Mercury's inner core size and core-crystallization regime. *Icarus*,
928 248, 254–268. <https://doi.org/https://doi.org/10.1016/j.icarus.2014.10.038>.
- 929 33. Ebel, D. S., & Grossman, L. (2000). Condensation in dust-enriched systems, 64(2), 339–366.
- 930 34. Elkins-Tanton, L. T., E. M. Parmentier, and P. C. Hess (2003), Magma ocean fractional crystallization
931 and cumulate overturn in terrestrial planets: Implications for Mars, *Meteoritics & Planetary Science*,
932 38(12), 1753-1771, doi:10.1111/j.1945-5100.2003.tb00013.x.

- 933 35. Fegley, B., & Cameron, A. G. W. (1987). A vaporization model for iron/silicate fractionation in the
934 Mercury protoplanet. *Earth and Planetary Science Letters*, 82(3–4), 207–222.
935 [https://doi.org/10.1016/0012-821X\(87\)90196-8](https://doi.org/10.1016/0012-821X(87)90196-8)
- 936 36. Floran, R.J., Prinz M., Hlava, P.F., Keil, K., Nehru, C.E. & Hinthorne. J. R. (1978). The Chassigny
937 meteorite: a cumulate dunitite with hydrous amphibole-bearing melt inclusions. *Geochimica et*
938 *Comsochimica acta*, 42 (8), 1213-1219. [https://doi.org/10.1016/0016-7037\(78\)90115-1](https://doi.org/10.1016/0016-7037(78)90115-1)
- 939 37. Freitas, D., Manthilake, G., Schiavi, F., Chantel, J., Bolfan-Casanova, N., Bouhifd, M.A., Andrault, D.,
940 (2017). Experimental evidence supporting a global melt layer at the base of the Earth’s upper mantle.
941 *Nat. Commun.* 8:2186. <https://doi.org/10.1038/s41467-017-02275-9>.
- 942 38. Freitas, D., Manthilake, G., Chantel, J., Bouhifd, M., Andrault, D., (2019). Simultaneous measurements
943 of electrical conductivity and seismic velocity of partially molten geological materials: Implications for
944 melt fraction in the upper mantle. *Phys. Chem. Miner.* 0, 0. doi:10.1007/s00269-019-01021-5.
- 945 39. Freitas, D., Manthilake, G., (2019b). Electrical conductivity of hydrous silicate melts : Implications for
946 the bottom-up hydration of Earth ’ s upper mantle *EPSL* 523, 1–9. doi:10.1016/j.epsl.2019.115712.
- 947 40. Frossard P., Boyet M., Bouvier A., Hammouda T., Monteux J. (2019). Evidence for anorthositic crust
948 formed on an inner solar system planetesimal. *Geochemical Perspectives Letters* vol.11, p.28-32,
949 DOI:10.7185/geochemlet.1921 .
- 950 41. Fujisawa, H., Fujii, N., Mizutani, H., Kanmori, H., Akimoto, S., (1968) Thermal Diffusivity of
951 Mg₂SiO₄, Fe₂SiO₄, and NaCl at High Pressures and Temperatures. *J. Geophys. Res.* 73:4727.
- 952 42. Garrick-Bethell, I., Weiss, B. P., Shuster, D. L., & Buz, J. (2009). Early lunar magnetism. *Science*,
953 323(5912), 356–359. <https://doi.org/10.1126/science.1166804>
- 954 43. Genova, A., Goossens, S., Mazarico, E., Lemoine, F. G., Neumann, G. A., Kuang, W., et al. (2018).
955 Geodetic evidence that Mercury has a solid inner core, 1–13. <https://doi.org/10.1029/2018GL081135>
- 956 44. Gillet, P., Richet, P., Guyot, F., Fiquet, G., (1991). High-temperature thermodynamic properties of
957 forsterite. *J. Geophys. Res.* B96:11805–11816.
- 958 45. Giordano, D., Nichols, A. R. L., & Dingwell, D. B. (2005). Glass transition temperatures of natural
959 hydrous melts: A relationship with shear viscosity and implications for the welding process. *Journal of*
960 *Volcanology and Geothermal Research*, 142(1-2 SPEC. ISS.), 105–118.
961 <https://doi.org/10.1016/j.jvolgeores.2004.10.015>
- 962 46. Glassmeier, K. H., Auster, H. U., & Motschmann, U. (2007). A feedback dynamo generating Mercury's
963 magnetic field. *Geophysical Research Letters*, 34, L22201. <https://doi.org/10.1029/2007GL031662>.
- 964 47. Grant, F. S., West, G. F., (1965). *Interpretation theory in applied geophysics*. 583 pp., McGraw-Hill
965 Book Co., New York.
- 966 48. ten Grotenhuis, S.M., Drury, M.R., Spiers, C.J., Peach, C.J., 2005. Melt distribution in olivine rocks
967 based on electrical conductivity measurements. *J. Geophys. Res. Solid Earth* 110, 1–11.
968 <https://doi.org/10.1029/2004JB003462>
- 969 49. Guérin G., (1983). *La thermoluminescence des plagioclases, méthode de datation du volcanisme.*
970 *Application au domaine volcanique français : chaîne des Puys, Mont-Dore et Cézallier, Bas-Vivarais,*
971 *thèse, université Paris-6, 258 p.*
- 972 50. Harte, B., (1977). Rock nomenclature with particular relation to deformation and recrystallisation
973 textures in olivine-bearing xenoliths. *The Journal of Geology*, 85(3), 279-288.
- 974 51. Hartlieb, P., Toifl, M., Kuchar, F., Meisels, R., & Antretter, T. (2016). Thermo-physical properties of
975 selected hard rocks and their relation to microwave-assisted comminution. *Minerals Engineering*, 91,
976 34–41. <https://doi.org/10.1016/j.mineng.2015.11.008>
- 977 52. Hauck, S. A., Margot, J. L., Solomon, S. C., Phillips, R. J., Johnson, C. L., Lemoine, F. G., et al. (2013).
978 The curious case of Mercury’s internal structure. *Journal of Geophysical Research E: Planets*, 118(6),
979 1204–1220. <https://doi.org/10.1002/jgre.20091>

- 980 53. Heimpel, M. H., Aurnou, J. M., Al-Shamali, F. M., & Gomez Perez, N. (2005). A numerical study of
981 dynamo action as a function of spherical shell geometry. *Earth and Planetary Science Letters*, 236(1–2),
982 542–557. <https://doi.org/10.1016/j.epsl.2005.04.032>.
- 983 54. Hernlund, J., Leinenweber, K., Locke, D., & Tyburczy, J. A. (2006). A numerical model for steady-state
984 temperature distributions in solid-medium high-pressure cell assemblies. *American Mineralogist*, 91(2–
985 3), 295–305. <https://doi.org/10.2138/am.2006.1938>
- 986 55. Hofmeister, A. M., & Branlund, J. M. (2015). Thermal Conductivity of the Earth. *Treatise on*
987 *Geophysics: Second Edition (Vol. 2)*. Elsevier B.V. [https://doi.org/10.1016/B978-0-444-53802-
988 4.00047-6](https://doi.org/10.1016/B978-0-444-53802-4.00047-6)
- 989 56. Hofmeister, Anne M., Whittington, A. G., & Pertermann, M. (2009). Transport properties of high albite
990 crystals, near-endmember feldspar and pyroxene glasses, and their melts to high temperature.
991 *Contributions to Mineralogy and Petrology*, 158(3), 381–400. [https://doi.org/10.1007/s00410-009-
992 0388-3](https://doi.org/10.1007/s00410-009-0388-3)
- 993 57. Hofmeister, Anne M., Sehlke, A., & Whittington, A. G. (2014). Thermal diffusivity of Fe-rich pyroxene
994 glasses and their melts. *Chemical Geology*, 384, 1–9. <https://doi.org/10.1016/j.chemgeo.2014.06.018>
- 995 58. Hood, L. L., Harrison, K. P., Langlais, B., Lillis, R. J., Poulet, F., & Williams, D. A. (2010). Magnetic
996 anomalies near Apollinaris Patera and the Medusae Fossae Formation in Lucus Planum, Mars. *Icarus*,
997 208(1), 118–131. <https://doi.org/10.1016/j.icarus.2010.01.009>
- 998 59. Johnson, Catherine L., & Phillips, R. J. (2005). Evolution of the Tharsis region of Mars: Insights from
999 magnetic field observations. *Earth and Planetary Science Letters*, 230(3–4), 241–254.
1000 <https://doi.org/10.1016/j.epsl.2004.10.038>
- 1001 60. Johnson, C. L., & McFadden, P. (2015). The Time-Averaged Field and Paleosecular Variation. *Treatise*
1002 *on Geophysics: Second Edition*, 5, 385–417. <https://doi.org/10.1016/B978-0-444-53802-4.00105-6>
- 1003 61. Johnson, C. L., Phillips, R. J., Purucker, M. E., Anderson, B. J., Byrne, P. K., Denevi, B. W., et al.
1004 (2015). Low-altitude magnetic field measurements by MESSENGER reveal Mercury's ancient crustal
1005 field. *Science*, 348(6237), 892–895. <https://doi.org/10.1126/science.aaa8720>.
- 1006 62. Kabin, K., Heimpel, M. H., Rankin, R., Aurnou, J. M., Gómez-Pérez, N., Paral, J., et al. (2008). Global
1007 MHD modeling of Mercury's magnetosphere with applications to the MESSENGER mission and
1008 dynamo theory. *Icarus*, 195(1), 1–15. <https://doi.org/10.1016/j.icarus.2007.11.028>
- 1009 63. Kanamori H., Mizutani, H., Fujii, N., (1969). Method of thermal diffusivity measurement. *J. Phys. Earth*.
1010 17:43–53.
- 1011 64. Katsura, T. (1993). Thermal diffusivity of silica glass at pressures up to 9 GPa. *Physics and Chemistry*
1012 *of Minerals*, 20(3), 201–208. <https://doi.org/10.1007/BF00200122>
- 1013 65. Keil, K. (2010) Enstatite achondrite meteorites (aubrites) and the histories of their asteroidal parent
1014 bodies, *Chemie der Erde*, 70 (4), 295-317. doi:10.1016/j.chemer.2010.02.002
- 1015 66. Laporte, D., Rapaille, C., Provost, A., (1997). Wetting angles, equilibrium melt geometry, and the
1016 permeability threshold of partially molten crustal protoliths BT. In: Bouchez JL, Hutton DHW, Stephens
1017 WE (eds) *Granite: from segregation of melt to emplacement fabrics*. Springer Netherlands, Dordrecht,
1018 pp 31–54.
- 1019 67. Laporte, D., Provost, A., 2000. Equilibrium geometry of a fluid phase in a polycrystalline aggregate
1020 with anisotropic surface energies: Dry grain boundaries. *J. Geophys. Res. Solid Earth* 105, 25937–
1021 25953. <https://doi.org/10.1029/2000jb900256>
- 1022 68. Lawrence, K., Johnson, C., Tauxe, L., & Gee, J. (2008). Lunar paleointensity measurements:
1023 Implications for lunar magnetic evolution. *Physics of the Earth and Planetary Interiors*, 168(1-2), 71-87.
- 1024 69. Li, L., Wentzcovitch, R. M., Weidner, D. J., & Da Silva, C. R. S. (2007). Vibrational and thermodynamic
1025 properties of forsterite at mantle conditions. *Journal of Geophysical Research: Solid Earth*, 112(5).
1026 <https://doi.org/10.1029/2006JB004546>

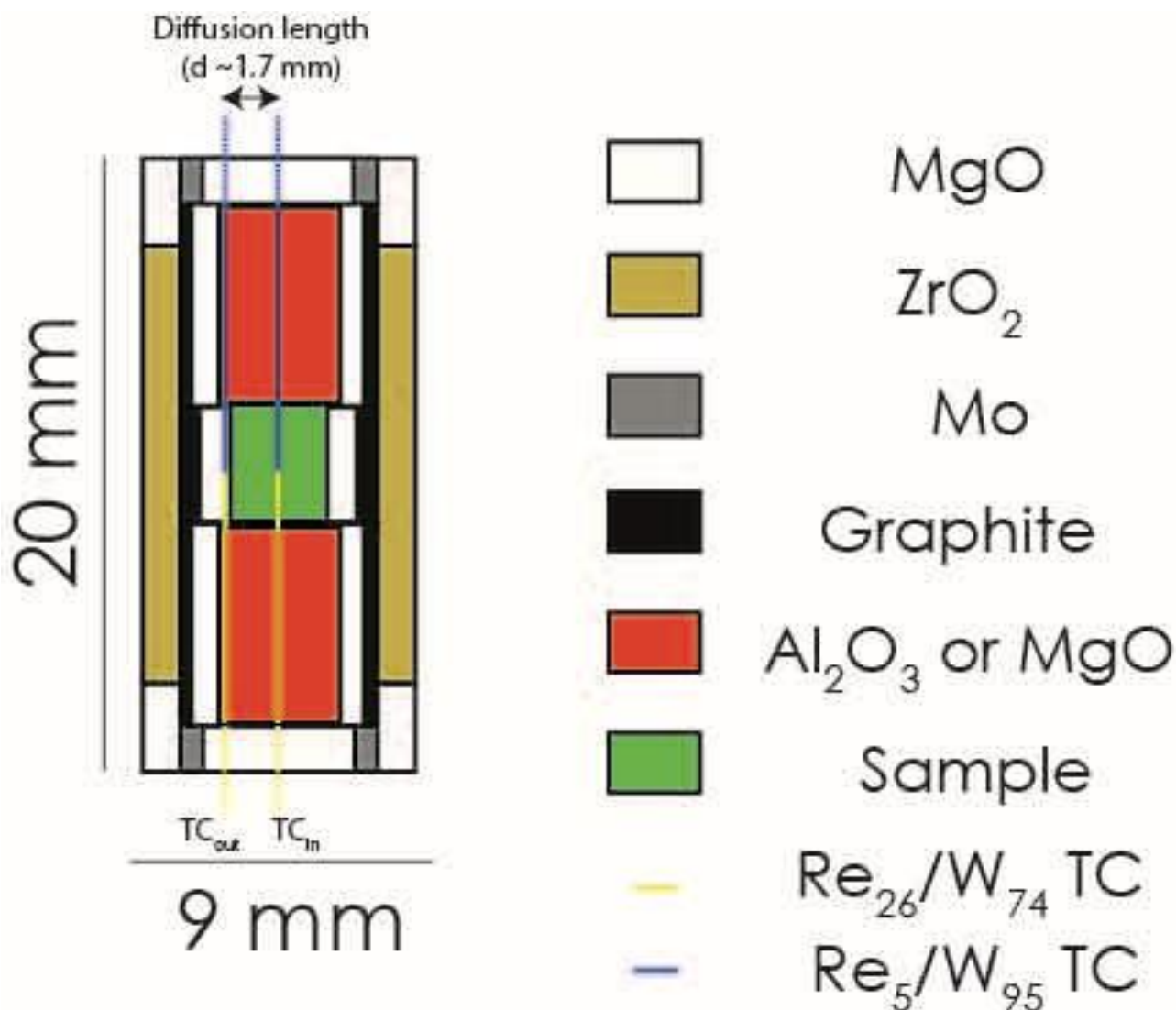
- 1027 70. Lillis, R. J., Frey, H. V., Manga, M., Mitchell, D. L., Lin, R. P., Acuña, M. H., & Bougher, S. W. (2008).
1028 An improved crustal magnetic field map of Mars from electron reflectometry: Highland volcano
1029 magmatic history and the end of the martian dynamo. *Icarus*, 194(2), 575–596.
1030 <https://doi.org/10.1016/j.icarus.2007.09.032>
- 1031 71. Lillis, R. J., Robbins, S., Manga, M., Halekas, J. S., & Frey, H. V. (2013). Time history of the Martian
1032 dynamo from crater magnetic field analysis. *Journal of Geophysical Research E: Planets*, 118(7), 1488–
1033 1511. <https://doi.org/10.1002/jgre.20105>
- 1034 72. Malavergne, V., Toplis, M. J., Berthet, S., & Jones, J. (2010). Highly reducing conditions during core
1035 formation on Mercury: Implications for internal structure and the origin of a magnetic field. *Icarus*,
1036 206(1), 199–209. <https://doi.org/10.1016/j.icarus.2009.09.001>
- 1037 73. Malfait, W. J., Seifert, R., Petitgirard, S., Mezouar, M., & Sanchez-Valle, C. (2014). The density of
1038 andesitic melts and the compressibility of dissolved water in silicate melts at crustal and upper mantle
1039 conditions. *Earth and Planetary Science Letters*, 393, 31–38. <https://doi.org/10.1016/j.epsl.2014.02.042>
- 1040 74. Manthilake, G., Chantel, J., Monteux, J., Andrault, D., Bouhifd, M.A., Casanova, N.B., Boulard, E.,
1041 Guignot, N., King, A., Itie, J.P., (2019). Thermal conductivity of FeS and its implications for Mercury’s
1042 long sustaining magnetic field. *J. Geophys. Res. E Planets*. doi:10.1029/2019JE005979
- 1043 75. Manthilake, M. A. G. M., De Koker, N., & Frost, D. J. (2011a). Thermal conductivity of CaGeO₃
1044 perovskite at high pressure. *Geophysical Research Letters*, 38(8), 3–6.
1045 <https://doi.org/10.1029/2011GL046882>
- 1046 76. Manthilake, G. M., De Koker, N., Frost, D. J., & McCammon, C. A. (2011b). Lattice thermal
1047 conductivity of lower mantle minerals and heat flux from Earth’s core. *Proceedings of the National
1048 Academy of Sciences of the United States of America*, 108(44), 17901–17904.
1049 <https://doi.org/10.1073/pnas.1110594108>
- 1050 77. Matsukage, K.N., Jing, Z., Karato, S., (2005). Density of hydrous silicate melt at the conditions of
1051 Earth’s deep upper mantle. *Nature* 438, 488–91. doi:10.1038/nature04241.
- 1052 78. Maumus, J., Bagdassarov, N., Schmeling, H., 2005. Electrical conductivity and partial melting of
1053 mafic rocks under pressure. *Geochim. Cosmochim. Acta* 69, 4703–4718.
1054 <https://doi.org/10.1016/j.gca.2005.05.010>
- 1055 79. Médard, E., McCammon, C. A., Barr, J. A., & Grove, T. L. (2008). Oxygen fugacity, temperature
1056 reproducibility, and H₂O contents of nominally anhydrous piston-cylinder experiments using graphite
1057 capsules. *American Mineralogist*, 93(11–12), 1838–1844. <https://doi.org/10.2138/am.2008.2842>
- 1058 80. Mercier, J.-C.C., (1972). Structure des péridotites en enclaves dans quelques basaltes d’Europe et
1059 d’Hawaii. vRegards sur la constitution du manteau supérieur. Thèse 3e cycle, Nantes, France, 229 p.
- 1060 81. Mercier, J. C. C., & Nicolas, A. (1975). Textures and fabrics of upper-mantle peridotites as illustrated
1061 by xenoliths from basalts. *Journal of Petrology*, 16(1), 454–487.
1062 <https://doi.org/10.1093/petrology/16.1.454>
- 1063 82. Mibe, K., Fujii, T., Yasuda, A., (1998). Connectivity of aqueous fluid in the Earth’s upper mantle. *Geo.
1064 Res. Lett.* 25, 1233–1236
- 1065 83. Mibe, K., Fujii, T., Yasuda, A., 1999. Control of the location of the volcanic front in island arcs by
1066 aqueous fluid connectivity in the mantle wedge. *Nature* 401, 259–262. <https://doi.org/10.1038/45762>
- 1067 84. Mighani, S., Wang, H., Shuster, D. L., Borlina, C. S., Nichols, C. I. O., & Weiss, B. P. (2020). The end
1068 of the lunar dynamo. *Science*, 1–8.
- 1069 85. Minarik, W.G., Watson, E.B., (1995). Interconnectivity of carbonate melt at low melt fraction. *Earth.
1070 Planet. Sci. Lett* 133(3–4):423–437. [https://doi.org/10.1016/0012-821X\(95\)00085](https://doi.org/10.1016/0012-821X(95)00085)
- 1071 86. Monteux, J., Jellinek, A. M., & Johnson, C. L. (2011). Why might planets and moons have early
1072 dynamos? *Earth and Planetary Science Letters*, 310(3–4), 349–359.
1073 <https://doi.org/10.1016/j.epsl.2011.08.014>

- 1074 87. Monteux, J., Arkani-Hamed, J., (2014). Consequences of giant impacts in early Mars: Core merging and
1075 Martian dynamo evolution. *J. Geophys. Res.* 119, 480–505.
- 1076 88. Monteux, J., Andrault, D., & Samuel, H. (2016). On the cooling of a deep terrestrial magma ocean.
1077 *Earth and Planetary Science Letters*, 448, 140–149. <https://doi.org/10.1016/j.epsl.2016.05.010>
- 1078 89. Monteux J., Andrault D., Guitreau M., Samuel H., Demouchy S. (2020). A mushy Earth's mantle for
1079 more than 500 Myr after the magma ocean solidification. *Geophysical Journal International*
1080 DOI:10.1093/gji/ggaa064
- 1081 90. Namur, O., Collinet, M., Charlier, B., Grove, T. L., Holtz, F., & McCammon, C. (2016). Melting
1082 processes and mantle sources of lavas on Mercury. *Earth and Planetary Science Letters*, 439, 117–128.
1083 <https://doi.org/10.1016/j.epsl.2016.01.030>
- 1084 91. Neuville, D. R., Courtial, P., Dingwell, D. B., & Richet, P. (1993). Thermodynamic and rheological
1085 properties of silicate melts. *Contributions to Mineralogy and Petrology*, 113, 572–581.
- 1086 92. Ni, H., Keppler, H., & Behrens, H. (2011). Electrical conductivity of hydrous basaltic melts:
1087 Implications for partial melting in the upper mantle. *Contributions to Mineralogy and Petrology*, 162(3),
1088 637–650. <https://doi.org/10.1007/s00410-011-0617-4>
- 1089 93. Ni, H., Hui H., & G. Steinle-Neumann (2015). Transport properties of silicate melts. *Rev. Geophys.*,
1090 53, 715–744, doi:10.1002/2015RG000485.
- 1091 94. Nimmo, F. (2002). Why does Venus lack a magnetic field? *Geology*, 30(11), 987–990.
1092 [https://doi.org/10.1130/0091-7613\(2002\)030<0987:WDVLAM>2.0.CO;2](https://doi.org/10.1130/0091-7613(2002)030<0987:WDVLAM>2.0.CO;2)
- 1093 95. Novella, D., Frost, D.J., Hauri, E.H., Bureau, H., Raepsaet, C., Roberge, M., (2014). The distribution of
1094 H₂O between silicate melt and nominally anhydrous peridotite and the onset of hydrous melting in the
1095 deep upper mantle. *Earth Planet Sci Lett* 400:1–13. <https://doi.org/10.1016/j.epsl.2014.05.006>.
- 1096 96. O'Rourke, J. G., Buz, J., Fu, R. R., & Lillis, R. J. (2019). Detectability of Remanent Magnetism in the
1097 Crust of Venus. *Geophysical Research Letters*, 46(11), 5768–5777.
1098 <https://doi.org/10.1029/2019GL082725>
- 1099 97. Pertermann, M., & Hofmeister, A. M. (2006). Thermal diffusivity of olivine-group minerals at high
1100 temperature. *American Mineralogist*, 91, 1747–1760.
- 1101 98. Picard, D., Arbaret, L., Pichavant, M., Champallier, R., & Launeau, P. (2013). The rheological transition
1102 in plagioclase-bearing magmas. *Journal of Geophysical Research: Solid Earth*, 118(4), 1363-1377.
- 1103 99. Pommier, A., Leinenweber, K., & Tran, T. (2019). Mercury 's thermal evolution controlled by an
1104 insulating liquid outermost core? *Earth and Planetary Science Letters*, 517, 125–134.
1105 <https://doi.org/10.1016/j.epsl.2019.04.022>
- 1106 100. Richet, P., & Fiquet, G. (1991). High-temperature heat capacity and premelting of minerals in the system
1107 MgO-CaO-Al₂O₃-SiO₂. *Journal of Geophysical Research*, 96(B1), 445–456.
1108 <https://doi.org/10.1029/90JB02172>
- 1109 101. Romine, W. L., Whittington, A. G., Nabelek, P. I., & Hofmeister, A. M. (2012). Thermal diffusivity of
1110 rhyolitic glasses and melts: Effects of temperature, crystals and dissolved water. *Bulletin of*
1111 *Volcanology*, 74(10), 2273–2287. <https://doi.org/10.1007/s00445-012-0661-6>
- 1112 102. Sakamaki, T., Suzuki, A., & Ohtani, E. (2006). Stability of hydrous melt at the base of the Earth's upper
1113 mantle. *Nature*, 439(7073), 192–194. <https://doi.org/10.1038/nature04352>
- 1114 103. Sambridge, M., Mosegaard, K., (2002). Monte Carlo methods in geophysical inverse problems. *Reviews*
1115 *of Geophysics*, 40(3), 3-1.
- 1116 104. Schiavi, F., Bolfan-Casanova, N., Withers, A. C., Médard, E., Laumonier, M., Laporte, D., et al. (2018).
1117 Water quantification in silicate glasses by Raman spectroscopy: Correcting for the effects of confocality,
1118 density and ferric iron. *Chemical Geology*, 483(February), 312–331.
1119 <https://doi.org/10.1016/j.chemgeo.2018.02.036>

- 1120 105. Schmeling, H., (1986). Numerical models on the influence of partial melt on elastic, anelastic and
 1121 electrical properties of rocks. Part II: electrical conductivity. *Phys. Earth Planet. Inter.* 43(2):123–136.
 1122 [https://doi.org/10.1016/0031-9201\(86\)90080-4](https://doi.org/10.1016/0031-9201(86)90080-4).
- 1123 106. Schubert, G., Chan, K. H., Liao, X., & Zhang, K. (2004). Planetary dynamos: Effects of electrically
 1124 conducting flows overlying turbulent regions of magnetic field generation. *Icarus*, 172(2), 305–315.
 1125 <https://doi.org/10.1016/j.icarus.2004.06.007>
- 1126 107. Shea, T., Houghton, B.F., Gurioli, L., Cashman, K.V., Hammer, J.E., Hobden, B.J., (2010). Textural
 1127 studies of vesicles in volcanic rocks: an integrated methodology. *J. Volcanol. Geotherm. Res.* 190(3–
 1128 4):271–289. <https://doi.org/10.1016/j.jvolgeores.2009.12.003>
- 1129 108. Smith, D. S., Fayette, S., Grandjean, S., Martin, C., Telle, R., & Tonnessen, T. (2003). Thermal
 1130 resistance of grain boundaries in alumina ceramics and refractories. *Journal of the American Ceramic
 1131 Society*, 86(1), 105–111. <https://doi.org/10.1111/j.1151-2916.2003.tb03285.x>
- 1132 109. Smith, D. S., Alzina, A., Bourret, J., Nait-Ali, B., Pennec, F., Tessier-Doyen, N., Otsu, K., Matsubara,
 1133 H., Elser, P., & Gonzenbach, U. T. (2013). Thermal conductivity of porous materials. *J. Mater. Res.*, 28
 1134 (17).
- 1135 110. Solomatov, V., (2007). Magma Oceans and Primordial Mantle Differentiation. Schubert, G. (Ed.),
 1136 *Treatise of Geophysics*, vol.9. Elsevier.
- 1137 111. Spera, F., J., (2000). Physical properties of magma. *Encyclopedia on Volcanoes*
- 1138 112. Stanley, S., Bloxham, J., Hutchison, W. E., & Zuber, M. T. (2005). Thin shell dynamo models consistent
 1139 with Mercury's weak observed magnetic field. *Earth and Planetary Science Letters*, 234(1–2), 27–38.
 1140 <https://doi.org/10.1016/j.epsl.2005.02.040>
- 1141 113. Stevenson, D. J., Spohn, T., & Schubert, G. (1983). Magnetism and thermal evolution of the terrestrial
 1142 planets. *Icarus*, 54(3), 466–489. [https://doi.org/10.1016/0019-1035\(83\)90241-5](https://doi.org/10.1016/0019-1035(83)90241-5).
- 1143 114. Stevenson, D. J. (1987). Mercury's magnetic field: A thermoelectric dynamo? *Earth and Planetary
 1144 Science Letters*, 82(1–2), 114–120. [https://doi.org/10.1016/0012-821X\(87\)90111-7](https://doi.org/10.1016/0012-821X(87)90111-7).
- 1145 115. Strom, R. G., & Sprague, A. L. (2003). *Exploring Mercury: the iron planet*. Springer Science & Business
 1146 Media.
- 1147 116. Tarduno, J.A., Cottrell, R.D., Watkeys, M.K., Hofmann, A., Doubrovine, P.V., Mamajek, E.E., Liu, D.,
 1148 Sibeck, D.G., Neukirch, L.P. and Usui, Y., (2010). Geodynamo, solar wind, and magnetopause 3.4 to
 1149 3.45 billion years ago. *science*, 327(5970), pp.1238-1240.
- 1150 117. Tian, Z., Zuber, M. T., & Stanley, S. (2015). Magnetic field modeling for Mercury using dynamo models
 1151 with a stable layer and laterally variable heat flux. *Icarus*, 260, 263–268.
 1152 <https://doi.org/10.1016/j.icarus.2015.07.019>.
- 1153 118. Tikoo, S. M., Weiss, B. P., Cassata, W. S., Shuster, D. L., Gattacceca, J. Ô., Lima, E. A., et al. (2014).
 1154 Decline of the lunar core dynamo. *Earth and Planetary Science Letters*, 404, 89–97.
 1155 <https://doi.org/10.1016/j.epsl.2014.07.010>
- 1156 119. Tyburczy, J. A., & Waff, H. S. (1983). electrical conductivity of molten basalt and andesite to 25 kilobars
 1157 pressures: geophysical significance and implications for charge transport and melt structure. *Journal of
 1158 Geophysical Research*, 88(2), 2413–2430.
- 1159 120. Von Bargen, N., Waff, H.S., (1986). Permeabilities, interfacial areas and curvatures of partially molten
 1160 systems: results of numerical computations of equilibrium microstructures. *J. Geophys. Res.* 91:9261–
 1161 9276.
- 1162 121. Waff, H.S., (1974). Theoretical consideration of electrical conductivity in a partially molten mantle and
 1163 implications for geothermometry. *J. Geophys. Res.* 79(26):4003–4010.
- 1164 122. Watanabe, H. (1982). Thermochemical properties of synthetic high-pressure compounds relevant to the
 1165 Earth's mantle.

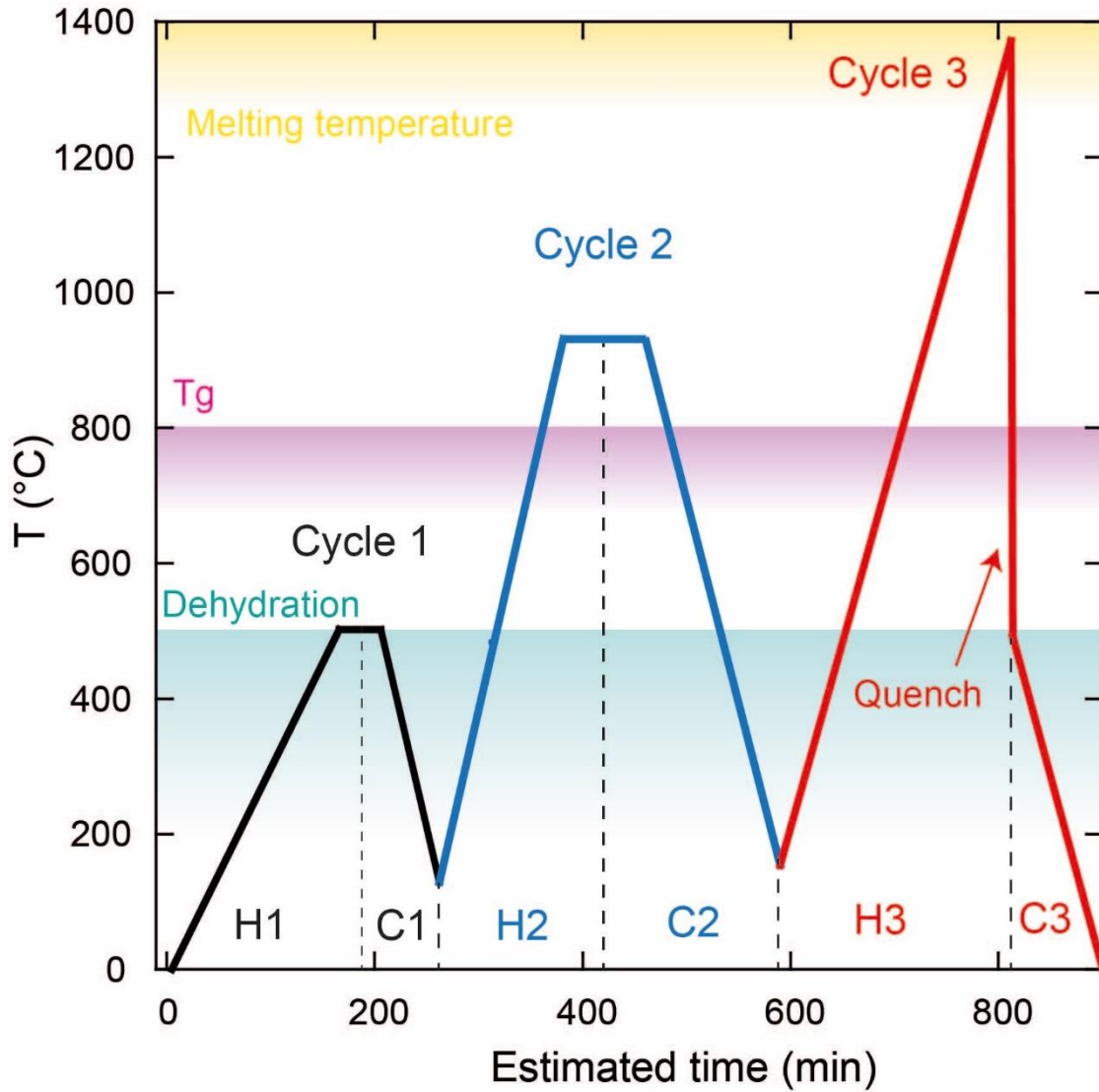
- 1166 123. Watanabe, T., Kurita, K. (1993). The relationship between electrical conductivity and melt fraction in a
1167 partially molten simple system: Archie's law behavior. *Physics of the earth and planetary interiors*, 78(1-
1168 2), 9-17.
- 1169 124. Watson, E.B., Brenan, J.M., Baker, D.R., (1991). *Continental Mantle*. Oxford Univeristy Press.
- 1170 125. Weidenschilling, S. J. (1978). Iron/silicate fractionation and the origin of Mercury. *Icarus*, 35(1), 99–
1171 111. [https://doi.org/10.1016/0019-1035\(78\)90064-7](https://doi.org/10.1016/0019-1035(78)90064-7)
- 1172 126. Weiss, B. P., Berdal, James, S., Elkins-Tanton, L., Stanley, S., Lima, E. A., & Carporzen, L. (2008).
1173 Magnetism on the Angrite Differentiation of Planetesimals. *Scientific Reports*, 322(October), 713–716.
- 1174 127. Wurm, G., Trieloff, M., & Rauer, H. (2013). Photophoretic separation of metals and silicates: The
1175 formation of mercury-like planets and metal depletion in chondrites. *Astrophysical Journal*, 769(1).
1176 <https://doi.org/10.1088/0004-637X/769/1/78>
- 1177 128. Xu, Y., Shankland, T. J., Linhardt, S., Rubie, D. C., Langenhorst, F., & Klasinski, K. (2004). Thermal
1178 diffusivity and conductivity of olivine, wadsleyite and ringwoodite to 20 GPa and 1373 K. *Physics of
1179 the Earth and Planetary Interiors*, 143(1–2), 321–336. <https://doi.org/10.1016/j.pepi.2004.03.005>
- 1180 129. Yoshino, T., Takei, Y., Wark, D.A., Watson, E.B., (2005). Grain boundary wetness of texturally
1181 equilibrated rocks, with implications for seismic properties of the upper mantle. *J. Geophys. Res. B
1182 Solid Earth* 110(8):1–16. <https://doi.org/10.1029/2004J B003544>.
- 1183 130. Yoshino, T., Nishihara, Y., Karato, S. I, 2007. Complete wetting of olivine grain boundaries by a
1184 hydrous melt near the mantle transition zone. *Earth Planet. Sci. Lett.* 256, 466–472.
1185 <https://doi.org/10.1016/j.epsl.2007.02.002>
- 1186 131. Zebib, A., Schubert, G., Dein, J.L., & Paliwal, R.C. (1983) Character and stability of axisymmetric
1187 thermal convection in spheres and spherical shells. *Geophys. Astrophys. Fluid Dyn.* 23, 1–42.
- 1188 132. Baohua, Z., Ge, J., Xiong, Z. & Zhai, S., (2019). Effect of water on the thermal properties of olivine
1189 with implications for lunar internal temperature. *J.Geo.Res. Planets*, 124 (12), 3469-3481. doi:
1190 10.1029/2019JE006194
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1192 **Figure 1:** Schematic cross-section of the high-pressure cell assembly used for Angstrom method with
 1193 multi-anvil apparatus. For glass and melt experiments, thermocouples were inserted in alumina tubes
 1194 of 0.6 mm diameter and 4.5 mm length in the sample zone.
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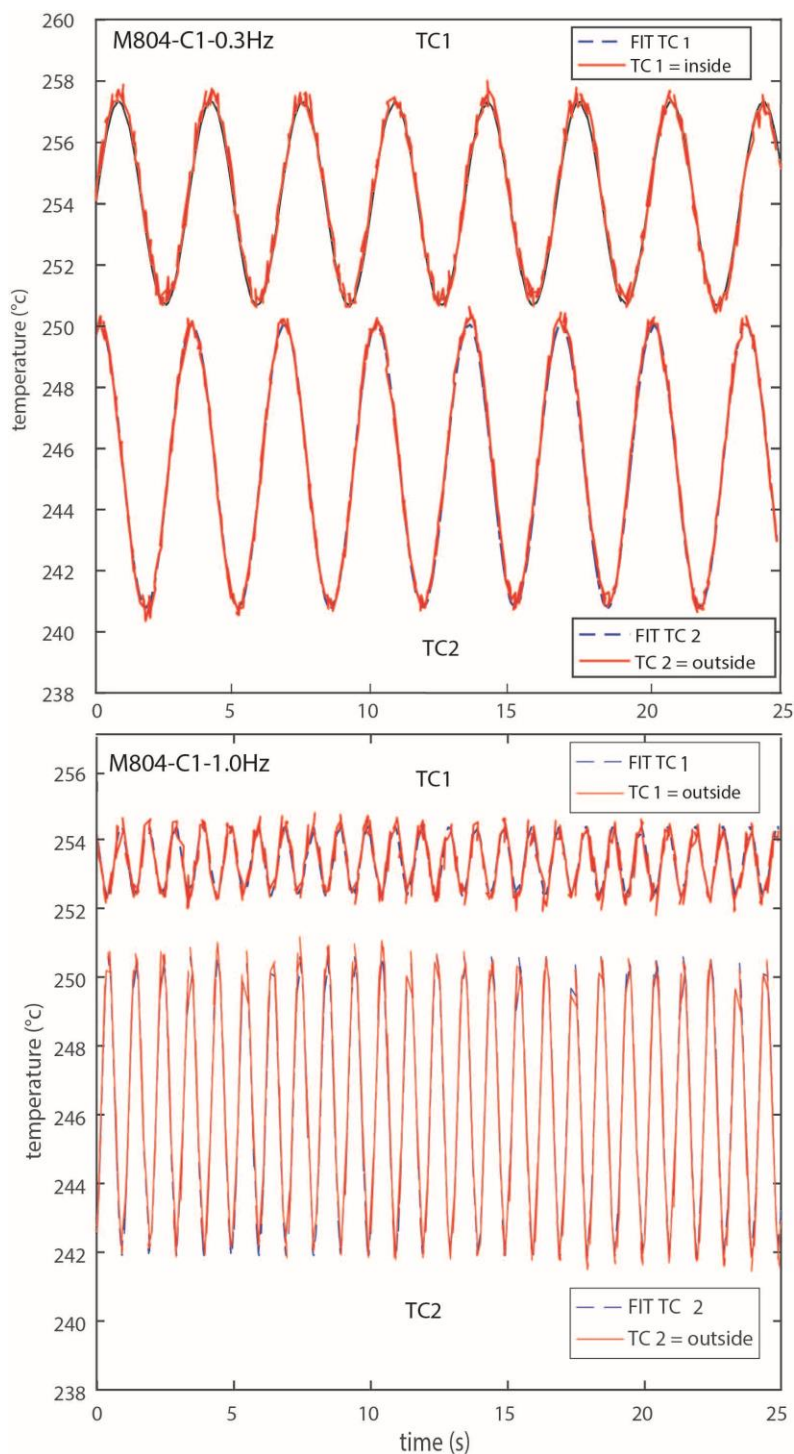
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1210 **Figure 2:** Experimental procedure for the measurement of thermal diffusivity of peridotites, glasses
1211 and melts: black, blue and red for cycles 1, 2 and 3 respectively. Heating (H) and cooling (C) parts
1212 of the cycles are separated by black dashed lines (for example H1: 1st cycle heating). The expected
1213 temperatures for sample dehydration (up to ~500°C), glass transition and melting are illustrated in
1214 shaded colors: green, pink and yellow respectively.
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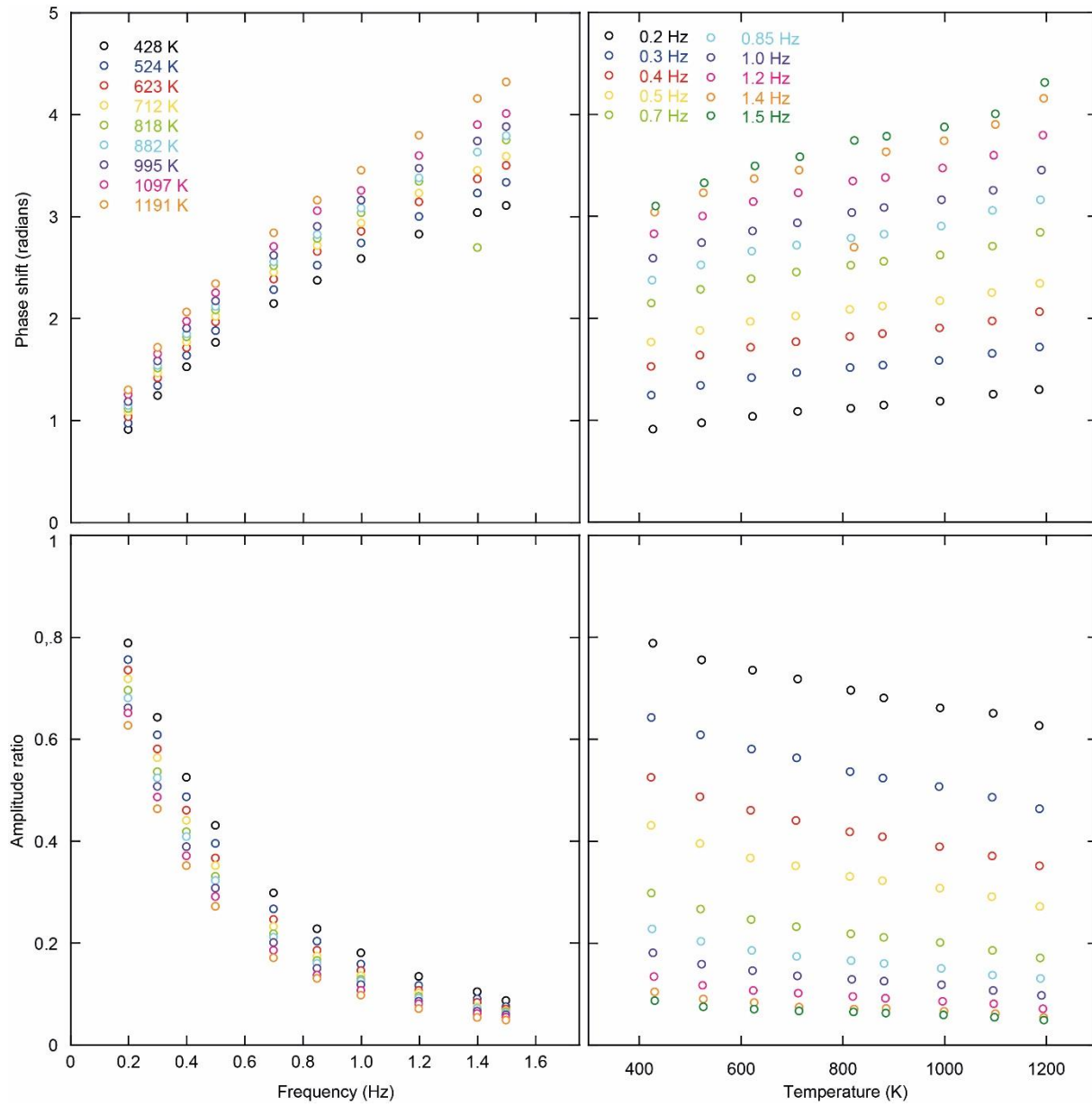
1220 **Figure 3:** Examples of temperature waves recorded at high temperature during M804 experiments.
1221 Such data set is acquired after the sample equilibration at target temperature conditions for a few
1222 minutes (here $T = 250^{\circ}\text{C}$). Acquisitions are performed at an interval ranging from 40 ms to 200 ms.
1223 At least 10 periods of the temperature oscillation are recorded to ensure a good extraction of phase
1224 shifts and amplitude ratio based on the fitting procedure described in Methods. Upper and lower
1225 frames correspond to temperature measurements performed at frequencies of 0.3 and 1 Hz,
1226 respectively. Red symbols are measured data and the dash thick lines represent the fitted sine waves.
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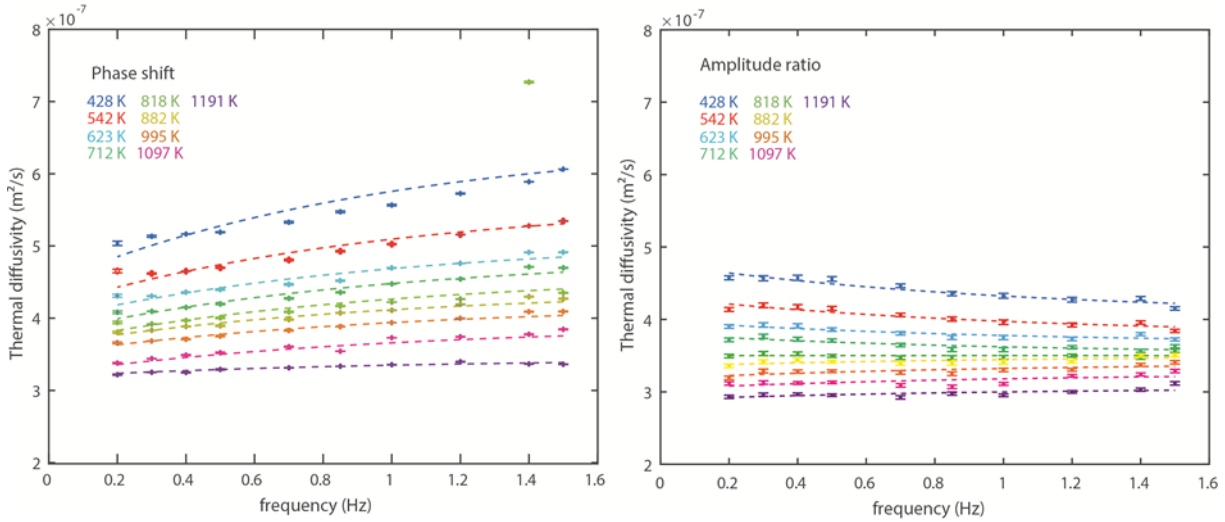
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Figure 4: Example of refined phase shifts (top panels) and amplitude ratio (bottom panels) as a function of frequency (from 0.2 to 1.5 Hz, left panels) and temperature (from 428 K to 1191 K, right panels) for sample M844.



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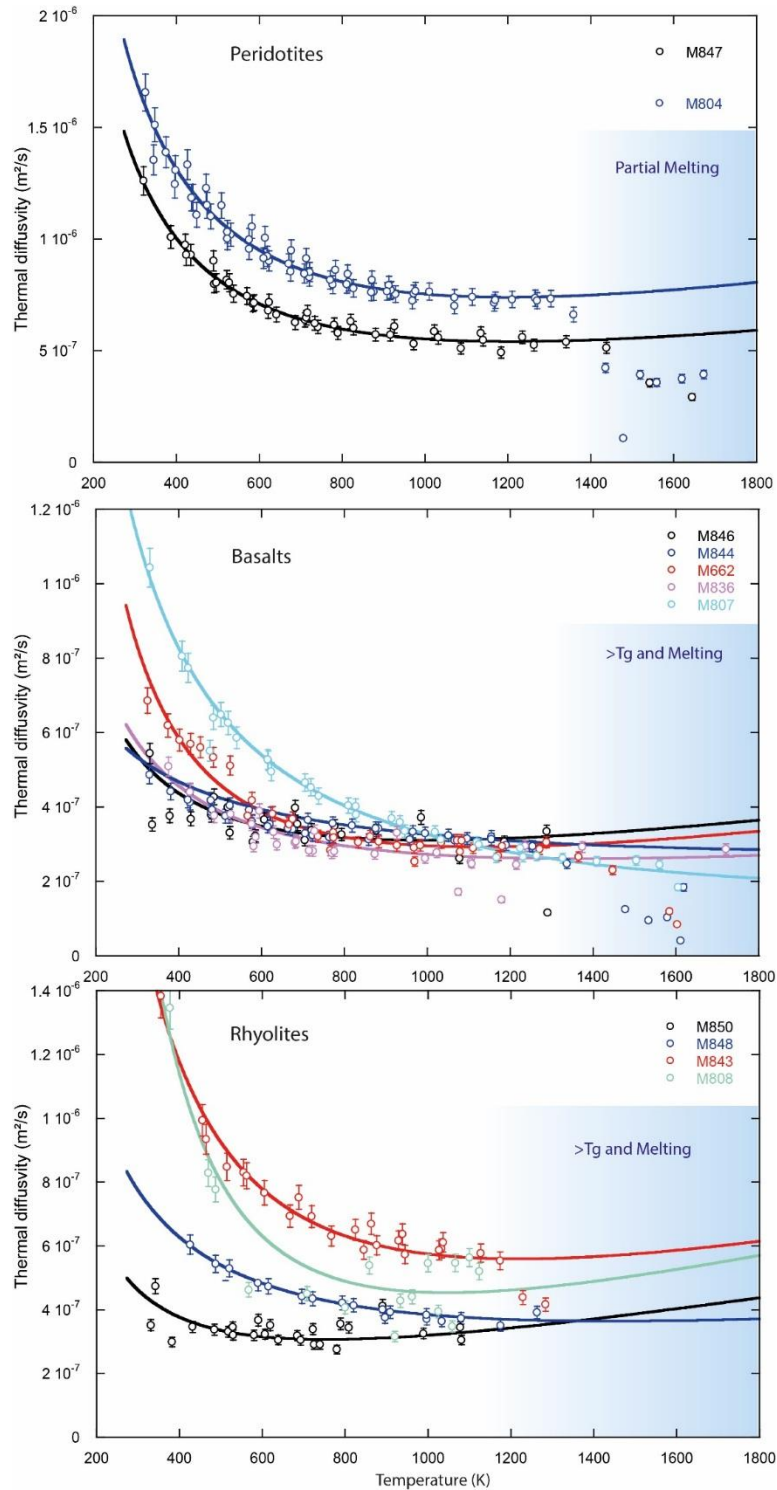
1236 **Figure 5:** Typical example of a set of raw values of the thermal diffusivity extracted from our
 1237 measurements for a basaltic glass, presented as a function of the frequency of the heat source.
 1238 Experimental temperatures range between ~430 and ~1200 K at a pressure of 2GPa (sample M844).
 1239 Left and right panels correspond to thermal diffusivities calculated from phase shifts and amplitude
 1240 ratios, respectively. The color code is kept similar in both panels. The fit through the data is made
 1241 following the equation $D = D_0 + A \exp(-f/f_0)$ (Xu et al., 2004), where D_0 is the high-frequency
 1242 asymptote of the diffusivity (see text). Error bars are generally within the size of the symbol.
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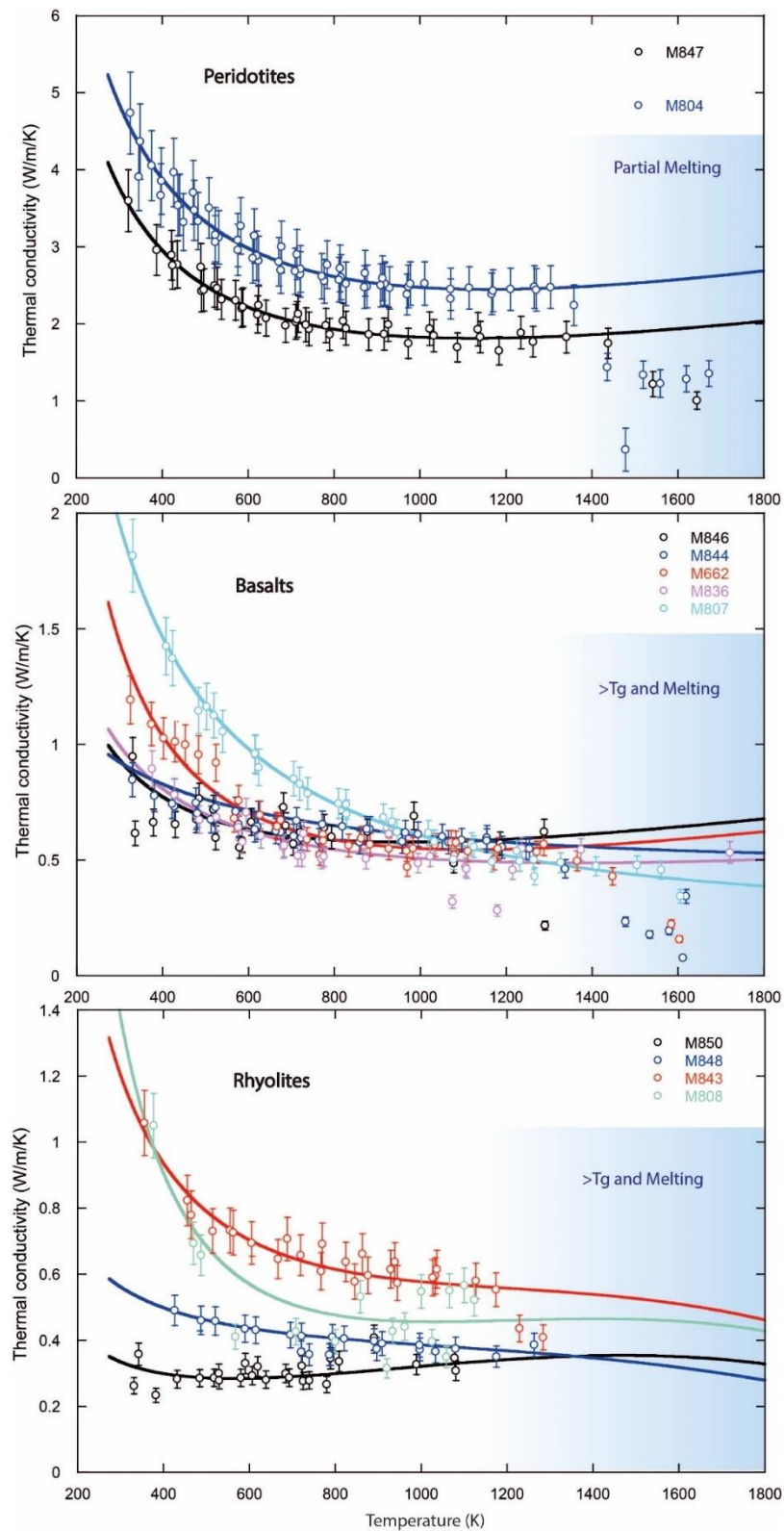
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Figure 6: Thermal diffusivities data refined for our peridotites, basalts and rhyolites samples (from top to bottom) as a function of temperature. Associated fits are represented with a solid line of the same color than the symbols. Error bars are shown when larger than the symbol size. Several basaltic (M807, M662) and rhyolite (M848, M808) samples were partially recrystallized during the annealing and dehydration procedures prior to the conductivity measurement, as indicated by their high conductivity values at low temperatures. Thermal diffusivities values converge at high temperatures and melting.



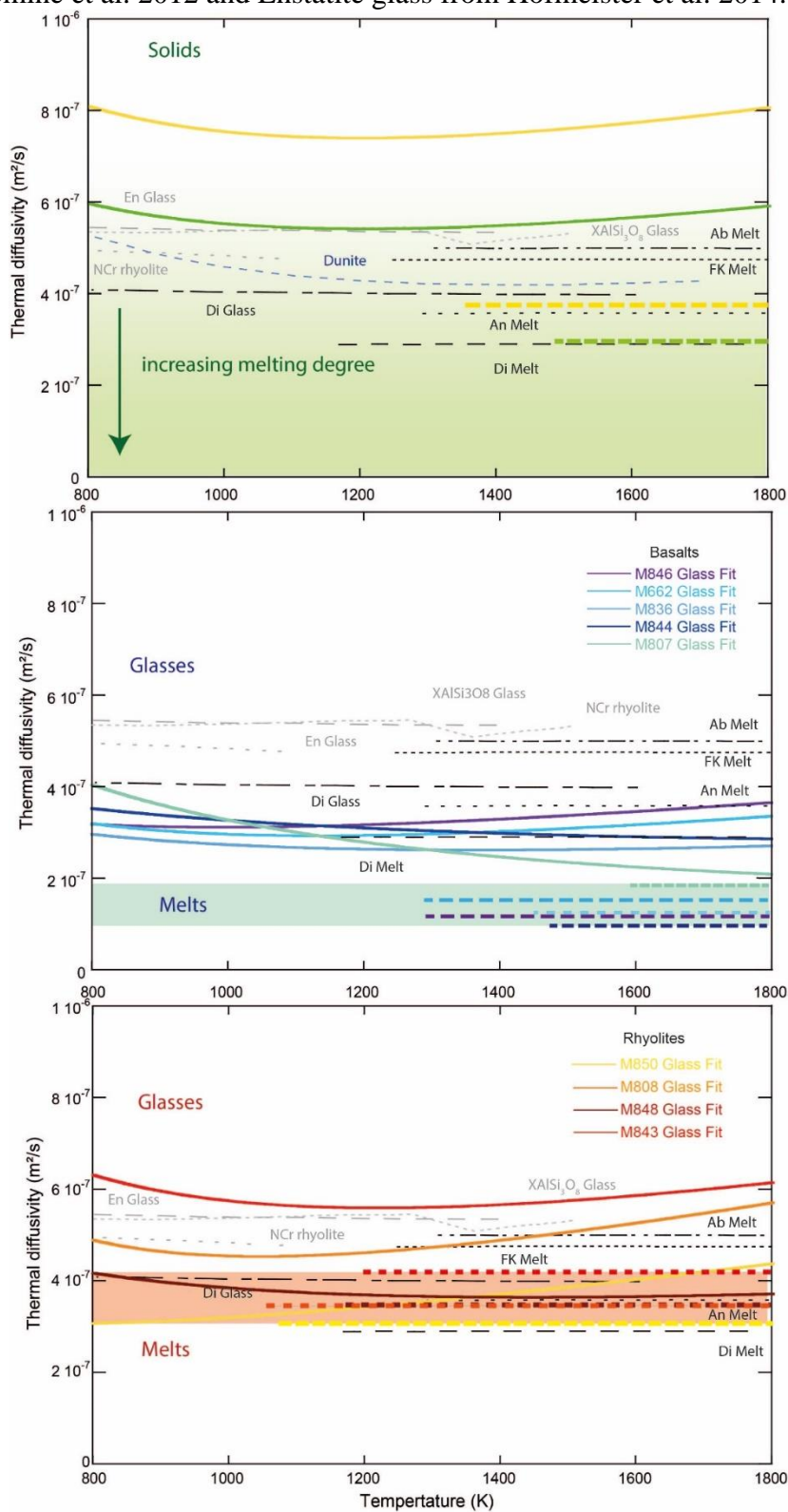
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1257 **Figure 7:** Thermal conductivities inferred from diffusivities (Figure 6), heat capacities and densities
1258 (see Figure S4), based on Eq. 17 of the main text.
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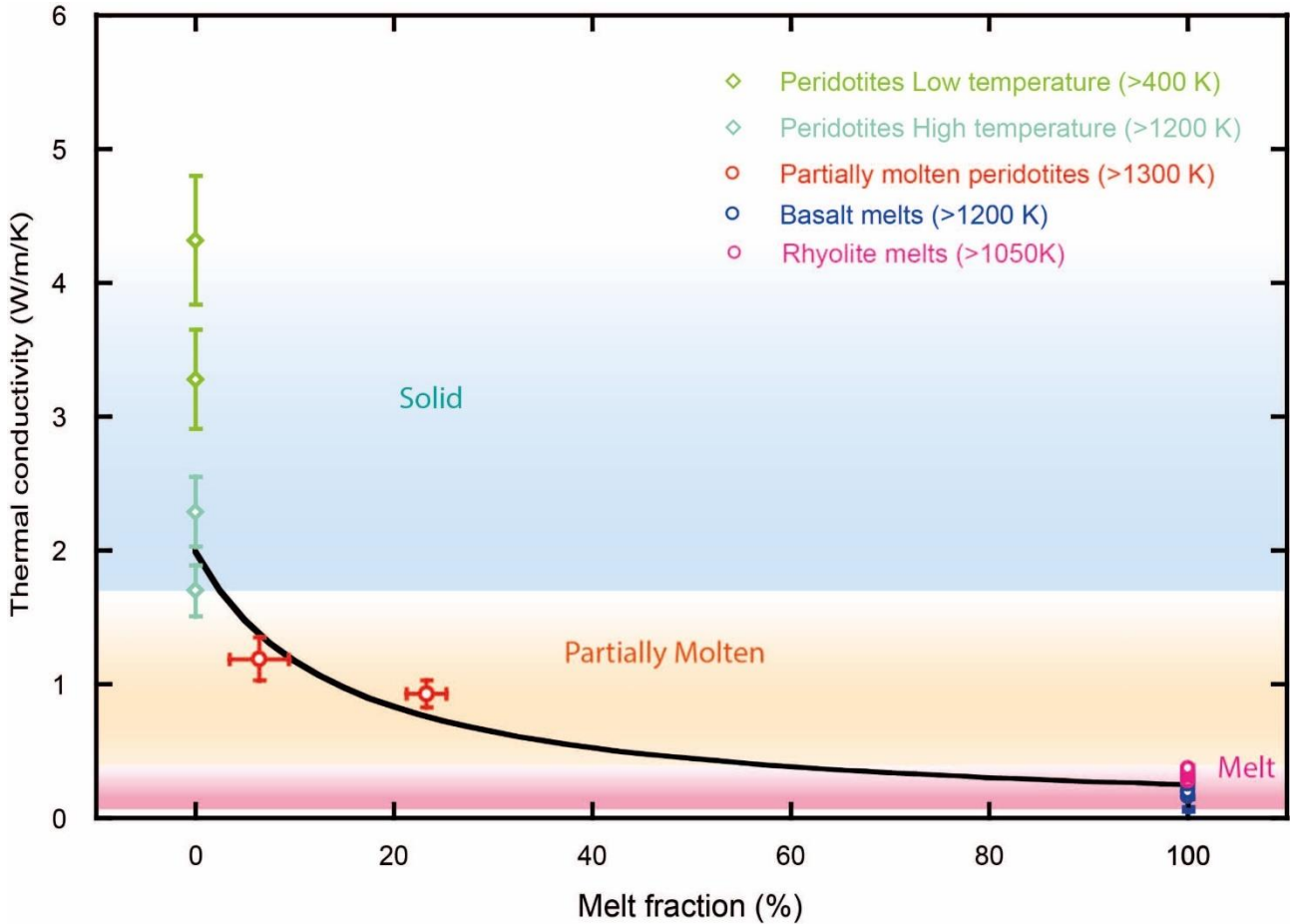
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1262 **Figure 8:** Comparison with the literature of our thermal diffusivities obtained our peridotite (upper panel), basaltic (center panel) and rhyolitic (lower panel) starting materials. Solids (peridotite and
 1263 glasses) and melts are represented in thick lines and thick dashed line, respectively. Literature data
 1264 are reported with thinner dashed lines (grey to black). Diopside (Di) glass and melt, XAlSi₃O₈ glass,
 1265 Anorthite melt (An), Albite melt (Ab) and orthose melt (FK) from Hofmeister et al. 2009, NCr
 1266 Rhyolite from Romine et al. 2012 and Enstatite glass from Hofmeister et al. 2014.
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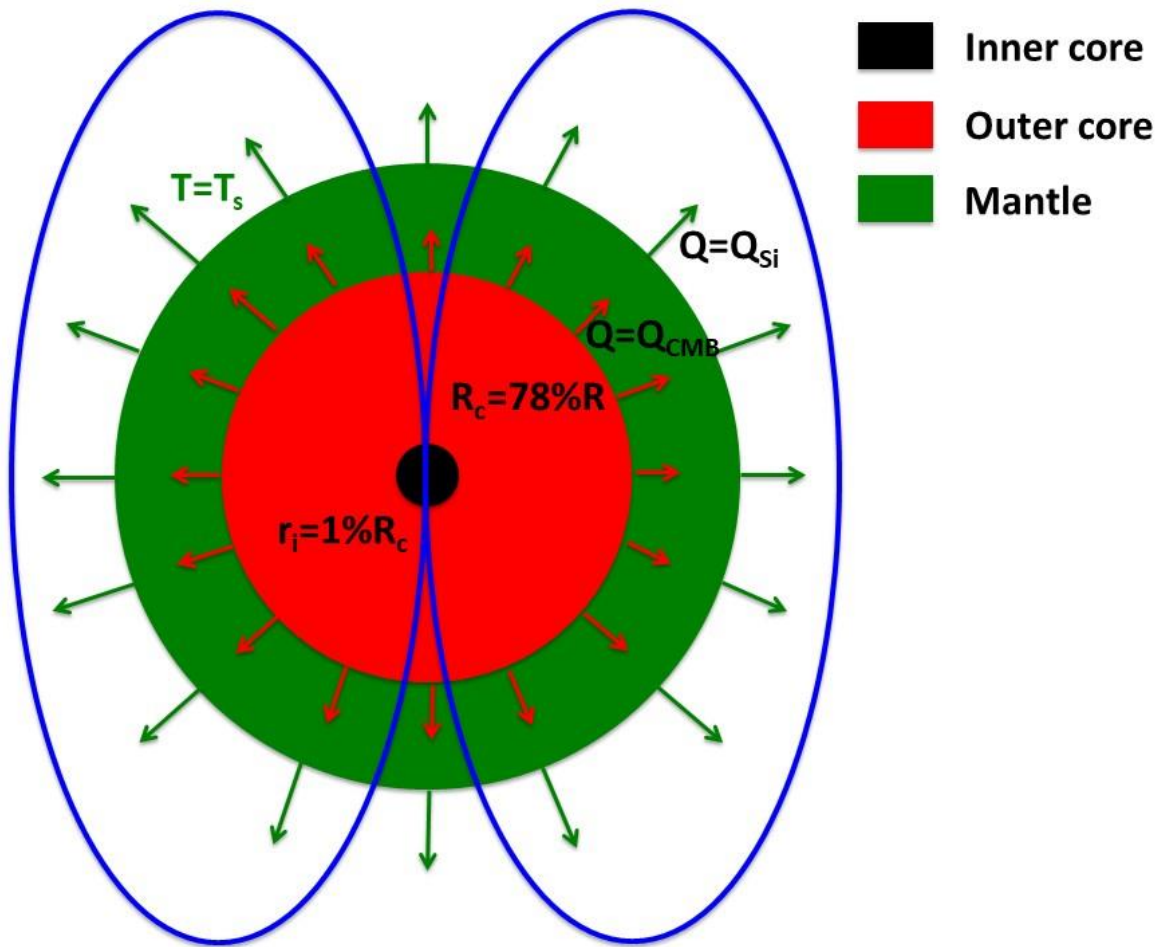
1269 **Figure 9:** Our best model of the evolution of thermal diffusivity as a function of the melt fraction at
1270 high temperature (>1200 K), based on all available data sets. Solid peridotites at low and high
1271 temperature are represented with green and blue diamonds, respectively. Partially molten peridotites
1272 with 6.4% and 23.3% of melt (see Supplementary Text S3) are represented with red circles and melts
1273 (basalts and rhyolites) are represented with blue and pink circles. All the errors represented are 1
1274 standard deviation. The fit represented is using a model of thermal resistors in series (Eq. 22), which
1275 fits our data set very well (see Figure S14).
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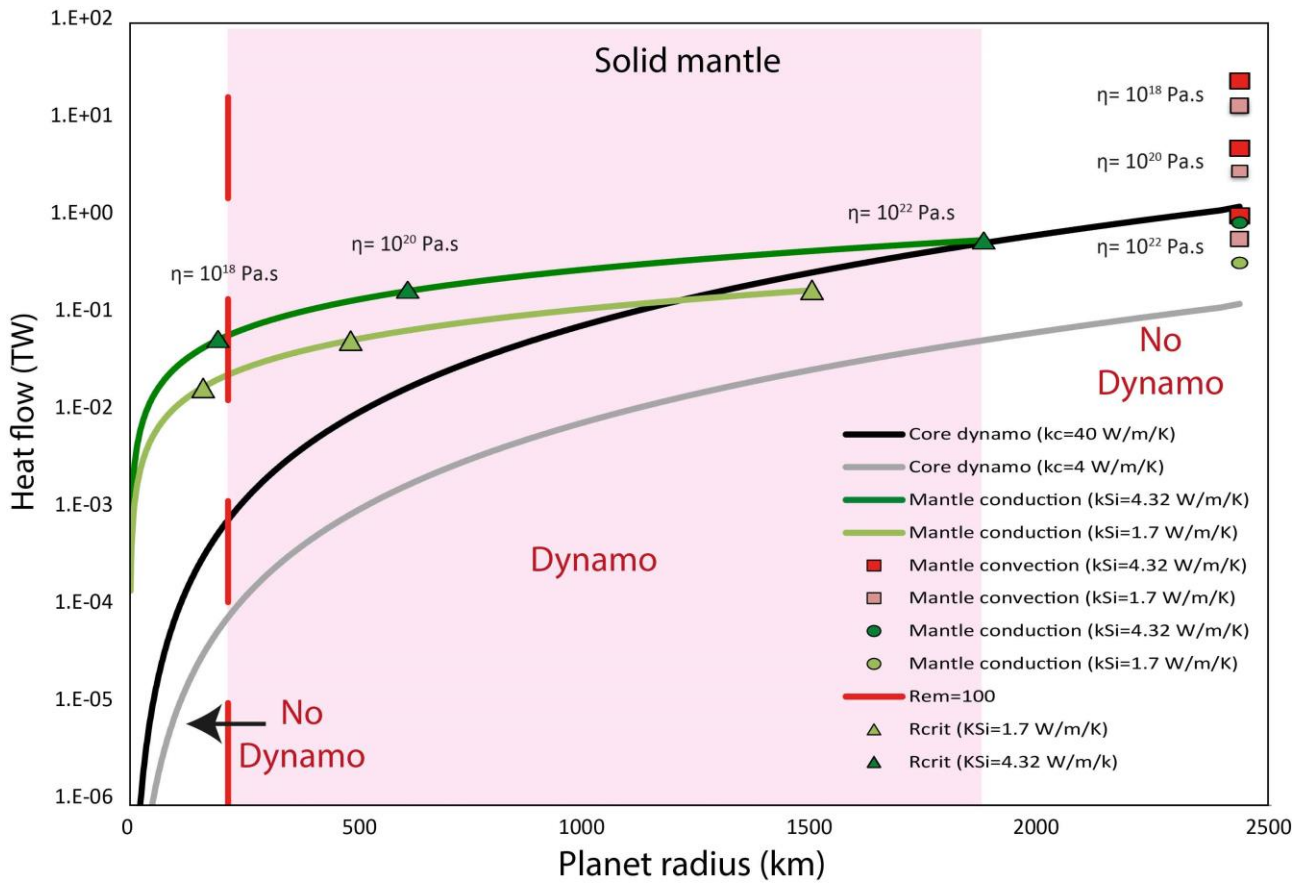
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Figure 10: Schematic representation of the geometry and of the heat flows computed in our model.



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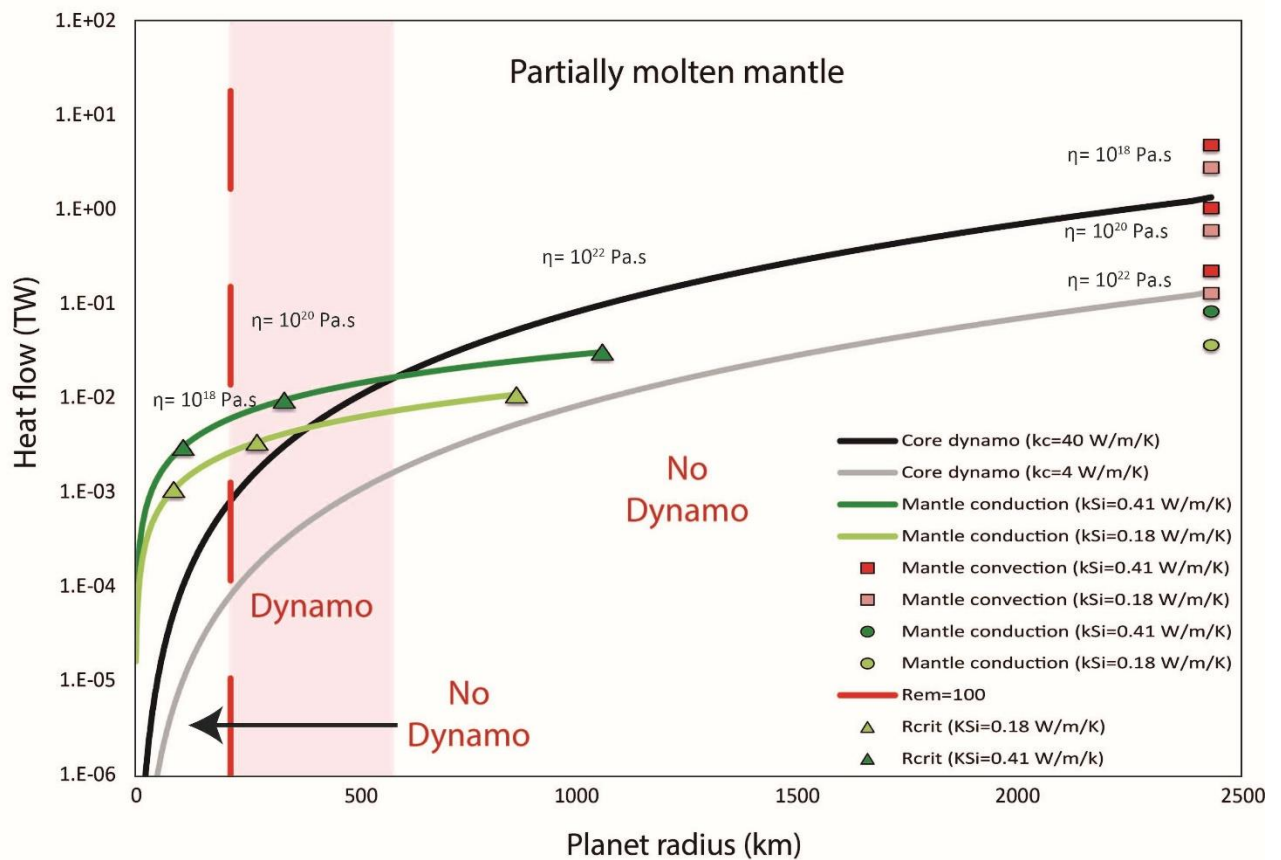
1282 **Figure 11:** Heat flows as a function of the planetary radius for a fully solid planetary mantle. The
 1283 black and grey lines represent the critical heat flow needed for a thermally driven dynamo for $k_c = 40$
 1284 and 4 W/m/K , respectively (Conditions 1 and 2, Eq. 31). The dark and light green lines represent the
 1285 mantle heat flow in a conductive regime for $\kappa_{Si} = 4.32$ and 1.70 W/m/K , respectively (Eq. 35). Green
 1286 triangles represent the critical planet size at which $Ra = Ra_c$, therefore corresponding to the minimum
 1287 heat flow at the CMB, for mantle viscosity ranging from 10^{18} to 10^{22} Pa.s . Extrapolated values at $R = R_{\text{Mercury}} = 2440 \text{ km}$
 1288 are also represented with green circles for conductive heat transfers and red and
 1289 pink squares for convective cases (eq. 36) with mantle viscosities indicated nearby. The red dashed
 1290 line represents the critical value for $Rem = 100$ (Condition 3). At the left side of this line, a dynamo
 1291 is unlikely. The likely zone for a planet radius capable of powering a dynamo is represented by the
 1292 pink shaded area in case $\kappa_{Si} = 4.32 \text{ W/m/K}$ and $\kappa_{\text{core}} = 40 \text{ W/m/K}$.



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1305 **Figure 12.**
 1306 Heat flows as a function of the planetary radius for a stagnant partially molten planetary mantle. The
 1307 black and grey lines represent the critical heat flow for a thermally driven dynamo for $k_c = 40$ and 4
 1308 W/m/K , respectively (Conditions 1 and 2, Eq. 31). The dark and light green lines represent the mantle
 1309 heat flow for $\kappa_{\text{Si}} = 0.41$ and 0.18 W/m/K , respectively (Eq. 35). Green triangles represent the critical
 1310 planet size at which $\text{Ra} = \text{Ra}_c$, for mantle viscosity ranging from 10^{18} to 10^{22} Pa.s. Extrapolated values
 1311 at $R = R_{\text{Mercury}} = 2440 \text{ km}$ are also represented with green circles for conductive heat transfers and red
 1312 and pink squares for convective cases (Eq. 36) with viscosities indicated nearby. The red dashed line
 1313 represents the critical value for $\text{Re}_m = 100$ (Condition 3). At the left side of this line, a dynamo is
 1314 unlikely. The likely zone for a planet radius capable of powering a dynamo is represented by the pink
 1315 shaded area in case $\kappa_{\text{Si}} = 0.41 \text{ W/m/K}$ and $\kappa_{\text{core}} = 40 \text{ W/m/K}$.



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1318 **Table 1:** Sum up of our Ansgtröm experiments with compositions, P-T ranges and associated
 1319 parameters.

Run#	M662	M804	M807	M808	M836	M843	M844	M846	M847	M848	M850
Sample name	EPR-MORB	RP4-S2	DR07-S2	GD-S1	Haplo-S1	ATHO-S1	DR11-S1	DR11-S2	RP4-S3	ATHO-S2	GD-S2
Composition	MORB	Peridotite	MORB	Rhyolite	haplobasalt	Rhyolite	MORB	MORB	Peridotite	Rhyolite	Rhyolite
Pressure (GPa)	2	2	2	2	2	2	2	2	2	2	2
length before (µm)	1400	1771.2	1061.9	1098.8	1124.6	1541.9	1141.3	1048.2	1652.9	1045.1	1336.4
length after (µm)	1334.2	1780.0	-	1462.7	1213.2	1666.1	1287.6	1324.9	1643.6	1309.1	1287.4
expected Melting T° (K)	>1600	>1523	>1600	>1173	>1723	>1173	>1600	>1600	>1523	>1173	>1173
Max T (K)	1614	1681	1612	1146	1769	1219	1609	1284	1573	1325	1073
Run duration (min)	544	1089	953	480	832	800	834	439	767	761	610

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Table 2: Fitting parameters obtained for the last cycle (C2-H3) of each experiment. Associated errors (1 STD) are given in parenthesis

Run#	a	b	c	R ²	R ² adj
Peridotites					
M804	7.0e-4 (2.2e-4)	1.06 (0.06)	2.97e-10 (2.8e-11)	0.9659	0.9647
M847	8.8e-4 (7e-4)	1.15 (0.05)	2.37e-10 (1.8e-11)	0.9763	0.9750
Basalts					
M846	8.6e-6 (1.5e-5)	0.5 (0.3)	9.7e-11 (8.4e-11)	0.3933	0.3294
M662	7.2e-4 (1.8e-4)	1.20 (0.04)	1.01e-10 (4.1e-12)	0.9964	0.9962
M836	5.6e-5 (7.6e-5)	0.84 (0.23)	1.03e-10 (3.5e-11)	0.5748	0.5421
M844	1.9e-5 (1.1e-5)	0.63 (0.10)	8.0e-11 (2.3e-11)	0.9608	0.9556
M807	8.1e-4 (1.9e-4)	2.15 (0.04)	4.1e-11 (9e-12)	0.9933	0.9929
Rhyolites					
M850	1.9e-5 (3.5e-5)	0.70 (0.32)	1.7e-10 (7e-11)	0.2788	0.2101
M808	1.8e-2 (7e-2)	1.7 (0.7)	2.3e-10 (1.4e-10)	0.8003	0.7696
M848	1.4e-4 (7e-5)	0.91 (0.09)	1.3e-10 (2.0e-11)	0.9851	0.9826
M843	5.1e-3 (3.3e-3)	1.42 (0.11)	3.18e-10 (3.6e-11)	0.9715	0.9684

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1338 **Table 3:** Thermal diffusivities and conductivities retrieved for our samples at low and high
 1339 temperatures for solids, above Tg and/or melting temperature (indicated for each group) for melts and
 1340 partially molten samples. Literature estimates are also provided on the bottom part: "Xu" refers to
 1341 polycrystalline olivine (dunite) measurements performed at ambient pressure (Xu et al., 2004).
 1342 "Per06" refers to average values from 3 oriented (100, 010, 001) single crystal measurements at room
 1343 pressure (Pertermann and Hofmeister, 2006) and some dunite values are also available in the same
 1344 study. "Beck78" refers to Dunite measurements performed at 2 GPa (Beck et al. 1978). Regarding
 1345 melts, literature data are taken from Hofmeister et al. (2009), Di: diopside, An : anorthite, Fk : orthose,
 1346 Ab : albite. Finally, data from a few rhyolites Ncar, NCr and SCE are selected from Romine et al.
 1347 (2012). When literature data are available in the form of thermal diffusivities, their conductivities are
 1348 obtained using either the Cp and ρ provided in the source study or from our own parameters if not
 1349 available (see Supplementary Materials).

SAMPLE NAME	Temperature range	Melt fraction	STD	κ	STD	D	STD
		%	%	W/m/K	W/m/K	m ² /s	m ² /s
SOLIDS							
M804	LT < 400K	0	0	4.32	0.48	1.66E-06	2.05E-09
M847	LT < 400K	0	0	3.28	0.37	1.26E-06	3.34E-09
M804	HT >1200 K	0	0	2.29	0.26	7.33E-07	1.80E-09
M847	HT >1200 K	0	0	1.70	0.19	5.39E-07	6.13E-10
Partially molten samples							
M804	>1300 K	6.44	2.98	1.19	0.16	3.74E-07	2.55E-08
M847	>1300 K	23.31	2	0.93	0.10	2.92E-07	1.22E-10
MELTS							
BASALTS	MELT >1200K						
M807	>1224 K	100	0	0.34	0.02	1.84E-07	1.51E-09
M844	>1326 K	100	0	0.18	0.01	9.52E-08	2.96E-11
M662	>1450 K	100	0	0.22	0.16	1.20E-07	8.73E-08
M846	>1200 K	100	0	0.22	0.12	1.16E-07	6.47E-08
M836	1180 K	100	0	0.28	0.02	1.51E-07	6.94E-10
rhyolite	Melt >1000 K						
M850	> 1080 K	100	0	0.31	0.03	3.05E-07	2.42E-08
M808	>1060 K	100	0	0.35	0.03	3.47E-07	5.95E-10
M848	>1260 K	100	0	0.39	0.03	3.91E-07	6.21E-10
M843	> 1280 K	100	0	0.41	0.03	4.16E-07	2.36E-10
literature							
SOLIDS							
Olivine Xu 0 GPa LT	294	0	0	4.15	0.42	4.10E-06	2.05E-07
Olivine Xu 0 GPa HT	1377	0	0	1.95	0.19	5.00E-07	2.50E-08
Olivine Hof 16 AVG	300 K	0	0	6.37	0.64	2.37E-06	4.75E-08
Olivine Hof AVG16	>1500 K	0	0	2.76	0.28	7.01E-07	1.40E-08
Dunite Per 06	300 K	0	0	4.22	0.42	1.57E-06	3.15E-08
Dunite 2 Per 06	300 K	0	0	5.35	0.53	1.99E-06	3.99E-08
Dunite Per 06	>1500 K	0	0	1.65	0.17	4.20E-07	8.40E-09
Dunite 2 Per 06	>1500 K	0	0	2.23	0.22	5.65E-07	1.13E-08
Dunite Beck 78 2Gpa	300 K	0	0	4.15	0.42		
Dunite Beck 78 2Gpa	300 K	0	0	5.49	0.55		
Dunite 2 GPa Dry Z19	300 K	0	0	5.21	0.26	1.71E-06	8.55E-08
Dunite 2 GPa 0.08% wt H ₂ O Z19	300 K	0	0	4.86	0.24	1.59E-06	7.95E-08
Melts							
Di Hof 09	>1140 K	100	0	1.25	0.13	2.89E-07	5.79E-09
An Hof 09	>1290 K	100	0	1.45	0.15	3.58E-07	7.16E-09
Fk Hof 09	>1245 K	100	0	1.45	0.15	4.75E-07	9.50E-09
Ab Hof 09	>1300 K	100	0	1.60	0.16	4.99E-07	9.99E-09
Rhyolite/obsidian							
Ncar Ro 12	1384	100	0	1.53	0.15	5.17E-07	1.03E-08
NCr Ro 12	1365	100	0	1.50	0.15	5.08E-07	1.02E-08
SCE Ro 12	1464	100	0	1.59	0.16	5.31E-07	1.06E-08

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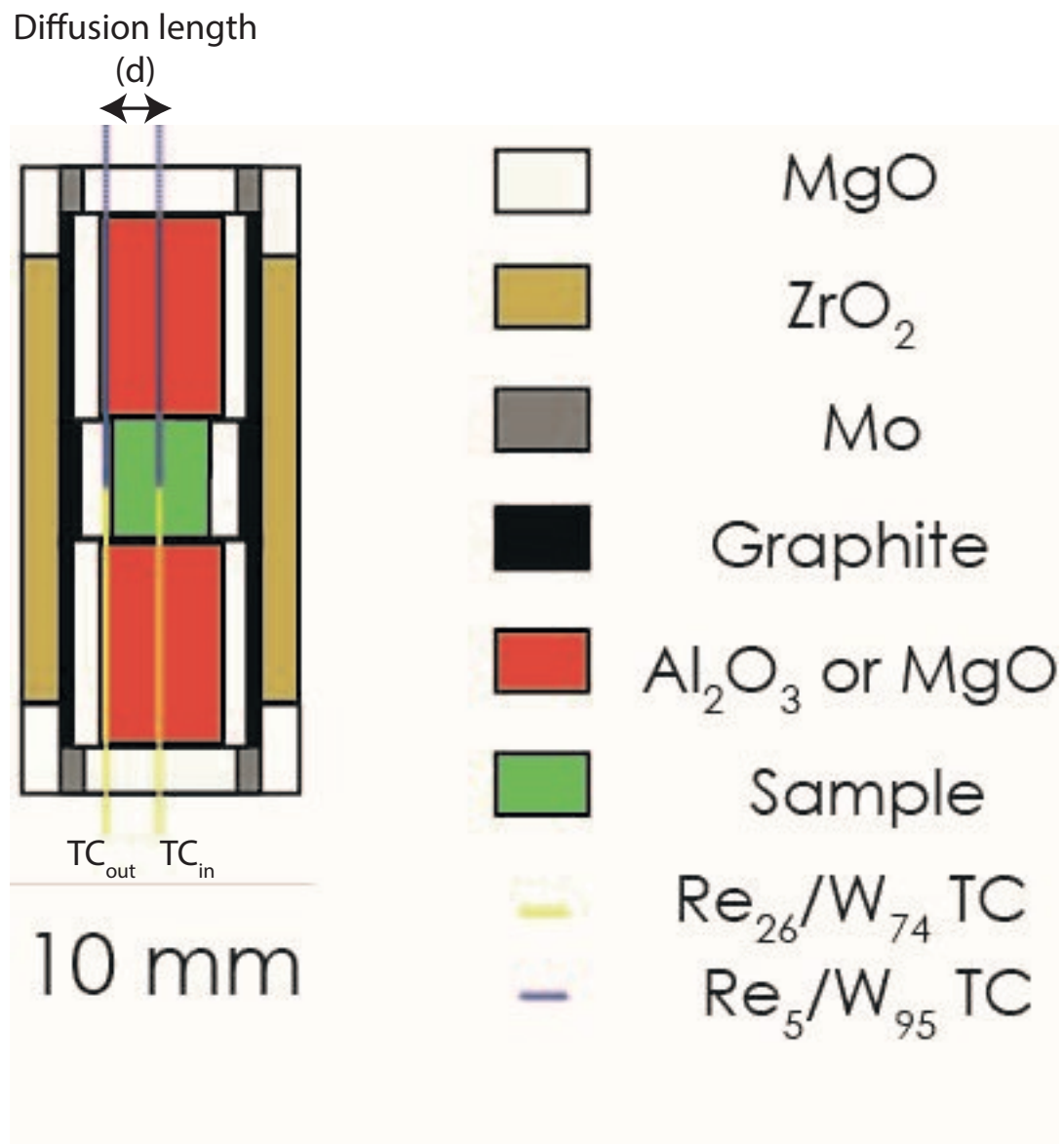
1351 **Table 4:** Typical parameter values for heat fluxes calculations displayed in Figures 11 and 12.

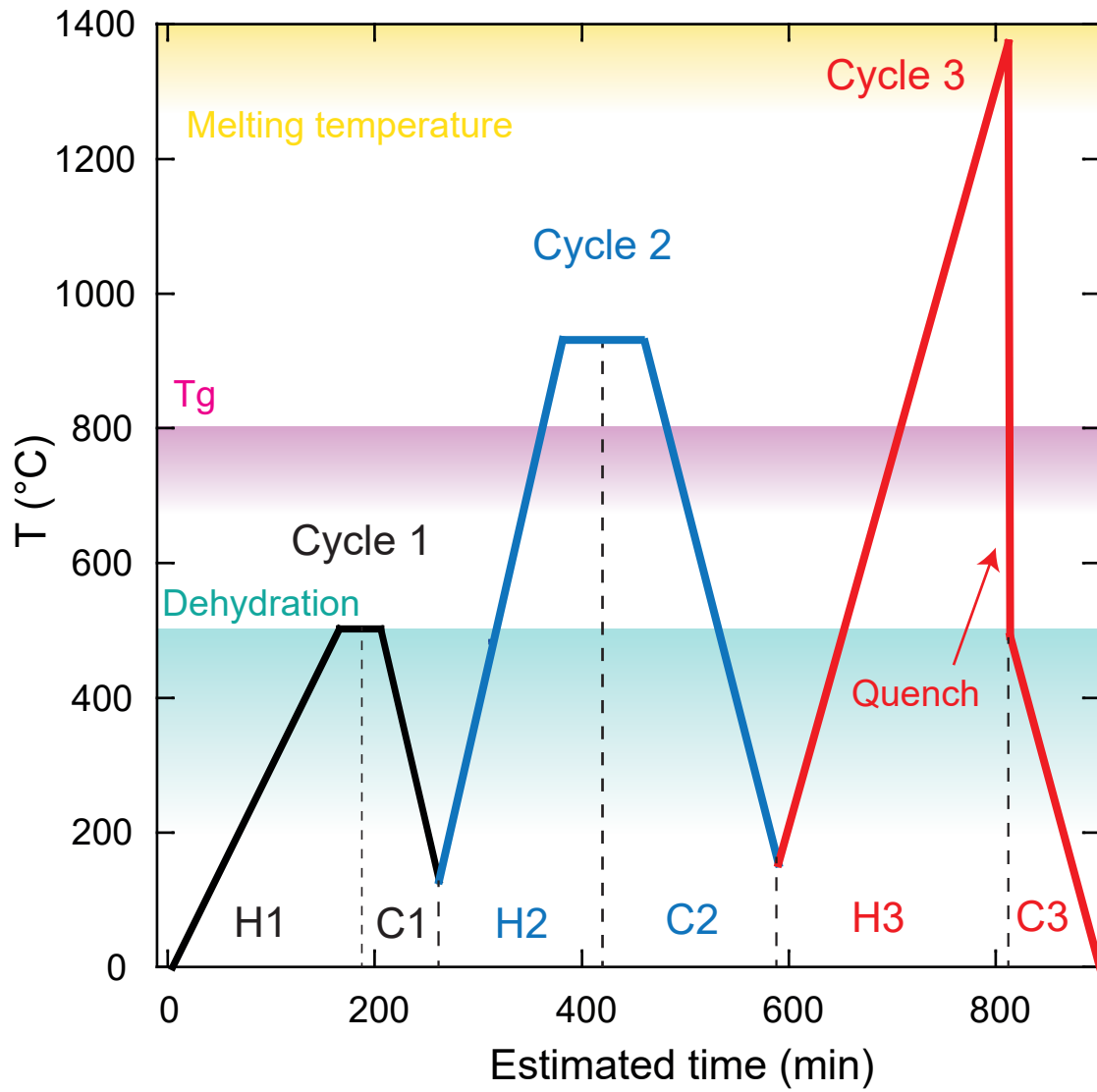
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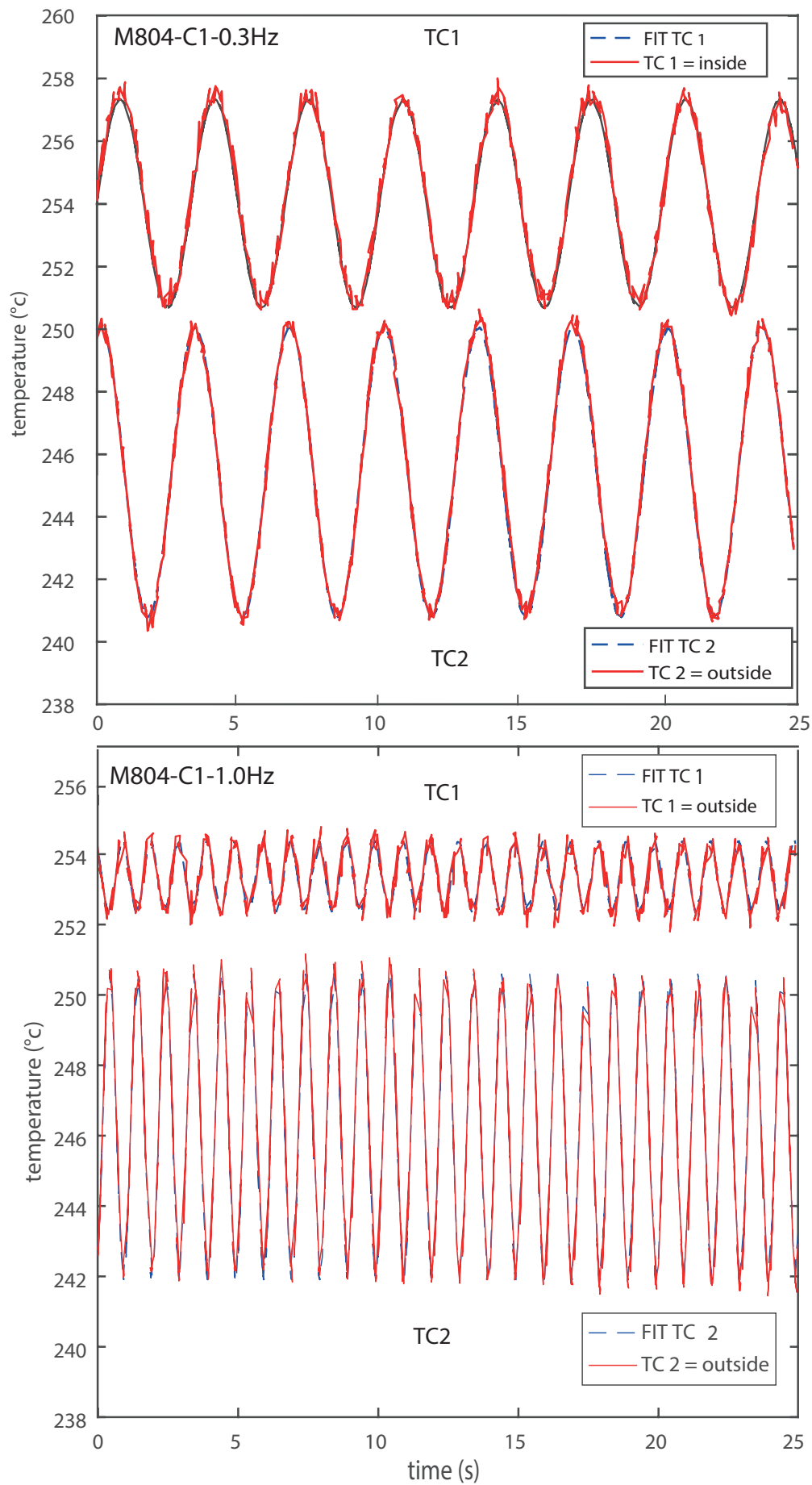
κ_c	Core thermal conductivity	4-40	$\text{W m}^{-1} \text{K}^{-1}$
α_c	Core thermal expansion	7.7×10^{-5}	K^{-1}
$C_{p,c}$	Core heat capacity	800	$\text{J kg}^{-1} \text{K}^{-1}$
ν_c	Core magnetic diffusivity	2	$\text{m}^2 \text{s}^{-1}$
μ_c	Core magnetic permeability	$4\pi \cdot 10^{-7}$	H m^{-1}
ρ_c	Core density	7500	kg m^{-3}
B	Average magnetic field strength	Eq. 3	T
R	Planet radius		m
R_c	Core radius	77% R	m
r_i	Inner core radius	1% R_c	m
G	Gravitational constant	6.67×10^{-11}	$\text{m}^3 \text{kg}^{-1} \text{s}^{-2}$
κ_{Si}	Silicates thermal conductivity	4.32-1.70	$\text{W m}^{-1} \text{K}^{-1}$
κ_{Si}	Molten silicates thermal conductivity	0.18-0.41	$\text{W m}^{-1} \text{K}^{-1}$
α_{Si}	Silicates thermal expansion	1×10^{-5}	K^{-1}
$C_{p,Si}$	Silicates heat capacity	1000	$\text{J kg}^{-1} \text{K}^{-1}$
T_s	Surface temperature	500	K
δ_{Si}	Mantle thickness	33% R	m

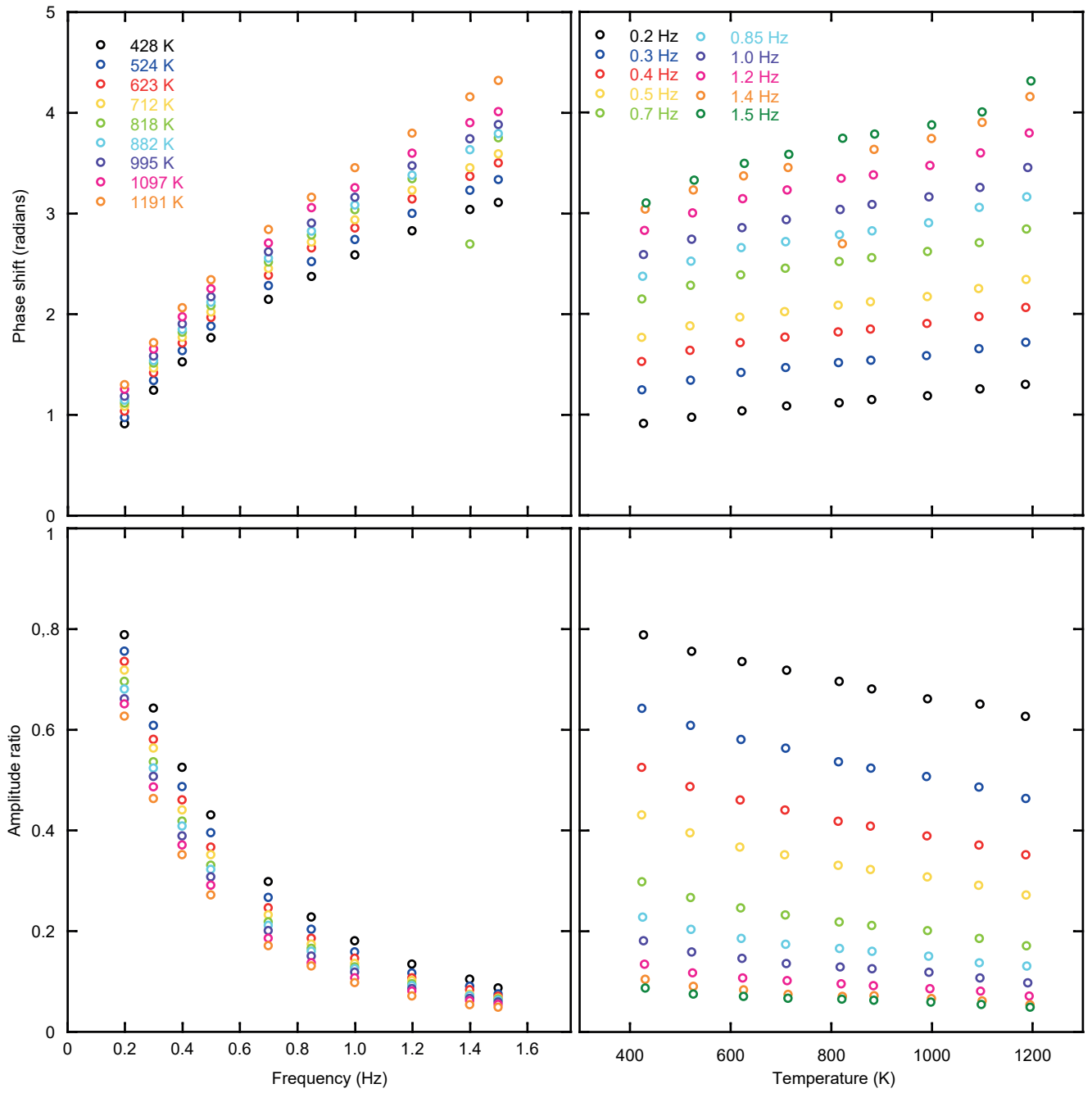
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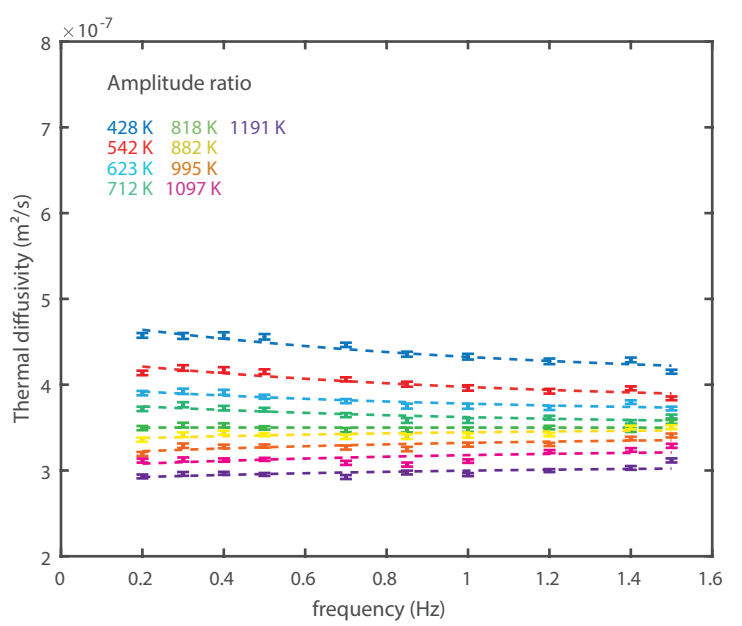
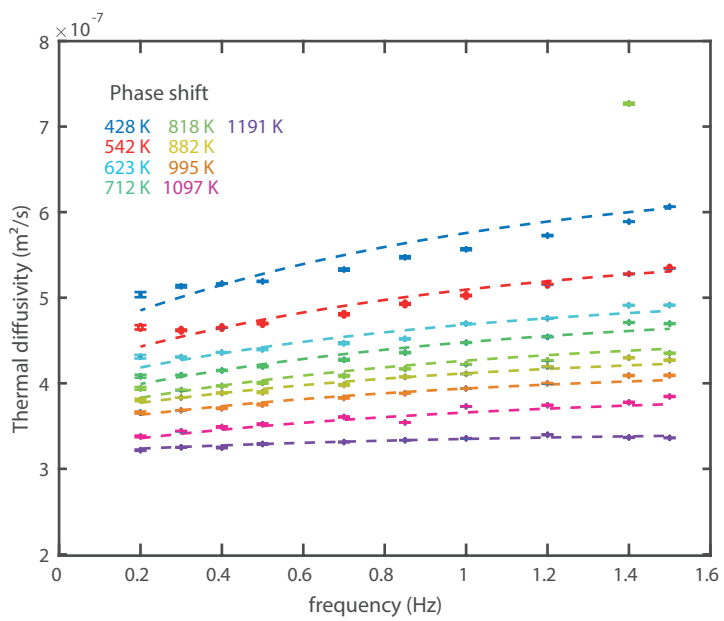
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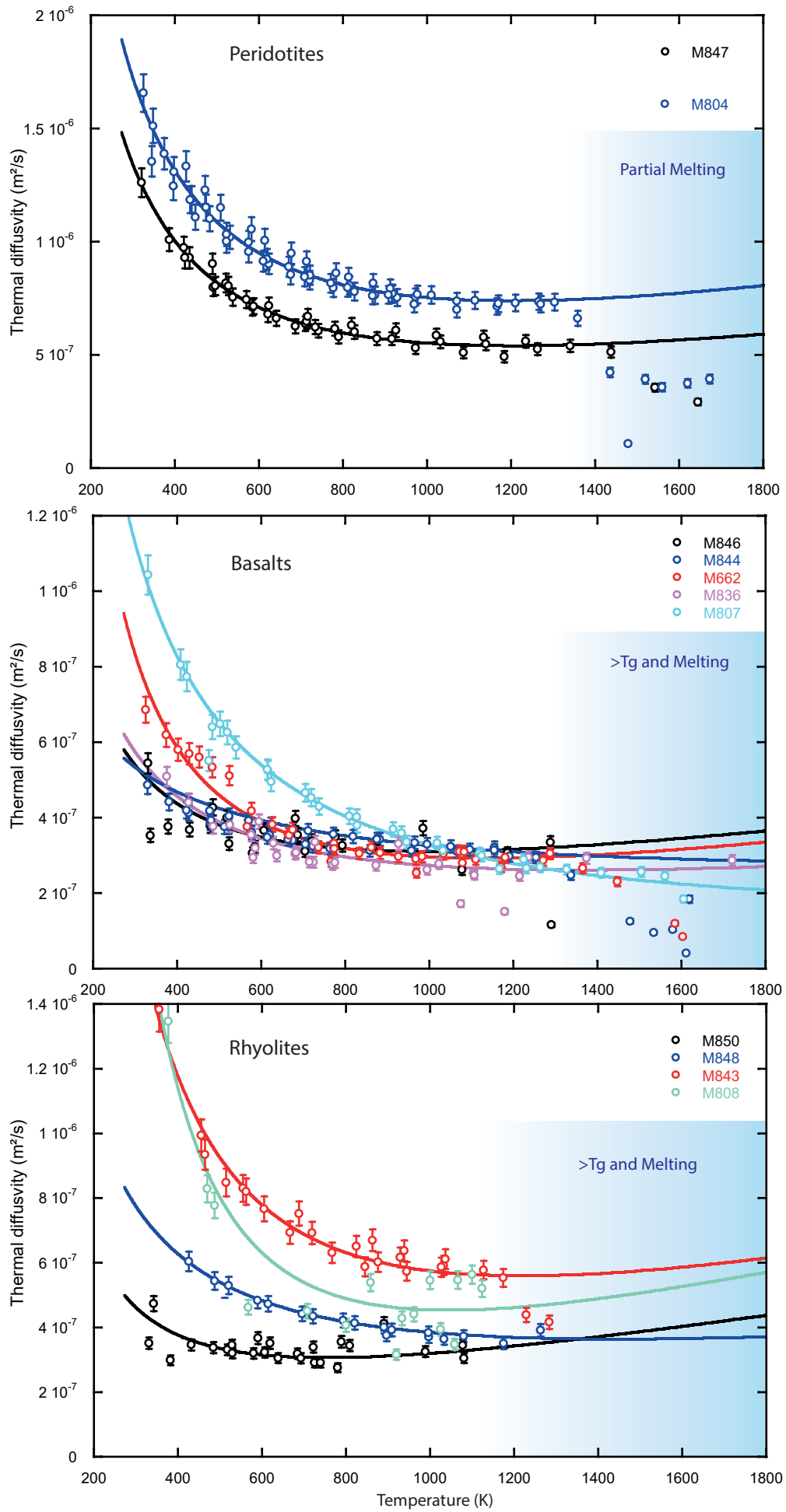












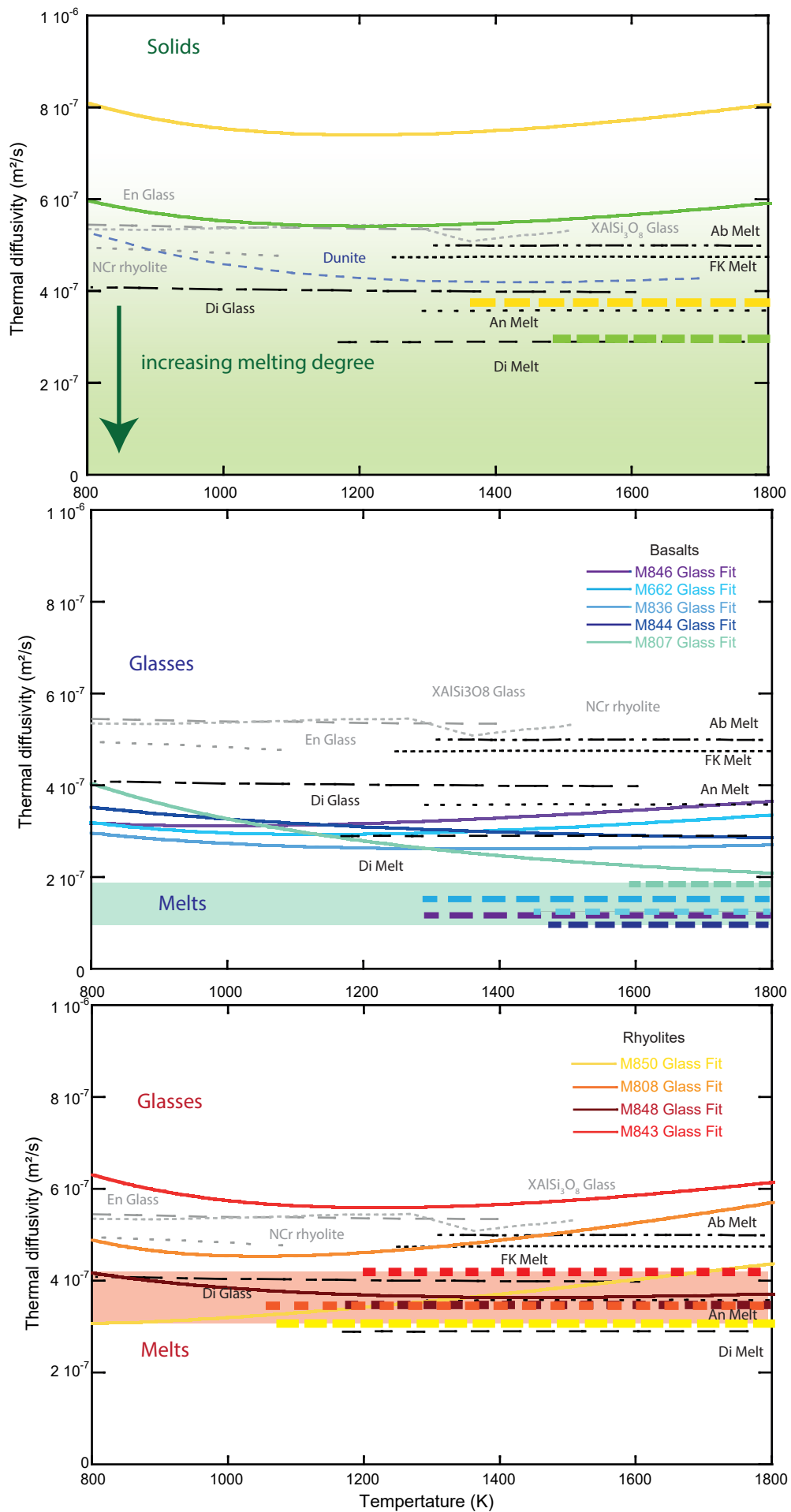


Figure 9

