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In-situ determination of Nd isotope ratios in apatite by LA-MC-ICPMS: challenges and limitations

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Abstract

In-situ measurement of Sm-Nd isotopes in rare-earth-element bearing minerals has been successfully used in the recent years to access the history of old Earth materials (e.g. Hammerli et al., 2019; Fisher et al., 2020). However, the analytical protocol of Laser Ablation-Multi Collector-Inductively Coupled Plasma Mass Spectrometry (LA-MC-ICPMS) for Sm-Nd measurements strongly depends on the instruments, the analytical parameters and the applied corrections. In this paper, we intend to summarize some of the difficulties to set up this technique. We present new results by evaluating the influence of laser parameters (spot size, fluence, frequency and He/N $_2$ gas flows) on 143 Nd/ 144 Nd, 145 Nd/ 144 Nd and 147 Sm/ 144 Nd ratios. We also report results on tests of cone geometry and interference and mass discrimination corrections on both the accuracy and precision of Nd isotopic analyzes. We report new Sm-Nd measurements performed by TIMS and LA-MC-ICPMS on Durango reference apatite (Mexico) and apatite crystals from 2 carbonatites (Fogo, Cape Verde and Phalaborwa, South Africa). We conclude from these measurements that the laser parameters have no influence on the ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd isotopic ratios. The value of the ¹⁴⁷Sm/¹⁴⁴Nd ratio, however, is correlated to the size of the spot (Fisher et al., 2020). More importantly, we show that 147 Sm/ 144 Nd ratio measurements can vary when the He/N₂ gas flows are changed even punctually (e.g. sample exchange) during an analytical session and decrease systematically during the day, which we relate to a systematic instrumental drift. We also conclude that our Durango crystal is homogeneous for Nd isotopic ratios, and slightly heterogeneous for Sm/Nd at the level of ±1-1.5% allowing accurate measurements when used as external standard. We retrieve expected isotopic ratios for both recent Fogo (4 Ma old) and ancient Phalaborwa (2060 Ma old) samples and we achieve precisions in the order of 125-150 ppm (1.3-1.5 ENdunits) for the 143 Nd/ 144 Nd ratio for a laser spot of 40 μ m, which can be considered as a reasonable size for apatite crystals in most geological samples. Finally, we present raster measurements allowing to improve the precision by a factor of ~2 with 70 ppm (0.7 ENd-units).

Keywords: Nd, isotopes, in-situ, analysis, LA-MC-ICPMS, apatite

1. Introduction

The Samarium-Neodymium (Sm-Nd) isotopic systematics relies on the radioactive decay of the long-lived 147 Sm isotope to the radiogenic 143 Nd (half-life of 106 ± 0.8 Gyr; Lugmair and Marti, 1978). Its application as a source tracer and/or a chronometer covers a wide range of geological subjects related to the differentiation and history of the Earth and other planets. The Sm-Nd pair involves two elements of the same family, the rare earth elements (REE), with a similar ionic radius, thus giving them similar chemical behaviors during magmatic processes. The small fractionation resulting from these characteristics and the long half-life of 147 Sm lead to relatively small variations of the Nd isotopic compositions with time. As a consequence, the 147 Sm- 143 Nd systematics does not allow to determine ages of young rocks with a great precision. However, the possibly low chemical disturbance of the Sm-Nd pair during alteration and low degree of metamorphism can allow to minimize any post-magmatic events. The Sm-Nd isotopic systematics is therefore a powerful source tracer for old and possibly altered, geological samples.

The Nd isotopic ratios are classically determined on bulk rocks and/or mineral separates by thermal ionization (TIMS) or plasma source mass spectrometry (MC-ICPMS), after acid digestion of the samples and chemical separation of Nd from the other elements by ion-exchange resin chromatography. However, insitu measurements are also possible by coupling the MC-ICPMS instrumentation to a laser ablation system (LA-MC-ICPMS; e.g. Foster and Vance, 2006; McFarlane and McCulloch, 2007; Fisher et al., 2011; Mitchell et al., 2011; Iizuka et al., 2011; Yang et al., 2014; Qian and Zhang, 2019; Fisher et al., 2020; and references therein). In this case, the determination of the isotopic compositions is quicker, being free from the steps of dissolution and chemical separation. More importantly, it gives the possibility of considering much finer scales of heterogeneity than bulk-rock analyzes, at the level of a few tens of micrometers.

The in-situ measurement of isotopic compositions of Nd is suitable in minerals rich in this element (e.g. monazite, titanite, apatite). Apatite is an ideal candidate as its Nd concentrations are generally greater than several hundreds of ppm, and it is an ubiquitous phase in most rocks (magmatic, metamorphic and sedimentary), giving wide field of application in geology. Several laboratories have already developed the in-situ analysis of Nd isotopic ratios in this mineral. Analytical protocols are nearly similar at some extent, but diverge on other points such as correction for instrumental Sm-Nd fractionation (Fisher et al., 2011; lizuka et al., 2011). Moreover, the accuracy and precision of isotopic measurements seem to be strongly dependent on the instrumentation used (laser system + MC-ICPMS), as well as the number and diversity of isotopic ratios analyzed on, leading to the necessity for each laboratory to justify for its own protocol. Here, we present new Sm-Nd results acquired on the Thermo Fisher Scientific™ Neptune Plus™ instrument at the Laboratoire Magmas et Volcans (LMV), Clermont-Ferrand, France, on apatite crystals from three different localities (Durango apatite and two carbonatite apatites). We detail the different corrections needed to obtain accurate and precise Nd isotopic ratios with a focus on the laser (size of the spot, fluence, frequency and He/N_2 gas flows) and MC-ICPMS parameters (X vs. H skimmer cones) and their influence on the measurements. From a more general point of view, this contribution aims to help geochemical laboratories to set up and implement an in-situ analysis protocol for Nd isotope ratios.

2. Durango and Phalaborwa apatites

Durango apatite is a fluorapatite (usually millimeter- to centimeter in size) which has been sampled from the Cerro de Mercado mine, Mexico. Its age is estimated to be 31.02 ± 0.22 Ma (U-Th-He thermochronology; McDowell *et al.*, 2005); it originated from hydrothermal processes in relation with Oligocene alkaline volcanic rocks (Lyons, 1988). Durango apatite is classically used as a reference standard for isotopic analysis of Nd by LA-MC-ICPMS (e.g. Foster and Vance, 2006; Fisher *et al.*, 2011; Yang *et al.*, 2014), and as a reference standard for halogens in microprobe analysis (Rønsbo, 1989; Hammouda *et al.*, 2010), for (U-Th)/He and fission track dating (Carlson and Lugmair, 1979; Zeitler *et al.*, 1987; Hasebe *et al.*, 2004; McDowell *et al.*, 2005; Flowers *et al.*, 2009), and for trace element analysis (e.g. Trotter and Eggins, 2006; Bruand *et al.*, 2014).

Using Durango apatite as a standard has several advantages: 1) it is present in large quantities in the form of multi-centimeter crystals, 2) it is relatively easy to obtain, and 3) it has been reported as homogeneous in Nd isotopes at the crystal scale with a restricted range of 143Nd/144Nd between 0.512434 and 0.512498 (Foster and Vance, 2006; Fisher et al., 2011; Yang et al., 2014; Xu et al., 2018). Punctually, some studies have highlighted compositional variations between different Durango apatite crystals (e.g. Fisher et al., 2011; Yang et al., 2014). Thus, we tested the chemical and isotopic characteristics of the crystal studied in this contribution (sample #144954-4, Smithsonian National Museum of Natural History). For this purpose, three fragments of the mineral, weighting each between 10 and 20 mg and respectively called D1, D2 and D3, were cleaned with milliQ water to remove any dust on their surfaces and then acid-digested in Teflon vessel (savillex©) with a mixture of HNO₃-HF followed by HNO₃-HClO₄. Aliquots of around 5% of the digestions were used for Sm and Nd content measurements on an Agilent 7500C ICPMS instrument, whereas aliquots between 10 and 20% were mixed with a 150Nd-149Sm spike and passed through TRU- and Ln-Spec columns in order to determine their isotopic and chemical ratios (143Nd/144Nd, 145Nd/144Nd and 147Sm/144Nd) by isotope dilution on a Thermo Electron™ Finnigan Triton™ thermo-ionization mass spectrometer (TIMS). All TIMS measurements were made in static mode with relay matrix rotation (also called the virtual amplifier) and double Re filaments. Isotope ratios were mass-fractionation corrected with 146Nd/144Nd = 0.7219 and normalized to 143 Nd/ 144 Nd = 0.512100 for the JNDi-1 Nd standard (Garçon et al., 2018). Isotope dilution also assumed 150 Nd/ 144 Nd = 0.236451, 149 Sm/ 147 Sm = 0.921629 and 152 Sm/ 147 Sm = 1.783135. Three analyzes of the JNDi-1 performed during the Durango TIMS session gave 143 Nd/ 144 Nd = 0.512106 \pm 0.000008 (2 σ). Results are presented in Table 1. They show reproducible values at the level of $\pm 10-20$ ppm for 143 Nd/ 144 Nd and 145 Nd/ 144 Nd (2σ -variation) and of ± 0.5 % for 147 Sm/ 144 Nd. This suggests that our Durango crystal is relatively homogeneous, isotopically and chemically, at a scale of about 10 mg. Such a conclusion cannot be extended, however, to lower scales such as those of spot sizes for which 2σ-variations reported here must be a priori considered as minimum heterogeneity. Measured ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd ratios are consistent with previous TIMS determinations (Foster and Vance, 2006; Fisher et al., 2011; Yang et al., 2014; Xu et al., 2018), but we observe notable differences for the 147Sm/144Nd ratio with some of the data available in the literature (e.g. Yang et al., 2014: their Durango crystal has a 147Sm/144Nd ratio of 0.0881), reinforcing the importance for each geochemical laboratory to characterize their Durango crystal by solution before to use it for external normalization.

Phalaborwa apatites were separated from an intrusive calcio-carbonatite sample (PhB09) coming from the Phalaborwa carbonatite Complex, South Africa. Their age is estimated to be 2060 ± 2 Ma (U-Pb isotopic age determinations on baddeleyites separated from Phalaborwa carbonatite samples similar to PhB09; Wu

et al., 2011). Phalaborwa apatites are small crystals compared to Durango, with maximum sizes in the order of 100 μm, so it was not possible to characterize a fragment of each of the crystals analyzed by LA-MC-ICPMS using TIMS. Instead, three sets of multiple apatite grains weighting between 7 and 12 mg were analyzed using the same protocol than Durango fragments D1, D2 and D3. Results are presented in Table 1. They show 2σ -variations of ±50 ppm for ¹⁴³Nd/¹⁴⁴Nd and of ±0.2 % for ¹⁴⁷Sm/¹⁴⁴Nd. Initial ¹⁴³Nd/¹⁴⁴Nd ratios back-calculated using an age of 2060 Ma ranges between 0.509668 and 0.509674 (εNd(t2060) between -5.8 and -6.0). These are consistent values with initial ratios (εNd(t2060) between -3.4 and -7.5) determined for the Phalaborwa primary carbonatitic magma (Yuhara et al., 2005; Wu et al., 2011). Finally, Fogo apatites were separated from a calcio-carbonatite sample collected in a dike on Fogo Island, Cape Verde archipelago. Their age is estimated to be ~4 Ma (cf. Mata et al., 2010) and TIMS analyzes performed on 50-mg of multiple grains gave ¹⁴³Nd/¹⁴⁴Nd = 0.512945 ± 0.000006 (Doucelance et al., 2010).

3. Instrumentation and protocol of in-situ analysis of Nd isotopes

Neodymium isotopic measurements were performed using a Thermo Fisher Scientific[™] Neptune Plus[™] mass spectrometer coupled with a 193-nm wavelength Resonetics[™] M-50E laser system. All analyzes were acquired at the LMV, Clermont-Ferrand, France. Prior to in-situ analyzes, apatite samples were embedded in epoxy resin blocks and polished. Cathodoluminescence (CL) images were also obtained using a Jeol JSM-5910 LM scanning electron microscope (SEM) in order to characterize internal structures of apatite grains and better decide for spot targets.

The MC-ICPMS was operated in static mode using 7 Faraday Cups. This allowed the simultaneous collection of masses 143 to 149 (Table 2). A typical run consisted of 70 cycles of measurements with an integration time of 1s, the first 15 cycles being used to evaluate the baseline level of each Faraday Cup. X skimmer cone coupled to a Jet interface sampler cone were systematically used unless reported. ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd isotopic ratios together with ¹⁴⁷Sm/¹⁴⁴Nd chemical ratios were determined in each case. Baseline and gain reductions were operated online; all other calculations were performed off-line using an in-house Microsoft Excel™ spreadsheet. Gain calibration was done at the beginning of each analytical session. Corrections for isobaric interferences, elemental and mass biases are detailed in the following subsections, whereas LA-MC-ICPMS parameters are summarized in Table 3.

3.1. Correction for isobaric interferences of Neodymium isotopes

Isotope ¹⁴⁴Nd interferes with ¹⁴⁴Sm and ¹⁴⁶Nd with ¹³⁰Ba¹⁶O (barium oxide) if the analyzed material contains barium (Table 2). Barium-130 is only 0.1% of total Barium, and oxide production has been minimized at the beginning of each session of analysis (¹⁴⁶Nd¹⁶O/¹⁴⁶Nd ratio < 1%) meaning that the ¹³²Ba¹⁶O signal is negligible with respect to ¹⁴⁶Nd intensity of apatite samples. For these reasons, no correction was applied to ¹⁴⁶Nd. For Sm interferences, however, we followed Yang *et al.* (2008) and Fisher *et al.* (2011) and used the 3-steps procedure described in McFarlane and McCulloch (2007) that considers an exponential law for MC-ICPMS mass fractionation (Russell *et al.*, 1978) together with the most recent determinations of Sm isotopic abundances (here illustrated for mass 144):

• Step 1 - The measured, interference-free 147 Sm/ 149 Sm ratio is used to determine the mass bias correction factor of Sm (β_{Sm}):

$$\beta_{Sm} = \frac{\ln\left[\frac{(^{147}Sm/^{149}Sm)_{true}}{(^{147}Sm/^{149}Sm)_{meas}}\right]}{\ln\left[Mass_{\perp}^{147}Sm/Mass_{\perp}^{149}Sm\right]} \tag{1}$$

where subscripts *true* and *meas* denote the assumed (1.08680 - Dubois *et al.*, 1992) and measured values of the 147 Sm/ 149 Sm isotopic ratio, respectively, and $Mass_^{147}$ Sm/ $Mass_^{149}$ Sm is the ratio of atomic masses;

• Step 2 - The intensity of 144 Sm is then calculated from that of 149 Sm, given the true value of 144 Sm/ 149 Sm (144 Sm/ 149 Sm and 148 Sm/ 149 Sm ratios normalized to 147 Sm/ 149 Sm value of Dubois and co-authors are 0.22332 and 0.81407, respectively - Isnard *et al.*, 2005):

$$^{144}Sm = ^{149}Sm \times (^{144}Sm/^{149}Sm)_{true} \times (Mass_{-}^{144}Sm/Mass_{-}^{149}Sm)^{\beta_{Sm}}$$
 (2)

• Step 3 - Finally, the ¹⁴⁴Sm interference is removed from the total intensity at mass 144:

$$^{144}Nd = 144_{total} - ^{144}Sm (3)$$

3.2. Correction for Neodymium mass bias

The mass bias correction factor of Nd (β_{Nd}) was deduced from the measured ¹⁴⁶Nd/¹⁴⁴Nd ratio, corrected for the interference of ¹⁴⁴Sm and compared to its true value (0.7219 - Wasserburg *et al.*, 1981) via the exponential law:

$$\beta_{Nd} = \frac{ln \left[\frac{(^{146}Nd/^{144}Nd)_{true}}{(^{146}Nd/^{144}Nd)_{meas}} \right]}{ln \left[\frac{mass_{146}Nd/^{144}Nd}{mass_{144}Nd} \right]}$$
(4)

 143 Nd/ 144 Nd and 145 Nd/ 144 Nd isotopic ratios were then calculated from their measured values corrected for 144 Sm:

$$^{143}Nd/^{144}Nd = (^{143}Nd/^{144}Nd)_{meas} \times (Mass_{-}^{143}Nd/Mass_{-}^{144}Nd)^{\beta_{Nd}}$$
 (5)

$$^{145}Nd/^{144}Nd = (^{145}Nd/^{144}Nd)_{meas} \times (Mass_{-}^{145}Nd/Mass_{-}^{144}Nd)^{\beta_{Nd}}$$
 (6)

3.3. Determination of the ¹⁴⁷Sm/¹⁴⁴Nd chemical ratio

Simultaneous and accurate determination of the ¹⁴⁷Sm/¹⁴⁴Nd chemical ratio allows recalculating initial Nd compositions of old apatites given that ages of the samples are independently known. An internal determination of ¹⁴⁷Sm/¹⁴⁴Nd, as for Nd and Sm isotope ratios, is impossible to obtain due to the numerous effects which can potentially fractionate Sm and Nd: different vaporization rates of the two elements (Kuhn and Günther, 2003; Košler *et al.*, 2005), different element ionization rates in the plasma (Guillong and Günther, 2002), instrumental drift (Cheatham *et al.*, 1993), and also, similarly to Nd or Sm isotope ratios, mass discrimination in the MC-ICPMS (Vance and Thirlwall, 2002). Depending on the laser system and cells, sample position in laser cell can also be problematic (Fisher *et al.*, 2011). All these issues are reinforced by the fact that elemental fractionation also varies with the nature of the sample matrices (Chen *et al.*, 1999). In their paper focusing on the ¹⁴⁷Sm/¹⁴⁴Nd and ¹⁴³Nd/¹⁴⁴Nd in-situ analysis of monazite, lizuka *et al.* (2011) proposed two distinct methods for acquiring the correction factor *f* between the true and measured ¹⁴⁷Sm/¹⁴⁴Nd ratios.

The first procedure consists to normalize unknown samples to a standard whose 147 Sm/ 144 Nd is uniform and has been independently determined (Foster and Vance, 2006). This is the way Yang *et al.* (2014) corrected their apatite Sm/Nd data, using AP2 apatite reference material as external standard. This is also the approach retained by Fisher *et al.* (2011), although they did not use a matrix-match standard but a glass enriched in light REE (LREE). The second method proposed by lizuka *et al.* (2011) is to use an ancient monazite standard for which a 147 Sm/ 144 Nd- 143 Nd/ 144 Nd isochron can be independently defined. Comparison between the standard isochron and a 143 Nd/ 144 Nd vs. measured 147 Sm/ 144 Nd plot for the same standard then allows estimating the correction factor *f.* Foster and Vance (2006) also proposed to correct the measured 147 Sm/ 144 Nd ratio for the instrumental mass bias using the mass bias correction factor of Nd (β_{Nd} , as defined in eq. 4). This leads to the following equation, providing that 144 Nd is corrected for 144 Sm contribution:

$$^{147}Sm/^{144}Nd = (^{147}Sm/^{144}Nd)_{meas} \times (Mass_{-}^{147}Sm/Mass_{-}^{144}Nd)^{\beta_{Nd}}$$
(7)

Mass bias correction factors of Nd and Sm are significantly different (McFarlane and McCulloch, 2007; Yang *et al.*, 2008) so that the choice of β_{Nd} to correct for instrumental mass bias is questioned. However, such a correction, even not accurate, allows discussing other reasons for Sm/Nd fractionation than mass bias. In the present study, we used both approaches: 147 Sm/ 144 Nd correction using eq. 7 and external calibration against the 147 Sm/ 144 Nd ratio of our Durango-apatite crystal that was measured regularly during the day-sessions of analysis of unknown apatite samples.

4. Instrumental drift at constant laser parameters

One important observation made during this study is the presence of a systematic, decreasing drift of measured 147 Sm/ 144 Nd values while keeping all instrumental parameters constant. To illustrate this key point, Figure 1 shows 48 runs of the same Durango chip standard that were all acquired with the same laser parameters (spot size = $40 \mu m$, fluence = 3.8 J/cm^2 and frequency = 6 Hz) over five hours. 143 Nd/ 144 Nd and 145 Nd/ 144 Nd ratios are all within uncertainties of each other and show respective, $2-\sigma$ reproducibilities (2σ -variation divided by the mean value) of 178 and 157 ppm, whereas 147 Sm/ 144 Nd ratios seem to decrease continuously from -0.0775 to -0.0760 before to settle. The total range corresponds to a variation of $\pm 1.2\%$ on the average 147 Sm/ 144 Nd value (see runs #18 to 65, day-session 2, in Supplementary Table). Correcting for the drift however gives a $2-\sigma$ reproducibility of 0.7%, which is comparable to the level of heterogeneity revealed by solution analyzes of Durango fragments D1, D2 and D3.

It may be argued that such a drift is due to zoning within Durango since the analyzes are reported in the order of acquisition. Several elements argue however against a major control of zoning on the ¹⁴⁷Sm/¹⁴⁴Nd ratio: 1) our Durango crystal has been broken independently of the c-axis, along which zoning can be potentially observed; 2) the spots reported on Figure 1 are all located in a limited area of our Durango crystal: they form a cluster but do not draw a traverse; and 3) cathodoluminescence imaging of our Durango grain did not show any concentric zoning (see Supplementary Fig. S1). Finally, the analysis at constant MC-ICPMS and laser parameters of an homogeneous, Sm-doped JNDi glass in order to avoid the possible contribution of an heterogeneity of our natural standard show similar results (Figs 2A,B,C). ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd ratios have reproducible values within uncertainties, whereas ¹⁴⁷Sm/¹⁴⁴Nd decrease before to settle. The drift induced a variation up to ±0.8% over five hours which has to be considered when trying to

evaluate the most accurately possible the ¹⁴⁷Sm/¹⁴⁴Nd ratio of the analyzed material. Such variations are correlated to the average ¹⁴⁶Nd ion beam signal (Fig. 2D) suggesting that the decreasing drift is a response to the decreasing ¹⁴⁶Nd signal that is generally observed when we keep instrumental parameters constant.

5. Influence of laser parameters on the determination of ¹⁴³Nd/¹⁴⁴Nd, ¹⁴⁵Nd/¹⁴⁴Nd and ¹⁴⁷Sm/¹⁴⁴Nd ratios

All the data presented in this section have been acquired during seven day-sessions over a period of one year. In the following, we discuss the possible influence of the laser parameters (the size of the spot, the fluence and the frequency of the laser range from 33 to 93 μ m, from 3.4 to 4.6 J/cm², and from 5 to 8 Hz, respectively) together with that of He/N₂ gas flows on Nd isotopic compositions and ¹⁴⁷Sm/¹⁴⁴Nd ratios. Attention is focused more specifically on 40- μ m laser spots since this corresponds to a reasonable size for apatite crystals in most geological samples.

5.1. Laser spot size

Figure 3 shows ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd isotope ratios together with ¹⁴⁷Sm/¹⁴⁴Nd chemical ones obtained for Durango apatite during day-session 1. The data are reported in chronological order of acquisition. Measurements reported with a red square have been performed with constant frequency and fluence (6 Hz, 3.8 J/cm²) but various sizes of spot (93, 40 and 33 μm). The three sets of data show reproducible values for ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd isotope ratios at the level of 79-85 ppm for 93-μm-spot measurements (79 ppm for ¹⁴³Nd/¹⁴⁴Nd and 85 ppm for ¹⁴⁵Nd/¹⁴⁴Nd), and 132-187 ppm and 158-247 ppm, for 40- and 33-μm spot sizes (Figs. 3A,B and Supplementary Table). These reprodubilities are not as precise as those obtained by TIMS for the three Durango fragments but all analysis performed at different spot sizes give ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd within uncertainties. This suggests that the weaker reproducibility of in-situ measurements results mainly from the small ion beam signals and consequently the size of the laser spot has no influence on the isotopic composition of the apatite sample.

On the other end, average 147 Sm/ 144 Nd ratios vary significantly, increasing from 0.0744 to 0.0784 when the spot size is decreasing from 93 to 40 µm (Fig. 3C). Such a variation of about $\pm 3\%$ on the average 147 Sm/ 144 Nd value (that was also observed to a lesser extent during day-session 3, with 147 Sm/ 144 Nd ratios equal to 0.0751-0.0755 and 0.0773-0.0777 for spot sizes of 93 and 40 µm, respectively; see Supplementary Table) is much greater than that resulting from the chemical heterogeneity of the Durango crystal. Comparison with analyzes operated at constant parameters (section 4, Figure 1) shows that the variation of the spot size is likely to be responsible for at least 2/3 of the variation of the 147 Sm/ 144 Nd ratio during day-session 1. This confirms previous results by Fisher *et al.* (2020) that showed that the spot size controls how Sm isotopes are sampled relative to Nd ones during ablation. Variations they reported for the 147 Sm/ 144 Nd ratio when changing the spot size between 120 and 40 µm are in the order of 5%.

5.2. Fluence of the laser

At constant spot-size and frequency, the fluence controls the amount of ablated material, and therefore the sampling of apatite crystals. In Figures 3A,B, 143 Nd/ 144 Nd and 145 Nd/ 144 Nd isotopic ratios measured at 6 Hz with a spot size of 40 μ m and a laser fluence varying from 3.4 to 4.6 J/cm² (around 0.5 J/cm² step;

red symbols) are identical within error-bars. The 147 Sm/ 144 Nd chemical ratio varies slightly between these different conditions but average values recorded during the session (large symbols) are also all within uncertainties (Fig. 3C). Analyzes performed during day-session 2 with the same conditions (40 μ m - 6 Hz - 3.4 to 4.6 J/cm²) show similar results. The 143 Nd/ 144 Nd and 145 Nd/ 144 Nd ratios display the same values within error-bars, whereas 147 Sm/ 144 Nd variations are observed but cannot be correlated with the intensity of the fluence (Fig. 4). All the results strongly suggest that the fluence range used here has no effect on the measurements of isotopic and chemical compositions of the apatites. The variations of the chemical ratio are of 1% during day-session 1 and of 2.5% during day-session 2 (2- σ variations considering individual runs that scatter slightly more than average values). As seen before, up to 1% of theses variations can be attributed to the instrumental drift observed during a day-session. The remaining scatter is most probably related to the slight chemical heterogeneity of the Durango chip.

5.3. Frequency of the laser

Similarly to the fluence, the frequency of the laser influences the amount of ablated material and thus can induce fractionation of Sm and Nd during ablation. Analyzes performed with 40-µm spots and 3.8-J/cm² fluence (reported with yellow- and red squares in Figure 3) show that a change from 6 to 8 Hz does not modify the Nd isotopic ratios significantly but results in a decrease of error-bars. This is also the case when considering 33-µm-spots and 6- to 7-Hz variations (blue- and red diamonds in Figure 3). Conversely, the associated, average ¹⁴⁷Sm/¹⁴⁴Nd ratios seem to vary with the frequency. However, in the first case, we observe an increase of the chemical ratio with decreasing frequency (8 to 6 Hz), and a decrease of the chemical ratio in the second case (7 to 6 Hz). In both cases, the ¹⁴⁷Sm/¹⁴⁴Nd individual measurements that compose average values are quite dispersed. However, the internal dispersion is of the same magnitude as the difference between the frequency sets for a given spot-size. Thus, with our current dataset, no relationship is observable between ¹⁴⁷Sm/¹⁴⁴Nd values and laser frequencies.

5.4. He and N₂ gas flows

Gas flows arriving at the cell (He, N_2) can be punctually modified during a day-session of analysis when changing a sample block or when optimizing the signal while keeping the oxides to a low level ($Nd^+O/Nd^+ < 1\%$). Furthermore, it cannot be excluded that the He gas tank also arrives to exhaustion, thus forcing to change it during a day-session. Such flow modifications are not of the same magnitude. A sample exchange will lead to nitrogen shut-down and helium flow diminution (down to ~100 ml/min) before all flows are set back to their initial values, whereas He and N_2 flows to optimize the signal and minimize oxides will be modified in a lesser extent (addition or removal < 150 ml/mn for He and ~1 ml/mn for N_2). In this contribution, we tune both gas flows and the z-torch position to control oxide formation. During the course of the study, we experienced all these gas flow changes and we observed that they have had a great impact on the $^{147}Sm/^{144}Nd$ ratios.

The helium tank was changed during day-session 2 after 17 Durango measurements (Fig. 4) and therefore N_2 and He were set to 0 for several minutes. The change had no consequence on isotope results. $^{143}Nd/^{144}Nd$ and $^{145}Nd/^{144}Nd$ ratios show identical values before and after helium-tank exhaustion. On the contrary, $^{147}Sm/^{144}Nd$ chemical ratios varied significantly: they slightly increased after the He tank change

and then stabilized (Fig. 4C). This observation strongly suggests that 147 Sm/ 144 Nd ratios are sensitive to extremely fine gas flow variations since N₂ and He flows were set back to initial values after tank change.

Such a conclusion is also supported by measurements performed on a Sm-doped JNDi glass during day-sessions 6 and 7, during which we proceeded to a sample exchange. In both cases, while 143 Nd/ 144 Nd and 145 Nd/ 144 Nd ratios show constant values within error-bars, 147 Sm/ 144 Nd systematically jump to higher values after the sample exchange (see the last run but one of day-session 6 for 53- μ m measurements and the last one of day-session 7, Figure 2C). The variation of the 147 Sm/ 144 Nd ratio reached 1.9% during day-session 7. The sample exchange consisted here in lowering He to 100 ml/mn and 142 Nd ratio 144 Nd ratio several minutes, and setting the He and 142 Slows up to their initial values once the new mounts were in. The results acquired during day-session 6 thus show that within the 10 mn following the sample exchange, the 147 Sm/ 144 Nd ratio measurements are affected before to settle back to the trend of the instrumental drift.

The previous observations were made while bringing back the gas flows back to their initial values. During several sessions, we also attempted to change the He flow values to lower oxides. Figure 5 show measurements that were operated alternatively in the same Durango grain and in a NIST SRM 610 glass sample before and after He flow adjustment operated in order to decrease the oxide level from 3% down to 0.4%. Correction factors determined for ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd ratios show identical values within errorbars for both samples, and also before and after the He flow modification, whereas correction factors for ¹⁴⁷Sm/¹⁴⁴Nd show significant variations up to 1%. Interestingly, ¹⁴⁷Sm/¹⁴⁴Nd correction factors move in opposite ways: values decrease for Durango, while they increase for NIST SRM 610.

6. Influence of laser parameters on the within-run variation of the ¹⁴⁷Sm/¹⁴⁴Nd chemical ratio

Once the corrections described in section 3 have been performed, the ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd isotopic ratios measured for the n cycles of a single spot analysis show constant values inferring that the corrections are robust. Conversely, variable values of the ¹⁴⁷Sm/¹⁴⁴Nd chemical ratio due to laser induced elemental fractionation or down-hole fractionation that occurs when measuring inter-element ratios (lizuka *et al.*, 2011; Fisher *et al.*, 2017; 2020) are expected. Correction for Sm/Nd fractionation indeed only considers Nd-isotope mass-discrimination that proceeds mainly in the interface region.

Figure 6 compiles cycle-to-cycle ¹⁴⁷Sm/¹⁴⁴Nd ratios that were measured in our Durango apatite grains for different laser parameters. Various evolutions can be observed within a single analysis: uniform increase, «V-shape» or «inverse V-shape» patterns. Such cycle-to-cycle variations are very reproducible as illustrated on Figure 7. This figure reports the ¹⁴⁷Sm/¹⁴⁴Nd patterns observed in ten consecutive analyzes for three distinct sets of laser parameter values. Relationships between laser parameters and cycle-to-cycle evolutions, however, remain unclear. Decrease of the spot-size for constant fluence and frequency values has impact on ¹⁴⁷Sm/¹⁴⁴Nd behavior from «inverse V-shape» to uniform increase (Fig. 6C to Fig. 6B to Fig. 6D). Decrease of the fluence (for constant spot-size and laser frequency) leads to «inverse V-shape» ¹⁴⁷Sm/¹⁴⁴Nd evolutions (Fig. 6B to Fig. 6A), whereas increase of the same parameter conducts to « V-shape» patterns (Fig. 6B to Fig. 6E). Lastly, increase of the laser frequency is able to change the ¹⁴⁷Sm/¹⁴⁴Nd pattern from monotonous increase to «V-shape» (Fig. 6B to Fig. 6G), although it can also have no significant effect (Fig. 5F to Fig. 5H). When analyzing other apatite crystals than Durango with the same laser parameters (namely Fogo Island, Cape Verde, and Phalaborwa, South Africa), different within-run variations of ¹⁴⁷Sm/¹⁴⁴Nd ratios

are observed. Figures 6J and 6K compare cycle-to-cycle results obtained in apatite grains from the two carbonatites that have been cross-measured with Durango chips. For identical laser parameters, Fogo and Phalaborwa apatite grains show clearly distinct patterns. Fisher *et al.* (2020) reported within-run ¹⁴⁷Sm/¹⁴⁴Nd evolutions measured in the Tory Hill apatite for spot sizes varying from 50 to 130 µm. All patterns show an uniform increase, except for the smallest spot size for which they are flat. They also noticed that spot sizes of 50 µm produce accurate ¹⁴⁷Sm/¹⁴⁴Nd ratios (in comparison to their TIMS value), whereas largest ones give lower, erroneous values. Such a relationship is not so clear for our Durango data. All measurements reported in Figure 6 and related to 40 and 33 µm spot sizes have average ¹⁴⁷Sm/¹⁴⁴Nd ratios between 0.0775 and 0.0785, which is higher than the TIMS value. On the contrary, 93- and 53-µm analyzes have average ¹⁴⁷Sm/¹⁴⁴Nd values slightly lower than our TIMS determination. All these observations show that the observed patterns depend on the laser parameters but also on the composition of the apatite. However, even if these patterns have been identified, we will see in the next section that their average value are reproducible within ~1% or better (see also Supplementary Table).

7. Reproducibility and accuracy of apatite analysis on different Durango chips

Figure 8 compiles ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd isotopic ratios together with ¹⁴⁷Sm/¹⁴⁴Nd chemical ratios measured in the same Durango grain (chip #1) during 3 day-sessions of acquisition (3 sessions over one month: 2 successive days and another one 30 days later, for which 85, 77 and 59 runs have been acquired; the data are reported in chronological order of acquisition). 143Nd/144Nd and 145Nd/144Nd ratios show comparable reproducibility with 2sd-values comprised between 139 and 179 ppm and between 111 and 170 ppm, respectively (Figs. 8A,B). While clear differences of mean 143Nd/144Nd ratios between sessions can be observed (up to 174 ppm), mean ¹⁴⁵Nd/¹⁴⁴Nd are one order of magnitude lower (17 ppm). However, these variations between mean values are all within uncertainties of each other. Finally, except for day-session one, the 143Nd/144Nd ratios mean values are within error of the Durango-chip TIMS values. It should be mentioned that individual 143Nd/144Nd are lower than TIMS values for most of the runs, while 145Nd/144Nd scatter around the TIMS determination. Mean ¹⁴⁷Sm/¹⁴⁴Nd values are reproducible at the level of 0.8% (Fig. 8C). However, the reproducibility within sessions is far lower and the 147Sm/144Nd variations are such that errorbars associated to mean values for each session are all overlapping with each other over the 3 days of acquisition. Considering these important uncertainties, they are also in agreement with TIMS values although calculated mean values are all higher than TIMS values. Such observation is likely to be a coincidence since: (i) individual ¹⁴⁷Sm/¹⁴⁴Nd ratios do not scatter around calculated mean values but show different evolutions depending of the day considered; and (ii) 147Sm/144Nd ratios reported in Figure 8C are not corrected for Sm/Nd elemental bias.

Measurements made on two other Durango chips (chips #2 and #3; spot size = 40 μm; fluence comprised between 4.5 and 4.8 J/cm²; and frequency between 6 and 10 Hz) during the third day-session confirm all the above observations (Fig. 9). Most of the 143 Nd/ 144 Nd ratios have a value lower than that determined by TIMS. On the contrary, 145 Nd/ 144 Nd determinations scatter around the TIMS value, whereas 147 Sm/ 144 Nd ones are significantly distinct from one Durango chip to another and also within a single chip. These measurements confirm that our Durango crystal is homogeneous for Nd isotopes: 143 Nd/ 144 Nd and 145 Nd/ 144 Nd ratios define similar ranges of variations of 150 and 120 ppm for the 3 chips. The variation of the chemical ratio is of 1.5% (2-σ variations considering all the individual runs) and it is likely to be controlled by the

heterogeneity of our Durango crystal, although small variations of measurements conditions explaining part of the ¹⁴⁷Sm/¹⁴⁴Nd jumps cannot be excluded. The difference in accuracy between ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd ratios does not result from an inaccurate TIMS value of the Durango apatite. Indeed, similar systematics are observed for the measurements made on apatite grains from Fogo (Cape Verde Islands) and Phalaborwa (South Africa; Fig. 10).

Systematic offsets between ¹⁴³Nd/¹⁴⁴Nd ratios measured by LA-MC-ICPMS and their corresponding TIMS value have been previously reported (e.g. Newman *et al.*, 2009; Newman, 2012; Xu *et al.*, 2015; Fisher *et al.*, 2011; Spandler *et al.*, 2016). Fisher *et al.* (2020) have recently suggested that the accuracy of ¹⁴³Nd/¹⁴⁴Nd, ¹⁴⁵Nd/¹⁴⁴Nd and ¹⁴⁷Sm/¹⁴⁴Nd measurements can be greatly influenced by the total sample gas flow and therefore the production of oxides. Here we explore if the difference of offset values (especially for the ¹⁴³Nd/¹⁴⁴Nd ratio) observed between day-sessions 1 to 3 (Fig. 8A) could be related to our sample gas flow. Our measurements were all performed with a flow in the range 0.974 to 0.984 l/mn. This approximatively corresponds to the threshold value from which inaccuracy of measured ratios is observed by Fisher *et al.* (2020; about 0.95 l/mn). Our sample gas flow may then explain why lower-than-TIMS ¹⁴³Nd/¹⁴⁴Nd ratios were systematically measured together with variable values from a day-session to another one. However, following Fisher *et al.* (2020), this would have implied inaccurate ¹⁴⁵Nd/¹⁴⁴Nd ratios which is not observed (Fig. 8B). Thus, the deviations described by Fisher *et al.* (2020) related to the sample gas flow are most certainly comprised in the error-bars of our measurements, but they do not explain our observations. The difference of offset values between day-sessions 1 to 3 is certainly related to a slight change in the instrument behavior.

Inaccurate 143 Nd/ 144 Nd ratios relative to 145 Nd/ 144 Nd may be also related to an inappropriate correction for mass discrimination. In this paper, we consider that the mass-discrimination factor β^{Nd} is the same for the three 146 Nd/ 144 Nd, 143 Nd/ 144 Nd and 145 Nd/ 144 Nd ratios (eq. 4). On the other hand, Thirlwall (2002) proposed that β^{Nd} may be a function of the considered isotopic ratio, so that $\beta^{143/144} < \beta^{145/144} < \beta^{146/144}$. In this case, determining the β^{Nd} factor with the 146 Nd/ 144 Nd ratio may result in an over-correction of the 143 Nd/ 144 Nd and 145 Nd/ 144 Nd ratios, with the bias being more important for 143 Nd/ 144 Nd. For instance, Vance and Thirlwall (2002) when normalizing MC-ICPMS solution data to 146 Nd/ 144 Nd ratio, observed inaccurate 143 Nd/ 144 Nd and 145 Nd/ 144 Nd ratios by 100 and 30 ppm, respectively. Such a factor of ~3 in the bias between 143 Nd/ 144 Nd and 145 Nd/ 144 Nd is not observable in our data. Furthermore, if the 143 Nd/ 144 Nd bias was related to an inappropriate mass-discrimination law such as proposed by Vance and Thirlwall (2002), we should expect to observe a relationship between the Δ^{143} Nd/ 144 Nd (where Δ^{143} Nd/ 144 Nd is the difference between the observed 143 Nd/ 144 Nd ratio and its true, TIMS determined value) and the β^{Nd} factor determined with 146 Nd/ 144 Nd, which is not the case (Supplementary Fig. S2).

As an alternative hypothesis, we also explored the effect of the ¹⁴⁴Sm correction on ¹⁴⁴Nd by considering distinct sets of values for the ¹⁴⁷Sm/¹⁴⁹Sm and ¹⁴⁴Sm/¹⁴⁹Sm ratios (section 3.1). Using ¹⁴⁴Sm/¹⁴⁹Sm = 1.08504 and ¹⁴⁷Sm/¹⁴⁹Sm = 0.22245 (Brandon *et al.*, 2009; Rankenburg *et al.*, 2006) has no influence on the Nd isotopic ratios. Using recent values published by Bouvier and Boyet (2016), i.e., ¹⁴⁴Sm/¹⁴⁹Sm = 1.08506 and ¹⁴⁷Sm/¹⁴⁹Sm = 0.22336, increases the ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd ratios by 100 and 30 ppm, respectively. The ¹⁴³Nd/¹⁴⁴Nd value, however, remains too low (see Supplementary Fig. S3). Interestingly, the effect of the Sm correction documented here is of the same order of magnitude than biases observed by Vance and Thirlwall (2002) and attributed to the use of an inappropriate mass-discrimination law.

Finally, we investigated the effect of the sampling system. Indeed, previous workers have tested the influence of sample- and skimmer-cone combination on Nd isotope measurements (e.g. Newman et al., 2009; Newman, 2012; Xu et al., 2015). In particular, Newman (2012) indicated that the association of Jetcone and X-skimmer allows to greatly improve the analytical sensitivity compare to the N-cone/H-Skimmer association. However, they also indicate that the X-skimmer introduces additional contributions to the instrumental mass fractionation that cannot be corrected for by using a classical exponential correction. These studies clearly show that the use of X-skimmer induces the formation of Nd⁺O which alters the accuracy of the measurements. Finally, results of these previous studies show that by minimizing the Nd⁺O/Nd⁺ (<1%), measured isotope ratios tend to approach the "true" value. Measurements done on our Durango chip with H-skimmer coupled to sampler cone (see Supplementary Fig. S4 and Supplementary Table) have ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd ratios scattering around the TIMS values (although average ¹⁴³Nd/¹⁴⁴Nd values remain slightly lower than those determined by TIMS) so that both isotope ratios are now accurate. Withinrun error-bars reported on isotopic ratios, however, are higher by a factor 2 to 3 than those determined with the reference protocol (X-skimmer and Jet-cone) for the same laser parameters and the 2-σ reproducibility of ¹⁴³Nd/¹⁴⁴Nd is degraded to 400 ppm (4 εNd-units). Considering the ¹⁴⁷Sm/¹⁴⁴Nd chemical ratio, all measured values are higher than the TIMS determination, in agreement with some of our "reference" measurements.

All of the above assumptions (high sample gas flow, inaccurate mass discrimination law, inappropriate Sm isotopic composition, influence of the sampling system) are not mutually exclusive. On the contrary, they surely all participate, to a certain extent, to the variations observed from spot to spot, or to the inaccuracy of some corrected ratios. Small variations in the setting of the instrumentation, from a day-session to another one, may also play a role. In any case, none of the tests realized during this study allows both accurate and precise isotope and chemical ratios. Therefore, the measured data must necessarily be normalized against an external, known reference.

8. External normalization against Durango apatite

All the recent studies on in-situ determination of ¹⁴⁷Sm/¹⁴⁴Nd in apatite have concluded that an external normalization of measured ratios is mandatory and that a matrix-matched standard is required (Fisher *et al.*, 2020; and references therein). Our intercalated NIST610/Durango data agree perfectly with this prerequisite (Fig. 5). Although both samples were measured using the same analytical conditions, ¹⁴⁷Sm/¹⁴⁴Nd correction factors determined for NIST SRM 610 and Durango are significantly different.

Figure 11 presents normalized Durango ratios (data acquired during day-session 2 at constant laser parameters: spot size = 40 μm, fluence = 3.8 J/cm², and frequency = 6 Hz) that are obtained when considering that one measurement every two is an unknow, and compares them with raw values (corrected for baseline, gain, Sm interference and mass discrimination). Recalculated ¹⁴³Nd/¹⁴⁴Nd, ¹⁴⁵Nd/¹⁴⁴Nd and ¹⁴⁷Sm/¹⁴⁴Nd ratios all scatter around their respective TIMS determination and show total reproducibilities of 235 ppm, 230 ppm and 1.36% (n = 46), respectively. These values are up to five times higher than those obtained by Fisher *et al.* (2011) and Yang *et al.* (2014) for ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd ratios (94 and 55 ppm, and 90 and 83 ppm, respectively), but surprisingly they are far better for ¹⁴⁷Sm/¹⁴⁴Nd. Such differences have to be considered, however, with care. Fisher *et al.* (2011) made their analyzes with a laser spot size of 148 μm and normalized data to a LREE glass, whereas measurements by Yang and collaborators (2014) were

performed with a laser spot size of 90 μ m and normalized to AP2 apatite. In our study we considered smaller spot sizes of 40 μ m that seemed to us more reasonable if we want to resolve zoning variations in apatite crystals in geological samples. Also, it has to be mentioned that the 90-ppm reproducibility published by Yang *et al.* (2014) for ¹⁴³Nd/¹⁴⁴Nd corresponds to the two-standard-deviation of 8 sessions during which the reproducibility was up to 120 ppm. As a better comparison, our 93- μ m analyzes give similar reproducibilities of 80 and 85 ppm for ¹⁴³Nd/¹⁴⁴Nd and ¹⁴⁵Nd/¹⁴⁴Nd.

Fogo and Phalaborwa apatite measurements were also normalized to the Durango standard. In the case of Fogo, apatites were separated from a 'zero'-age carbonatite sample and therefore normalized Nd ratios are expected to fit with TIMS determinations that were performed on both apatite and whole carbonatite samples (Doucelance et al., 2010). For Paleoproterozoic Phalaborwa apatites, however, comparison between normalized data and TIMS values may be less pertinent. It can be reasonably assumed that the initial ¹⁴³Nd/¹⁴⁴Nd ratios of apatites were all identical but we cannot exclude small variations of their initial ¹⁴⁷Sm/¹⁴⁴Nd, so that radioactive decay of ¹⁴⁷Sm to ¹⁴³Nd over 2060 Ma may have led to actual Nd compositions and ¹⁴⁷Sm/¹⁴⁴Nd ratios varying from one crystal to another. Since the TIMS measurements were performed on crystals distinct from those analyzed in-situ, some discrepancies may thus appear between the two determinations. This tends to agree with the poorer reproducibility of PhB09 143Nd/144Nd and 147Sm/144Nd TIMS measurements (Table 1) and also with the large range of variations of the 146Nd ion beam signal that was recorded in Phalaborwa apatite grains (Supplementary Table). A better way to interpret these values consists then in comparing initial 143Nd/144Nd ratios recalculating from both TIMS values and Durango-normalized, in-situ data. Results for Fogo and Phalaborwa apatites are reported in Figure 12. Normalization to Durango standard allows to retrieve accurate, measured (Fogo) and age-corrected (Phalaborwa), 143Nd/144Nd ratios relative to the TIMS values. Associated reproducibilities are better than those reported for Durango apatite for 40-µm spot sizes with 123 ppm (1.2 ENd-units) for Phalaborwa initial Nd ratio and between 82 and 144 ppm (0.8 and 1.4 εNd-units) for Fogo ¹⁴³Nd/¹⁴⁴Nd ratio, respectively. Such a gain in reproducibility most likely relates to the higher Nd contents of Fogo and Phalaborwa apatite grains relative to Durango (~2000 ppm against ~1000 ppm, see Table 1 and Doucelance et al., 2010). We are aware that Fogo and Phalaborwa apatites represent a limited set of samples, but these two last comparisons strongly suggest that our Durango crystal can be used as an external standard for recent and ancient (2 Ga) samples. This conclusion is another argument to confirm that the slight Sm/Nd heterogeneity of our standard is limited to maximum ±1-1.5%. Fisher et al. (2020) have indeed shown that significant ENd deviations (> 1 ENd-unit) from the true value are related to overestimate of at least 2% of the 147Sm/144Nd ratio for an age of 2 Ga.

9. Application to natural apatites

In this contribution, we focused our analyzes in apatite crystals extremely rich in LREE (several thousands of ppm of Nd) such as commonly referred in carbonatite complexes (e.g. Djeddi, 2019). Here, we show that the Sm-Nd in-situ technique applied to apatites is particularly powerful for carbonatite samples even at a small spot size, thus allowing the analysis of the finer zoning when this is required. However, the size of the spot needed for such analysis is greatly depending on the igneous apatite content and therefore on the magma from which it grew up. Recent, Sm-Nd in-situ papers (e.g. Hammerli *et al.*, 2019; Fisher *et al.*, 2020) focused on apatite grains from Archean TTG samples that have low REE contents compared to that of Durango and therefore necessitate a much bigger spot than the 40-µm spot size used in this study.

In such material, precise Sm-Nd measurements are challenging. Sm/Nd-zoned regions have resulted in contrasted Nd isotope compositions with time (Fisher *et al.*, 2017) and different zones are more likely to be mixed in a single big spot. On the other hand, apatites from other magmas occurring through geological times such as sanukitoids or typical Basalt Andesite Diorite Rhyolite suite (BADR) can have LREE contents comparable to that of Durango (Bruand *et al.*, 2020). They are also good candidates for Sm-Nd in-situ analysis.

From a general point of view, measurements done with a spot-size of 40 microns in apatite with a Nd content lower than Durango will certainly result in important error-bars on both isotopic and chemical ratios thus limiting their interpretations. In this case, programmed-raster analyzes on cathodoluminescence-imaged apatite crystals allow to increase the counting statistics and thus decrease within-run errors. Figure 13 compares raster (length ~150 μ m) and spot measurements operated one after the other during day-session 2 with the following laser parameters (40 μ m - 3.8 J/cm² - 6 Hz). ¹⁴³Nd/¹⁴⁴Nd, ¹⁴⁵Nd/¹⁴⁴Nd and ¹⁴⁷Sm/¹⁴⁴Nd ratios show fully consistent values between the two sets of data (the ¹⁴⁷Sm/¹⁴⁴Nd variation between the two raster runs is included in the Sm/Nd heterogeneity of Durango). More importantly, within-run error-bars are divided by a factor 2-3 for each ratio (see also Supplementary Table, day-session 2).

10. Summary

In this contribution, we show as previously done (Fisher *et al.*, 2011; Spandler *et al.*, 2016) that results obtained using Jet-interface and X-skimmer reveal a systematic shift between ¹⁴³Nd/¹⁴⁴Nd ratios corrected for Sm contribution and for mass discrimination and their true values, while ¹⁴⁵Nd/¹⁴⁴Nd values agree well. We also show that the spot size, the fluence and the frequency of the laser have no influence on the accuracy of Nd ratios. At most, the values used for these parameters can lead to better signals, and therefore more precise data. We confirm as previously proposed (Newman, 2012; Xu *et al.*, 2015) that accurate Nd ratios can be obtained by using N-cone and H-Skimmer which results in a much lower precision.

In this study, we focused our attention on testing the laser parameters and their influence on the ¹⁴⁷Sm/¹⁴⁴Nd ratio during LA-MC-ICPMS measurements on Durango apatite and both NIST SRM 610 and Sm-doped JNDi synthetic glasses when it was necessary. We confirm the recent observation that the ¹⁴⁷Sm/¹⁴⁴Nd ratio correlates negatively with the spot size (Fisher *et al.*, 2020), but more importantly we show the different contributions that the other laser parameters or a simple sample exchange can have on this ratio. Our main conclusions reveal that the fluence and the frequency of the laser in the range of values that were tested have no influence on the measurement of the ¹⁴⁷Sm/¹⁴⁴Nd ratio. However, we identified that our instrument presents a systematic decreasing drift with time that can account for up to 50% of the variation observed on Durango. This drift, if there is no change in the analysis conditions, settles half-way through five hours of a day-session before to define a plateau. Measured ¹⁴⁷Sm/¹⁴⁴Nd ratios can be also affected by very punctual gas flow changes induced for example by a sample exchange. In this case, it is wise to monitor the ¹⁴⁷Sm/¹⁴⁴Nd values on standards in the 10 minutes following the sample exchange and wait for the stabilization of the chemical ratio. More generally, the chemical ratio is extremely sensitive to the settings made during the analysis session (oxide minimization, z optimization, ...). Thus, settings have to be chosen at the beginning of the session and maintained constant all along.

Considering all these results, we favor the use of higher sensitivity sampling system (Jet-interface and X-skimmer) with a control on the Nd^+O/Nd^+ production (<1%) and the use of a matrix-matched, well-

characterized standard measured with the same laser and MC-ICPMS parameters than samples as external normalization. Our Durango crystal is homogeneous for Nd isotopic ratios, and slightly heterogeneous for Sm/Nd at the level of ± 1 -1.5%. Using it as reference material we retrieve accurate 143 Nd/ 144 Nd ratios for recent (Fogo) and 2060-Ma-old (Phalaborwa) carbonatitic apatites. Precisions determined for a laser spot of 40 μ m (reasonable size for apatite crystals in most geological samples) are in the order of 125-150 ppm (1.3-1.5 ϵ Nd-units) for 143 Nd/ 144 Nd. Such values allow to discuss the source origin of altered samples, whatever their age, that would not be possible to document as a whole. Although Durango normalization is appropriate for the Paleoproterozoic carbonatitic material studied in this contribution, the slight heterogeneity of this natural apatite may also be a limit in back-calculating accurate initial Nd ratios of older samples. Carbonatite apatites have moreover extremely high Nd contents of several thousands of ppm allowing good precision with a 40- μ m spot size. Such a spot size may be a limit, however, to resolve zoning in some igneous apatites such as the ones that are more depleted in LREE (e.g. Archean TTG, Bruand *et al.*, 2020). Raster analyzes for a spot size of 40 μ m and a length of 150 μ m in imaged apatite crystals are able to improve the precision of the measurements by a factor 2-3. This would be a good alternative to improve the precision of less concentrated igneous apatites.

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 $\begin{tabular}{ll} \textbf{Table 1} \\ \textbf{Sm/Nd chemical and isotopic characteristics of Durango crystal available at the LMV and Phalaborwa PHB09 apatites. \end{tabular}$

Sample	[Sm] ppm	[Nd] ppm	¹⁴⁷ Sm/ ¹⁴⁴ Nd	¹⁴³ Nd/ ¹⁴⁴ Nd	¹⁴³ Nd/ ¹⁴⁴ Nd
Durango 1-1 ¹	128	1020	0.0760 (±5)	0.512470 (±6)	0.348407 (±3)
Durango 1-2 ¹	128	1021	0.0760 (±5)	0.512476 (±3)	0.348414 (±2)
Durango 2	131	1038	0.0760 (±5)	0.512473 (±4)	0.348407 (±2)
Durango 3	135	1066	0.0764 (±5)	0.512475 (±3)	0.348407 (±2)
Mean Durango	130 ± 6 ²	1036 ± 43 ²	0.0761 ± 0.0004 ²	0.512473 ± 0.000005 ²	0.348409 ± 0.000007 ²
PhB09-1	413	2108	0.1185 (±8)	0.511274 (±4)	0.348398 (±3)
PhB09-2	415	2110	0.1189 (±8)	0.511286 (±4)	0.348394 (±3)
PhB09-3	398	2049	0.1172 (±8)	0.511261 (±4)	0.348396 (±3)
Mean PhB09	409 ± 19 ²	2089 ± 69 ²	0.1182 ± 0.0017 ²	0.512473 ± 0.000025 ²	0.348396 ± 0.000004 ²

¹ Sample D1 was separated into subsamples D1-1 and D1-2 after acid-digestion and addition of a ¹⁵⁰Nd-¹⁴⁹Sm spike. The good reproducibility between D1-1 and D1-2 attests to the quality of the chemical separation. ² Uncertainty in calculated means is 2 standard deviations.

Table 2 MC-ICPMS cup configuration and potential interferences.

Faraday Cup	L4	L3	L2	L1	Axial	H1	H2
Main isotopes	¹⁴³ Nd	¹⁴⁴ Nd	¹⁴⁵ Nd	¹⁴⁶ Nd	¹⁴⁷ Sm	¹⁴⁸ Nd	¹⁴⁹ Sm
Interferences		¹⁴⁴ Sm		¹³⁰ Ba ¹⁶ O		¹⁴⁸ Sm+ ¹³² Ba ¹⁶ O	

Table 3 LA-MC-ICPMS parameters.

MC-ICI	PMS	Laser ablation		
Model	Neptune plus	Туре	Excimer 193-nm M-50E	
Forward power	1200 W	Model	Resonetics	
Mass resolution	400	Mode	1 spot ~50 cycles	
Integration time	4.004 -	Frequency	5-10 Hz	
	1.094 s	Spot size	33-93 microns	
Gas flows (l/min)		Laser fluence	3.5-5 J/cm ²	
Cool / plasma	15.50	Gas flows (ml/mn)		
Auxiliary (Ar)	0.72	Helium	650-800	
Sample (Ar)	0.974-0.984	Nitrogen	2-3.5	

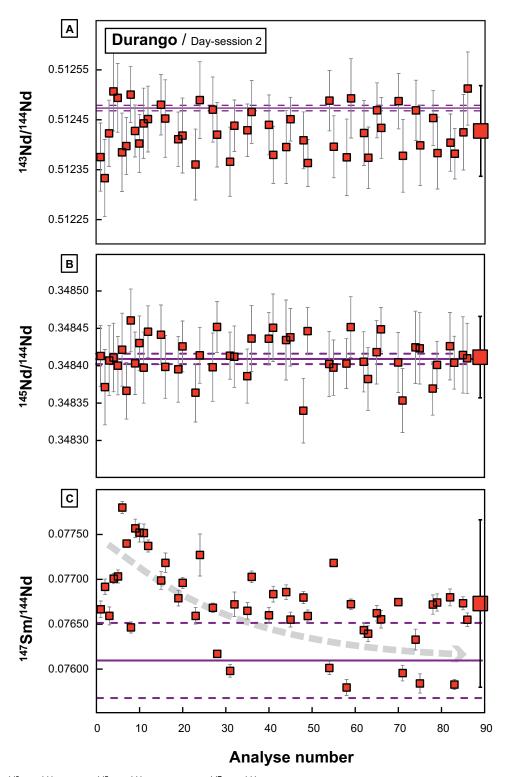
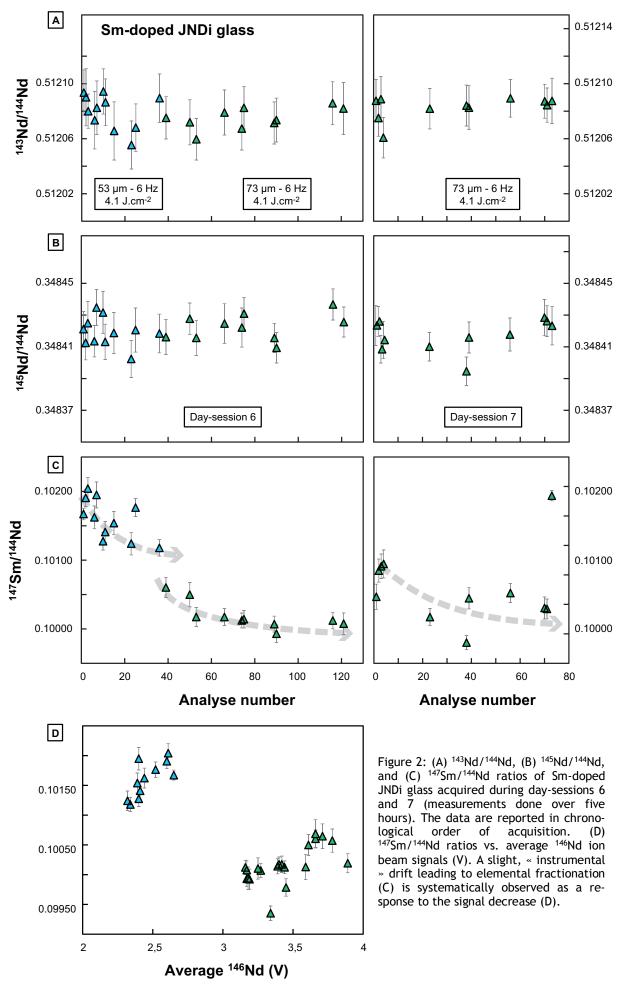


Figure 1: (A) 143 Nd/ 144 Nd, (B) 145 Nd/ 144 Nd, and (C) 147 Sm/ 144 Nd ratios of Durango apatite acquired during day-session 2 (measurements done for five hours) at constant laser parameters (spot size = 40 μ m, fluence = 3.8 J/cm², and frequency = 6 Hz). The data are reported in chronological order of acquisition. Also plotted are the TIMS determinations (purple lines) \pm 2 s.d. (dashed, purple lines). A slight, « instrumental » drift leading to elemental fractionation (C) cannot be excluded.



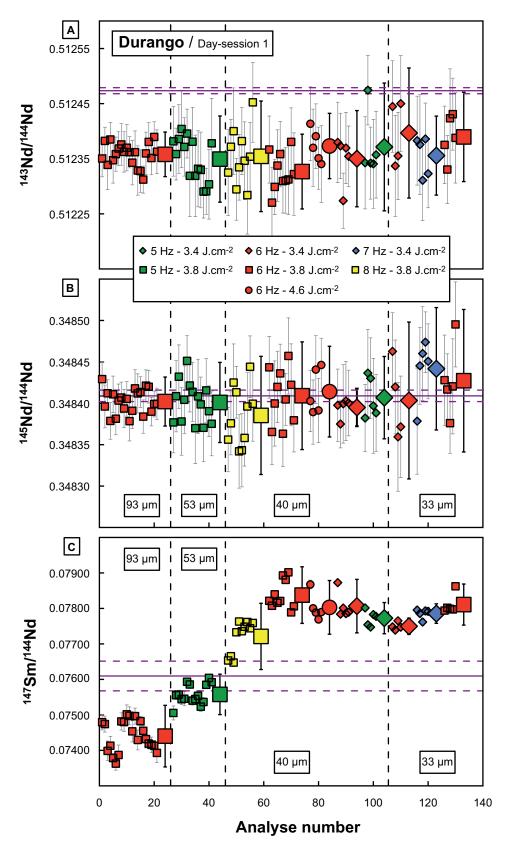


Figure 3: (A) 143 Nd/ 144 Nd, (B) 145 Nd/ 144 Nd, and (C) 147 Sm/ 144 Nd ratios of Durango apatite measured by LA-MC-ICPMS during day-session 1. The data are reported in chronological order of acquisition. Each symbol corresponds to a given set of frequency and fluence (see inset between A and B); values inside white rectangles refer to the size of the spot. Also reported are the TIMS determinations (purple lines) ± 2 s.d. (dashed, purple lines).

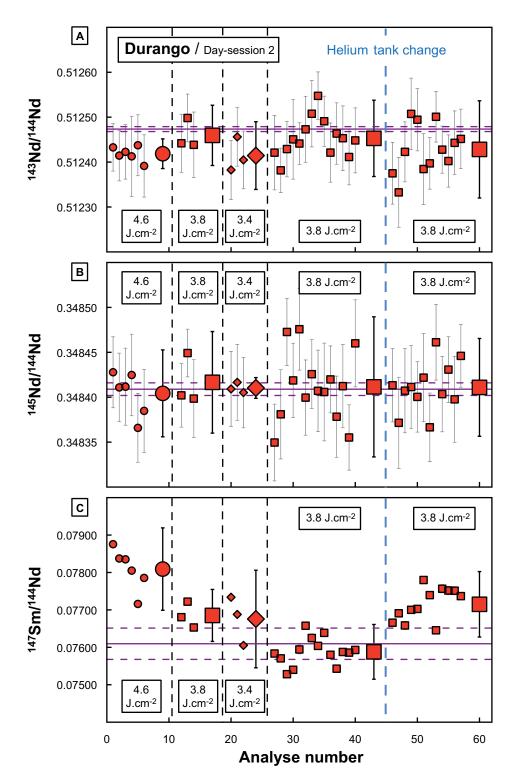


Figure 4: Effect of laser fluence on (A) 143 Nd/ 144 Nd, (B) 145 Nd/ 144 Nd, and (C) 147 Sm/ 144 Nd ratios of Durango apatite (spot-size = 40 µm, frequency = 6 Hz). Analyzes were performed during day-session 2. The dashed, blue line indicates when the helium tank was changed. The data are reported in chronological order of acquisition (note that the 12 runs performed after the tank change correspond to the first 12 runs reported on Figure 1). Also plotted are the TIMS determinations (purple lines) \pm 2 s.d. (dashed, purple lines).

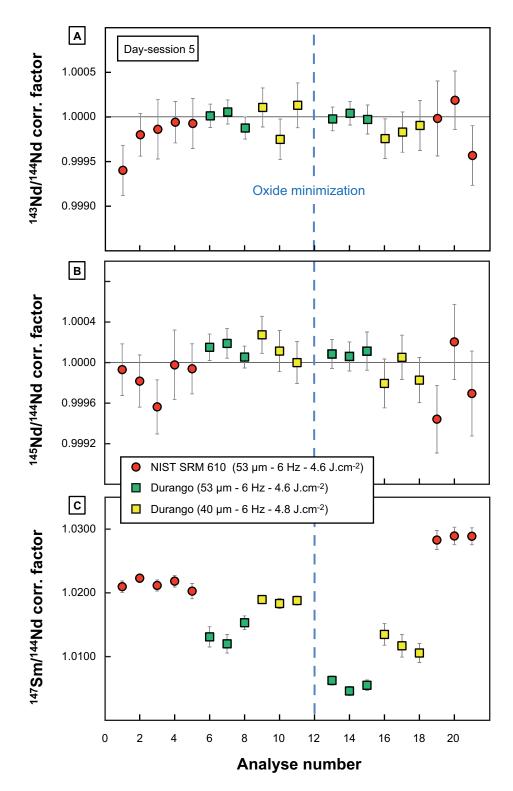


Figure 5: Effect of oxide minimization on (A) ¹⁴³Nd/¹⁴⁴Nd, (B) ¹⁴⁵Nd/¹⁴⁴Nd, and (C) ¹⁴⁷Sm/¹⁴⁴Nd ratios of Durango apatite and NIST SRM 610. Measurements were done with H-skimmer coupled to sampler cone. Correction factors corresponds to the ratios of measured values over reference ones. Values for NIST SRM 610 are from Yang *et al.* (2014). Analyzes were performed during day-session 5; they are reported in chronological order of acquisition. The dashed, blue line indicates when the minimization was operated.

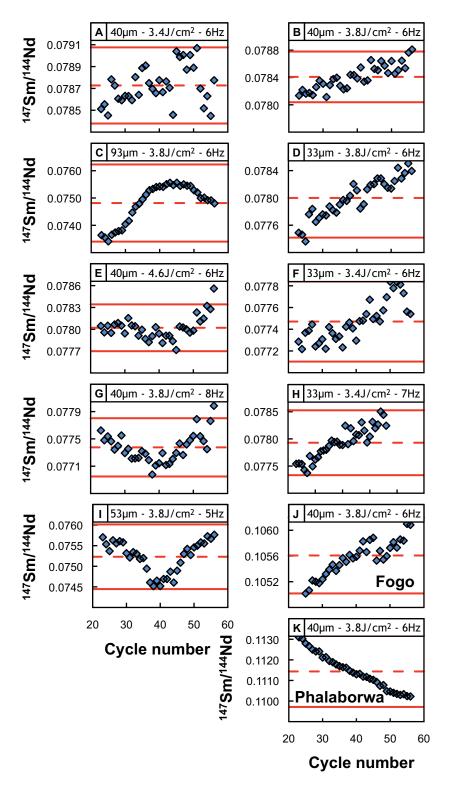


Figure 6: (A) to (K) report cycle-to-cycle 147 Sm/ 144 Nd ratios measured in Durango, Fogo and Phalaborwa apatites for different sets of laser parameters (values indicated inside white rectangles). Also plotted are the mean value (dashed, red lines) ± 2 s.d. (red lines) for each analysis. All data were acquired during successive day-sessions 1, 2 and 3.

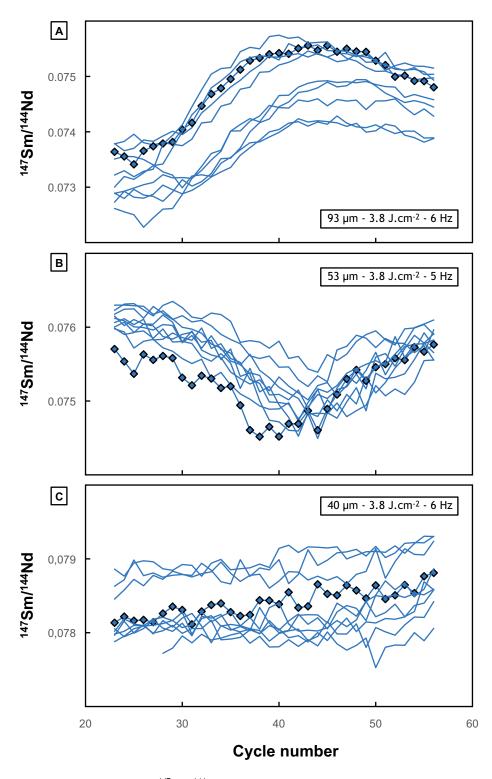


Figure 7: Reproducibility of cycle-to-cycle ¹⁴⁷Sm/¹⁴⁴Nd variations (10 consecutive analyzes). Values inside white rectangles give the laser parameters. Patterns with blue diamonds are those reported in figures 6C, 6I and 6B, respectively.

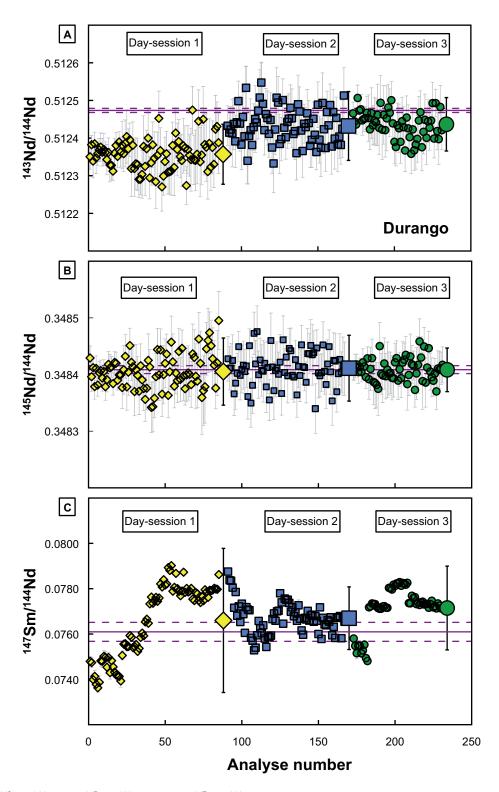


Figure 8: (A) 143 Nd/ 144 Nd, (B) 145 Nd/ 144 Nd, and (C) 147 Sm/ 144 Nd ratios of Durango apatite measured in the same chip during 3 day-sessions of acquisition (3 sessions over one month: 2 successive days and another one 30 days later) for various laser parameter values. The data are reported in chronological order of acquisition. Also plotted are the TIMS determinations (purple lines) ± 2 s.d. (dashed, purple lines).

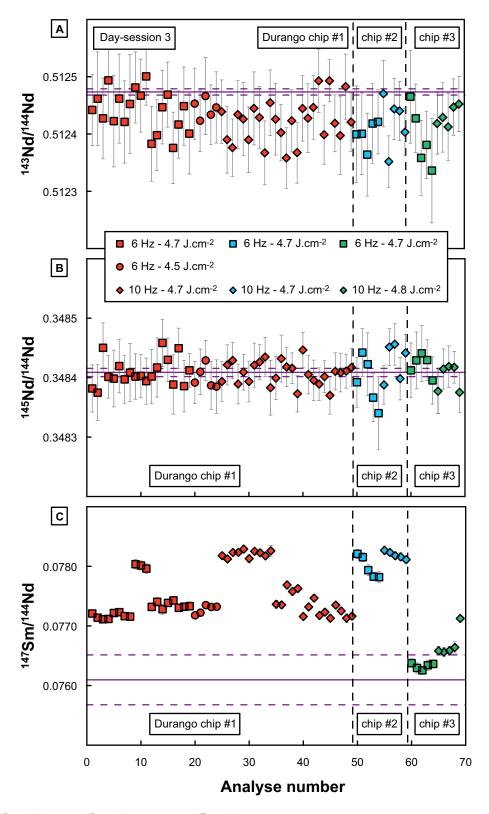


Figure 9: (A) 143 Nd/ 144 Nd, (B) 145 Nd/ 144 Nd, and (C) 147 Sm/ 144 Nd ratios of Durango apatite measured in three distinct chips during the third day-session (spot size = 40 μ m, fluence and frequency are given in the inset). The data are reported in chronological order of acquisition. Also plotted are the TIMS determinations (purple lines) ± 2 s.d. (dashed, purple lines).

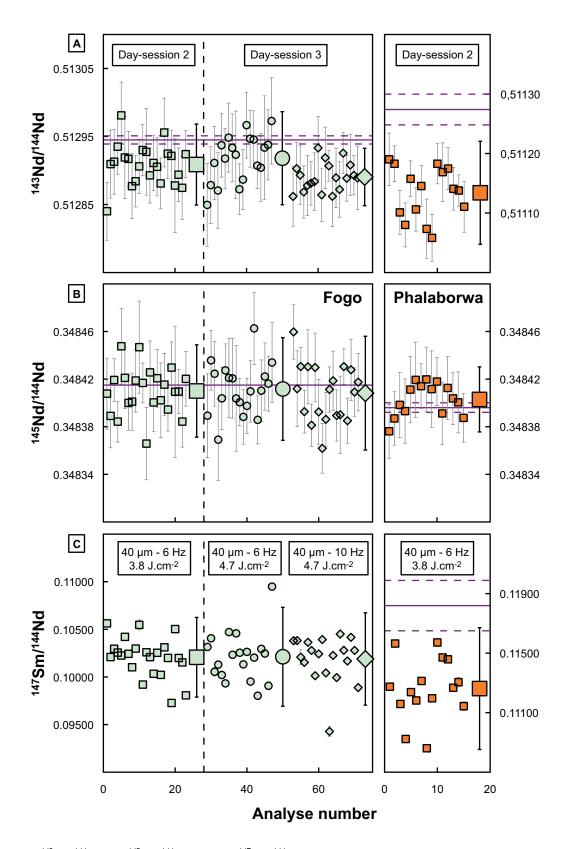


Figure 10: (A) 143 Nd/ 144 Nd, (B) 145 Nd/ 144 Nd, and (C) 147 Sm/ 144 Nd ratios measured in apatite grains from Fogo, Cape Verde Islands (left column), and Phalaborwa, South Africa (right column). The data are reported in chronological order of acquisition. Also plotted are the corresponding TIMS determinations (purple lines) ± 2 s.d. (dashed, purple lines). The accuracy and precision of Fogo analyzes are few dependent on the energy or the frequency of the laser.

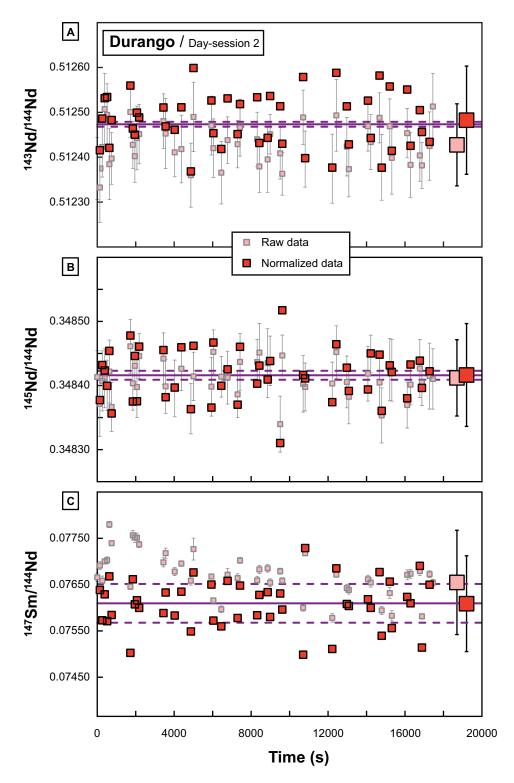


Figure 11: External normalization of (A) 143 Nd/ 144 Nd, (B) 145 Nd/ 144 Nd, and (C) 147 Sm/ 144 Nd ratios of Durango apatite (40 µm, 3.8 J/cm², 6 Hz). Each measurement (at given t) is normalized to the mean value of (t-1) and (t+1) analyzes assuming a linear, step-by-step, time-depending variation of measured ratios and TIMS determinations. Pink symbols are measured data, and red squares are normalized ones. Also plotted are the corresponding TIMS determinations (purple lines) \pm 2 s.d. (dashed, purple lines).

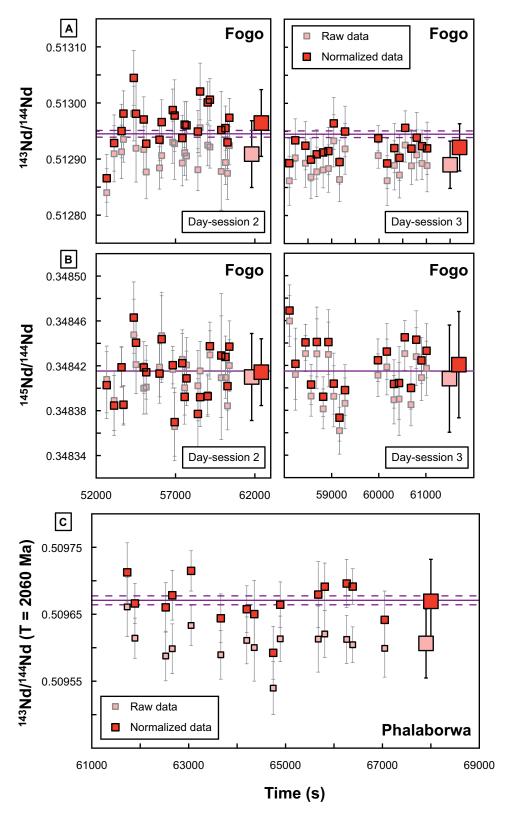


Figure 12: External normalization of (A) 143 Nd/ 144 Nd and (B) 145 Nd/ 144 Nd ratios of Fogo grains, and (C) initial 143 Nd/ 144 Nd ratios of Phalaborwa grains against Durango apatite. Pink symbols are measured data, and red squares are normalized ones. Also plotted are the corresponding TIMS determinations (purple lines) \pm 2 s.d. (dashed, purple lines).

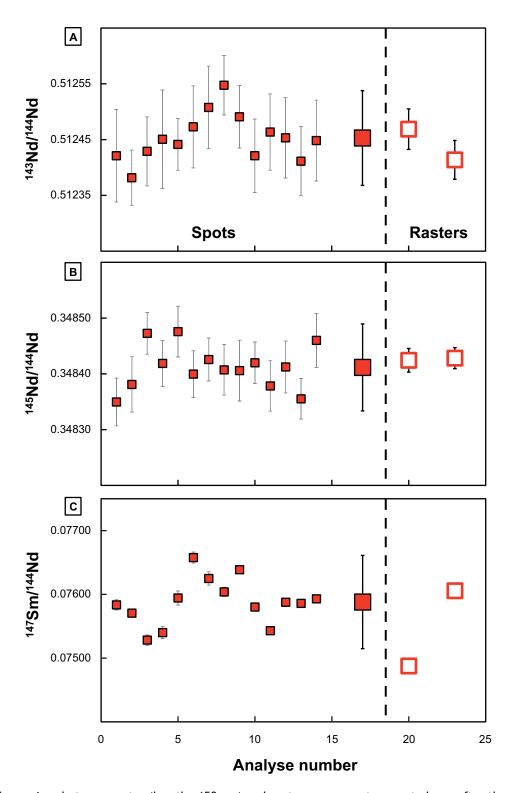


Figure 13: Comparison between raster (length ~150 μ m) and spot measurements operated one after the other during day-session 2 in Durango Apatite. A) 143 Nd/ 144 Nd, (B) 145 Nd/ 144 Nd, and (C) 147 Sm/ 144 Nd. Laser parameters are as follows: 40 μ m - 3.8 J/cm² - 6 Hz. Within-run error-bars are divided by a factor 2-3 for each ratio.