

# Zircon U-Pb geochronology, major-trace elements and Sr-Nd isotope geochemistry of Mashhad granodiorites (NE Iran) and their mafic microgranular enclaves: Evidence for magma mixing and mingling

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 Zircon U-Pb geochronology, major-trace elements and Sr–Nd isotope geochemistry of Mashhad granodiorites (NE Iran) and their mafic microgranular enclaves: Evidence for magma mixing and mingling

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Mashhad granitoids and associated mafic microgranular enclaves (MMEs), in NE Iran record late early Mesozoic magmatism which was related to the Paleo-Tethys closure and Iran-Eurasia collision. These represent ideal rocks to explore magmatic processes associated with Late Triassic closure of the Paleo-Tethyan ocean and post-collisional magmatism. In this study, new geochronological data, whole-rock geochemistry, and Sr-Nd isotope data are presented for Mashhad granitoids and MMEs. LA-ICP-MS U-Pb dating of zircon yields crystallisation ages of 205.0±1.3 Ma for the MMEs, indicating their formation during the Late Triassic. This age is similar to the host granitoids. Our results including the major and trace elements discrimination diagrams, in combination with field and petrographic observations (such as ellipsoidal MMEs with feldspar megacrysts, disequilibrium textures of plagioclase) as well as mineral chemistry, suggest that MMEs formed by mixing of mafic and felsic magmas. The host granodiorite is a felsic, high K calc-alkaline I-type granitoid, with  $SiO_2 = 67.5-69.4$  wt%, high K<sub>2</sub>O (2.4-4.2 wt%), and low Mg# (42.5–50.5). Normalized abundances of LREEs and LILEs are enriched relative to HREEs and HFSEs (e.g. Nb, Ti). Negative values of whole-rock  $\epsilon Nd_{(1)}$  (-3 to -2.3) from granitoids indicate that the precursor magma was generated by partial melting of enriched lithospheric mantle with some contributions from old lower continental crust. In the MMEs, SiO<sub>2</sub> (53.4–58.2 wt%) is lower and Ni (3.9–

49.7 ppm), Cr (0.8–93.9 ppm), Mg# (42.81–62.84), and  $\varepsilon Nd_{(1)}$  (-2.3 to +1.4) are higher than those in the host granodiorite, suggesting a greater contribution of mantle-derived mafic melts in the genesis of MMEs.

**Keywords:** Mafic microgranular enclaves; Mashhad granodiorite; Magma mixing; Paleo-Tethys ocean; NE Iran.

#### **1. Introduction**

Mafic microgranular enclaves (MMEs; Didier and Barbarin 1991) are common in most calc-alkaline granitoids (Xiong et al. 2012; Jiang et al. 2013; Chen et al. 2015) and have been documented in most Cordilleran-type granitoid batholiths (e.g. Pitcher 1983; Didier and Barbarin 1991; Barbarin 1999). MMEs have mafic to intermediate compositions, and occur as angular to rounded enclaves within host intrusive rocks whose silica contents range from intermediate to felsic. The formation of MMEs has been attributed to mixing and/or mingling of mafic and felsic magmas (e.g. Barbarin 2005; Wang et al. 2012), pieces of refractory rock or of melt residues (e.g. Chappell et al. 1987; Chappell and Wyborn 2012), or as chilled margins or cumulates of early-formed crystals (e.g. Donaire et al. 2005; Shellnutt et al. 2010). Numerous studies have confirmed that MMEs in granitoids are formed due to injection of a mafic magmatic component into a felsic magma chamber via a mixing process (e.g. Blundy and Sparks 1992; Barbarin 2005; Jiang et al. 2013). 

The Mashhad granitoids and associated MMEs are part of the Mesozoic granitoids of northwestern Pamir-Mashhad magmatic arc, situated between Mashhad in Iran and Paropamisus Mountains in Afghanistan. These granitoids were formed as a result of the northward subduction of the Paleo-Tethys oceanic lithosphere beneath the Turan plate and during the collision of Iran with Eurasia after the closure of Paleo-Tethys ocean (Golonka 2003; Mirnejad et al. 2013) 

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1	(Figure 1). The Mashhad area contains the Mashhad batholith, Dehnow stock and their MMEs
2	(Figure 2; Samadi et al. 2014). Mirnejad et al. (2013) reported three distinct zircon U-Pb ages for
3	the Mashhad granitoids, ranging from 217 to 200 Ma. They believe these ages correspond to
4	subduction of Paleo-Tethys under the Turan plate (Eurasia) which was prevailed during Late
5	Triassic. Karimpour et al. (2010a) obtained U-Pb zircon ages of 215-217 Ma for the isolated and
6	small quartz diorite-tonalite-granodiorite bodies that outcrop in the Dehnow-Vakilabad regions,
7	northwest of the Mashhad batholith. The other U-Pb zircon ages reported by Karimpour et al.
8	(2010b, 2011) and Abbasi (2018) from the main Mashhad batholith are $205.9 \pm 4.1$ Ma and
9	$200.1 \pm 3.8$ Ma for the collision- and arc-related granitoids, respectively. However, no data exist
10	on the nature of the relationship between granitoid rocks and their MMEs. The MMEs, which
11	occur with various shapes in Mashhad granitoids, present a unique opportunity to describe the
12	petrogenesis, origin and geodynamic setting of granitoid magmas (e.g. Didier and Barbarin 1991;
13	Barbarin 2005; Ventura et al. 2006). MMEs can also provide evidence for crust-mantle
14	interaction, which is not available from their host granitoids (e.g. Griffin et al. 2002; Zhao et al.
15	2012). Thus, a detailed study of MMEs in the Mashhad granodiorite can improve our
16	understanding of igneous processes related to the Paleo-Tethys subduction. This paper reports
17	new zircon U-Pb ages, whole-rock chemical compositions, and Sr-Nd isotope data from MMEs
18	and their host granodiorite in the Mashhad batholith. Our findings, placed in the context of other
19	published tectonomagmatic data and stratigraphic relationships in the region, offer insight about
20	the nature and genesis of MMEs and associated host granodiorite as well as their tectonic
21	significance during the closure of the Paleo-Tethys ocean.

### 22 2. Geologic setting, field observations and petrologic descriptions

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During the Late Paleozoic and early Mesozoic eras, a north-dipping subduction zone was present at the southern boundary of the Turan platform (Eurasia), resulting in an extensive magmatic arc that stretched from what is today the Greater Caucasus through the Paropamisus Range, and into the northern Pamir Mountains (Figure 1(a)) (Aharipour et al. 2010; Rossetti et al. 2017; Shafaii Moghadam et al. 2017). It has been proposed that drifting of a series of continental blocks such as Lhasa, Turan, etc. from Gondwana and their collision with Eurasia during Late Paleozoicearly Mesozoic led to the establishment of the Paleo-Tethys orogenic belt throughout the southern margin of Eurasia (Zanchetta et al. 2009). The Paleo-Tethys orogenic belt extends into northern Iran, along the Alborz Mountains from the Binalood Mountains in the east to the Talysh Mountains in the west (Figure 1(b)) (Alavi 1991; Zanchetta et al. 2009). The Mashhad area in northeastern Iran records the geodynamic evolution of the Paleo-Tethys

orogenic belt in the western Pamir-Mashhad magmatic arc (Alavi 1991; Zanchi et al. 2009), which consists of three parallel NW-SE zones: (1) Kopet Dagh, (2) a transitional zone and (3) Binalood (Alavi 1991; Samadi et al. 2014) (Figure 1(b)). The transitional zone, situated between Binalood and Kopet Dagh (Figure 1(b)), was formed by collision between the Eurasian Turan domain in the north, and a Gondwana-derived central Iran block in the south (Alavi 1991; Mirnejad et al. 2013; Samadi et al. 2014). Late Neoproterozoic crust has also been reported in the transitional zone, between the Gondwana and Turan plates (Ghavi et al. 2018).

The Mashhad granodiorite (59°22'-59°45'N, 36°00'-36°22'E) intruded into the transitional zone along major faults of the Paleo-Tethys suture. The oldest units (Figure 2) comprise Late Paleozoic Mashhad–Fariman ophiolites, including ultramafic to mafic-ultramafic lavas, isotropic gabbros, pelagic limestone, interbedded radiolarite, phyllite, and slate (interpreted as an accretionary prism) (Alavi 1991; Ghazi et al. 2001). They are unconformably overlain by Upper 

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1	Triassic to Jurassic siliciclastic deposits including quartz conglomerate, sandstone, crystallized
2	limestone, shale, and phyllite (Figure 2) (Alavi 1991; Eftekharnezhad and Behrozi 1991).
3	<sup>40</sup> Ar/ <sup>39</sup> Ar dating of hornblende gabbros in the Mashhad ophiolite, near Mashhad granitoids
4	(Figure 2(a) and (b)) provides ages of ca. 288 to 282 Ma (Ghazi et al. 2001). The Mashhad
5	batholith consists of three distinct granitoid intrusions: (1) a quartz diorite-tonalite-granodiorite
6	emplaced from $217 \pm 4$ to $215 \pm 4$ Ma (Karimpour et al. 2010), (2) a second granodiorite from
7	$212.0 \pm 5.2$ to $200.1 \pm 3.8$ Ma (Mirnejad et al. 2013; Abbasi 2018), and (3) a monzogranite
8	intrusion around $199.8 \pm 3.7$ Ma (Mirnejad et al. 2013). They were intruded into Late Paleozoic
9	metamorphic rocks in the northwestern Mashhad ophiolite (Figures 1(b) and 2) which resulted in
10	a narrow contact metamorphic aureole consisting of almandine- and andalusite-rich pelites,
11	quartzite, and calc-silicate rocks (Iranmanesh and Sethna 1998). The aforementioned rocks are
12	overlain by Cretaceous conglomerate, arkose, and limestone (Figure 2). The NW-SE trend of the
13	Mashhad granodiorite (Figure 2(a)) agrees with emplacement of the Mashhad granitoids during
14	NE-SW directed extension and major faults relating to the Paleo-Tethys suture. The
15	granodiorites are volumetrically the most important part of the granitoids, and contain abundant
16	MMEs (Figure 3), which are scarce or absent in the quartz diorite, and monzogranite.

Abundant MMEs in the granodiorite have rounded to ellipsoidal forms that range in size from 4 cm to 5 m on their longest axis (Figure 3(a-c)). Rare small sigmoidal and elongated enclaves occur locally, reflecting deformation during flow of the host magma (Figure 3(b)). In some places, MMEs with massive (Figure 3(c)) and banded structures (Figure 3(d)), or with linear (Figure 3(e)) or irregular boundaries (Figure 3(f)) enriched in fine-grained mafic material, are present in the host rocks. The MMEs are abundant, mostly dark-colored, fine-grained, and generally massive and show variable shapes (e.g. rounded, oval, ellipsoidal) (Figure 3). Notably, enclaves partially enclose megacrysts of the host rocks (Figure 3(g)), and MMEs intruded by
granitoid magma are common. Margins of some MMEs exhibit a light-coloured zone enriched in
feldspar (Figure 3(h)).

# 4 3. Petrography

The host granodiorite has a medium-grained, holocrystalline texture and contains mainly euhedral to subhedral phenocrysts of K–feldspar (30–42%), plagioclase (~35%), quartz (15– 25%), and amphibole and/or biotite (4–5%) (Figure 4(a)). Subhedral to anhedral K–feldspar (0.5–2 mm) is mostly perthitic. Plagioclase occurs as subhedral to anhedral crystals with compositions ranging from andesine to oligoclase. Magnetite, ilmenite, zircon, apatite  $\pm$  titanite are the principal accessory minerals.

MMEs generally contain the same mineral assemblage but with higher contents of amphibole, biotite and plagioclase, and lower contents of quartz and K-feldspar. They contain plagioclase (50-56%), K-feldspar (10-16%), hornblende (23-26%), biotite (15-20%), minor quartz, and accessory magnetite, needle-shaped apatite, titanite, and zircon. The textures vary from fine-grained to porphyritic (Figure 4(b)). Plagioclase occurs as either large megacrysts (Figure 4(c)), or as elongated laths within large megacrysts (Figure 4(d)) possibly with complex oscillatory zoning (Figure 4(e)). Minor quartz occurs either as megacrysts or as anhedral crystals that are interstitial to other minerals (Figure 4(f)). Hornblende and biotite are the most abundant mafic minerals in the MMEs, occurring as subhedral to anhedral grains (Figure 4(b,e)), or as anhedral crystals interstitial to plagioclase (Figure 4(f)). Biotite shows greenish brown to reddish-brown pleochroism. Apatite, a common accessory mineral in the MMEs, occurs as acicular crystals (Figure 4(c)). 

# 1 4. Sampling and analytical methods

Two hundred samples were collected from MMEs and their host granodiorite. From these samples, 140 representative thin sections were prepared and studied by optical microscopy as a basis to select the most suitable samples for geochemical and isotopic analyses. Mineral chemistry analyses for selected minerals of the enclaves and the host granodiorite were performed using the JEOL JXA-8530F field-emission electron microprobe at the Laboratoire Magmas & Volcans (LMV), Clermont–Ferrand, France. The minerals were analyzed at an accelerating voltage of 15 kV using a 10 nA probe current with a spot size of 2 µm to reduce volatilization. Matrix corrections were performed using the ZAF correction scheme.

Major- and trace-element contents of the twenty-five samples were measured at the Laboratoire Magmas & Volcans (LMV) in Clermont–Ferrand, France. For major element analysis, powdered samples (100 mg) were mixed with 300 mg of LiBO<sub>2</sub> in a graphite crucible and melted in an induction oven at 1050 °C for 4 minutes. The glass bead was then dissolved in dilute nitric acid, and the resulting solution was further diluted to reach a dilution factor of 2000 (solution/rock weight ratio). The final solutions were analysed by inductively-coupled-plasma-optical-emission-spectrometry (ICP-OES, HORIBA Jobin Yvon Ultima C). Trace element and Sr-Nd isotope analysis were performed on the same dissolution: 50 mg of powder were mixed with 200 mg of purified ammonium bifluoride ( $NH_4HF_2$ ) in 5 ml Savillex screw-top Teflon vials and placed in an oven at 230°C for 24 hours, which ensures total dissolution of the most resistant minerals. Once cooled, the excess of digestion reagent was eliminated by adding and evaporating 2 ml of 7M HNO<sub>3</sub>. The final residue was then totally dissolved in ca. 20-24 g of 7M HNO<sub>3</sub>. An aliquot of this solution (0.5 g hosting ca. 1 mg of rock) was diluted by a factor of ca. 12 with 0.05M HF to reach a rock dilution factor between 5000 and 6000, and trace element abundances 

were determined by inductively coupled-plasma mass spectroscopy (ICPMS, Agilent 7500). The analyses were performed in plasma robust mode (1550 W). The reaction cell (He mode) was used to reduce interferences on masses ranging from 45 (Sc) to 75 (As). The signal was calibrated externally (every 4 samples) with the BE-N basaltic standard (CRPG Nancy) dissolved in the same manner as the samples, and using the recommended elemental values of Chauvel et al. (2011). The external reproducibility ( $2\sigma$  error) of the method is 6% or less for lithophile elements (K, Rb, Ba, ...) and 15% or less for chalcophile elements (Pb, Cu, As, ...). Three analyses of the USGS G2 granite standard yielded concentrations that agree, within error, with the GeoRem recommended values (<u>http://georem.mpch-mainz.gwdg.de/</u>).

The remaining solution was reserved for Sr and Nd isotope analysis. After evaporation, samples were taken up in 6 M HCl and evaporated before final dissolution in 1.25 M HCl. Iron was removed from the samples using a strong cation exchanger (Bio-Rad AG50-X4, 200-400 mesh) (Pin and Zalduegui 1997). A tandem column elution scheme using Sr.Spec and Tru.Spec chromatographic materials was then used to achieve the concomitant separation of Sr and LREE (Pin et al. 1994). Nd was then purified from the adjacent lanthanides, especially Sm, using the Ln.Spec chromatographic material (Pin and Zalduegui 1997). The isotopic compositions were analyzed on a TRITON thermal ionisation mass spectrometer (for Sr) and a NEPTUNE plasma source mass spectrometer (for Nd), either in static or dynamic mode. <sup>87</sup>Sr/<sup>86</sup>Sr and <sup>143</sup>Nd/<sup>144</sup>Nd ratios were normalized to  ${}^{86}$ Sr/ ${}^{88}$ Sr = 0.1194 and  ${}^{146}$ Nd/ ${}^{144}$ Nd = 0.7219, respectively. Sr and Nd isotope ratios were further normalized relative to  ${}^{87}\text{Sr}/{}^{86}\text{Sr} = 0.710245 \pm 13$  (2 SD, 12 measurements) for NBS SRM 987, and to  ${}^{143}$ Nd/ ${}^{144}$ Nd = 0.512115 ± 20 (2 SD, 7 measurements) for JNdi. Initial <sup>87</sup>Sr/<sup>86</sup>Sr and <sup>143</sup>Nd/<sup>144</sup>Nd were subsequently obtained by back-calculation to ages of igneous crystallization determined by U-Pb on zircon. 

U-Th-Pb isotope data on zircon grains were obtained at the Laboratoire Magmas & Volcans
(LMV), Clermont-Ferrand, France by Laser-Ablation Inductively-Coupled Plasma MassSpectrometry (LA-ICP-MS) with a Thermo Element XR Sector Field high-resolution ICP-MS,
coupled to a Resonetics M50E 193 nm excimer laser system. The analytical method for isotope
dating with LA-ICP-MS is detailed in Mullen et al. (2018).

## 6 5. Analytical Results

# 5.1 Mineral chemistry

8 Representative chemical compositions of plagioclase, biotite and amphibole from the 9 granodiorite and associated MMEs are given in Supplementary Table 1. Plagioclase from the 10 Mashhad MMEs are albite-andesine in composition (An = 10.0%-31.6%). Plagioclases are 11 zoned, with a core of albitic composition and relative low An content (10.0%). The An content 12 of plagioclases shows a distinct increase to 31.6% in the rim with a rhythmic zonation, indicating 13 a reverse and complex oscillatory zonings (Figure 5a).

The Mg cations of biotite in granodiorite and MMEs are 1.8-2.5 and 1.5-3.4 atoms per formula unit (a.p.f.u.), whereas their Fe<sup>2+</sup> cations are 2.2-2.7 and 1.7-2.7 a.p.f.u., respectively. Biotites from granitoids have  $X_{Mg}$  (Mg/Mg + Fe) and Ti cations ranging from 0.41 to 0.52 and 0.19 to 0.41 a.p.f.u., respectively, while those from MMEs range from 0.35-0.65 and 0.11-0.43 a.p.f.u, respectively. The amphiboles are characterized by low TiO<sub>2</sub> (0.097-1.36 wt. %) and alkali (Na<sub>2</sub>O + K<sub>2</sub>O) contents (0.60-3.24 wt. %). The amphibole crystals in the enclaves usually have higher MgO and lower  $FeO_T$  contents than those in the host granodiorite (Supplementary Table 1). The amphiboles in the granodiorite and its MMEs have high Ca<sub>B</sub> values of 1.66–1.96, low Na<sub>B</sub> values 

of 0.23–0.52, and moderate Mg# values of 41.29–58.85, indicating they have mainly calcic and
ferrohornblendes to magnesiohornblende composition (Figure 5(e)).

### 3 5.2 Bulk-rock major and trace elements

Supplementary Table 2 provides whole-rock chemical compositions of representative (twenty-five) samples from Mashhad granodiorite and its MMEs. The granodiorite shows a restricted range in terms of major and trace element concentrations (Supplementary Table 2). It has moderate SiO<sub>2</sub> (67.5-69.4 wt%), Al<sub>2</sub>O<sub>3</sub> (15.2-17.6 wt%), Na<sub>2</sub>O (3.5-6.2 wt%) and K<sub>2</sub>O (2.4-4.2 wt%), and low MgO (0.9–1.6 wt%) contents. Bulk-rock compositions are mainly granodiorite (Figure 6(a)). In the quartz-alkali feldspar-plagioclase (QAP) diagram after (Streckeisen 1979), the Mashhad intrusive rocks show granodiorite to monzogranite compositions (Figure 6b), with high K calc-alkaline affinity (Figure 6(c)). Granodiorite is metaluminous to weakly peraluminous with A/CNK [molar Al<sub>2</sub>O<sub>3</sub> / (CaO + Na<sub>2</sub>O + K<sub>2</sub>O)] less than 1.2 (i.e., regarded as an upper limit for I-type granites as proposed by Chappell (1974) (Figure 6(d)). 

Relative to the host granodiorite, the MMEs exhibit more variable SiO<sub>2</sub> (47.98–65.97 wt%), K<sub>2</sub>O (1.96–3.53 wt%) and Al<sub>2</sub>O<sub>3</sub> (11.66–19.42 wt%), and higher MgO (1.43–6.00 wt%) and Na<sub>2</sub>O (2.01–6.60 wt%) contents. In a SiO<sub>2</sub> vs. Na<sub>2</sub>O + K<sub>2</sub>O classification diagram, MME compositions fall in monzonite, monzodiorite, or diorite fields (Figures 6(a, b)), and high-K calc-alkaline to shoshonite compositions on a K<sub>2</sub>O vs. SiO<sub>2</sub> diagram (Figure 6(c)). Accordingly, the MMEs, with intermediate composition, range from metaluminous to slightly peraluminous, with an aluminasaturation index below 1.2 (Figure 6(d)).

Figures 7(a) and (b) exhibit chondrite-normalized REE patterns for Mashhad granodiorite and its
MMEs (Sun and McDonough 1989). REE patterns for the Mashhad granodiorite show

enrichment in light REEs (LREEs) compared to heavy REEs (HREEs) ((La/Yb)n=13.67-62.26) and minor negative Eu anomalies (Eu/Eu\*=0.84-0.90). MMEs show LREE-fractionated patterns with (La/Yb)n=15.77-70.31 and slightly steeper patterns relative to host granodiorite. Total REE concentrations in granodiorite range from 198 to 370 ppm, while those from MMEs vary from 235 to 720 ppm. The primitive mantle-normalized (Sun and McDonough 1989) multi-element variation diagrams of the granodiorite and associated MMEs from the Mashhad (Figure 7 (c) and (d)) resemble those of arc-related magmas (Pearce and Peate, 1995) with enrichments in large-ion lithophile elements (LILEs; e.g. Rb, Ba and Sr), relative to high field strength elements (HFSEs; i.e., Nb, P and Ti) and HREEs (i.e., Yb). The studied granodiorite and MMEs display marked negative Nb, Zr, P, and Ti anomalies and a positive Sr anomaly relative to the primitive mantle (Figure 7). 

When plotted on Harker diagrams, data for the granodiorite and MMEs exhibit negative
correlations of Al<sub>2</sub>O<sub>3</sub>, CaO, Fe<sub>2</sub>O<sub>3</sub>, MgO, TiO<sub>2</sub>, P<sub>2</sub>O<sub>5</sub>, Mg#, La, and a positive correlation of Pb
(Figure 8).

# 15 5.3 U-Pb geochronology

Supplementary Table 3 and Figures 9 and 10 display cathodoluminescence (CL) images and U– Pb isotope data of selected zircon grains from MMEs, respectively. Zircon grains are typically colorless, euhedral to subhedral prismatic in form, and exhibit regular concentric zonation in CL images (Figure 9). Grain lengths range from 100 to 200  $\mu$ m, with length:width ratios between 1:1 and 2:1. U and Th contents of these zircons range from 383–1575 ppm and 12–2948 ppm, respectively, and their Th/U ratios vary between 0.02 and 2.30 with a mean value of 0.5. U-Pb isotope data for thirty MME grains give weighted mean ages of 205.0 ± 1.3 Ma (MSWD = 0.14)

(Figure 10). This date, considered to be the age of MMEs, falls in the age interval of the granitoid (217 to 200 Ma; Karimpour et al. 2010b, 2011; Mirnejad et al. 2013; Abbasi 2018).

### 5.4 Bulk-rock Sr–Nd isotopes

Supplementary Table 4 lists Sr and Nd isotope data for the Mashhad granodiorite and MMEs. Initial <sup>87</sup>Sr/<sup>86</sup>Sr and <sup>143</sup>Nd/<sup>144</sup>Nd are calculated based on the average age of the studied samples. In granodiorite, <sup>87</sup>Sr/<sup>86</sup>Sr and <sup>143</sup>Nd/<sup>144</sup>Nd occupy extremely narrow ranges from 0.7057 to 0.7059 and 0.51222 to 0.51225, respectively. Correspondingly,  $\epsilon Nd_{(t)}$  values are tightly grouped between -3 and -2.3. The MMEs display similar restricted initial <sup>87</sup>Sr/<sup>86</sup>Sr (0.7057 to 0.7061), but slightly higher initial <sup>143</sup>Nd/<sup>144</sup>Nd (0.51226 to 0.51230), and  $\varepsilon$ Nd<sub>(t)</sub> (-2.3 to -1.4) relative to host granodiorite. The Mashhad granodiorite and MMEs lie on the mantle array and in the enriched quadrant on <sup>87</sup>Sr/<sup>86</sup>Sr vs.  $\epsilon$ Nd<sub>(t)</sub> diagram (Figure 11). These resemble values for Triassic granitoids and associated MMEs of the western Kunlun orogeny (Jiang et al. 2013), East Kunlun Orogen (Xia et al. 2015) and Tagong pluton in Songpan–Ganze terrane (Chen et al. 2017) from the eastern part of the Paleo-Tethys suture zone in China, and are different from the primitive mafic and ultramafic rocks from the Mashhad ophiolite (Shafaii Moghadam et al. 2015), whose isotopic composition is similar to Pacific N-MORB (Figure 11).

17 6. Discussion

# 18 6.1 Implications for magma mixing

MMEs are suitable tracers to understand the petrogenesis and origin of magmas (e.g. Barbarin 2005; Yang et al. 2015). Several hypotheses have been suggested for the origin of MMEs, the most common ones being that they represent restites from the granite source (Chappell et al. 1987), cognate fragments or early-formed crystals from the host magma (e.g. Donaire et al. Page 13 of 61

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2005: Shellnutt et al. 2010), xenoliths (Maas et al. 1997) or mafic magma that was injected and mixed with host felsic magma (Griffin et al. 2002; Barbarin 2005). In the restite model, it has been suggested that enclaves are refractory solid residues of partial melting from the source region (e.g. Chappell 1996; Barbarin and Didier 1992; Chappell et al. 2000). Therefore, they must be older than the host rocks (Liu et al. 2013). Our new and compiled zircon U-Pb ages show the Mashhad MMEs ( $205 \pm 1.3$  Ma) are identical to the age of host granitoids (217 to 200Ma; Karimpour et al. 2010b, 2011; Mirnejad et al. 2013; Abbasi 2018). They also have fine-grained textures relative to the host granitoids without recrystallization, deformation, and metamorphic or residual sedimentary fabrics, which preclude "residues" model. The MMEs from the granodiorites have distinct contacts with their host rocks (Figure 3), and include chilled margins against granodiorites, which is in agreement with magma mixing model. In addition, due to the absence of angular enclaves (Figure 3), cognate model is thus dismissed. Moreover, Vernon (1983) proposed that early-formed minerals by fractional crystallization are commonly coarse grained with euhedral shapes and lack of cumulate textures in the MMEs. Therefore, we propose that the Mashhad MMEs are not mafic mineral cumulates and therefore, a cognate model is ruled out. 

Field and petrological observations include: (i) banded structures (Figure 3(d)), and the presence of irregular, fine-grained mafic material in host granodiorite rocks (Blundy and Sparks 1992; Barbarin 2005); (ii) spheroidal to ellipsoidal/ovoidal shapes of enclaves (Figure 3 (g, h)) that resemble pillows of undisputed injection origin in other plutons ((Baxter and Feely 2002; Liu et al. 2013); (iii) enclaves partially enclose megacrysts of the host rocks (Figure 3(g)), suggesting the intermingling of two magmas (Liu et al. 2013) (iv) Feldspar-rich zones at contacts (Liu et al. 2013) (Figure 3(h)); (v) acicular apatite and the presence of resorption surfaces in plagioclase

(Figure 4(c)) and small plagioclase inclusions in a larger plagioclase (Figure 4(d)); (vi) complex
oscillatory zoning (Figure 4(e)), with repeated resorption surfaces, marking changes in
plagioclase stability brought about by magma mixing (Müller et al. 2005) and (vii) inclusions of
biotite and hornblende in plagioclase (Figure 4(f)) (Baxter and Feely. 2002) are evidence for
chemical or thermal changes in the melt during crystal growth due to magma mixing.

Composition of amphibole, biotite and plagioclase can be used to distinguish magma sources. The reverse zoning of plagioclases from the MMEs is a disequilibrium texture that indicates magma mixing (Fig. 5a). In the TiO<sub>2</sub>-FeO<sub>total</sub>-MgO diagram, biotites show geochemical similarities to the re-equilibrated to primary biotites (Figure 5(b)) as well as calc-alkaline rocks (Figure5(c)). Moreover, in the TFeO/(TFeO+MgO) versus MgO diagram (after Zhou 1986), granodiorites and MMEs seems to be crystallized from a melt with crust to mantle geochemical signatures (Figure5(d)). This idea is also confirmed with the biotite chemistry, in a Ca-Fe<sub>tot</sub>-Mg discrimination diagram (after Xie and Zhang (1990). Moreover, the composition of biotites in the enclaves is similar to those in the host which confirms enclaves generated by magma mixing during extensive equilibration processes. We believe these observations are not compatible with a cognate model which proposed by Didier and Barbarin (1991) for the formation of MMEs. Almost all the amphibole and biotite data plot in the crust-mantle mixed category (Figure 5 (d-f)). These features imply that the granodioritic magma and its MMEs were derived from mantle and mixed with crustal components during the mixing/mingling process. Moreover, the geochemical characteristics of the Mashhad granitoids show a linear relationship between major and trace elements and SiO<sub>2</sub> content for both the MMEs and their host granodiorite relative to mafic end-member reported by Shafaii Moghadam et al. 2015 (Figure 8), which can be interpreted as the product of magma mixing (Castro et al. 1990; Chappell 1996; Chen et al. 

2017). It should be mentioned that these features (Figure 8) may also be associated with crystallization fractionation (He et al. 2018). However, the role of a fractional crystallization trend of basaltic magma can be excluded for the formation of MME magmas, as they plot near the magma mixing line in the  $(Fe_2O_3)_T$ -MgO and  $\varepsilon Nd_{(1)}$ -SiO<sub>2</sub> diagrams (e.g. basaltic and felsic; Castro et al. 1990; Chappell 1996) (Figure 12(a,b)). Shafaii Moghadam et al. (2015), on the basis of Hf isotope data and whole-rock Sr-Nd-Pb isotopic results for the basaltic rocks of Mashhad, suggest the MME magmas were generated from depleted mantle-derived magmas, which conferred primitive signatures of initial  ${}^{87}$ Sr/ ${}^{86}$ Sr = 0.703-0.704 and  $\varepsilon$ Nd<sub>(1)</sub> = +3.7-5.4. 

9 Element-ratio/element-ratio and element/element-ratio plots can support a mixing process if the
10 data plot on a hyperbolic curve in a geochemical discrimination diagram (Langmuir et al. 1978;
11 Blundy and Sparks 1992; Zhou 1994) as noted, for example, in certain suites in the Muchen
12 pluton, China (Liu et al. 2013) (Figure 12(c,d)).

The studied MMEs show similarity in the whole-rock <sup>87</sup>Sr/<sup>86</sup>Sr (0.70575-0.70618) isotopic compositions with respect to host granodiorite (0.70578-0.70593), which can be interpreted as chemical and radiogenic isotopic equilibrium during magma mixing (Lesher 1990; Poli et al. 1996) and/or origination from the same source without isotopic fractionation during magma mixing. In addition, MMEs and granodiorite have similar whole-rock <sup>143</sup>Nd/<sup>144</sup>Nd. However, the MMEs have relatively higher <sup>143</sup>Nd/<sup>144</sup>Nd (0.51226-0.512230) than those of the host granodiorite  $(^{143}Nd/^{144}Nd = 0.51222-0.51225)$ , with  $\epsilon Nd(t)$  differences of up to 1–2 units. Lesher (1990) and Poli et al. (1996) suggested that isotopes of an element are more readily equilibrated and homogenized during a mixing process than are bulk major elements. Furthermore, Holden et al. (1991) proposed that Sr isotopes are equilibrated and homogenized more quickly than Nd isotopes, which is in agreement with Sr and Nd isotopic results for the Mashhad MMEs and 

associated host granodiorite. In fact, when a more mafic magma is injected into a felsic magma chamber and then breaks up into enclaves, both magma mingling and chemical diffusion will change the initial composition of the enclave melt. Major elements such as Al and Si in the silicate melt are network-forming components (mainly in tetrahedral form), and thus it is difficult for them to diffuse and homogenize during magma mixing. However, Ca, Fe and Mg as non-network components have relatively different behavior. Moreover, trace elements and isotopic elements can diffuse more rapidly toward homogeneous compositions after mixing. Consequently, we suggest that the similar trace element and isotopic compositions of the MMEs and host granitoids and differences in major element composition may reflect different diffusion rates and partial re-equilibration of the elements during magma mixing, due to the mixing/mingling process (e.g. Lesher 1990; Xiong et al. 2012; Zhao et al. 2012).

# 12 6.2 Origin of granodiorite

The granodiorites from Mashhad that were emplaced in the Late Triassic (217-200 Ma; Karimpour et al. 2010, 2011; Mirnejad et al. 2013; Abbasi 2018) are characterized by relatively high SiO<sub>2</sub> contents compared to the MMEs, and have high-K calc-alkaline and metaluminous to slightly peraluminous geochemical signatures (Figure 6(d)). The P<sub>2</sub>O<sub>5</sub> contents of the Mashhad granodiorite vary from 0.2% to 0.5%, and decrease with increasing  $SiO_2$  (Figure 8(f)), similar to I-type granitoids. Moreover, Rb/Sr ratios of up to 0.8 in Mashhad granodiorites are similar to those in I-type granites (<0.9; e.g. Wang et al. 1993). The I-type granitic signatures of Mashhad granodiorites are supported by the presence of hornblende and lack of aluminous primary minerals, such as muscovite, cordierite, tourmaline, and alusite, and garnet, with variable Al<sub>2</sub>O<sub>3</sub> (15.16–15.97 wt%), and low K<sub>2</sub>O (2.38–4.18 wt%) and Na<sub>2</sub>O (3.49–6.18 wt%) (Chappell and White 2001). In addition, their Ba/La (9.99–19.36) is higher, and Nb/La (0.45-0.64) is lower than 

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those in magmas originating from MORB, OIB, and intra-plate environments (Sun and McDonough 1989).

The geochemical data presented here show that Mashhad granodiorites are I-type, subductionrelated calc-alkaline intrusions with isotopic signatures (relatively low <sup>87</sup>Sr/<sup>86</sup>Sr ratios of 0.7057– 0.7059), indicative of having been derived from the mantle. They are also depleted in Nb–Ti, and somewhat enriched in LILEs (e.g. Rb, Ba and Sr) and LREEs, and have high Ba/Zr ratios (0.30-6.43), similar to arc-like rocks formed in the subduction zone (with calc-alkaline affinity; cf. Pearce and Peate 1995). However, syn- to post-collisional rocks are also characterized by HFSE depletions and LILE enrichments.

The subduction process creates distinctive patterns of LILE/LREE, LILE/HFSE and REE/HFSE (Schiano et al. 1995) because during dehydration of a slab, HFSEs are not fractionated whereas LILEs and LREEs are fractionated (Münker et al. 2004). Taken together, the LILE and LREE patterns are compatible with a subduction-modified lithospheric mantle source (Pearce and Peate 1995). Depletion of HFSE (Nb-Ti) relative to LILEs may be ascribed principally to two processes: (1) the addition of an LILE-enriched, Nb–Ti-poor fluid component derived from the subducted slab to the mantle wedge, which in turn served as a more immediate source of magma or (2) retention of Nb–Ti in amphibole relative to Nb and Ti in other phases in the mantle source. These features argue that a subduction related component is present in the mantle source region (i.e., Pearce and Peate 1995). 

A plot of Th/Yb vs. Ta/Yb (Pearce et al. 1990) (Figure 13(a)) confirms enriched features of mantle source in the Mashhad granodiorite. In Figure 13(a), granodiorite data displays a trend sub-parallel to the mantle array, but shifted towards higher Th/Yb ratios, suggesting that the

1	melt(s) were formed from a source that had been previously metasomatized by fluids derived
2	from a subduction zone. The flat normalized HREE patterns in granodiorite suggest that garnet
3	was a residual phase in the mantle source region. Since Nd is more incompatible than Sm, the
4	granodiorite thus inherited low Sm/Nd ratios (0.13-0.16) consistent with small degrees of partial
5	melting in the presence of garnet. A low degree of partial melting is supported by a negative
6	correlation between La/Yb vs. Sm/Nd (Figure 13(b)) and positive correlation between Th/Yb vs.
7	Ta/Yb (Figure 13(a)). Rb/Sr ratios are regarded as another important tracer of mantle-crust
8	interaction. Mantle rocks have very low Rb/Sr ratios (0.01-0.1), compared to lower continental
9	crust (Rb/Sr = 0.12) and middle continental crust (Rb/Sr = 0.22) (Hofmann 1988; Rudnick and
10	Fountain 1995). The Rb/Sr ratios of the Mashhad granodiorite varies widely between 0.13 and
11	0.49, much higher than in mantle-derived rocks. Nb/La and Nb/Ce ratios in the mantle are
12	respectively about 1.01 and 0.39 (Sun and McDonough 1989) compared to 0.46 and 0.23 in the
13	crust (Weaver and Tarney 1984). Nb/La and Nb/Ce ratios in Mashhad granodiorite that range,
14	respectively, from 0.45 to 0.65 and 0.26 to 0.35, are similar to crustal ratios. Th/Ta ratios in
15	lower continental crust-derived magma (7.9; Shellnutt et al. 2009) are comparable to those in
16	Mashhad granodiorite (5.18 to 12.03, mean 7.95). A lower Mg# and higher SiO <sub>2</sub> wt.% content in
17	granodiorite, relative to the mantle, point additionally to the involvement of a crustal source in
18	the genesis of the Mashhad granodiorite.

### 19 6.3 Origin of MMEs

20 MMEs from the Mashhad granodiorites most likely represent remnants of a mafic magma 21 injected to a felsic magma. Consequently, they provide the best opportunity to study the source 22 of the mafic magma. The Mashhad MMEs are characterized by lower SiO<sub>2</sub>, higher Mg#, and 23 higher MgO, MnO, CaO relative to those in Mashhad granodiorite. These geochemical features along with low (<sup>87</sup>Sr/<sup>86</sup>Sr)i (0.7057–0.7061) and high εNd(t) values (-2.1 to -1.4) (Supplementary
 Table 4; Figure 11) show the MMEs formed from mantle-derived magmas.

The Mashhad MMEs are also characterized by high K to shoshonitic signatures (Figure 6(c)). Previous workers have established that potassic mafic magma end members might be generated from either small degree of partial melting of enriched lithospheric mantle at depths below the level of amphibole stability (Foley and Peccerillo 1992), and/or metasomatic processes in the subduction zone (Edgar et al. 1980). The similarity between the MMEs and Nb enriched arc basalts (NEBs) (Figure 13(c)) and their trends on the Ba/Th vs. Th/Nb plot (Figure 11(d)) imply that a mantle source metasomatized by slab fluids was involved in their petrogenesis. However, the relatively low initial  ${}^{87}$ Sr/ ${}^{86}$ Sr isotope ratios, high La/Nb (1.8 – 4.8) and low La/Ba (0.05 – 0.30) ratios, as well as enrichment of LILE (e.g. Rb, Ba, Th, and K) and LREEs and depletion of HFSEs in the studied MMEs advocates partial melting of an enriched lithospheric mantle metasomatized by slab fluid. 

A possible implication of lower crust endmembers can be inferred from trace-element ratios such as Rb/Sr, Nb/La, Nb/Ce, and Th/Ta. The Rb/Sr ratios of the MMEs vary widely between a 0.03 and 0.86, much higher than those in rocks derived from mantle, but similar to these ratios in host granodiorite. Nb/La, and Nb/Ce ratios in the MMEs (0.2–1.19 and 0.09–0.66, respectively) are much higher than those in mantle-derived rocks, suggesting possible participation of a lower crust source in addition to a mantle source. Most of Th/Ta ratios of the MMEs, ranging between 3.89–22.09 (mean, 8.93) show considerable contributions of lower crust materials in the source. Lower Mg# and higher SiO<sub>2</sub> are regarded as crustal geochemical signatures. These data, taken together, imply the involvement of both lower crust and an enriched lithospheric mantle that had been metasomatized by subducted slab fluid or/and decompression melting of enriched 

lithospheric mantle metasomatized by previous slab fluids following the post-orogenic collapse
extension (Barbarin 1999). However, in diagrams of Rb vs. Yb+Ta (Pearce et al. 1984) and R2
(6Ca + 2Mg + Al) vs. R1 (4Si - 11(Na + K) - 2(Fe + Ti)) (Batchelor and Bodwen 1985), the data
plot in the uppermost corner of the volcanic arc granite and pre-collision to syn-collision fields
(Figure 6(e, f)).

# 6 6.4 Geodynamic implications

The Pamir-Mashhad magmatic arc consists of a Mesozoic east-west-trending magmatic belt (Figure 1). This belt developed in response to the subduction of the Paleo-Tethyan plate beneath the Eurasian plate. Triassic igneous rocks in the western and central Pamir-Mashhad magmatic arc, including Mashhad granitoids (Karimpour et al. 2010, Mirnejad et al. 2013), Torbat Jam (Ghavi et al. 2018), the Feroz Koh unit, Paropamisus (Debon et al. 1987); Hindu Kush, Western Badakhshan (Debon et al. 1987) and central Badakhshan (Desio 1964) are related to the northward subduction of the Paleo-Tethys ocean under the southern Eurasian plate margin. Ages of the MMEs fall in the age interval of the Late Triassic Mashhad granitoids (217-200 Ma; Karimpour et al. 2010, 2011; Mirnejad et al. 2013; Abbasi 2018), Kashafroud granodiorite  $(210.9 \pm 2.5 \text{ Ma}; \text{Zanchetta et al. 2013})$ , Torbat Jam granites  $(217 \pm 1.7 \text{ Ma}; \text{Zanchetta et al.})$ 2013). These ages from the granitoids suggest that plutonism has occurred from  $\sim 217$  to 200 Ma in the Mashhad. LA-ICP-MS zircon U-Pb dating indicates that the Mashhad MMEs crystallised at 205±1.3 Ma. These ages clearly show that the MMEs and the host granodiorite were formed during the Late Triassic. 

The Sr–Nd isotopic signatures and trace element characteristics and mineral chemistry of the Mashhad granodiorite and its MMEs show interaction of the mantle–derived melts with

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continental crust during ascent, storage and evolution of the Mashhad magmas. Isotopically, the mantle-derived melts would be similar to the melts of the Mashhad ophiolite in the south of Mashhad granodiorite (Figure 2). Previous studies show that gabbro of Mashhad ophiolites have <sup>40</sup>Ar/<sup>39</sup>Ar plateau ages of 281–277 Ma (Late Pennsylvanian – Early Permian; Ghazi et al. 2001), similar to other ages for Paleo-Tethys oceanic crust (Ghazi et al. 2001). Shafaii Moghadam et al. (2015), on the basis of Hf isotope data and whole-rock Sr–Nd–Pb isotopic results, suggested that magma of the gabbroic and basaltic rocks were generated from depleted mantle-derived magmas, which conferred primitive signatures of initial  ${}^{87}$ Sr/ ${}^{86}$ Sr = 0.703-0.704 and  $\epsilon$ Nd<sub>(1)</sub> = +3.7+5.4, that was either affected by subducted sediments or was contaminated by continental crust.

Mashhad granodiorite and associated MMEs are located at the transitional zone (Figure 1(b)). Ghavi et al. (2018) suggested that the Kope-Dagh and transitional zone (Figure 1(b)) belong to Gondwana-derived blocks, which are characterized by a Neoproterozoic crust, and perhaps the Turan plate consists mostly of early Neoproterozoic crust. Therefore, the continental crust involved might be Neoproterozoic lower crust with initial <sup>87</sup>Sr/<sup>86</sup>Sr = 0.714 (Shafaii Moghadam et al. 2015).

When closure of Paleo-Tethys began in early Carnian time (c.a. 225 Ma), the northern margin of the Iran plate changed into a peripheral foreland basin (Zanchetta et al. 2013). In the Rb vs. Y + Ta and R2 vs. R1 diagrams (Figure 6(e,f)), data from the granodiorite plot between the uppermost corner of the volcanic arc granitoids, pre-collision setting, to syn-collision granitoids fields, suggesting that they may be generated in a pre-collisional (subduction-related) to collisional arc setting. These late Triassic ages, ranging from 217 to 200 Ma (Karimpour et al. 2010, 2011; Zanchetta et al. 2013; Mirnejad et al. 2013; Abbasi 2018), and the MME enclave age (205 Ma), pertain to arc-related granodioritic rocks (Mashhad and to surrounding areas including 

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Torbat Jam, the and Kashafroud Formation), all of them formed during continental collision
 between the Iran and Turan plates through a subduction to collisional setting.

At the end of subduction of the Paleo-Tethyan ocean (~225 Ma), the continent-continent collision between the Eurasian (Turan) and Iran plates (c.a. 225 Ma) took place, accompanied by crustal thickening (Zanchi et al. 2006; Wilmsen et al. 2009). The extension following crustal thickening led to decompression melting of enriched lithospheric mantle, which is metasomatized by previous slab fluids. The extension following the collision-collision continental crust can be confirmed by the occurrence of extensional faulting with NW-SE trending faults (Figure 2). The basaltic melts derived from this metasomatized mantle underplated the overlying mafic lower crust, where magma stayed in the magma chamber and provided heat for partial melting of Neoproterozoic lower crust. Underplating of basaltic magma under lower crust can be established by existence of basic rocks from Mashhad ophiolites, which have primitive isotopic signatures (initial <sup>87</sup>Sr/<sup>86</sup>Sr =0.703; Shafaii Moghadam et al. 2015). This basic magma underplated the lower crust, and produced partial melts of felsic composition, which formed the Mashhad calc-alkaline granodiorite with mixed mantle component. The mafic magma derived from the enriched lithospheric mantle mixed with the granodioritic magma to form MMEs (Figure 14). 

## 18 7. Conclusions

19 (1) U-Pb zircon dating indicates that the Mashhad MMEs crystallized at  $205.0 \pm 1.3$  Ma, 20 implying their formation during the Late Triassic.

(2) Field and petrographic observations, such as ellipsoidal and spherical shapes, K-feldspar
 megacrysts, the presence of acicular apatite, oscillatory zoned plagioclase, together with mineral

chemistry, geochemical and isotopic evidences indicate that MMEs and associated granodiorite were generated by mixing/mingling of felsic and mafic magmas.

3 (3) The relatively similar emplacement age and whole-rock εNd<sub>(t)</sub> values between the calc4 alkaline I-type granodiorite and enclosed MMEs indicate that the pluton was generated by the
5 subduction of the Paleo-Tethys oceanic lithosphere beneath the Turan plate during the Late
6 Triassic, but contained various degrees of contributions from an enriched lithospheric mantle and
7 lower crustal components.

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37 38 39	18	10. Figure captions
40 41	19	Figure 1. (a) the North Pamir–Mashhad arc passes south of Mashhad city and extends along the
42 43 44	20	Alborz Mountain to southern Azerbaijan (modified after Natal'in and Şengör 2005), and (b)
45 46 47	21	geological map of the Paleo-Tethys orogenic belt in northern Iran (after Zanchi et al. 2009).
48 49	22	Figure 2. Geological map of the Mashhad batholith (modified after Mirnejad et al. 2013).
50 51 52	23	Figure. 3. Relationships between MMEs and the Mashhad granodiorite host. (a) Cross cutting of
53 54 55	24	rounded to ellipsoid MMEs and host granodiorite by late stage aplitic phase; (b) sigmoidal and
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elongated enclaves within Mashhad granodiorite, reflecting deformations during flow of the host
magma; A view of (c) massive; (d) banded structures; (e) liner and (f) irregular boundaries
patches enriched in fine-grained mafic material in the granodiorite host rocks; (g) Enclosing of
feldspar megacrysts from the host by enclaves and (h) Enclosing of enclaves by feldspar
megacrysts from the host.

Figure 4. Thin section photographs (cross-polarised light) of representative MMEs from the
Mashhad granodiorite. (a) Granular texture in the granodiorite with associated minerals; (b)
Microgranular texture in the MMEs with associated minerals; (c) Acicular apatite in the
plagioclase with resorption texture in the MMEs; (d) small lath-shaped plagioclase inclusions
hosted by a large plagioclase in the MMEs; (e) oscillatory zoning in the plagioclase from MMEs;
(f) inclusions of quartz, biotite and hornblende in plagioclase. Abbreviations: Amp = amphibole;
Ap = Apatite; Bi = biotite; K-fls = K-feldspar; Plg = plagioclase; Qtz = quartz.

Figure 5. Compositional plots of plagioclase, biotites and hornblende. (a) The reverse zoning and compositional variation of a plagioclase crystal of the MME sample; The panel shows photomicrograph taken under cross-polarized light (left column), and An profile (right panel). The biotites from the granodiorite and associated MMEs are mainly plotted (b) in the re-equilibrated primary biotite field of the 10 \* TiO<sub>2</sub>-FeO<sub>total</sub>-MgO diagram (Nachit et al. 2005), (c) in the field of calc-alkaline magma of the MgO-FeO\*-Al<sub>2</sub>O<sub>3</sub> diagram (Abdel-Rahman 1994); (d) in the field of mantle-crust magma (after Zhou, 1986); TFeO indicates total iron as FeO; (e) the amphiboles from granodiorite and associated MMEs are mainly ferrohornblende to magnesiohornblende in composition (Leake et al., 1997); and (f) in the field of mantle-crust magma (Xie and Zhang 1990).

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Figure 6. Position of the Mashhad granodiorite and enclosed MMEs on (a) TAS (Na<sub>2</sub>O+K<sub>2</sub>O vs. SiO<sub>2</sub>) classification diagram (Middlemost 1994); (b) Quartz-Alkali feldspar–Plagioclase (QAP) normative classification diagram for Mashhad granodiorite and associated MMEs.; (c) K<sub>2</sub>O vs. SiO<sub>2</sub> diagram with lines separating tholeiitic, calc-alkaline, high-K calc-alkaline and shoshonitic series (Peccerillo and Taylor 1976), and (d) A/NK (Al<sub>2</sub>O<sub>3</sub>/Na<sub>2</sub>O+K<sub>2</sub>O) vs. A/CNK (Al<sub>2</sub>O<sub>3</sub>/CaO+Na<sub>2</sub>O+K<sub>2</sub>O) plots (Shand 1943); (e) Discrimination diagram of Rb vs. Yb+Ta for the Mashhad granodiorite and associated MMEs (after Pearce et al. 1984). WPG: within-plate granitoids; VAG: volcanic arc granitoids; ORG: ocean ridge granitoids; syn-COLG: syncollisional granitoids. Pinkish and orange fields show granitoid rocks from Mashhad and Dehnow, respectively (Karimpour et al. 2010; Mirnejad et al. 2013). (f) Discrimination diagram of R2 (6Ca + 2Mg + Al) vs. R1 (4Si - 11(Na + K) - 2(Fe + Ti)) for the Mashhad granodiorite (Batchelor and Bodwen 1985). 

Figure 7 (a) and (b) Chondrite-normalized REE patterns (Sun and McDonough 1989 for the Mashhad granodiorite and their MMEs, respectively; (c) and (d) Primitive-mantle normalized trace element patterns (Sun and McDonough 1989) for the Mashhad granodiorite and their MMEs, respectively.

Figure 8. Harker variation diagrams for major and some trace element compositions of the
Mashhad granodiorite and their MMEs. The variation of samples from the MMEs and granitoids
are continuous and display linear trends as might be expected if the enclaves formed by mixing.
Published data of mafic end-member in the study area are shown, with filled square, come from
Shafaii Moghadam et al. (2015).

22 Figure 9. Cathodoluminescence images for the MME in Mashhad granodiorite (sample GE-1)

1 Figure 10. Zircon U–Pb concordia diagram of the MMEs of the Mashhad granodiorite.

Figure 11. Initial <sup>87</sup>Sr/<sup>86</sup>Sr vs. ɛNd<sub>(t)</sub> for the Mashhad granodiorite and MMEs. Data of Triassic
granitoids and associated MMEs of the western Kunlun orogeny (Jiang et al. 2013), East Kunlun
Orogen (Xia et al. 2015) and Tagong pluton in Songpan–Ganze terrane (Chen et al. 2017), from
China, the eastern part of Paleo-Tethys suture zone.

Figure 12. (a) (Fe<sub>2</sub>O<sub>3</sub>)<sub>T</sub> vs. MgO and (b) εNd<sub>(t)</sub> vs. SiO<sub>2</sub> variations of host granitoids and MMEs
from the Mashhad (Castro et al. 1990; Chappell 1996). (c) Geochemical correlation diagrams of
Fe<sub>2</sub>O<sub>3</sub>/Al<sub>2</sub>O<sub>3</sub> vs. Na<sub>2</sub>O/CaO; (d) Rb (ppm) vs. Rb/Sr; data for gray and black symbols come from
granitoid and associated MMEs from Buchan, China (Liu et al. 2013) (Langmuir et al. 1978;
Blundy and Sparks 1992; Zhou 1994). Published data of mafic end-members in the study area
come from Shafaii Moghadam et al. (2015) and are shown with filled squares.

Figure 13. (a) Th/Yb vs. Ta/Yb diagram (Pearce et al. 1990) for the Mashhad granodiorite and their MMEs; (b) The La/Yb vs. Sm/Nd diagram display negative correlation; (c) Nb/U vs. Nb diagram (Kepezhinskas et al. 1996). (d) Th/Nb vs. Ba/Th diagram with sediment melt and fluids trends from Ribeiro et al. (2013). N-MORB composition come from Sun and McDonough (1989). Subduction enrichment and mantle metasomatism are from Pearce et al. (1990). Nb enriched arc basalts (NEBs).

Figure 14. Schematic model explaining the development of magma mixing between mantle
derived magmas and lower mafic crust-derived magmas within the magma chamber above a
subduction zone, modified after Miller et al. (2007). Abbreviations: SCLM = sub-continental
lithospheric mantle; PMMA= Pamir-Mashhad Magmatic Arc.

22 11. Table captions

1 2		
- 3 4	1	Table 1. Representative EPMA analyses of plagioclases, biotites and amphiboles of the Mashhad
5 6 7	2	granodiorite and its MMEs
8 9	3	Table 2. Major oxides (wt%) and trace element (ppm) contents of Mashhad granodiorite and its
10 11 12	4	MMEs
13 14 15	5	Table 3. LA-ICP-MS U-Pb data on zircons and ages for The MMEs in Mashhad granodiorite
16 17 18	6	Table 4. Results of whole rock Sr-Nd isotope ratios of the Mashhad granodiorite and its MMEs
19 20 21	7	
22 23 24		
25 26		
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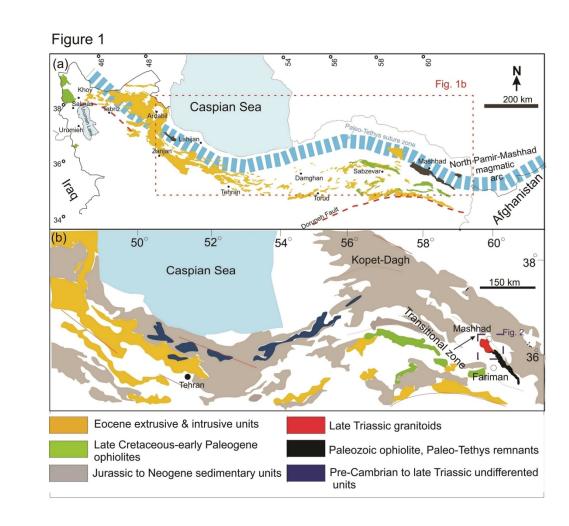
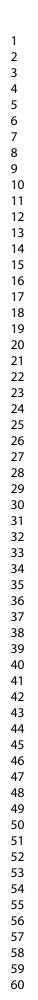


Figure 1. (a) the North Pamir–Mashhad arc passes south of Mashhad city and extends along the Alborz Mountain to southern Azerbaijan (modified after Natal'in and Şengör 2005), and (b) geological map of the Paleo-Tethys orogenic belt in northern Iran (after Zanchi et al. 2009).

151x144mm (300 x 300 DPI)



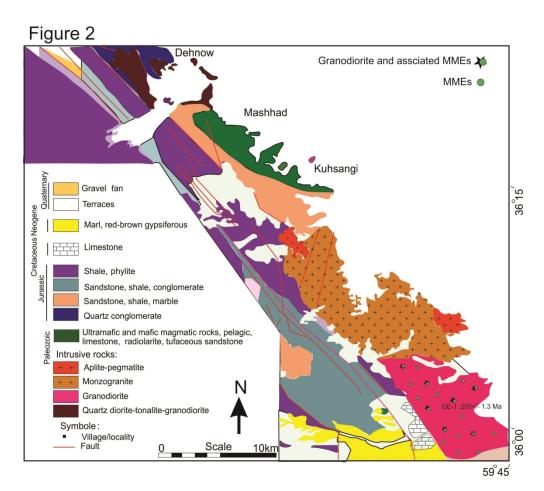


Figure 2. Geological map of the Mashhad batholith (modified after Mirnejad et al. 2013).

233x209mm (300 x 300 DPI)



Figure. 3. Relationships between MMEs and the Mashhad granodiorite host. (a) Cross cutting of rounded to ellipsoid MMEs and host granodiorite by late stage aplitic phase; (b) sigmoidal and elongated enclaves within Mashhad granodiorite, reflecting deformations during flow of the host magma; A view of (c) massive; (d) banded structures; (e) liner and (f) irregular boundaries patches enriched in fine-grained mafic material in the granodiorite host rocks; (g) Enclosing of feldspar megacrysts from the host by enclaves and (h) Enclosing of enclaves by feldspar megacrysts from the host.

67x57mm (300 x 300 DPI)

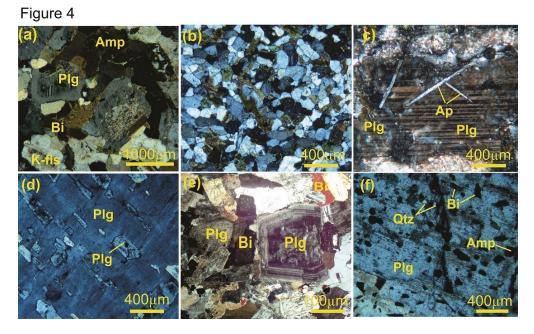
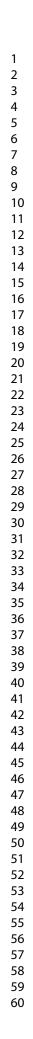


Figure 4. Thin section photographs (cross-polarised light) of representative MMEs from the Mashhad granodiorite. (a) Granular texture in the granodiorite with associated minerals; (b) Microgranular texture in the MMEs with associated minerals; (c) Acicular apatite in the plagioclase with resorption texture in the MMEs; (d) small lath-shaped plagioclase inclusions hosted by a large plagioclase in the MMEs; (e) oscillatory zoning in the plagioclase from MMEs; (f) inclusions of quartz, biotite and hornblende in plagioclase. Abbreviations: Amp = amphibole; Ap = Apatite; Bi = biotite; K-fls = K-feldspar; Plg = plagioclase; Qtz = quartz.

93x58mm (300 x 300 DPI)





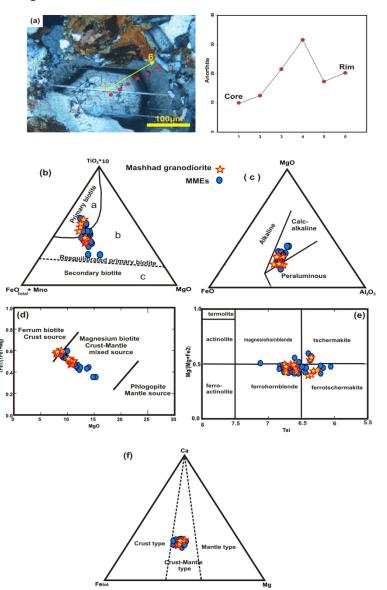


Figure 5. Compositional plots of plagioclase, biotites and hornblende. (a) The reverse zoning and compositional variation of a plagioclase crystal of the MME sample; The panel shows photomicrograph taken under cross-polarized light (left column), and An profile (right panel). The biotites from the granodiorite and associated MMEs are mainly plotted (b) in the re-equilibrated primary biotite field of the 10 \* TiO2-FeOtotal-MgO diagram (Nachit et al. 2005), (c) in the field of calc-alkaline magma of the MgO-FeO\*-Al2O3 diagram (Abdel-Rahman 1994); (d) in the field of mantle-crust magma (after Zhou, 1986); TFeO indicates total iron as FeO; (e) the amphiboles from granodiorite and associated MMEs are mainly ferrohornblende to magnesiohornblende in composition (Leake et al., 1997); and (f) in the field of mantle-crust magma (Xie and Zhang 1990).

164x260mm (300 x 300 DPI)

60

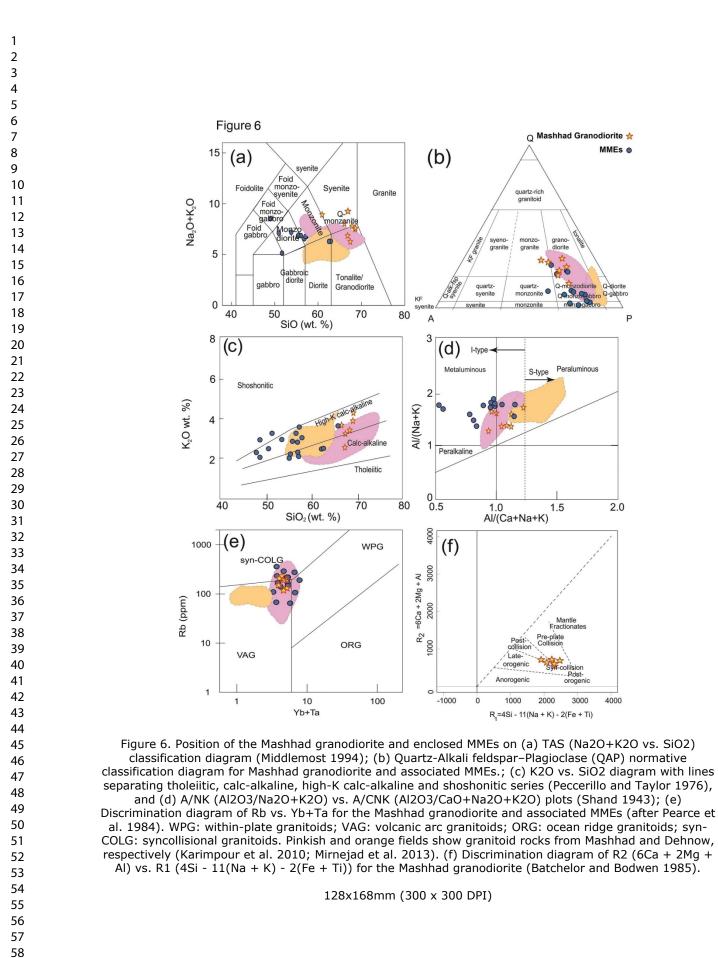
MMEs

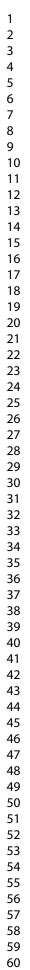
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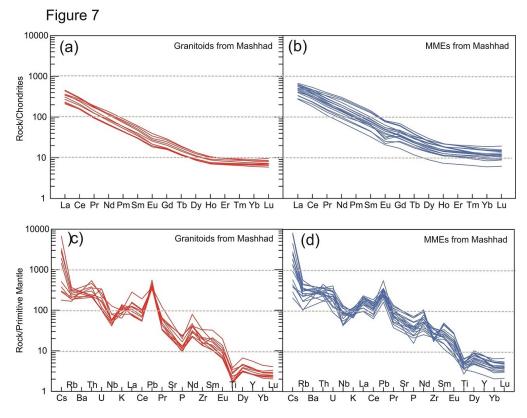
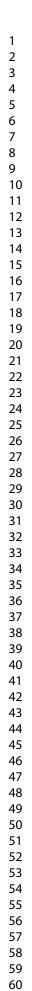


Figure 7 (a) and (b) Chondrite-normalized REE patterns (Sun and McDonough 1989 for the Mashhad granodiorite and their MMEs, respectively; (c) and (d) Primitive-mantle normalized trace element patterns (Sun and McDonough 1989) for the Mashhad granodiorite and their MMEs, respectively.

106x83mm (300 x 300 DPI)



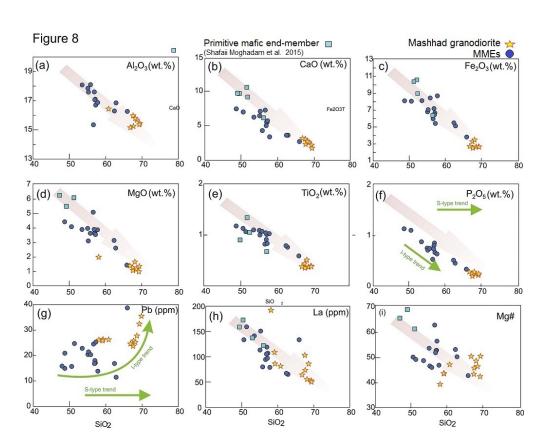


Figure 8. Harker variation diagrams for major and some trace element compositions of the Mashhad granodiorite and their MMEs. The variation of samples from the MMEs and granitoids are continuous and display linear trends as might be expected if the enclaves formed by mixing. Published data of mafic endmember in the study area are shown, with filled square, come from Shafaii Moghadam et al. (2015).

86x68mm (300 x 300 DPI)

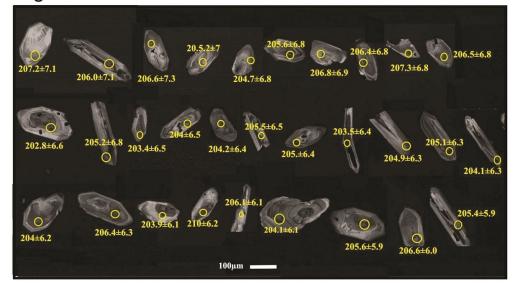
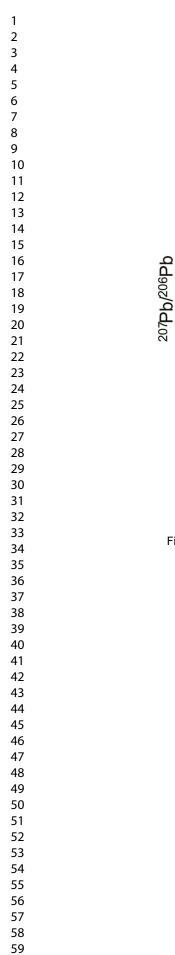
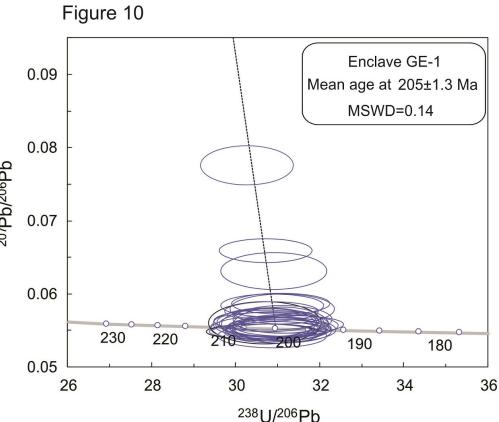
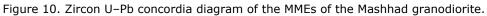


Figure 9. Cathodoluminescence images for the MME in Mashhad granodiorite (sample GE-1)

101x60mm (300 x 300 DPI)







106x88mm (300 x 300 DPI)

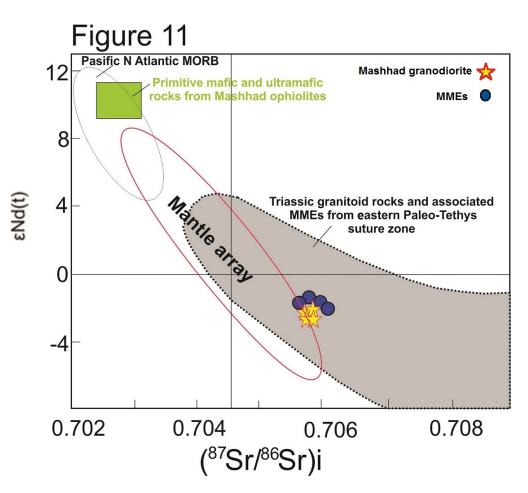


Figure 11. Initial 87Sr/86Sr vs. εNd(t) for the Mashhad granodiorite and MMEs. Data of Triassic granitoids and associated MMEs of the western Kunlun orogeny (Jiang et al. 2013), East Kunlun Orogen (Xia et al. 2015) and Tagong pluton in Songpan–Ganze terrane (Chen et al. 2017), from China, the eastern part of Paleo-Tethys suture zone.

94x85mm (300 x 300 DPI)

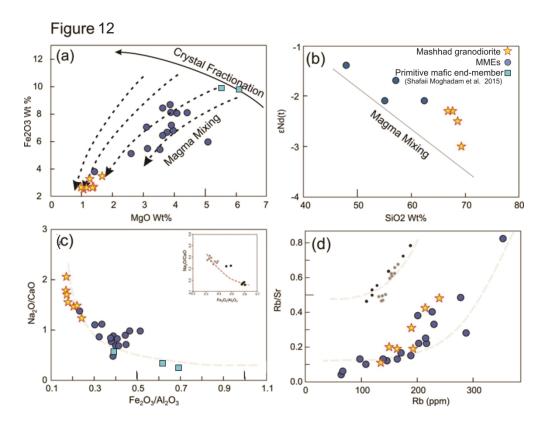


Figure 12. (a) (Fe2O3)T vs. MgO and (b) εNd(t) vs. SiO2 variations of host granitoids and MMEs from the Mashhad (Castro et al. 1990; Chappell 1996). (c) Geochemical correlation diagrams of Fe2O3/Al2O3 vs. Na2O/CaO; (d) Rb (ppm) vs. Rb/Sr; data for gray and black symbols come from granitoid and associated MMEs from Buchan, China (Liu et al. 2013) (Langmuir et al. 1978; Blundy and Sparks 1992; Zhou 1994).
Published data of mafic end-members in the study area come from Shafaii Moghadam et al. (2015) and are shown with filled squares.

283x217mm (300 x 300 DPI)

<sup>80</sup> (b)

70

60

50

40

30

20 └─ 0.13

3

2

Th/Nb

(d)t

related enrichr

melt

Sediment

0.14

0.15

Fluid related enrichmen

50

Sm/Nd

La/Yb

ı ı ı ıııı

100

\*

Mashhad granodiorite

0.16

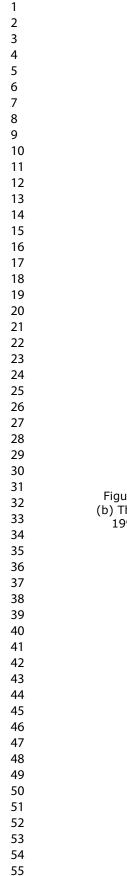
100

Ba/Th

0.17

150

MMEs O



60

Figure 13

N-MORB

.1

MORB

Island Arc Basalt

1

1.1.1111

1

Ta/Yb

ມມັນແມ

OIB.

10

Nb

10

NEB:

T T T T HILL

(a)

10

1

.1

.01

100

N/dN

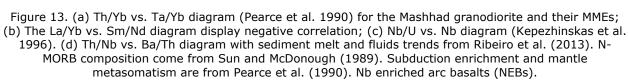
10

0.1

.01

(c)

Th/Yb



100

0

222x172mm (300 x 300 DPI)

Ν

SCLM

SCLM

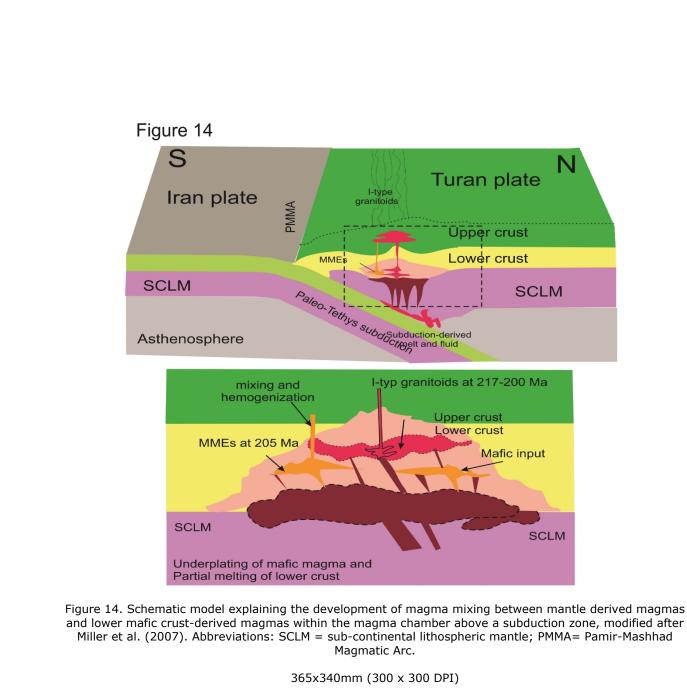


Table 1 Representative EPMA analyses of biotites of the Mashhad granodiorite and its MMEs in Wt.% oxide.

						Gran	odiorite						MN	ЛEs
Sample	B1.5	B1.6	B1.10	B1.11	B3.70	B3.71	G11	G11.1	RE10.6	RE10.7	RE10.8	RE10.9	BE1.30	BE
SiO <sub>2</sub>	37.05	36.77	37.30	36.71	36.26	36.62	38.04	38.23	36.91	36.18	36.75	36.23	37.01	36
TiO <sub>2</sub>	2.02	2.03	1.86	1.95	2.72	3.08	1.68	1.74	3.52	3.36	3.65	3.42	2.41	2
Al <sub>2</sub> O <sub>3</sub>	14.57	14.78	14.65	15.10	15.41	14.69	14.64	14.65	16.69	16.39	16.80	17.49	15.19	15
FeO	19.65	19.19	19.21	18.87	21.35	21.42	17.81	17.55	19.94	20.41	20.53	19.44	17.95	18
MnO	0.61	0.59	0.71	0.76	0.56	0.61	0.61	0.70	0.56	0.58	0.60	0.53	0.39	0
MgO	10.51	10.19	10.97	10.97	8.67	8.72	10.89	10.89	8.38	8.42	8.21	7.87	11.96	11
CaO	0.01	0.02	0.02	0.03	0.01	0.01	0.05	0.05	0.02	0.04	0.06	0.03	0.02	0
Na <sub>2</sub> O	0.02	0.07	0.10	0.11	0.05	0.09	0.02	0.07	0.05	0.05	0.07	0.07	0.05	0
K <sub>2</sub> O	9.63	9.58	9.62	9.43	9.59	9.45	9.87	9.76	10.00	9.63	9.75	9.58	9.72	9
Total	94.13	93.28	94.51	94.00	94.69	94.76	93.67	93.70	96.13	95.12	96.47	94.73	94.75	94
					94.09	94.70	95.07	95.70	90.15	95.12	90.47	94.75	94.75	24
Number of cat <sup>IV</sup> Si					5 ( )	5 (0)	5.07	5.00	5 (1		5 50	5.50	5.65	~
	5.75	5.75	5.75	5.69	5.64	5.69	5.87	5.89	5.61	5.57	5.58	5.56	5.65	5.
<sup>IV</sup> Al	2.25	2.25	2.25	2.31	2.36	2.31	2.13	2.11	2.39	2.43	2.42	2.44	2.35	2.
<sup>VI</sup> Al	0.42	0.47	0.42	0.44	0.47	0.39	0.53	0.54	0.60	0.55	0.58	0.73	0.39	0.
Ti	0.24	0.24	0.22	0.23	0.32	0.36	0.19	0.20	0.40	0.39	0.42	0.39	0.28	0.
Fe	2.53	2.49	2.46	2.42	2.76	2.76	2.28	2.24	2.51	2.61	2.58	2.48	2.27	2.
Mn	0.08	0.08	0.09	0.10	0.07	0.08	0.08	0.09	0.07	0.08	0.08	0.07	0.05	0.
Mg	2.43	2.37	2.52	2.53	2.01	2.02	2.51	2.50	1.90	1.93	1.86	1.80	2.73	2.
Ca	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.01	0.00	0.01	0.01	0.01	0.00	0.
		0.00	0.03	0.03	0.02	0.03	0.01	0.02	0.01	0.01	0.02	0.02	0.01	0.
Na	0.01	0.02	0.05	0.00										
Na K	0.01 1.91	0.02 1.91	1.89	1.86	1.90	1.88	1.94	1.92	1.94	1.89	1.89	1.88	1.89	1.
K Mg/(Mg+Fe)	1.91 0.49	1.91 0.49	1.89 0.50	1.86 0.51	1.90 0.42	0.42	0.52 tites of th	0.53 he Mash	0.43	0.42	0.42	0.42	1.89 0.54 V <b>t.% oxid</b>	1. 0. e.
K Mg/(Mg+Fe)	1.91 0.49	1.91 0.49	1.89 0.50	1.86 0.51	1.90 0.42	0.42	0.52 tites of th	0.53	0.43	0.42	0.42	0.42	0.54	0. e.
K Mg/(Mg+Fe) Table 1	1.91 0.49 (continu	1.91 0.49 1ed) Rep	1.89 0.50 resentat	1.86 0.51 ive EPM	1.90 0.42 A analys	0.42	0.52 tites of the	0.53 he Mash 1MEs	0.43 had gran	0.42 odiorite a	0.42 and its M	0.42 MEs in W	0.54 W <b>t.% oxid</b>	0. e. RE
K Mg/(Mg+Fe) Table 1 Sample	1.91 0.49 (continu GE10	1.91 0.49 1ed) Rep GE11	1.89 0.50 resentat KE4	1.86 0.51 ive EPM KE5	1.90 0.42 A analys KE6	0.42 ses of bio KE17	0.52 tites of th M KE18	0.53 he Mash 1MEs KE25	0.43 had gran KE26	0.42 odiorite a	0.42 and its M KE23	0.42 MEs in W RE11	0.54 Vt.% oxid RE13	0. e. RE 35
K Mg/(Mg+Fe) Table 1 Sample SiO <sub>2</sub>	1.91 0.49 (continu GE10 37.85	1.91 0.49 med) Rep GE11 38.18	1.89 0.50 resentat KE4 35.22	1.86 0.51 ive EPM KE5 36.01	1.90 0.42 A analys KE6 36.93	0.42 ses of bio KE17 38.12	0.52 tites of th M KE18 38.09	0.53 he Mash 1MEs KE25 36.00	0.43 had gran KE26 36.25	0.42 odiorite a KE22 36.99	0.42 and its M KE23 37.34	0.42 MEs in W RE11 35.67	0.54 Vt.% oxid RE13 36.04	0. e. RE 35
K Mg/(Mg+Fe) Table 1 Sample SiO2 TiO2	1.91 0.49 (continu GE10 37.85 1.59	1.91 0.49 1ed) Rep GE11 38.18 1.71	1.89 0.50 resentat KE4 35.22 1.16	1.86 0.51 ive EPM KE5 36.01 1.21	1.90 0.42 A analys KE6 36.93 1.52	0.42 ses of bio KE17 38.12 1.00	0.52 tites of th KE18 38.09 0.96	0.53 he Mash 1MEs KE25 36.00 3.27	0.43 had gram KE26 36.25 3.11	0.42 odiorite : KE22 36.99 2.57	0.42 and its M KE23 37.34 2.61	0.42 MEs in W RE11 35.67 3.78	0.54 Vt.% oxid RE13 36.04 2.86	0. e. RE 35 2
K Mg/(Mg+Fe) Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57	1.91 0.49 <b>GE11</b> 38.18 1.71 14.50 16.70 0.55	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56	1.86 0.51 ive EPM KE5 36.01 1.21 14.41 19.08 0.44	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51	0.42 ses of bio KE17 38.12 1.00 15.10 14.14 0.29	0.52 tites of the second secon	0.53 he Mash MEs KE25 36.00 3.27 15.54 19.53 0.44	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49	0.42 odiorite a KE22 36.99 2.57 15.36 18.00 0.44	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55	0. e. RE 35 2 15 20 0
K Mg/(Mg+Fe) Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57 11.97	1.91 0.49 <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56 13.82	1.86 0.51 ive EPM KE5 36.01 1.21 14.41 19.08 0.44 13.72	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44	0.42 ses of bio KE17 38.12 1.00 15.10 14.14 0.29 14.86	0.52 tites of the second secon	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00	0.42 odiorite a 36.99 2.57 15.36 18.00 0.44 11.38	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47	0. <b>RF</b> 35 2 15 20 0 9
K Mg/(Mg+Fe) Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57 11.97 0.27	1.91 0.49 <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56 13.82 0.04	1.86 0.51 ive EPM KE5 36.01 1.21 14.41 19.08 0.44 13.72 0.08	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07	0.42 ses of bio KE17 38.12 1.00 15.10 14.14 0.29 14.86 0.18	0.52 tites of the second secon	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05	0.42 odiorite a 36.99 2.57 15.36 18.00 0.44 11.38 0.06	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47 0.00	0. e. RE 35 20 0 9 0
K Mg/(Mg+Fe) Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07	1.91 0.49 <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03	1.86 0.51 ive EPM KE5 36.01 1.21 14.41 19.08 0.44 13.72	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44	0.42 ses of bio KE17 38.12 1.00 15.10 14.14 0.29 14.86 0.18 0.07	0.52 tites of the second secon	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00	0.42 odiorite a 36.99 2.57 15.36 18.00 0.44 11.38	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47	0. RE 35 2 15 20 0 9 0 0 0
K Mg/(Mg+Fe) Table 1 Sample SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57 11.97 0.27	1.91 0.49 <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56 13.82 0.04	1.86 0.51 ive EPM KE5 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06	0.42 ses of bio KE17 38.12 1.00 15.10 14.14 0.29 14.86 0.18	0.52 tites of the second seco	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08	0.43 had gran KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10	0.42 odiorite : KE22 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08	0. <b>REE</b> 355 20 0 9 0 0 9 0 0 9
K Mg/(Mg+Fe) Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51	1.91 0.49 <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75	1.86 0.51 <b>ive EPM</b> <b>KE5</b> 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27	0.42 <b>EXENT</b> <b>38.12</b> 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81	0.52 tites of the KE18 38.09 0.96 14.96 14.96 14.30 0.24 15.04 0.18 0.09 8.90	0.53 he Mash MEs KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96	0.42 odiorite a KE22 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72	0.54 <b>RE13</b> 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93	0. <b>RE</b> 35 2 15 20 0 9 0 0 9 0 0 9
K Mg/(Mg+Fe) Table 1 Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O Total Number of cat <sup>IV</sup> Si	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51	1.91 0.49 <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75	1.86 0.51 <b>ive EPM</b> <b>KE5</b> 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27	0.42 <b>EXENT</b> <b>38.12</b> 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81	0.52 tites of the KE18 38.09 0.96 14.96 14.96 14.30 0.24 15.04 0.18 0.09 8.90	0.53 he Mash MEs KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96	0.42 odiorite a KE22 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72	0.54 <b>RE13</b> 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93	0. <b>RF</b> 35 2 15 20 0 9 0 0 9 9 9 3 3 3 3 5 2 2 1 5 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 5 2 2 1 5 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 2 1 5 2 1 5 2 2 1 5 5 5 5 2 1 5 5 5 5 5 5 5 5 5 5 5 5 5
K Mg/(Mg+Fe) Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O Total Number of cat <sup>IV</sup> Si <sup>IV</sup> Al	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51 (cions on the second	1.91 0.49 <b>ied) Rep</b> <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12 <b>the basis</b>	1.89 0.50 resentat KE4 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75 of 11 ox	1.86 0.51 ive EPM KE5 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05 ygens	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27 92.97	0.42 <b>Set of bio</b> <b>KE17</b> 38.12 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81 92.62	0.52 tites of the second seco	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65 94.10	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96 94.35	0.42 odiorite a 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67 94.64	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79 95.56	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72 95.55	0.54 <b>RE13</b> 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93 94.61	0. RE 35 2 15 20 0 9 0 0 9 9 3 5.
K Mg/(Mg+Fe) Table 1 Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O Total Number of cat <sup>IV</sup> Si <sup>IV</sup> Al <sup>VI</sup> Al	1.91 0.49 (continu 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51 tions on t 5.96 2.04 0.47	1.91 0.49 <b>ied) Rep</b> <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12 <b>the basis</b> 5.87 2.13 0.50	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75 of 11 ox 5.54 2.46 0.26	1.86 0.51 <b>ive EPM</b> <b>KE5</b> 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05 <b>ygens</b> 5.63 2.37 0.28	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27 92.97 5.75 2.25 0.34	0.42 <b>EXEND</b> <b>Set of bio</b> <b>KE17</b> 38.12 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81 92.62 5.80 2.20 0.51	0.52 tites of tl N KE18 38.09 0.96 14.96 14.96 14.30 0.24 15.04 0.18 0.09 8.90 92.82 5.79 2.21 0.47	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65 94.10 5.60 2.40 0.44	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96 94.35 5.62 2.38 0.53	0.42 odiorite a KE22 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67 94.64 5.66 2.34 0.43	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79 95.56 5.65 2.35 0.43	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72 95.55 5.49 2.51 0.70	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93 94.61 5.61 2.39 0.38	0. <b>RF</b> 35 2 15 20 0 9 9 0 0 9 9 3 5. 2. 0.
K Mg/(Mg+Fe) Table 1 Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O Total Number of cat <sup>IV</sup> Si <sup>IV</sup> Al V <sup>I</sup> Al Ti	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51 cions on t 5.96 2.04 0.47 0.19	1.91 0.49 <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12 <b>the basis</b> 5.87 2.13 0.50 0.20	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75 of 11 ox 5.54 2.46 0.26 0.14	1.86 0.51 <b>KE5</b> 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05 <b>ygens</b> 5.63 2.37 0.28 0.14	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27 92.97 5.75 2.25 0.34 0.18	0.42 <b>EXEND</b> <b>Set of bio</b> <b>KE17</b> 38.12 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81 92.62 5.80 2.20 0.51 0.11	0.52 tites of the second seco	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65 94.10 5.60 2.40 0.44 0.38	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96 94.35 5.62 2.38 0.53 0.36	0.42 odiorite a 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67 94.64 5.66 2.34 0.43 0.30	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79 95.56 5.65 2.35 0.43 0.30	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72 95.55 5.49 2.51 0.70 0.44	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93 94.61 5.61 2.39 0.38 0.34	0. <b>RF</b> 35 20 0 9 0 0 9 9 35 5. 2. 0. 0.
K Mg/(Mg+Fe) Table 1 Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O Total Number of cat <sup>IV</sup> Si <sup>IV</sup> Al Ti Fe	1.91 0.49 (continu 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51 cions on t 5.96 2.04 0.47 0.19 2.14	1.91 0.49 <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12 <b>the basis</b> 5.87 2.13 0.50 0.20 2.13	1.89 0.50 resentat KE4 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75 of 11 ox 5.54 2.46 0.26 0.14 2.53	1.86 0.51 <b>ive EPM</b> <b>KE5</b> 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05 <b>ygens</b> 5.63 2.37 0.28 0.14 2.47	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27 92.97 5.75 2.25 0.34 0.18 2.33	0.42 ses of bio KE17 38.12 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81 92.62 5.80 2.20 0.51 0.11 1.78	0.52 tites of th KE18 38.09 0.96 14.96 14.30 0.24 15.04 0.18 0.09 8.90 92.82 5.79 2.21 0.47 0.11 1.80	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65 94.10 5.60 2.40 0.44 0.38 2.52	0.43 had gran KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96 94.35 5.62 2.38 0.53 0.36 2.49	0.42 odiorite a KE22 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67 94.64 5.66 2.34 0.43 0.30 2.28	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79 95.56 5.65 2.35 0.43 0.30 2.24	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72 95.55 5.49 2.51 0.70 0.44 2.74	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93 94.61 5.61 2.39 0.38 0.34 2.66	0. <b>RF</b> 35 2 2 2 0 9 0 0 9 0 0 9 3 5 2 0
K Mg/(Mg+Fe) Table 1 Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O Total Number of cat <sup>IV</sup> Si <sup>IV</sup> Al <sup>VI</sup> Al Ti Fe Mn	1.91 0.49 (continu 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51 tions on t 5.96 2.04 0.47 0.19 2.14 0.08	1.91 0.49 <b>ied) Rep</b> <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12 <b>the basis</b> 5.87 2.13 0.50 0.20 2.13 0.07	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75 of 11 ox 5.54 2.46 0.26 0.14 2.53 0.07	1.86 0.51 <b>KE5</b> 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05 <b>ygens</b> 5.63 2.37 0.28 0.14 2.47 0.06	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27 92.97 5.75 2.25 0.34 0.18 2.33 0.07	0.42 <b>EXENT</b> 38.12 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81 92.62 5.80 2.20 0.51 0.11 1.78 0.04	0.52 tites of the second state of the second	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65 94.10 5.60 2.40 0.44 0.38 2.52 0.06	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96 94.35 5.62 2.38 0.53 0.36 2.49 0.06	0.42 odiorite a KE22 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67 94.64 5.66 2.34 0.43 0.30 2.28 0.06	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79 95.56 5.65 2.35 0.43 0.30 2.24 0.06	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72 95.55 5.49 2.51 0.70 0.44 2.74 0.07	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93 94.61 5.61 2.39 0.38 0.34 2.66 0.07	0. <b>RE</b> 355 20 0 9 0 0 9 93 5. 2. 0. 0. 2. 0. 0. 2. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0
K Mg/(Mg+Fe) Table 1 Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O Total Number of cat <sup>IV</sup> Si <sup>IV</sup> Al V <sup>I</sup> Al Ti Fe Mn Mg	1.91 0.49 (continu 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51 cions on ( 5.96 2.04 0.47 0.19 2.14 0.08 2.81	1.91 0.49 <b>ied) Rep</b> <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12 <b>ibe basis</b> 5.87 2.13 0.50 0.20 2.13 0.07 2.77	1.89 0.50 resentat 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75 of 11 ox 5.54 2.46 0.26 0.14 2.53 0.07 3.24	1.86 0.51 <b>KE5</b> 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05 <b>ygens</b> 5.63 2.37 0.28 0.14 2.47 0.06 3.20	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27 92.97 5.75 2.25 0.34 0.18 2.33 0.07 2.89	0.42 <b>EXEND</b> <b>Set of bio</b> <b>KE17</b> 38.12 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81 92.62 5.80 2.20 0.51 0.11 1.78 0.04 3.37	0.52 tites of tl N KE18 38.09 0.96 14.96 14.30 0.24 15.04 0.18 0.09 8.90 92.82 5.79 2.21 0.47 0.11 1.80 0.03 3.41	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65 94.10 5.60 2.40 0.44 0.38 2.52 0.06 2.19	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96 94.35 5.62 2.38 0.53 0.36 2.49 0.06 2.08	0.42 odiorite a 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67 94.64 5.66 2.34 0.43 0.30 2.28 0.06 2.59	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79 95.56 5.65 2.35 0.43 0.30 2.24 0.06 2.65	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72 95.55 5.49 2.51 0.70 0.44 2.74 0.07 1.53	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93 94.61 5.61 2.39 0.38 0.34 2.66 0.07 2.20	0. <b>RE</b> 355 2 155 20 0 9 0 0 9 0 0 9 0 0 9 3 5 2 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
K Mg/(Mg+Fe) Table 1 Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O Total Number of cat <sup>IV</sup> Si <sup>IV</sup> Al V <sup>I</sup> Al Ti Fe Mn Mg Ca	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51 cions on t 5.96 2.04 0.47 0.19 2.14 0.08 2.81 0.05	1.91 0.49 red) Rep GE11 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12 the basis 5.87 2.13 0.50 0.20 2.13 0.07 2.77 0.01	1.89 0.50 resentat KE4 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75 of 11 ox 5.54 2.46 0.26 0.14 2.53 0.07 3.24 0.01	1.86 0.51 ive EPM KE5 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05 ygens 5.63 2.37 0.28 0.14 2.47 0.06 3.20 0.01	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27 92.97 5.75 2.25 0.34 0.18 2.33 0.07 2.89 0.01	0.42 ses of bio KE17 38.12 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81 92.62 5.80 2.20 0.51 0.11 1.78 0.04 3.37 0.03	0.52 tites of the second seco	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65 94.10 5.60 2.40 0.44 0.38 2.52 0.06 2.19 0.01	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96 94.35 5.62 2.38 0.53 0.36 2.49 0.06 2.08 0.01	0.42 odiorite a 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67 94.64 5.66 2.34 0.43 0.30 2.28 0.06 2.59 0.01	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79 95.56 5.65 2.35 0.43 0.30 2.24 0.06 2.65 0.00	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72 95.55 5.49 2.51 0.70 0.44 2.74 0.07 1.53 0.00	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93 94.61 5.61 2.39 0.38 0.34 2.66 0.07 2.20 0.00	0. <b>RE</b> 35 2 15 20 0 9 0 0 9 9 0 0 9 9 3 5. 2. 0. 0. 2. 0. 0. 2. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0
K Mg/(Mg+Fe) Table 1 Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O Total Number of cat <sup>IV</sup> Si <sup>IV</sup> Al <sup>VI</sup> Al VIAl Ti Fe Mn Mg Ca Mn Mg Ca Na	1.91 0.49 (continu 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51 tions on t 5.96 2.04 0.47 0.19 2.14 0.08 2.81 0.05 0.02	1.91 0.49 <b>ied) Rep</b> <b>GE11</b> 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12 <b>the basis</b> 5.87 2.13 0.50 0.20 2.13 0.07 2.77 0.01 0.03	1.89 0.50 resentat KE4 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75 of 11 ox 5.54 2.46 0.26 0.14 2.53 0.07 3.24 0.01 0.01	1.86 0.51 <b>ive EPM</b> <b>KE5</b> 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05 <b>ygens</b> 5.63 2.37 0.28 0.14 2.47 0.06 3.20 0.01 0.00	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27 92.97 5.75 2.25 0.34 0.18 2.33 0.07 2.89 0.01 0.02	0.42 ses of bio KE17 38.12 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81 92.62 5.80 2.20 0.51 0.11 1.78 0.04 3.37 0.03 0.02	0.52 tites of th KE18 38.09 0.96 14.96 14.30 0.24 15.04 0.24 15.04 0.24 15.04 0.18 0.09 8.90 92.82 5.79 2.21 0.47 0.11 1.80 0.03 3.41 0.03 0.03	0.53 he Mash IMEs KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65 94.10 5.60 2.40 0.44 0.38 2.52 0.06 2.19 0.01 0.02	0.43 had gran KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96 94.35 5.62 2.38 0.53 0.36 2.49 0.06 2.08 0.01 0.03	0.42 odiorite : KE22 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67 94.64 5.66 2.34 0.43 0.30 2.28 0.06 2.59 0.01 0.03	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79 95.56 5.65 2.35 0.43 0.30 2.24 0.06 2.65 0.00 0.02	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72 95.55 5.49 2.51 0.70 0.44 2.74 0.07 1.53 0.00 0.02	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93 94.61 5.61 2.39 0.38 0.34 2.66 0.07 2.20 0.00 0.02	0. <b>RE</b> 355 20 0 9 0 0 9 9 0 0 9 9 35 20 0 0 9 9 35 20 0 9 9 0 0 9 9 35 20 0 9 9 0 0 9 0 0 0 9 0 0 0 9 0 0 0 0 0 0 0 0 0 0 0 0 0
K Mg/(Mg+Fe) Table 1 Table 1 Sample SiO <sub>2</sub> TiO <sub>2</sub> Al <sub>2</sub> O <sub>3</sub> FeO MnO MgO CaO Na <sub>2</sub> O K <sub>2</sub> O Total Number of cat <sup>IV</sup> Si <sup>IV</sup> Al V <sup>I</sup> Al Ti Fe Mn Mg Ca	1.91 0.49 (continu GE10 37.85 1.59 13.55 16.41 0.57 11.97 0.27 0.07 8.16 90.51 cions on t 5.96 2.04 0.47 0.19 2.14 0.08 2.81 0.05	1.91 0.49 red) Rep GE11 38.18 1.71 14.50 16.70 0.55 12.09 0.04 0.09 9.21 93.12 the basis 5.87 2.13 0.50 0.20 2.13 0.07 2.77 0.01	1.89 0.50 resentat KE4 35.22 1.16 14.68 19.37 0.56 13.82 0.04 0.03 6.82 91.75 of 11 ox 5.54 2.46 0.26 0.14 2.53 0.07 3.24 0.01	1.86 0.51 ive EPM KE5 36.01 1.21 14.41 19.08 0.44 13.72 0.08 0.01 7.04 92.05 ygens 5.63 2.37 0.28 0.14 2.47 0.06 3.20 0.01	1.90 0.42 A analys KE6 36.93 1.52 14.06 18.06 0.51 12.44 0.07 0.06 9.27 92.97 5.75 2.25 0.34 0.18 2.33 0.07 2.89 0.01	0.42 ses of bio KE17 38.12 1.00 15.10 14.14 0.29 14.86 0.18 0.07 8.81 92.62 5.80 2.20 0.51 0.11 1.78 0.04 3.37 0.03	0.52 tites of the second seco	0.53 he Mash MES KE25 36.00 3.27 15.54 19.53 0.44 9.46 0.06 0.08 9.65 94.10 5.60 2.40 0.44 0.38 2.52 0.06 2.19 0.01	0.43 had gram KE26 36.25 3.11 15.93 19.41 0.49 9.00 0.05 0.10 9.96 94.35 5.62 2.38 0.53 0.36 2.49 0.06 2.08 0.01	0.42 odiorite a 36.99 2.57 15.36 18.00 0.44 11.38 0.06 0.11 9.67 94.64 5.66 2.34 0.43 0.30 2.28 0.06 2.59 0.01	0.42 and its M KE23 37.34 2.61 15.56 17.90 0.48 11.76 0.00 0.05 9.79 95.56 5.65 2.35 0.43 0.30 2.24 0.06 2.65 0.00	0.42 MEs in W RE11 35.67 3.78 17.66 21.42 0.54 6.64 0.00 0.05 9.72 95.55 5.49 2.51 0.70 0.44 2.74 0.07 1.53 0.00	0.54 Vt.% oxid RE13 36.04 2.86 15.06 20.56 0.55 9.47 0.00 0.08 9.93 94.61 5.61 2.39 0.38 0.34 2.66 0.07 2.20 0.00	0. <b>RE</b> 355 2 155 20 0 9 0 0 9 0 0 9 9 3 5. 2. 0. 0. 2. 0. 0. 2. 0. 0. 0. 0. 0. 0. 0. 0. 0. 0

TOTAL	99.6	96.5	99.3	99.6	99.0	99.5
K2O	0.2	0.1	0.2	0.1	0.1	0.1
Na2O	10.1	7.2	8.7	7.6	9.2	8.9
CaO	2.1	1.9	4.4	6.4	3.5	4.2
Al2O3 FeO	21.2 0.0	21.1 0.0	23.2 0.1	24.8 0.1	22.3 0.0	22.8 0.0
SiO2	65.9	66.1	62.9	60.7	63.8	63.5

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Table 1 (continued) Representative EPMA analyses of amphiboles of the Mashhad granodiorite and its MMEs in Wt.% oxide.

								grano	diorite					
sample	G11.7	G11.8	G11.9	G11.10	K4.44	K4.45	K4.46	K4.51	K4.52	K17.28	K17.29	K17.33	K17.34	K17.3
SiO2	45.43	45.22	44.94	43.25	40.99	41.26	42.23	44.03	42.92	43.82	44.04	42.94	43.73	43.97
TiO2	0.97	0.93	0.48	0.41	1.08	0.95	0.73	1.01	0.74	1.05	1.11	1.15	1.20	0.99
Al2O3	7.01	7.25	7.50	8.94	10.09	9.97	9.71	8.70	9.64	9.48	9.66	9.81	9.00	9.02
FeO	18.54	18.79	19.76	20.31	21.22	21.12	20.83	19.71	20.42	19.18	19.38	19.55	19.00	19.08
MnO	1.12	1.11	1.14	1.04	0.71	0.82	0.80	0.73	0.78	0.73	0.75	0.68	0.71	0.74
MgO	10.02	9.75	9.41	8.56	7.90	7.72	8.22	9.20	8.54	9.44	9.51	9.06	9.93	9.78
CaO	11.46	11.21	11.21	11.38	11.13	10.82	10.82	11.14	11.29	11.48	11.36	11.23	11.63	11.44
Na2O	1.53	1.45	1.20	1.44	1.54	1.61	1.52	1.54	1.59	1.56	1.55	1.55	1.60	1.54
K2O	0.96	1.01	0.94	1.28	1.62	1.61	1.45	1.27	1.38	1.28	1.22	1.44	1.26	1.23
					1.02	1.01	1.45	1.2/	1.56	1.20	1.22	1.44	1.20	1.23
Number of					(75	( )5	( 22	(7)	( 75	( 50	( 50	( 57		( )7
Si	6.75	6.57	6.76	6.56	6.75	6.25	6.32	6.73	6.75	6.59	6.59	6.57	6.66	6.37
Al (iv)	1.25	1.43	1.24	1.44	1.25	1.75	1.68	1.27	1.25	1.41	1.41	1.43	1.34	1.63
Al (vi)	0.35	0.32	0.37	0.35	0.58	0.36	0.37	0.50	0.48	0.59	0.52	0.49	0.60	0.37
Ti	0.10	0.11	0.11	0.07	0.06	0.27	0.19	0.06	0.09	0.11	0.13	0.10	0.06	0.11
Cr	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.00	0.01	0.00	0.01	0.00	0.00	0.00
Fe	2.42	2.55	2.48	2.52	2.35	2.52	2.53	2.63	2.56	2.34	2.31	2.32	2.35	1.80
Mn	0.09	0.11	0.10	0.11	0.05	0.06	0.06	0.06	0.06	0.09	0.09	0.09	0.08	0.08
Mg	2.26	2.00	2.20	2.08	2.30	1.96	1.97	2.04	2.11	2.17	2.18	2.11	2.21	2.05
Ca	1.89	1.87	1.86	1.88	1.92	1.88	1.90	1.93	1.89	1.90	1.96	1.92	1.95	1.88
Na	0.45	0.49	0.46	0.45	0.40	0.47	0.46	0.42	0.43	0.34	0.35	0.34	0.33	0.40
К	0.25	0.27	0.25	0.27	0.27	0.36	0.36	0.25	0.25	0.28	0.26	0.23	0.25	0.28
Mg#	48.33	43.97	47.06	45.26	49.44	43.72	43.85	43.75	45.23	48.12	48.56	47.61	48.47	53.25
	<b>T 11 1</b>				EDMA		e 1.						XX// 0/	• •
	Table I (	continue	a) Repre	esentative	EPMA	· ·	<u> </u>	boles of t	he Mashha	d granodic	orite and it	S MINIES I	n wt.% (	xide.
samnle	BF1 26	RF1 27	BF1 28	RF1 31	RF1 32		MMEs BF4 13	<b>BF4 14</b>	GF10 52	GF10 59	GF10.60	CF10.61	KF4 27	KF4 2
sample						BE4.12	BE4.13		GE10.52					
SiO2	41.63	41.84	41.51	41.81	42.35	<b>BE4.12</b> 42.71	<b>BE4.13</b> 42.50	42.54	43.66	44.62	43.23	43.47	42.39	42.25
SiO2 TiO2	41.63 0.99	41.84 1.23	41.51 0.93	41.81 0.93	42.35 0.80	<b>BE4.12</b> 42.71 1.30	<b>BE4.13</b> 42.50 1.36	42.54 0.81	43.66 0.60	44.62 0.54	43.23 0.60	43.47 0.55	42.39 1.23	42.25 0.96
SiO2 TiO2 Al2O3	41.63 0.99 11.12	41.84 1.23 10.90	41.51 0.93 11.02	41.81 0.93 11.06	42.35 0.80 10.55	<b>BE4.12</b> 42.71 1.30 9.63	<b>BE4.13</b> 42.50 1.36 9.77	42.54 0.81 9.96	43.66 0.60 9.03	44.62 0.54 8.20	43.23 0.60 9.42	43.47 0.55 9.06	42.39 1.23 10.03	42.25 0.96 10.28
SiO2 TiO2 Al2O3 FeO	41.63 0.99 11.12 18.60	41.84 1.23 10.90 18.53	41.51 0.93 11.02 18.94	41.81 0.93 11.06 18.71	42.35 0.80 10.55 18.20	<b>BE4.12</b> 42.71 1.30 9.63 19.41	<b>BE4.13</b> 42.50 1.36 9.77 20.40	42.54 0.81 9.96 21.18	43.66 0.60 9.03 18.81	44.62 0.54 8.20 18.15	43.23 0.60 9.42 18.92	43.47 0.55 9.06 18.60	42.39 1.23 10.03 18.82	42.25 0.96 10.28 19.05
SiO2 TiO2 Al2O3 FeO MnO	41.63 0.99 11.12 18.60 0.61	41.84 1.23 10.90 18.53 0.61	41.51 0.93 11.02 18.94 0.56	41.81 0.93 11.06 18.71 0.60	42.35 0.80 10.55 18.20 0.53	<b>BE4.12</b> 42.71 1.30 9.63 19.41 0.65	<b>BE4.13</b> 42.50 1.36 9.77 20.40 0.73	42.54 0.81 9.96 21.18 0.67	43.66 0.60 9.03 18.81 0.95	44.62 0.54 8.20 18.15 0.96	43.23 0.60 9.42 18.92 0.92	43.47 0.55 9.06 18.60 0.90	42.39 1.23 10.03 18.82 0.68	42.25 0.96 10.28 19.05 0.68
SiO2 TiO2 Al2O3 FeO	41.63 0.99 11.12 18.60	41.84 1.23 10.90 18.53	41.51 0.93 11.02 18.94	41.81 0.93 11.06 18.71	42.35 0.80 10.55 18.20	<b>BE4.12</b> 42.71 1.30 9.63 19.41	<b>BE4.13</b> 42.50 1.36 9.77 20.40	42.54 0.81 9.96 21.18	43.66 0.60 9.03 18.81	44.62 0.54 8.20 18.15	43.23 0.60 9.42 18.92	43.47 0.55 9.06 18.60	42.39 1.23 10.03 18.82	42.25 0.96 10.28 19.05 0.68 9.16
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O	41.63 0.99 11.12 18.60 0.61 9.00	41.84 1.23 10.90 18.53 0.61 9.03	41.51 0.93 11.02 18.94 0.56 9.25	41.81 0.93 11.06 18.71 0.60 9.24	42.35 0.80 10.55 18.20 0.53 9.37	<b>BE4.12</b> 42.71 1.30 9.63 19.41 0.65 9.13	<b>BE4.13</b> 42.50 1.36 9.77 20.40 0.73 8.25	42.54 0.81 9.96 21.18 0.67 8.12	43.66 0.60 9.03 18.81 0.95 9.67	44.62 0.54 8.20 18.15 0.96 10.10	43.23 0.60 9.42 18.92 0.92 9.26	43.47 0.55 9.06 18.60 0.90 9.57	42.39 1.23 10.03 18.82 0.68 9.72	42.25 0.96 10.28 19.05 0.68 9.16
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34	42.35 0.80 10.55 18.20 0.53 9.37 11.46	<b>BE4.12</b> 42.71 1.30 9.63 19.41 0.65 9.13 11.24	<b>BE4.13</b> 42.50 1.36 9.77 20.40 0.73 8.25 11.34	42.54 0.81 9.96 21.18 0.67 8.12 11.02	43.66 0.60 9.03 18.81 0.95 9.67 11.56	44.62 0.54 8.20 18.15 0.96 10.10 11.56	43.23 0.60 9.42 18.92 0.92 9.26 11.32	43.47 0.55 9.06 18.60 0.90 9.57 11.45	42.39 1.23 10.03 18.82 0.68 9.72 11.37	42.25 0.96 10.28 19.05 0.68 9.16 11.22
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations of	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>1 the bas</b>	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b>	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b>	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32	<b>BE4.12</b> 42.71 1.30 9.63 19.41 0.65 9.13 11.24 1.61 1.21	<b>BE4.13</b> 42.50 1.36 9.77 20.40 0.73 8.25 11.34 1.41 1.26	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of Si	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations or 6.37	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>1 the bas</b> 6.35	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b> 6.40	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b> 6.49	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32 6.44	<b>BE4.12</b> 42.71 1.30 9.63 19.41 0.65 9.13 11.24 1.61 1.21 6.59	<b>BE4.13</b> 42.50 1.36 9.77 20.40 0.73 8.25 11.34 1.41 1.26 6.76	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18 6.60	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30 6.70	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07 6.57	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29 6.55	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25 6.59	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50 6.54	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52 7.11
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of Si Al (iv)	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations of 6.37 1.63	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>h the bass</b> 6.35 1.65	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b> 6.40 1.60	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b> 6.49 1.51	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32 6.44 1.56	<b>BE4.12</b> 42.71 1.30 9.63 19.41 0.65 9.13 11.24 1.61 1.21 6.59 1.41	<b>BE4.13</b> 42.50 1.36 9.77 20.40 0.73 8.25 11.34 1.41 1.26 6.76 1.24	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18 6.60 1.40	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30 6.70 1.30	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07 6.57 1.43	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29 6.55 1.45	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25 6.59 1.41	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50 6.54 1.46	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52 7.11 0.89
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of Si Al (iv) Al (vi)	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations of 6.37 1.63 0.37	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>h the bas</b> 6.35 1.65 0.34	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b> 6.40 1.60 0.39	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b> 6.49 1.51 0.25	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32 6.44 1.56 0.22	<b>BE4.12</b> 42.71 1.30 9.63 19.41 0.65 9.13 11.24 1.61 1.21 6.59 1.41 0.62	<b>BE4.13</b> 42.50 1.36 9.77 20.40 0.73 8.25 11.34 1.41 1.26 6.76 1.24 0.69	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18 6.60 1.40 0.67	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30 6.70 1.30 0.33	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07 6.57 1.43 0.40	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29 6.55 1.45 0.43	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25 6.59 1.41 0.40	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50 6.54 1.46 0.32	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52 7.11 0.89 0.37
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of Si Al (iv) Al (vi) Ti	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations of 6.37 1.63 0.37 0.11	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>1 the bas</b> 6.35 1.65 0.34 0.11	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b> 6.40 1.60 0.39 0.11	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b> 6.49 1.51 0.25 0.09	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32 6.44 1.56 0.22 0.10	<b>BE4.12</b> 42.71 1.30 9.63 19.41 0.65 9.13 11.24 1.61 1.21 6.59 1.41 0.62 0.09	<b>BE4.13</b> 42.50 1.36 9.77 20.40 0.73 8.25 11.34 1.41 1.26 6.76 1.24 0.69 0.17	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18 6.60 1.40 0.67 0.17	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30 6.70 1.30 0.33 0.05	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07 6.57 1.43 0.40 0.14	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29 6.55 1.45 0.43 0.11	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25 6.59 1.41 0.40 0.13	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50 6.54 1.46 0.32 0.08	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52 7.11 0.89 0.37 0.10
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of Si Al (iv) Al (vi) Ti Cr	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations or 6.37 1.63 0.37 0.11 0.00	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>1.44</b> 0.35 1.65 0.34 0.11 0.00	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b> 6.40 1.60 0.39 0.11 0.00	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b> 6.49 1.51 0.25 0.09 0.00	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32 6.44 1.56 0.22 0.10 0.00	<b>BE4.12</b> 42.71 1.30 9.63 19.41 0.65 9.13 11.24 1.61 1.21 6.59 1.41 0.62 0.09 0.00	<b>BE4.13</b> 42.50 1.36 9.77 20.40 0.73 8.25 11.34 1.41 1.26 6.76 1.24 0.69 0.17 0.00	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18 6.60 1.40 0.67 0.17 0.00	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30 6.70 1.30 0.33 0.05 0.01	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07 6.57 1.43 0.40 0.14 0.00	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29 6.55 1.45 0.43 0.11 0.00	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25 6.59 1.41 0.40 0.13 0.00	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50 6.54 1.46 0.32 0.08 0.00	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52 7.11 0.89 0.37 0.10 0.00
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of Si Al (iv) Al (vi) Ti Cr Fe	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations of 6.37 1.63 0.37 0.11 0.00 1.80	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>1 the bas</b> 6.35 1.65 0.34 0.11 0.00 1.64	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b> 6.40 1.60 0.39 0.11 0.00 1.47	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b> 6.49 1.51 0.25 0.09 0.00 2.19	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32 6.44 1.56 0.22 0.10 0.00 2.51	BE4.12           42.71           1.30           9.63           19.41           0.65           9.13           11.24           1.61           1.21           6.59           1.41           0.62           0.09           0.00           2.41	BE4.13           42.50           1.36           9.77           20.40           0.73           8.25           11.34           1.41           1.26           6.76           1.24           0.69           0.17           0.00           2.55	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18 6.60 1.40 0.67 0.17 0.00 2.56	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30 6.70 1.30 0.33 0.05 0.01 2.63	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07 6.57 1.43 0.40 0.14 0.00 2.44	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29 6.55 1.45 0.43 0.11 0.00 2.47	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25 6.59 1.41 0.40 0.13 0.00 2.41	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50 6.54 1.46 0.32 0.08 0.00 2.70	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52 7.11 0.89 0.37 0.10 0.00 2.41
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of Si Al (iv) Al (vi) Ti Cr Fe Mn	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations or 6.37 1.63 0.37 0.11 0.00 1.80 0.08	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>n the bas</b> 6.35 1.65 0.34 0.11 0.00 1.64 0.07	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b> 6.40 1.60 0.39 0.11 0.00 1.47 0.08	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b> 6.49 1.51 0.25 0.09 0.00 2.19 0.11	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32 6.44 1.56 0.22 0.10 0.00 2.51 0.10	BE4.12           42.71           1.30           9.63           19.41           0.65           9.13           11.24           1.61           1.21           6.59           1.41           0.62           0.09           0.00           2.41           0.08	BE4.13 42.50 1.36 9.77 20.40 0.73 8.25 11.34 1.41 1.26 6.76 1.24 0.69 0.17 0.00 2.55 0.07	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18 6.60 1.40 0.67 0.17 0.00 2.56 0.08	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30 6.70 1.30 0.33 0.05 0.01 2.63 0.14	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07 6.57 1.43 0.40 0.14 0.00 2.44 0.09	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29 6.55 1.45 0.43 0.11 0.00 2.47 0.09	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25 6.59 1.41 0.40 0.13 0.00 2.41 0.09	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50 6.54 1.46 0.32 0.08 0.00 2.70 0.10	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52 7.11 0.89 0.37 0.10 0.00 2.41 0.11
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of Si Al (iv) Al (vi) Ti Cr Fe Mn Mg	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations or 6.37 1.63 0.37 0.11 0.00 1.80 0.08 2.05	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>n the bass</b> 6.35 1.65 0.34 0.11 0.00 1.64 0.07 2.11	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b> 6.40 1.60 0.39 0.11 0.00 1.47 0.08 2.11	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b> 6.49 1.51 0.25 0.09 0.00 2.19 0.11 1.91	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32 6.44 1.56 0.22 0.10 0.00 2.51 0.10 1.86	BE4.12           42.71           1.30           9.63           19.41           0.65           9.13           11.24           1.61           1.21           6.59           1.41           0.62           0.09           0.00           2.41           0.08           2.21	BE4.13 42.50 1.36 9.77 20.40 0.73 8.25 11.34 1.41 1.26 6.76 1.24 0.69 0.17 0.00 2.55 0.07 1.90	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18 6.60 1.40 0.67 0.17 0.00 2.56 0.08 1.82	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30 6.70 1.30 0.33 0.05 0.01 2.63 0.14 1.98	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07 6.57 1.43 0.40 0.14 0.00 2.44 0.09 2.25	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29 6.55 1.45 0.43 0.11 0.00 2.47 0.09 2.12	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25 6.59 1.41 0.40 0.13 0.00 2.41 0.09 2.23	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50 6.54 1.46 0.32 0.08 0.00 2.70 0.10 1.90	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52 7.11 0.89 0.37 0.10 0.00 2.41 0.11 2.40
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of Si Al (iv) Al (vi) Ti Cr Fe Mn Mg Ca	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations of 6.37 1.63 0.37 0.11 0.00 1.80 0.08 2.05 1.88	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>1 the bas</b> 6.35 1.65 0.34 0.11 0.00 1.64 0.07 2.11 1.89	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b> 6.40 1.60 0.39 0.11 0.00 1.47 0.08 2.11 1.90	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b> 6.49 1.51 0.25 0.09 0.00 2.19 0.11 1.91 1.87	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32 6.44 1.56 0.22 0.10 0.00 2.51 0.10 1.86 1.83	BE4.12           42.71           1.30           9.63           19.41           0.65           9.13           11.24           1.61           1.21           6.59           1.41           0.62           0.09           0.00           2.41           0.08           2.21           1.66	BE4.13 42.50 1.36 9.77 20.40 0.73 8.25 11.34 1.41 1.26 6.76 1.24 0.69 0.17 0.00 2.55 0.07 1.90 1.92	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18 6.60 1.40 0.67 0.17 0.00 2.56 0.08 1.82 1.87	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30 6.70 1.30 0.33 0.05 0.01 2.63 0.14 1.98 1.89	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07 6.57 1.43 0.40 0.14 0.00 2.44 0.09 2.25 1.89	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29 6.55 1.45 0.43 0.11 0.00 2.47 0.09 2.12 1.86	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25 6.59 1.41 0.40 0.13 0.00 2.41 0.09 2.23 1.88	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50 6.54 1.46 0.32 0.08 0.00 2.70 0.10 1.90 1.80	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52 7.11 0.89 0.37 0.10 0.00 2.41 0.11 2.40 1.91
SiO2 TiO2 Al2O3 FeO MnO MgO CaO Na2O K2O Number of Si Al (iv) Al (vi) Ti Cr Fe Mn Mg	41.63 0.99 11.12 18.60 0.61 9.00 11.50 1.36 1.42 cations or 6.37 1.63 0.37 0.11 0.00 1.80 0.08 2.05	41.84 1.23 10.90 18.53 0.61 9.03 11.47 1.44 1.41 <b>n the bass</b> 6.35 1.65 0.34 0.11 0.00 1.64 0.07 2.11	41.51 0.93 11.02 18.94 0.56 9.25 11.52 1.49 1.43 <b>is of 23 o</b> 6.40 1.60 0.39 0.11 0.00 1.47 0.08 2.11	41.81 0.93 11.06 18.71 0.60 9.24 11.59 1.43 1.34 <b>xygens</b> 6.49 1.51 0.25 0.09 0.00 2.19 0.11 1.91	42.35 0.80 10.55 18.20 0.53 9.37 11.46 1.49 1.32 6.44 1.56 0.22 0.10 0.00 2.51 0.10 1.86	BE4.12           42.71           1.30           9.63           19.41           0.65           9.13           11.24           1.61           1.21           6.59           1.41           0.62           0.09           0.00           2.41           0.08           2.21	BE4.13 42.50 1.36 9.77 20.40 0.73 8.25 11.34 1.41 1.26 6.76 1.24 0.69 0.17 0.00 2.55 0.07 1.90	42.54 0.81 9.96 21.18 0.67 8.12 11.02 1.39 1.18 6.60 1.40 0.67 0.17 0.00 2.56 0.08 1.82	43.66 0.60 9.03 18.81 0.95 9.67 11.56 1.44 1.30 6.70 1.30 0.33 0.05 0.01 2.63 0.14 1.98	44.62 0.54 8.20 18.15 0.96 10.10 11.56 1.38 1.07 6.57 1.43 0.40 0.14 0.00 2.44 0.09 2.25	43.23 0.60 9.42 18.92 0.92 9.26 11.32 1.48 1.29 6.55 1.45 0.43 0.11 0.00 2.47 0.09 2.12	43.47 0.55 9.06 18.60 0.90 9.57 11.45 1.49 1.25 6.59 1.41 0.40 0.13 0.00 2.41 0.09 2.23	42.39 1.23 10.03 18.82 0.68 9.72 11.37 1.65 1.50 6.54 1.46 0.32 0.08 0.00 2.70 0.10 1.90	42.25 0.96 10.28 19.05 0.68 9.16 11.22 1.73 1.52 7.11 0.89 0.37 0.10 0.00 2.41 0.11 2.40

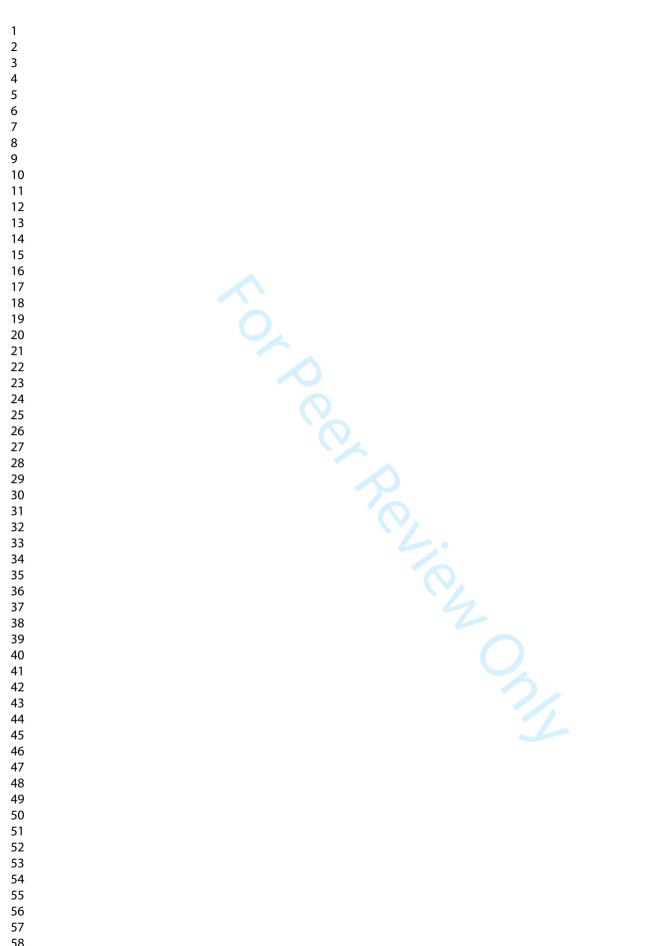


Table 1 (continued) Representative EPMA analyses of amphiboles of the Mashhad granodiorite and its MMEs in Wt.% oxide

K17.36	sample	KE4.30	KE4.37	KE4.38	KE4.39	KE15.58	KE15.62	KE15.75	KE17.4	KE17.5	KE17.9	KE17.1
43.08	SiO2	43.57	42.41	43.62	42.34	42.83	43.54	47.06	44.645	42.94	51.11	41.48
0.89	TiO2	0.89	0.98	0.94	0.62	1.26	0.54	0.25	0.384	0.73	0.10	0.56
10.00	Al2O3	8.75	9.57	8.81	9.77	10.10	10.05	6.66	9.960	11.15	4.22	11.62
19.56	FeO	18.69	19.68	19.13	19.41	18.64	18.17	18.25	15.943	16.64	13.98	18.75
0.72	MnO	0.72	0.83	0.78	0.81	0.44	0.39	0.46	0.316	0.27	0.38	0.37
9.11	MgO	9.80	8.66	9.54	9.00	9.24	9.96	10.79	11.240	10.63	13.84	9.14
11.37	CaO	11.40	11.25	11.17	11.30	11.42	11.59	11.97	11.753	11.73	12.65	11.77
1.50	Na2O	1.498	1.642	1.53	1.49	1.36	1.33	0.96	1.485	1.583	0.623	1.440
1.30	K2O	1.286	1.389	1.28	1.38	1.41	1.37	0.73	1.233	1.495	0.384	1.671
	Number	of cations o	on the basi	s of 23 oxy	gens							
6.35	Si	6.651	6.239	6.21	6.64	6.06	6.21	6.43	6.171	6.538	6.790	6.825
1.65	Al (iv)	1.349	1.761	1.79	1.36	1.94	1.79	1.57	1.829	1.462	1.210	1.175
0.34	Al (vi)	0.413	0.366	0.34	0.48	0.39	0.65	0.55	0.200	0.499	0.521	0.590
0.11	Ti	0.086	0.223	0.28	0.15	0.21	0.26	0.06	0.131	0.089	0.122	0.130
0.00	Cr	0.000	0.000	0.00	0.00	0.00	0.00	0.00	0.000	0.000	0.000	0.000
1.64	Fe	2.646	2.514	2.56	2.42	2.28	2.29	2.43	2.391	2.236	2.485	2.51
0.07	Mn	0.102	0.060	0.06	0.06	0.04	0.04	0.05	0.044	0.048	0.095	0.098
2.11	Mg	1.973	1.911	1.80	2.13	2.01	2.09	2.11	2.010	2.321	2.182	2.197
1.89	Ca	1.875	1.881	1.92	1.90	1.95	1.93	1.95	1.954	1.927	1.905	1.886
0.44	Na	0.479	0.507	0.47	0.41	0.45	0.47	0.43	0.412	0.445	0.470	0.460
0.28	К	0.272	0.374	0.39	0.28	0.41	0.40	0.33	0.345	0.286	0.253	0.242
56.27	Mg#	42.72	43.18	41.35	46.90	46.79	47.70	46.48	45.66	50.93	46.75	46.60

Table 1 (continued) Representative EPM	A analyses of amphiboles of the Ma	ashhad granodiorite and its MMEs in Wt.% oxid

							MMEs			
KE4.29	sample	KE17.14	KE17.15	KE17.16	KE17.20	KE25.49	KE25.50	KE25.51	KE25.52	
42.54	SiO2	51.48	53.10	39.82	42.19	42.54	42.53	42.43	42.96	
1.13	TiO2	0.36	0.00	1.13	0.76	0.97	1.08	0.85	0.50	
9.91	Al2O3	2.68	0.98	11.11	10.74	10.91	10.57	10.50	10.67	
18.59	FeO	10.43	9.37	18.45	17.25	18.06	17.82	17.89	18.14	
0.68	MnO	0.36	0.40	0.33	0.36	0.69	0.70	0.68	0.58	
9.65	MgO	11.07	12.14	8.70	10.05	9.39	9.43	9.12	9.57	
11.33	CaO	22.75	23.34	11.77	11.61	11.42	11.83	11.58	11.74	
1.62	Na2O	0.67	0.56	1.37	1.48	1.12	1.15	1.12	1.08	
1.38	K2O	0.031	0.050	1.74	1.45	1.43	1.31	1.18	1.27	
	Number	of cations o	on the basis	s of 23 oxy	gens					
6.82	Si	6.654	6.777	6.81	6.68	6.60	6.44	6.53	6.52	
1.18	Al (iv)	1.346	1.223	1.19	1.32	1.40	1.56	1.47	1.48	
0.41	Al (vi)	0.446	0.421	0.46	0.50	0.47	0.26	0.30	0.18	
0.12	Ti	0.134	0.140	0.12	0.10	0.06	0.07	0.06	0.10	
0.00	Cr	0.006	0.000	0.00	0.00	0.00	0.00	0.00	0.00	
2.55	Fe	2.534	2.462	2.47	2.53	1.51	2.21	1.97	2.34	
0.10	Mn	0.089	0.093	0.10	0.09	0.11	0.10	0.09	0.11	
2.13	Mg	2.093	2.294	2.26	2.11	1.89	1.91	1.88	1.90	
1.85	Ca	1.864	1.931	1.90	1.89	1.86	1.84	1.88	1.84	
0.46	Na	0.465	0.480	0.46	0.45	0.41	0.41	0.40	0.40	
0.25	К	0.284	0.249	0.24	0.26	0.24	0.26	0.24	0.24	
45.43	Mg#	45.24	48.24	47.76	45.38	55.61	46.37	48.80	44.75	

Table 2	Major oxides (wt%) and trace element (ppm) contents of Mashhad granodiorite and its MMEs
I ubic 2	major oxides (werve) and trace element (ppin) contents or masimud frandulorite and its minitize

Rock Type						MMEs								
Sample No.	BE1	KE24	KE26	KE27	KE4	BE3	GE1	GE10	GE4	RE11	<b>RE12</b>	RE5	RE7	RF
SiO <sub>2</sub>	48.75	56.65	55.12	53.55	55.30	50.64	57.16	57.03	57.29	65.98	55.85	62.88	57.85	62.
$Al_2O_3$	18.18	15.34	17.84	18.07	17.58	19.42	17.09	17.05	16.69	16.26	18.09	16.84	16.87	16
Fe <sub>2</sub> O <sub>3</sub> <sup>T</sup>	8.09	5.96	6.79	8.11	7.04	8.05	6.66	6.43	5.41	3.79	8.43	5.11	8.68	5.
MgO	4.42	5.09	3.97	3.89	3.11	4.09	3.77	3.63	3.54	1.43	3.63	2.61	3.87	3.
CaO	7.40	7.22	6.39	6.19	7.05	6.95	5.47	5.49	5.67	2.65	4.24	3.56	3.32	3.
Na <sub>2</sub> O	6.61	3.43	4.91	4.38	4.84	4.75	4.81	4.59	4.88	3.63	4.14	3.92	3.24	3
$K_2O$	2.00	3.18	1.97	2.87	2.14	2.42	2.04	2.24	3.47	3.53	2.78	2.45	2.97	2
TiO <sub>2</sub>	1.17	0.82	1.09	1.10	0.99	1.12	1.01	0.91	0.83	0.59	1.07	0.76	1.02	0
MnO	0.18	0.14	0.14	0.20	0.24	0.17	0.21	0.22	0.13	0.10	0.22	0.13	0.22	0
P2O5	1.13	0.53	0.75	0.88	0.71	1.10	0.85	0.80	0.67	0.34	0.75	0.46	0.64	0
$H_2O+$	0.27	0.53	-	0.00	0.19	0.30	0.09	0.32	0.53	0.07	0.01	0.22	0.01	
H <sub>2</sub> O-	0.94	0.54	0.63	0.78	0.28	0.61	0.69	0.42	0.42	0.57	0.59	0.73	0.80	0
Total	99.39	99.63	99.74	100.23	99.72	99.86	99.98	99.33	99.82	99.06	99.88	99.77	99.58	9
V	136	93.5	118	119	174	113	122	120	84.1	34.2	126.5	72.9	112.0	9
Cr	4.26	93.97	3.77	8.68	1.74	1.70	9.38	6.67	21.2	4.04	1.68	12.14	19.15	1
Co	29.9	35.2	28.5	25.1	39.0	24.9	31.1	30.2	26.9	35.8	28.3	30.1	28.5	2
Ni	4.82	49.75	8.49	7.02	7.54	3.91	11.2	10.9	20.2	2.09	2.40	8.47	15.0	1
Rb	165	105	203	189	64.1	146	217	288	139	201	272	230	356	2
Sr	1290	786	923	1241	1609	1188	992	1029	1084	526	553	704	413	:
Y	29.2	19.5	23.9	33.7	25.3	27.8	22.3	20.2	17.8	23.9	34.9	11.5	17.7	1
Zr	400	222	379	412	300	374	232	285	303	304	315	237	250	
Nb	60.0	79.0	62.4	88.7	82.4	63.2	66.3	54.0	44.5	58.5	95.1	38.9	68.6	5
Ва	1195	1192	1220	1313	1200	1140	354	1031	1855	942	298	317	258	-
La	148	94	113	141	150	133	97	119	107	133	79.6	64.9	78.7	6
Ce	270	169	206	257	287	240	193	223	194	230	144	111	141	
Pr Nd	27.0 91.1	17.0 57.5	20.8 71.7	26.4 90.8	30.6 106.3	24.1 81.8	20.5 71.8	23.4 81.8	19.5 66.1	21.6 68.4	14.5 50.3	10.4 34.2	13.8 46.3	1
Sm	13.1	8.71	10.7	90.8 13.6	100.5	12.1	10.6		9.47	10.0	50.5 9.66	54.2 4.98	40.5 6.87	6
Eu	3.19	2.20	2.60	3.16	3.65	2.92	2.60	2.82	2.34	1.63	1.74	1.20	1.30	1
Tb	1.14	0.77	0.94	1.27	1.13	1.08	0.88	0.88	0.77	0.93	1.21	0.436	0.689	0
Gd	8.98	6.03	7.51	9.70	9.34	8.49	7.12	7.33	6.18	6.99	8.39	3.47	5.20	4
Dy	5.81	3.91	4.86	6.60	5.11	5.63	4.41	4.14	3.66	4.66	6.72	2.26	3.52	2
Но	1.04	0.714	0.889	1.23	0.869	1.04	0.799	0.716	0.649	0.826	1.23	0.411	0.644	0
Er	2.89	1.94	2.37	3.38	2.33	2.88	2.17	1.95	1.82	2.25	3.25	1.14	1.72	1
Yb	2.63	1.85	2.15	3.18	2.02	2.54	1.96	1.70	1.59	2.02	2.80	1.01	1.43	1
Lu	0.381	0.285	0.334	0.489	0.300	0.370	0.299	0.244	0.231	0.296	0.399	0.157	0.223	0
Hf	8.14	5.17	8.37	9.17	5.98	8.09	5.35	6.14	6.46	6.51	7.12	5.47	5.81	4
Та	2.73	4.95	2.86	4.71	3.76	2.81	3.42	2.98	2.63	3.30	3.88	2.74	2.28	3
Pb	14.9	24.5	20.4	22.8	21.4	15.7	17.1	18.0	25.2	38.6	20.3	11.4	13.8	1
Th	23.7	19.3	23.6	30.6	37.5	22.9	22.2	29.5	26.1	32.5	30.4	19.5	21.5	2
U	3.48	4.67	5.22	5.91	6.06	2.68	4.03	6.32	6.46	3.53	7.23	1.86	2.47	4
∑REE	575	365	444	557	614	516	413	478	413	483	324	235	301	2
(La/Yb)N	70.31	24.35	41.39	44.60	32.19	44.85	25.90	41.21	42.53	57.11	26.83	15.77	52.74	3
Eu/Eu*	0.90	0.93	0.89	0.84	0.94	0.88	0.92	0.94	0.93	0.60	0.59	0.88	0.66	(
Ba/La	8.10	12.65	10.80	9.31	7.97	8.55	3.66	8.68	17.39	7.07	3.74	4.87	3.28	3
Nb/La	0.41	0.84	0.55	0.63	0.55	0.47	0.69	0.45	0.42	0.44	1.19	0.60	0.87	0
Ba/Zr	2.98	5.38	3.22	3.19	4.00	3.05	1.53	3.62	6.13	3.10	0.94	1.34	1.03	1
Rb/Sr	0.13	0.13	0.22	0.15	0.04	0.12	0.22	0.28	0.13	0.38	0.49	0.33	0.86	0
Nb/Ce	0.22	0.47	0.30	0.35	0.29	0.26	0.34	0.24	0.23	0.25	0.66	0.35	0.49	0
Th/Ta	8.71	3.90	8.25	6.50	9.98	8.12	6.50	9.92	9.90	9.86	7.83	7.12	9.43	5

D.L. = Detectin Limit; \* Dilution Factor of 2000; \*\* Dilution Factor of 4000

Rock Type		Μ	MEs				0	Granodio	rite			Stand	lard
Sample No.	<b>KE25</b>	KE15	KE16E	KE17E	B1	G11	G3	KE17H	R2	<b>R</b> 7	<b>RE10</b>	BHVO-1	D.L.*
SiO <sub>2</sub>	56.91	51.40	47.98	48.75	67.63	67.53	68.57	66.78	69.26	68.14	69.36	50.32	5.00
$Al_2O_3$	17.59	11.66	12.98	13.51	15.25	15.97	15.68	15.16	15.36	15.78	15.45	13.41	5.00
Fe <sub>2</sub> O <sub>3</sub> <sup>T</sup>	7.18	11.89	12.12	10.35	2.51	3.27	2.65	2.71	2.66	3.48	2.64	12.14	5.00
MgO	3.91	5.81	5.73	6.01	1.07	1.27	1.16	1.39	0.99	1.67	1.36	7.09	3.00
CaO	4.96	11.07	12.54	11.69	2.26	3.02	2.54	3.11	2.14	2.67	1.70	11.25	3.00
Na <sub>2</sub> O	4.08	2.02	2.83	2.59	6.18	4.43	4.54	4.54	3.77	3.96	3.49	2.37	2.50
K <sub>2</sub> O	2.75	3.18	2.23	2.85	3.16	2.50	3.34	3.58	3.79	2.38	4.18	0.55	2.50
TiO <sub>2</sub>	1.04	1.06	1.40	1.34	0.35	0.44	0.37	0.38	0.39	0.51	0.38	2.74	2.50
MnO	0.17	0.27	0.20	0.22	0.06	0.10	0.08	0.09	0.08	0.10	0.08	0.17	2.50
P <sub>2</sub> O <sub>5</sub>	0.74	1.42	1.43	1.55	0.20	0.25	0.24	0.29	0.22	0.28	0.25	0.28	2.50
H <sub>2</sub> O+	0.01	0.22	0.08	0.06	0.24	0.18	-0.07	0.27	0.27	0.02	0.04		
H <sub>2</sub> O-	0.00	0.49	0.59	0.66	0.62	0.23	0.38	0.53	0.37	0.61	0.64		
Total	99.47	100.7	100.3	99.85	99.71	99.37	99.66	99.03	99.45	99.72	99.69		D.L.*
V	105	292	304	237	30.4	44.2	35.8	43.0	28.0	39.4	25.0	332.05	0.12
Cr	7.61	0.83	3.51	2.41	2.41	2.78	4.31	5.66	5.03	6.33	4.77	292.50	0.20
Со	39.4	57.1	49.6	44.3	62.4	51.9	59.4	37.8	40.2	49.0	35.0	44.11	0.20
Ni	11.7	7.59	8.02	15.6	1.51	2.18	2.71	3.81	2.54	4.02	2.68	118.02	0.01
Rb	215	168	66.9	108	150	192	164	117	231	190	2.00	9.65	0.03
Sr	868	950	1116	1054	749	1020	856	859	468	608	474	416.20	0.37
Y	22.1	35.1	26.9	28.3	11.7	15.2	11.9	13.6	12.1	11.7	11.7	25.02	0.02
Zr	387	207	190	227	224	229	214	178	155	183	151	184.04	0.94
Nb	58.4	31.9	30.4	29.9	40.7	57.5	38.9	39.9	32.2	35.7	29.2	18.09	0.01
Ba	645	1712	1240	2097	1099	1029	943	1124	793	746	971	129.28	0.28
La	105	159	121	130	73.4	102.9	86.0	64.3	52.8	55.3	50.1	14.40	0.03
Ce	183	326	259	286	127	178	149	118	95	100	95	36.80	0.05
Pr	18.5	37.4	31.4	34.8	11.9	16.6	13.6	11.8	8.92	9.39	9.00	4.94	0.01
Nd	63.3	142	123	132	37.8	52.9	42.6	40.2	29.6	31.0	29.7	24.22	0.03
Sm	9.42	21.9	19.5	20.4	5.30	7.21	5.72	6.15	4.57	4.65	4.71	5.99	0.02
Eu	1.94	4.70	4.41	4.57	1.26	1.60	1.25	1.49	1.07	1.14	1.14	2.01	0.00
Tb	0.857	1.64	1.41	1.45	0.455	0.584	0.445	0.540	0.450	0.430	0.454	0.87	0.00
Gd	6.80	14.18	12.37	12.83	3.50	4.58	3.55	4.18	3.29	3.31	3.41	6.21	0.01
Dy	4.45	7.50	6.32	6.48	2.24	2.86	2.21	2.75	2.36	2.27	2.29	5.21	0.01
Но	0.815	1.25	1.03	1.07	0.406	0.523	0.402	0.501	0.435	0.415	0.399	0.91	0.00
Er	2.21	3.23	2.55	2.65	1.14	1.47	1.15	1.35	1.22	1.18	1.11	2.51	0.01
Yb	2.02	2.53	1.90	1.96	1.12	1.44	1.18	1.27	1.20	1.16	1.04	1.93	0.01
Lu	0.310	0.354	0.263	0.276	0.164	0.219	0.181	0.185	0.184	0.171	0.150	0.25	0.00
Hf	8.88	4.70	5.24	5.71	4.81	4.73	4.54	4.17	3.94	4.28	3.66	4.31	0.03
Ta	3.27	1.34	1.77	1.37	2.80	3.67	2.67	3.35	3.18	3.50	2.97	1.00	0.00
Pb Th	16.8 25.4	24.7 14.8	15.8 14.4	20.9 30.3	25.7 25.4	23.7 38.9	27.7 32.2	24.6 20.6	32.4 18.6	25.2 18.2	39.2 20.3	1.39 1.06	0.07
U	5.00	4.77	3.57	30.3 8.49	3.27	5.47	2.47	4.19	2.23	2.13	20.3	0.35	0.01
∑REE	399	721	583	634	266	371	307	253	201	210	198	0.55	0.0
		53.43		46.49						31.40			
(La/Yb)N	45.45		32.40		13.68	62.26	40.58	31.24	28.22		29.25		
Eu/Eu*	0.74	0.82	0.87	0.86	0.89	0.85	0.85	0.90	0.84	0.89	0.87		
Ba/La	6.13	10.78	10.28	16.10	14.97	10.00	10.96	17.49	15.01	13.48	19.36		
Nb/La	0.56	0.20	0.25	0.23	0.55	0.56	0.45	0.62	0.61	0.64	0.58		
Ba/Zr	1.67	8.28	6.54	9.22	4.91	4.50	4.40	6.33	5.11	4.08	6.43		
Rb/Sr	0.25	0.18	0.06	0.10	0.20	0.19	0.19	0.14	0.49	0.31	0.44		
Nb/Ce	0.32	0.10	0.12	0.10	0.32	0.32	0.26	0.34	0.34	0.36	0.31		
Th/Ta	7.77	11.03	8.13	22.09	9.07	10.61	12.04	6.14	5.85	5.18	6.83		

Table 3	LA-ICP-MS U-Pb data on zin	rcons and ages of the MMEs	in Mashhad granodiorite

Analysis GE1 (Mean A02 A04 A08 A09 B01-1 B01-2		Th ppm 5±1.3 Ma 699 38 167 30		0.68	0.2249	<sup>207</sup> Pb/ <sup>235</sup> U 2 δ error	<sup>206</sup> Pb/ <sup>238</sup> U	<sup>206</sup> Pb/ <sup>238</sup> U 2 δ error	KIIO	<sup>206</sup> Pb/ <sup>238</sup> U Age (Ma)	<sup>206</sup> Pb/ <sup>238</sup> U 2 δ error
A02 A04 A08 A09 B01-1	40 12 15 15	699 38 167	1034 383		0 2240	200100				Age (Ma)	
A02 A04 A08 A09 B01-1	40 12 15 15	699 38 167	1034 383		0 2240						
A04 A08 A09 B01-1	12 15 15	38 167	383		0.2249	0.0088	0.03266	0.00114	0.89	207.2	7.1
A08 A09 B01-1	15 15	167			0.2285	0.0110	0.03248	0.00114	0.73	206.0	7.1
A09 B01-1	15			0.40	0.2297	0.0133	0.03258	0.00117	0.62	206.6	7.3
B01-1			517	0.06	0.2271	0.0099	0.03235	0.00111	0.79	205.2	7.0
	20	12	676	0.02	0.2223	0.0095	0.03226	0.00108	0.79	204.7	6.8
	15	18	493	0.04	0.2204	0.0097	0.03240	0.00111	0.78	205.6	6.8
B02	14	14	458	0.03	0.2263	0.0099	0.03260	0.00111	0.78	206.8	6.9
B03	42	1012	957	1.06	0.2302	0.0089	0.03254	0.00108	0.86	206.4	6.8
B06-1	58	1138	1452	0.78	0.2249	0.0084	0.03267	0.00108	0.89	207.3	6.8
B06-2	76	2934	1273	2.30	0.2306	0.0095	0.03254	0.00108	0.81	206.5	6.8
B07	49	1014	1244	0.81	0.2244	0.0085	0.03195	0.00105	0.87	202.8	6.6
B09	19	97	589	0.16	0.2598	0.0126	0.03235	0.00108	0.69	205.2	6.8
B10	37	1096	757	1.45	0.2244	0.0090	0.03206	0.00105	0.81	203.4	6.5
C10	19	23	641	0.04	0.2352	0.0107	0.03215	0.00105	0.72	204.0	6.5
D01-1	55	1484	1233	1.20	0.2234	0.0084	0.03218	0.00102	0.84	204.2	6.4
D01-2	52	1577	1097	1.44	0.2387	0.0096	0.03239	0.00105	0.80	205.5	6.5
D03	37	286	1123	0.25	0.2225	0.0085	0.03231	0.00102	0.83	205.0	6.4
D04-1	21	86	689	0.13	0.2221	0.0092	0.03208	0.00102	0.77	203.5	6.4
D04-2	50	241	1575	0.15	0.2235	0.0081	0.03229	0.00102	0.87	204.9	6.3
D06-1	21	118	652	0.18	0.2238	0.0090	0.03232	0.00102	0.78	205.1	6.3
D06-2	21	128	688	0.19	0.2235	0.0089	0.03217	0.00099	0.77	204.1	6.3
D07	42	961	1035	0.93	0.2386	0.0090	0.03218	0.00099	0.81	204.2	6.2
D09	14	25	451	0.06	0.2295	0.0104	0.03253	0.00102	0.69	206.4	6.3
E03	18	107	594	0.18	0.2285	0.0098	0.03213	0.00099	0.72	203.9	6.1
E06	30	59	894	0.07	0.3444	0.0143	0.03311	0.00099	0.72	210.0	6.2
E07	15	49	474	0.10	0.2283	0.0101	0.03249	0.00096	0.67	206.1	6.1
E08-1	15	127	453	0.28	0.2255	0.0112	0.03217	0.00096	0.60	204.1	6.1
E08-2	43	281	1309	0.21	0.2355	0.0084	0.03241	0.00096	0.83	205.6	5.9
E09-1	36	186	1104	0.17	0.2737	0.0101	0.03256	0.00096	0.80	206.6	6.0
E11-1	78	2948	1370	2.15	0.2257	0.0081	0.03237	0.00093	0.80	205.4	5.9



Table 4Results of whole rock Sr-Nd isotope ratios for the Mashhad granodiorite and their MMEs

Sample No.	Rock types	Sample age (y)	<sup>87</sup> Rb/ <sup>86</sup> Sr	<sup>87</sup> Sr/ <sup>86</sup> Sr	2 StdErr (Abs)	147Sm/144Nd	143Nd/144Nd	$({}^{87}{ m Sr}/{}^{86}{ m Sr})_i$	( <sup>143</sup> Nd/ <sup>144</sup> Nd)T	εNd(t
B1	Granodiorite	2.05E+08	0.579	0.707475	4.3E-06	0.085	0.512372	0.705788	0.512258	-2.3
G3		2.05E+08	0.554	0.707546	5.2E-06	0.081	0.512356	0.705931	0.512247	-2.5
KE16H		2.05E+08	0.330	0.706869	5.7E-06	0.092	0.512377	0.705906	0.512254	-2.3
R2		2.05E+08	1.427	0.709978	7.1E-06	0.093	0.512345	0.705817	0.512220	-3.0
GE1	MMEs	2.05E+08	0.633	0.707594	5.3E-06	0.089	0.512405	0.705749	0.512285	-1.7
KE16E		2.05E+08	0.174	0.706355	6.0E-06	0.096	0.512432	0.705849	0.512303	-1.4
KE25		2.05E+08	0.718	0.708278	6.5E-06	0.090	0.512385	0.706184	0.512265	-2.1
RE1		2.05E+08	1.216	0.709640	5.1E-06	0.095	0.512397	0.706095	0.512269	-2.1
					907					

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## Highlights:

- Mafic microgranular enclaves (MMEs) are commonly present in Mashhad granodiorite, Northeastern Iran
- The mafic and felsic magmas are derived from partial melting of an enriched lithospheric mantle metasomatized by slab fluid
- The Mashhad granodiorite and its MMEs are the result of the mixing between mafic and felsic magmas

magmas