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R. Alheit, Andrea I. Flossmann, H. Pruppacher. A Theoretical Study of the Wet Removal of Atmospheric Pollutants. Part IV: The Uptake and Redistribution of Aerosol Particles through Nucleation and Impaction Scavenging by Growing Cloud Drops and Ice Particles. Journal of the Atmospheric Sciences, 1990, 47 (7), pp.870 - 887. 10.1175/1520-0469(1990)0472.0.CO; hal-01905576

HAL Id: hal-01905576 https://uca.hal.science/hal-01905576

Submitted on 2 Feb 2021

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A Theoretical Study of the Wet Removal of Atmospheric Pollutants. Part IV: The Uptake and Redistribution of Aerosol Particles through Nucleation and Impaction Scavenging by Growing Cloud Drops and Ice Particles

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(Manuscript received 27 March 1989, in final form 16 October 1989)

ABSTRACT

A theoretical model has been formulated which allows the study of the effects of an ice phase on the removal of atmospheric aerosol particles by nucleation and impaction scavenging in a convective cloud. This microphysical model-although in principle applicable to higher dimensional cloud dynamic models-was tested by using a simple parcel model with entrainment as the dynamic framework. The present model has been applied to and numerically evaluated for a convective cloud in which the cloud particles grow via vapor deposition, collision, and coalescence and riming. The computations were carried out for a rural-background aerosol of given particle size distribution. Two different chemical compositions of the aerosol particles and two different modes of ice initiation were considered. Our study shows for in-cloud scavenging: (i) scavenging of aerosol particles by drop nucleation dominates impaction scavenging by drops as well as all other scavenging mechanisms; (ii) scavenging of aerosol particles by nucleation of snow crystals via drop freezing or any other nucleation mechanism dominates impaction scavenging by snow crystals; (iii) impaction scavenging of aerosol particles by snow crystals is, for the conditions studied, the least efficient scavenging mechanism; (iv) scavenging of aerosol particles by riming of graupel is an extremely efficient process due to the prominent scavenging of aerosol particles by nucleation of drops and the efficient uptake of drops by graupel; (v) the transfer of aerosol mass into the ice phase by riming of graupel and by freezing of drops dominates all other transfer mechanisms, (vi) inside mixed icewater clouds the aerosol mass becomes redistributed in such a manner that the main aerosol mass is associated with the main graupel mass if riming is the dominant process of precipitation formation, and with the main water mass if collision and coalescence of drops in the dominant process of precipitation formation.

1. Introduction

In three recent publications (Flossmann et al. 1985, 1987; Flossmann and Pruppacher 1988) we have attempted to shed some light on the removal mechanisms of particulate and gaseous atmospheric sulfur by convective clouds and precipitation. In our first two studies (Flossmann et al. 1985, 1987) we have used our air parcel model with detailed microphysics. From sensitivity studies we were able to point out some interesting facts on the redistribution of the particulate and gaseous pollutants inside a cloud, on the significance of particle removal by nucleation scavenging, and on the relative efficiency with which sulfate aerosol particles and SO₂ gas become incorporated in the cloud water depending on various parameters. In a subsequent study (Flossmann and Pruppacher 1988) we have coupled our particle scavenging model with the two-dimensional model for convective clouds by Clark and coworkers (Clark 1977, 1979; Clark and Gall 1982; Clark and Farley

* Part of a Ph.D. thesis.

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1984; and Hall 1980) and applied this model to a specific case in Hawaii.

All three model studies assumed that the clouds under investigation consisted entirely of liquid water, and that no ice phase was present. Most midlatitude clouds, however, contain ice particles. Thus the question arises as to how the ice phase affects the uptake, the redistribution inside the cloud, and the deposition of the pollutants. We, therefore, consider in the present paper the uptake and redistribution of aerosol particles inside a mixed ice-water cloud.

Considerable knowledge is available in literature which documents the capability of snow crystals to scavenge aerosol particles either by direct pick-up or indirectly by colliding with cloud drops which contain scavenged aerosol particles. Thus, field measurements to this effect have been carried out by Magono et al. (1974), Graedel and Franey (1975), Magono et al. (1975a,b), Magono et al. (1979), Scott (1981), Delmas et al. (1982), Borys et al. (1983), Mayewski and Lyons (1983), Murakami et al. (1983), Raynor and Hayes (1983), and Borys et al. (1988). Laboratory studies which determined the scavenging efficiency of aerosol particles by ice crystals were carried out by Starr and Mason (1966), Vittori and Prodi (1967), Sood and Jackson (1970), Stavitskaya (1972), Vittori (1973),

Knutson et al. (1976), Murakami et al. (1981), Prodi et al. (1981), Prodi (1983), Murakami and Magono (1983), Murakami et al. (1985), and Sauter and Wang (1989). Theoretical studies which attempted to determine the aerosol particle scavenging efficiency of ice crystals were performed by Martin et al. (1980), Wang and Pruppacher (1980), Wang and Chuang (1983), and Wang (1985).

Up to now several theoretical studies have been performed using models which have been formulated for mixed phase ice-water clouds to determine the rate at which these clouds take up aerosol particles and gaseous pollutants (Scott 1982a,b; Easter and Hales 1983; Molenkamp 1983; Hegg et al. 1984; Rutledge et al. 1986). However, in all of these models some form of bulk parameterization was used to describe the microphysical processes. This did not allow following the uptake and redistribution of the scavenged pollutants as the drop and ice particle size distributions evolve.

The present study attempts to broaden the results obtained by the above mentioned bulk parameterized models with the results from our own spectral model which we extended to mixed ice-water clouds. However, the complex nature of the mixed ice-water clouds prevented us from considering all microphysical processes involved in such clouds. Thus, we considered, as a first step, a simple convective cloud in which drops and snow crystals were nucleated, and then grew by diffusion of water vapor. Subsequently, the drops were growing by collision and coalescence and the snow crystals by riming. Aggregation of snow crystals and splintering mechanisms were not considered in this first step of model performance. We also confined ourselves to the in-cloud scavenging processes. The effects of fallout of precipitation particles, and the effects of belowcloud scavenging were not considered. Although these restrictions limit the applicability of our results we, nevertheless, were able to deduce from our model some results of general applicability, such as to determine (1) the relative importance of aerosol particle scavenging via ice nucleation as compared to the uptake of aerosol particles by snow crystals via impaction scavenging, (2) the relative importance of the uptake of aerosol particles by snow crystals as compared to the uptake by drops, (3) the relative importance of the uptake of aerosol particles by snow crystals and graupel via riming as compared to other uptake mechanisms, and (4) the effects of the presence of the ice phase on the redistribution of the scavenged aerosol particles inside the cloud.

2. The present model

a. General

For the present study on the *in-cloud scavenging* of aerosol particles by drops and ice particles we used the air parcel model of Flossmann et al. (1985) which was extended to a mixed phase ice—water cloud. This model allows entrainment of air and gives a detailed descrip-

tion of many pertinent cloud microphysical processes. At present the model includes: the nucleation of drops and snow crystals, their growth by diffusion of water vapor, the growth of the drops by collision and coalescence, the breakup of drops, the growth of the snow crystals by riming, and the growth of the graupel by diffusion of water vapor and riming. Below-cloud scavenging and the effects of ice particle melting and aggregation will be considered at a later time in our two-dimensional cloud model with spectral microphysics previously applied to warm clouds (Flossmann and Pruppacher 1988). This work is presently in preparation. Furthermore, the present model makes the assumption that ice nucleation proceeds foremostly in the temperature range -10° to -20°C where planar type snow crystals are nucleated. Alternatively, ice particles were assumed to result from drop freezing. After nucleation, both ice particles were assumed to grow by vapor diffusion and riming. Field observations support the assumption that in convective type clouds most graupel appear to have originated on planar snow crystals or frozen drops rather than on columnar or needle crystals.

b. Specifics

A complete list of all equations necessary to describe the model would go beyond the allowed length of a paper. Thus, we decided to discuss only those equations which are *new*. For all other equations we shall refer to our previous article (Flossmann et al. 1985). For the nomenclature used we refer to the list of symbols given in the Appendix of this paper.

The time rate of change of the cloud drop number density distribution function, f_d , is given by the relation

$$\frac{\partial f_d(m)}{\partial t} = -\mu f_d(m) U + \frac{\partial f_d(m)}{\partial t} \Big|_{d,\text{coal}} + \frac{\partial f_d(m)}{\partial t} \Big|_{AP,\text{coll},d} + \frac{\partial f_d(m)}{\partial t} \Big|_{con/\text{eva}} + \frac{\partial f_d(m)}{\partial t} \Big|_{act} + \frac{\partial f_d(m)}{\partial t} \Big|_{d,\text{break}} + \frac{\partial f_d(m)}{\partial t} \Big|_{t,\text{rim}} + \frac{\partial f_d(m)}{\partial t} \Big|_{t,\text{r$$

Term 1 on the right-hand side of Eq. (1) is the entrainment term, the terms 2 to 5 are given by Eqs. (5), (10)-(11), (13) and (15), and (19)-(21), respectively, in Flossmann et al. (1985). Term 6 was evaluated after Hall (1980). The last three terms are new and are given by the equations

$$\frac{\partial f_d(m)}{\partial t}\bigg|_{i,\text{rim}} = -f_d(m) \int_0^\infty f_i(a_i) K_i(a_i, a(m)) da_i$$
(2)

$$\frac{\partial f_d(m)}{\partial t} \bigg|_{g, \text{rim}} = -f_d(m) \int_0^\infty f_g(a_g) K_g(a_g, a(m)) da_g$$

$$\frac{\partial f_d(m)}{\partial t} \bigg|_{frd} = -f_d(m) B' \frac{m}{\rho_w} \exp(b'(T_o - T)).$$
(4)

The time rate of change of the mass density distribution function of aerosol particle mass in the drops, g_{APd} , is given by the relation

$$\frac{\partial g_{APd}(m)}{\partial t} = -\mu g_{APd}(m) U + \frac{\partial g_{APd}(m)}{\partial t} \bigg|_{d,\text{coal}} + \frac{\partial g_{APd}(m)}{\partial t} \bigg|_{d,\text{coal}} + \frac{\partial g_{APd}(m)}{\partial t} \bigg|_{d,\text{coal}} + \frac{\partial g_{APd}(m)}{\partial t} \bigg|_{d,\text{break}} + \frac{\partial g_{APd}(m)}{\partial t} \bigg|_{d,\text{break}} + \frac{\partial g_{APd}(m)}{\partial t} \bigg|_{d,\text{trim}} + \frac{\partial g_{APd}(m)}{\partial t} \bigg|_{g,\text{rim}} + \frac{\partial g_{APd}(m)}{\partial t} \bigg|_{frd} \tag{5}$$

Term 1 on the right-hand side of Eq. (5) is the entrainment term, the terms 2 to 5 are given by Eqs. (6)–(7), (12) and (14), and (19)–(21), respectively, in Flossmann et al. (1985). Again, term 6 was evaluated after Hall (1980). The last three terms are new and are given by the equations

$$\frac{\partial g_{APd}(m)}{\partial t}\bigg|_{i,\text{rim}} = -g_{APd}(m) \int_0^\infty f_i(a_i) K_i(a_i, a(m)) da_i \qquad \frac{\partial f_{APa}(m_{AP})}{\partial t}\bigg|_{AP,\text{coll},i} = -f_{APa}(m_{AP})$$
(6)

$$\frac{\partial g_{\text{AP}d}(m)}{\partial t} \bigg|_{g,\text{rim}} = -g_{\text{AP}d}(m) \times \int_0^\infty f_g(a_g) K_g(a_g, a(m)) da_g \quad (7)$$

$$\left. \frac{\partial g_{\text{AP}d}(m)}{\partial t} \right|_{\text{frd}} = -g_{\text{AP}d}(m)B' \frac{m}{\rho_w} \exp(b'(T_o - T)) \tag{8}$$

with

$$K_i(a_i, a) = \pi E_i(a_i, a)(a_i + a)^2 |u_{i,\infty}(a_i) - u_{d,\infty}(a)|$$
(9)

$$K_g(a_g, a) = \pi E_g(a_g, a)(a_g + a)^2$$

 $\times |u_{g,\infty}(a_g) - u_{d,\infty}(a)|.$ (10)

In Eq. (10) $E_g(a_g, a)$ was computed after Hall (1980) and Rasmussen and Heymsfield (1985) who replaced in the scheme of Beard and Grover (1974) the Stokes inertia parameter by the mixed Froud number for the case that $40 \ \mu m \le a_g \le 600 \ \mu m$, and that $a/a_g \le 0.1$.

For all other combinations the collision efficiencies summarized by Hall (1980) were used. For $u_{g,\infty}(a_g)$ we used the terminal velocities for lump graupel III given by Locatelli and Hobbs (1974). The fall speeds of the drops $u_{d,\infty}(a)$ were computed following the scheme of Beard (1976). The collision efficiencies $E_i(a_i, a)$ of snow-crystal plates colliding with drops in Eq. (9) were those given by Martin et al. (1981), and the terminal velocities of unrimed and rimed snow crystals were those of Heymsfield (1972) and Locatelli and Hobbs (1974), respectively.

The time rate of change of the number density distribution function for aerosol particles remaining in the air, f_{APa} , is given by the relation

$$\frac{\partial f_{\text{AP}a}(m_{\text{AP}})}{\partial t} = -\mu (f_{\text{AP}a}(m_{\text{AP}}) - f'_{\text{AP}a}(m_{\text{AP}}))U$$

$$+ \frac{\partial f_{\text{AP}a}(m_{\text{AP}})}{\partial t} \bigg|_{\text{con/eva}} + \frac{\partial f_{\text{AP}a}(m_{\text{AP}})}{\partial t} \bigg|_{\text{AP,coll},d}$$

$$+ \frac{\partial f_{\text{AP}a}(m_{\text{AP}})}{\partial t} \bigg|_{\text{act}} + \frac{\partial f_{\text{AP}a}(m_{\text{AP}})}{\partial t} \bigg|_{\text{AP,coll},i}$$

$$+ \frac{\partial f_{\text{AP}a}(m_{\text{AP}})}{\partial t} \bigg|_{\text{AP,coll},g}. (11)$$

Term 1 on the right-hand side of Eq. (11) again is the entrainment term; the terms 2 to 4 are given by Eqs. (8), (16) and (18), and (19)-(21), respectively, in Flossmann et al. (1985). The last two terms are new and given by the equations

$$\frac{\partial f_{\text{AP}a}(m_{\text{AP}})}{\partial t} \bigg|_{\text{AP,coll},i} = -f_{\text{AP}a}(m_{\text{AP}})$$

$$\times \int_{0}^{\infty} f_{i}(a_{i}) K_{i,AP}(a_{i}, r_{\text{AP}}(m_{\text{AP}})) da_{i} \quad (12)$$

$$\frac{\partial f_{\text{AP}a}(m_{\text{AP}})}{\partial t} \bigg|_{\text{AP,coll},g} = -f_{\text{AP}a}(m_{\text{AP}})$$

$$\times \int_{0}^{\infty} f_{g}(a_{g}) K_{g,\text{AP}}(a_{g}, r_{\text{AP}}(m_{\text{AP}})) da_{g} \quad (13)$$

in which

$$K_{i,AP}(a_i, r_{AP}) = \pi E_{i,AP}(a_i, r_{AP})(a_i + r_{AP})^2 u_{i,\infty}(a_i)$$
(14)

$$K_{g,AP}(a_g, r_{AP}) = \pi E_{g,AP}(a_g, r_{AP})(a_g + r_{AP})^2 u_{g,\infty}(a_g).$$
 (15)

The efficiencies $E_{i,AP}(a_i, r_{AP})$ for the collision of snow crystals with aerosol particles were taken from the theoretical values of Martin et al. (1980) for a water saturated environment supplemented by the experimental values of Murakami et al. (1985). The efficiencies for the collision of graupel particles with aerosol particles were computed on the assumption $E_{g,AP}(a_g, r_{AP}) = E_{d,AP}(a = a_g, r_{AP})$. Values for $E_{d,AP}(a, r_{AP})$

 $r_{\rm AP}$) were derived from a combination of the theoretically computed values of Grover et al. (1977) with the experimental values of Wang et al. (1978), as used in our earlier computations (Flossmann et al., 1985).

The time rate of change of the mass density distribution function for the aerosol particle mass remaining in the air, g_{APa} , is given by the relation

$$\frac{\partial g_{APa}(m_{AP})}{\partial t} = -\mu (g_{APa}(m_{AP}) - g'_{APa}(m_{AP})) U
+ \frac{\partial g_{APa}(m_{AP})}{\partial t} \bigg|_{con/eva} + \frac{\partial g_{APa}(m_{AP})}{\partial t} \bigg|_{AP,coll,d}
+ \frac{\partial g_{APa}(m_{AP})}{\partial t} \bigg|_{act} + \frac{\partial g_{APa}(m_{AP})}{\partial t} \bigg|_{AP,coll,i}
+ \frac{\partial g_{APa}(m_{AP})}{\partial t} \bigg|_{AP,coll,g} . (16)$$

Term 1 on the right-hand side of Eq. (16) again is the entrainment term; the terms 2 to 4 are given by Eqs. (9), (17) and (18), and (19)-(21), respectively, in Flossmann et al. (1985). The last two terms are new and are given by the equations

$$\frac{\partial g_{APa}(m_{AP})}{\partial t} \bigg|_{AP,coll,i} = -g_{APa}(m_{AP})$$

$$\times \int_{0}^{\infty} f_{i}(a_{i})K_{i,AP}(a_{i}, r_{AP}(m_{AP}))da_{i} \quad (17)$$

$$\cdot \frac{\partial g_{APa}(m_{AP})}{\partial t} \bigg|_{AP,coll,g} = -g_{APa}(m_{AP})$$

$$\times \int_{0}^{\infty} f_{g}(a_{g})K_{g,AP}(a_{g}, r_{AP}(m_{AP}))da_{g}. \quad (18)$$

The time rate of change of the number density distribution function for snow crystals, which describes the size distribution of the snow crystals, can be expressed in a form analogous to the function for drops

$$\frac{\partial f_i(a_i)}{\partial t} = -\mu f_i(a_i) U + \frac{\partial f_i(a_i)}{\partial t} \Big|_{\text{act}} + \frac{\partial f_i(a_i)}{\partial t} \Big|_{\text{frd}} + \frac{\partial f_i(a_i)}{\partial t} \Big|_{\text{dep}} + \frac{\partial f_i(a_i)}{\partial t} \Big|_{\text{conv}}.$$
(19)

The time rate of change of the mass density distribution function for snow crystals is given by

$$\frac{\partial g_{i}(a_{i})}{\partial t} = -\mu g_{i}(a_{i})U + \frac{\partial g_{i}(a_{i})}{\partial t} \bigg|_{\text{act}} + \frac{\partial g_{i}(a_{i})}{\partial t} \bigg|_{\text{frd}} + \frac{\partial g_{i}(a_{i})}{\partial t} \bigg|_{\text{frd}} + \frac{\partial g_{i}(a_{i})}{\partial t} \bigg|_{\text{conv}} + \frac{\partial g_{i}(a_{i})}{\partial t} \bigg|_{\text{AP,coll},i} + \frac{\partial g_{i}(a_{i})}{\partial t} \bigg|_{i,\text{rim}}. (20)$$

The time rate of change of the mass density distribution function of aerosol particle mass in the snow crystals is given by

$$\frac{\partial g_{APi}(a_i)}{\partial t} = -\mu g_{APi}(a_i) U + \frac{\partial g_{APi}(a_i)}{\partial t} \bigg|_{act} + \frac{\partial g_{APi}(a_i)}{\partial t} \bigg|_{frd} + \frac{\partial g_{APi}(a_i)}{\partial t} \bigg|_{dep} + \frac{\partial g_{APi}(a_i)}{\partial t} \bigg|_{conv} + \frac{\partial g_{APi}(a_i)}{\partial t} \bigg|_{AP, coll, i} + \frac{\partial g_{APi}(a_i)}{\partial t} \bigg|_{i, tim}. \tag{21}$$

The number density distribution function for graupel, f_g , is written as

$$\frac{\partial f_g(a_g)}{\partial t} = -\mu f_g(a_g) U + \frac{\partial f_g(a_g)}{\partial t} \bigg|_{\text{frd}} + \frac{\partial f_g(a_g)}{\partial t} \bigg|_{\text{conv}} + \frac{\partial f_g(a_g)}{\partial t} \bigg|_{\text{dep}} + \frac{\partial f_g(a_g)}{\partial t} \bigg|_{\text{conv}} + \frac{\partial f_g(a_g)}{\partial t} \bigg|_{\text{AP-coll},g} + \frac{\partial f_g(a_g)}{\partial t} \bigg|_{\text{g,rim}}. \tag{22}$$

The time rate of change of the mass density distribution function for graupel, g_g , is given by

$$\frac{\partial g_{g}(a_{g})}{\partial t} = -\mu g_{g}(a_{g}) U + \frac{\partial g_{g}(a_{g})}{\partial t} \bigg|_{\text{frd}} + \frac{\partial g_{g}(a_{g})}{\partial t} \bigg|_{\text{dep}} + \frac{\partial g_{g}(a_{g})}{\partial t} \bigg|_{\text{dep}} + \frac{\partial g_{g}(a_{g})}{\partial t} \bigg|_{\text{conv}} + \frac{\partial g_{g}(a_{g})}{\partial t} \bigg|_{\text{AP,coll},g} + \frac{\partial g_{g}(a_{g})}{\partial t} \bigg|_{g,\text{rim}}.$$
(23)

The corresponding time rate of change of the mass density distribution function of aerosol particle mass in the graupel, g_{APg} , is

$$\frac{\partial g_{APg}(a_g)}{\partial t} = -\mu g_{APg}(a_g) U + \frac{\partial g_{APg}(a_g)}{\partial t} \bigg|_{frd}
+ \frac{\partial g_{APg}(a_g)}{\partial t} \bigg|_{dep} + \frac{\partial g_{APg}(a_g)}{\partial t} \bigg|_{conv}
+ \frac{\partial g_{APg}(a_g)}{\partial t} \bigg|_{AP,coll,g} + \frac{\partial g_{APg}(a_g)}{\partial t} \bigg|_{g,rim}.$$
(24)

The first term on the right-hand side of Eqs. (19) to (24) expresses the effects of entrainment. In the following we shall discuss briefly each of the remaining terms. Actually, on the right-hand side of Eq. (19) one would expect the additional term $\{[\partial f_i(a_i)]/\partial t\}|_{i,\text{rim}}$ to describe the time rate of change of the number density distribution function of snow crystals due to riming. However, this term is neglected since we assume that the attached drops are uniformly distributed on the basal face of the snow crystal and therefore do not con-

tribute to the growth of the snow crystal along the a-axis, i.e., a change of the number density distribution function. This assumption is justified as studies [e.g. by Pflaum et al. (1978) and Lew et al. (1986a,b)] have shown that riming of crystals contributes most to the basal faces of the crystal plates. Riming and impaction scavenging, however, contribute to the time rate of change of the mass density distribution functions of snow crystals and aerosol particle mass in snow crystals, g_i and g_{APi} , respectively.

The second term on the right-hand side of Eq. (19) can be computed by assuming a certain mechanism for ice nucleation. In our model nucleation of snow crystals was assumed to proceed either by means of drop freezing or by nucleation of snow crystals from the vapor. In the latter case the formulation of Heymsfield et al. (1979) was used

$$N_{IN}(m^{-3}) = 130 \left(\frac{T_o - T}{16}\right)^{3/2} \tag{25}$$

with $T_o = 273.15$ K, based on measurements in Colorado *cumulus congestus*. This equation was assumed to hold up to almost 0° C.

Another relationship for the dependence of the snow crystal concentration on temperature and broadness of the droplet distribution has been given by Hobbs and Rangno (1985) for clouds in the Pacific North West area. They found, that in most of the observed convective clouds with top temperatures $\leq -6^{\circ}$ C the maximum ice particle concentrations had values of roughly 0.1% of the values of the total number concentration of drops larger than 20 μ m in diameter. Since this relationship—when used in our model—gave values for the maximum ice particle concentrations which were in between those computed from Eq. (25) and Eq. (29) for drop freezing, we did not include in this report our results with the Hobbs-Rangno relationship.

The change in time of the snow crystal number concentration function due to *nucleation of snow crystals* from the vapor then is given by

$$\left. \frac{\partial f_i(a_i)}{\partial t} \right|_{\text{act}} = \frac{N_{IN} - (N_i + N_g)}{\Delta t} \tag{26}$$

where Δt is the time step in the numerical computation. From Eq. (26) the corresponding change of the aerosol particle mass in the snow crystals was computed from

$$\left. \frac{\partial g_{APi}(a_i)}{\partial t} \right|_{act} = \frac{\partial f_i(a_i)}{\partial t} \left|_{act} m_{IN}. \right|$$
 (27)

For simplicity it was assumed that the aerosol particles, on which ice nucleation from the vapor occurred, had a typical radius of 1 μ m and a density of 2 g cm⁻³. Such a value is justified considering the field observations of Rosinski et al. (1980) and of Stein and Georgii

(1983). The corresponding change of the mass density distribution function for snow crystals was computed from

$$\frac{\partial g_i(a_i)}{\partial t}\bigg|_{\text{act}} = \frac{\partial f_i(a_i)}{\partial t}\bigg|_{\text{act}} m_i(a_i)$$
 (28)

where m_i is the mass of the planar snow crystal.

When ice nucleation was considered to proceed via drop freezing the change in the number and mass distribution functions of snow crystals and graupel were computed from the formula of Bigg (1953) (see also Pruppacher and Klett 1978, 274–281):

$$\frac{dN_f}{dt} = N_d B' V_d \exp[b'(T_o - T)]$$
 (29)

where N_f is the number of frozen drops and N_d is the number of unfrozen drops of volume V_d , and with $B' = 10^{-4}$ cm⁻³ sec⁻¹ and b' = 0.66 after a suggestion of Orville and Kopp (1977).

For computing the change in the mass-density distribution function of aerosol particle mass in the frozen drops we assumed that the aerosol mass captured by the drop prior to freezing was incorporated into the resulting ice particles. In addition, we arbitrarily assigned all frozen drops with radii smaller than 100 μ m to the category of *snow crystals* and assumed that they would continue to grow into hexagonal plate-like snow crystals (see Pruppacher and Klett 1978, Fig. 16-4). Drops with radii larger than 100 µm we arbitrarily assigned to the category of graupel. With this working hypothesis we were able to compute the change in the number and mass distribution functions for graupel and snow crystals due to drop freezing $\partial f_g(a_g)/\partial t|_{frd}$, $\partial f_i(a_i)/\partial t|_{\text{frd}}$, $\partial g_g(a_g)/\partial t|_{\text{frd}}$ and $\partial g_i(a_i)/\partial t|_{\text{frd}}$, respectively, and the change in the mass distribution function of the aerosol particle mass in the graupel and snow crystals, $\partial g_{APg}(a_g)/\partial t|_{frd}$ and $\partial g_{APi}(a_i)/\partial t|_{frd}$, respectively.

The growth of snow crystals by vapor deposition was computed from

$$\frac{dm_i(a_i)}{dt} \bigg|_{\text{dep}} = \left[4\pi C^*(a_i) s_{v/i}(T) f_{v,L^*} \right] \bigg/ \left[\frac{L_s}{k_a T} \left(\frac{L_s}{R_v T} - 1 \right) + \frac{1}{D_v^* \rho_{\text{sat},i}(T)} \right] \quad (30)$$

where

$$f_{\nu,L^*} = \begin{cases} 1.00 + 0.14\chi^2, & \chi < 1.0 \\ 0.86 + 0.28\chi^2, & \chi \ge 1.0, \end{cases}$$

and where the shape factor C^* was computed according to Pruppacher and Klett [1978, Eqs. (13-71) and (13-87)]. With Eq. (30) it was possible to compute the effect of vapor deposition on the time rate of change of the *number and mass* density distribution functions for *snow crystals*, and on the time rate of change of

the mass density distribution function of aerosol particle mass in the snow crystals from

$$\frac{\partial f_i(a_i)}{\partial t}\bigg|_{t=0} = -\frac{\partial}{\partial \ln a_i} \left\{ \frac{d \ln a_i}{dt} \bigg|_{t=0} f_i(a_i) \right\} \quad (31)$$

$$\frac{\partial g_i(a_i)}{\partial t}\bigg|_{dep} = f_i(a_i) \frac{dm_i(a_i)}{dt}\bigg|_{dep}$$

$$-\frac{\partial}{\partial \ln a_i} \left\{ \frac{d \ln a_i}{dt} \bigg|_{\text{den}} g_i(a_i) \right\} \quad (32)$$

$$\frac{\partial g_{APi}(a_i)}{\partial t}\bigg|_{dep} = -\frac{\partial}{\partial \ln a_i} \left\{ \frac{d \ln a_i}{dt} \bigg|_{dep} g_{APi}(a_i) \right\} \quad (33)$$

with

$$\frac{d \ln a_i}{dt} \bigg|_{\text{dep}} = \frac{1}{a_i} \frac{da_i}{dm_i} \bigg|_{\text{dep}} \frac{dm_i(a_i)}{dt} \bigg|_{\text{dep}}.$$
 (34)

Following Hall and Pruppacher (1976) [their Eqs. (34) and (35)] the vapor mass acquired by a snow crystal was deposited such that the relationship between the thickness h_i and the diameter d_i of the unrimed planar snow crystals followed Auer and Veal (1970) according to

$$h_i = 0.01263 \, d_i^{0.449} \tag{35}$$

where h_i and d_i are given in cm. Together with Eq. (35) we determined

$$\frac{da_i}{dm_i}\bigg|_{\text{dep}} = \left(\frac{\partial m_i}{\partial h}\frac{\partial h}{\partial a_i} + \frac{\partial m_i}{\partial a_i}\right)^{-1}.$$
 (36)

Once the snow crystals have reached a size which allows them to collide with supercooled drops with finite collision efficiency they were assumed to begin to grow by riming while continuing to grow by diffusion of water vapor. Thereby, as mentioned above, the attachment of supercooled drops was assumed to take place only on the ice crystal's basal planes and not on the sides. The continued diffusional growth was assumed to proceed by preserving the relationship $\partial b_i/\partial a_i = \partial h_i/\partial a_i$ according to Eq. (35), where b_i is the thickness of the rimed snow crystal. Once the rimed snow crystals have reached a size where the axis ratio b_i/a_i exceeds a value of 0.25 the ice particle was assumed to be a graupel.

The growth of the graupel by deposition of water vapor was treated analogously to Eq. (30), assuming the graupel to be spherical. Then the time rate-of-change of the number and mass density distribution functions of graupel and mass density distribution function of aerosol particle mass in the graupel is given by

$$\frac{\partial f_g(a_g)}{\partial t}\bigg|_{\text{dep}} = -\frac{\partial}{\partial \ln a_g} \left\{ \frac{d \ln a_g}{dt} \bigg|_{\text{dep}} f_g(a_g) \right\} \quad (37)$$

$$\frac{\partial g_g(a_g)}{\partial t} \bigg|_{\text{dep}} = f_g(a_g) \frac{dm_g(a_g)}{dt} \bigg|_{\text{dep}} - \frac{\partial}{\partial \ln a_g} \left\{ \frac{d \ln a_g}{dt} \bigg|_{\text{dep}} g_g(a_g) \right\} \quad (38)$$

$$\frac{\partial g_{APg}(a_g)}{\partial t}\bigg|_{dep} = -\frac{\partial}{\partial \ln a_g} \left\{ \frac{d \ln a_g}{dt} \bigg|_{dep} g_{APg}(a_g) \right\}$$
(39)

$$\frac{d \ln a_g}{dt} \bigg|_{\text{dep}} = \frac{d \ln a_g}{dm_g(a_g)} \bigg|_{\text{dep}} \frac{dm_g(a_g)}{dt} \bigg|_{\text{dep}} \tag{40}$$

with

$$\left. \frac{d \ln a_g}{d m_g(a_g)} \right|_{\text{dep}} = \frac{1}{3 m_g(a_g)} \left(\frac{\rho_g}{\rho_{g,\text{dep}}} \right)^{1/3}. \tag{41}$$

For these computations we have assumed the density of the water vapor mass deposited on the graupel $\rho_{g,dep}$ = 0.2 g cm⁻³, according to Heymsfield and Pflaum (1985, Fig. 5). The density ρ_g of the graupel was calculated from the mass of the graupel according to

$$m_g(a_g) = \frac{g_g(a_g)}{f_g(a_g)}$$
 (42)

The time rate-of-change of the number density distribution function for graupel due to collision of supercooled drops with the graupel we computed from

$$\frac{\partial f_g(a_g)}{\partial t}\bigg|_{g \text{ rim}} = -\frac{\partial}{\partial \ln a_g} \left\{ \frac{d \ln a_g}{dt} \bigg|_{g \text{ rim}} f_g(a_g) \right\}$$
(43)

$$\frac{d \ln a_g}{dt} \bigg|_{g, \text{rim}} = \frac{d \ln a_g}{dm_g(a_g)} \bigg|_{g, \text{rim}} \frac{dm_g(a_g)}{dt} \bigg|_{g, \text{rim}}$$
(44)

$$\frac{d \ln a_g}{dm_g(a_g)}\bigg|_{q, \text{rim}} = \frac{1}{3m_g(a_g)} \left(\frac{\rho_g}{\rho_{g, \text{accr}}}\right)^{1/3}$$
(45)

$$\left. \frac{dm_g(a_g)}{dt} \right|_{g,\text{rim}} = \int_0^\infty m f_d(m) K_g(a_g, a(m)) dm. \quad (46)$$

The density $\rho_{g,accr}$ of the rime deposited by attachment of supercooled drops on the graupel was computed following the scheme by Heymsfield and Pflaum (1985) where

$$\rho_{g,\text{accr}}(a_g) = \begin{cases} 0.30\psi^{0.44}, \\ T_g \le -5^{\circ}\text{C} \\ 0.30\psi^{0.44}, \\ T_g > -5^{\circ}\text{C} \text{ and } \psi < 1.60 \text{ (47)} \\ e^{-0.03115 - 1.7030\psi + 0.9116\psi^2 - 0.1224\psi^3}, \\ T_g > -5^{\circ}\text{C} \text{ and } \psi \ge 1.60 \end{cases}$$

with

$$\psi = -\frac{\bar{a}u_{\text{impact}}(a_g, \bar{a})}{T_g(a_g)} \tag{48}$$

in which the surface temperature $T_g(a_g)$ of the graupel was computed analogously to the surface temperature of snow crystals growing by diffusion of water vapor (Hall and Pruppacher 1978). The radial impact velocity u_{impact} was calculated using the parameterization scheme of Rasmussen and Heymsfield (1985).

The time rate of change of the mass density distribution functions for graupel, snow crystals, and aerosol particle mass in the graupel and snow crystals, respectively, is found from

$$\left. \frac{\partial g_i(a_i)}{\partial t} \right|_{i,\text{rim}} = f_i(a_i) \int_0^\infty m f_d(m) K_i(a_i, a(m)) dm$$
(49)

$$\frac{\partial g_{APi}(a_i)}{\partial t}\bigg|_{i,\text{rim}} = f_i(a_i) \int_0^\infty g_{APd}(m) K_i(a_i, a(m)) dm$$
(50)

$$\left. \frac{\partial g_g(a_g)}{\partial t} \right|_{g,\text{rim}} = f_g(a_g) \int_0^\infty m f_d(m) K_g(a_g, a(m))$$

$$\times dm - \frac{\partial}{\partial \ln a_g} \left\{ \frac{d \ln a_g}{dt} \bigg|_{g, \text{rim}} g_g(a_g) \right\} \quad (51)$$

$$\frac{\partial g_{APg}(a_g)}{\partial t}\bigg|_{g, \text{rim}} = f_g(a_g) \int_0^\infty g_{APd}(m) K_g(a_g, a(m))$$

$$\times dm - \frac{\partial}{\partial \ln a_g} \left\{ \frac{d \ln a_g}{dt} \Big|_{g, \text{rim}} g_{APg}(a) \right\}.$$
 (52)

The time rate of change of the number and mass density distribution functions of snow crystals and graupel due to impaction of aerosol particles was computed analogously to the collision of supercooled drops on the particles; i.e. $\{[\partial f_g(a_g)]/\partial t\}|_{AP,coll,g}$ from Eqs. (43)–(46) where $K_{g,AP}(a_g, r_{AP})$ now replaces $K_g(a_g, a)$, $\{[\partial g_i(a_i)]/\partial t\}|_{AP,coll,i}$ from Eq. (49) where $K_{i,AP}(a_i, r_{AP})$ replaces $K_i(a_i, a)$, $\{[\partial g_{AP_i}(a_i)]/\partial t\}|_{AP,coll,i}$ from Eq. (50) where $K_{i,AP}(a_i, r_{AP})$ replaces $K_i(a_i, a)$, $\{[\partial g_g(a_g)]/\partial t\}|_{AP,coll,g}$ from Eq. (51) where $K_{g,AP}(a_g, r_{AP})$ replaces $K_g(a_g, a)$, and $\{[\partial g_{AP_g}(a_g)]/\partial t\}|_{AP,coll,g}$ from Eq. (52) where $K_{g,AP}(a_g, r_{AP})$ replaces $K_g(a_g, a)$.

The time rate of change of the number and mass density distribution functions of the snow crystals and graupel due to densely rimed snow crystals converting to graupel were computed from the relations

$$\frac{\partial f_g\left(a_g=a_i\right)}{\partial t} = -\frac{\partial f_i(a_i)}{\partial t}$$
 (53)

$$\frac{\partial g_g\left(a_g=a_i\right)}{\partial t}\bigg|_{\text{conv}} = -\frac{\partial g_i(a_i)}{\partial t}\bigg|_{\text{conv}}$$
 (54)

$$\frac{\partial g_{APg}(a_g = a_i)}{\partial t} \bigg|_{conv} = -\frac{\partial g_{APi}(a_i)}{\partial t} \bigg|_{conv}.$$
 (55)

c. Dynamic framework

The dynamic model used for the present computations was the air-parcel model with entrainment discussed in detail by Flossmann et al. [1985, Eqs. (22)–(31)]. Due to the presence of ice particles in the cloud only their Eqs. (25)–(28) and (30)–(31) remain unchanged. Equations (22)–(24) and (29), however, are now

$$\frac{dT}{dt} = -\frac{gU}{c_{pa}} - \mu(T - T')U + \frac{L_e}{c_{pa}}C_{ph,e} + \frac{L_m}{c_{pa}}C_{ph,m} + \frac{L_s}{c_{pa}}C_{ph,s}$$
 (56)

with

$$C_{ph,e} = \frac{dw_L^*}{dt} \bigg|_{\text{con/eva}} \tag{57}$$

$$C_{ph,m} = \frac{dw_i + dw_g}{dt} \bigg|_{\text{frd/rime/AP-coll}}$$
 (58)

$$C_{ph,e} = \frac{dw_i + dw_g}{dt} \bigg|_{\text{dep}} \tag{59}$$

$$\frac{dw_{v}}{dt} = -\mu(w_{v} - w'_{v})U - C_{ph,e} - C_{ph,s}, \quad (60)$$

and with the entrainment parameter

$$\mu = \frac{0.5}{R_o} \,. \tag{61}$$

d. Initialization and boundary conditions

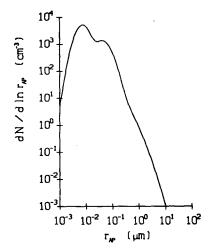
As initial condition for solving the model outlined above we chose the aerosol particle-size distribution as suggested by Jaenicke (1988) for a rural background aerosol, resulting from a superposition of three lognormal distributions

$$f_{APa} \left(\ln r_{AP}, t = 0 \right)$$

$$= \sum_{j=1}^{3} \left[\left(\frac{n_j}{\sqrt{2\pi \log \sigma_j \ln 10}} \right) \exp \left[-0.5 \frac{\left(\log \frac{r_{AP}}{r_j} \right)^2}{\left(\log \sigma_j \right)^2} \right]$$
(62)

TABLE 1. Parameters to characterize the "rural background" aerosol particle distribution given in Eq. (62) after Jaenicke (1988); n_j = total number of aerosol particles per cm³, r_j = geometric mean aerosol particle radius in μ m, σ_j = standard deviation in mode j.

Mode j	n_j	r _j	$\log \sigma_j$
1	6650	0.00739	0.225
2	147	0.02690	0.557
3	1990	0.04190	0.266



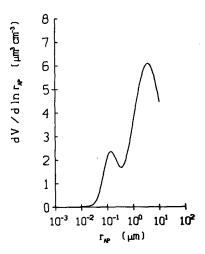


FIG. 1. Number (left-hand side) and volume (right-hand side) distribution of aerosol particles after Jaenicke (1988) [see Eq. (62) and Table 1]; total number of particles = 8786 cm⁻³, total aerosol particle mass = $28.17 \mu g m^{-3}$.

with n_j , r_j and $\log \sigma_j$ given in Table 1 as the starting condition at cloud base. The distribution is plotted in Fig. 1. The aerosol particles were assumed to consist of either 50% or 15% water-soluble material; i.e. $\epsilon = 0.5$ or $\epsilon = 0.15$, where $\epsilon = m_s/m_{AP}$. The aerosol composition $\epsilon = 0.5$ represents a typical value for aerosol particles over Europe, while the composition $\epsilon = 0.15$ represents a value measured by Hobbs et al. (1985, Table 1) over the High Plains in the United States. To take into account the decrease of aerosol particle number concentrations with height in the environmental air we assumed

$$f'_{APa}(\ln r_{AP}, Z) = f_{APa}(\ln r_{AP}, t = 0)e^{-Z/H}$$
 (63)

with a vertical attenuation height of H = 3500 m.

In the present paper we study three model cases with different initialization of the aerosol particle composition and different ice-nucleation mechanisms. In case-1 the aerosol particles were assumed to consist of 50% of water soluble material; i.e., $\epsilon=0.5$, and ice nucleation proceeded via freezing according to Eq. (29) following the condition of Orville and Kopp (1977) with $B'=10^{-4}~{\rm cm}^{-3}~{\rm sec}^{-1}$ and b'=0.66. In case-2 the aerosol particles consisted of 15% of water-soluble substance ($\epsilon=0.15$) while the ice-nucleation mechanism remained the same as in case-1. In case-3 we chose the ice nucleation to proceed according to Eq. (25), and the aerosol particles were set to consist of 50% of soluble material ($\epsilon=0.5$), as in case-1.

The air parcel was assumed to start with an initial updraft speed of 1 m s⁻¹ at an altitude of 750 m above ground at a temperature of 6°C, water saturation and a pressure of 925 hPa. The vertical temperature and humidity distribution in the atmosphere was that measured at Essen (FRG) on 17 April 1986 at 0000 UTC (see Fig. 2). On this day graupel and rain showers were observed in Central Europe.

3. Results

In Fig. 3a-c (case-1) we have plotted the change in time during the parcel's ascent of the drop mass-, snow crystal mass-, graupel mass-distribution function, and of the distribution functions of the aerosol mass in the cloud water, in the snow crystal mass and in the graupel mass for the case that the aerosol particle distribution on which drops were nucleated had a solubility of $\epsilon = 0.5$, and for the case that ice nucleation proceeded via drop freezing [Eq. (29)].

Figure 3a (left-hand side) shows that most of the cloud-water mass remains in the small drops and that collision-coalescence shifts the spectrum only a little, with small precipitation-sized drops developing only very late during the model computation. Accordingly, the main mass of the aerosol captured by nucleation scavenging remains confined to the small drops (Fig. 3a, right-hand side).

Figure 3b (left-hand side) and Table 2 (line 1) show

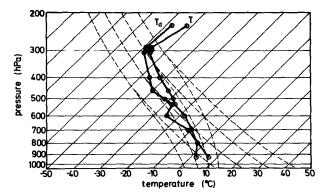


FIG. 2. Vertical temperature and humidity distribution on 17 April 1986, 00 UTC over Essen (FRG). Temperature and pressure at cloud base 6°C and 925 hPa, respectively.

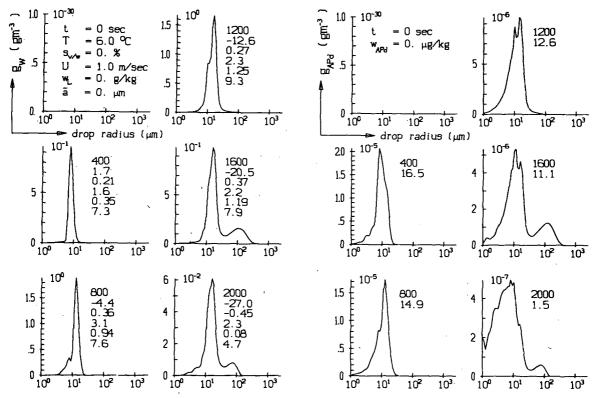


FIG. 3a. Evolution in time of the mass distribution function g_w of the cloud water mass (left-hand side) and of the distribution function g_{APd} of the aerosol mass in the cloud water (right-hand side); for ice nucleation via drop freezing according to Eq. (29) with $B' = 10^{-4}$ cm⁻³ sec⁻¹ and b' = 0.66, and for $\epsilon = 0.5$ (case-1).

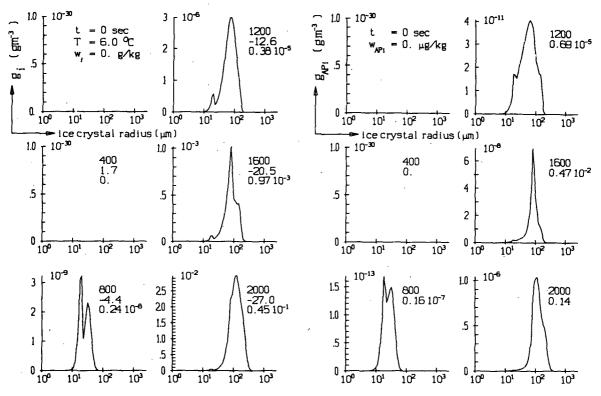


FIG. 3b. Evolution in time of the snow crystal mass distribution function g_i (left-hand side) and of the distribution function g_{APi} of the aerosol mass in the snow crystals (right-hand side) for the conditions in Fig. 3a.

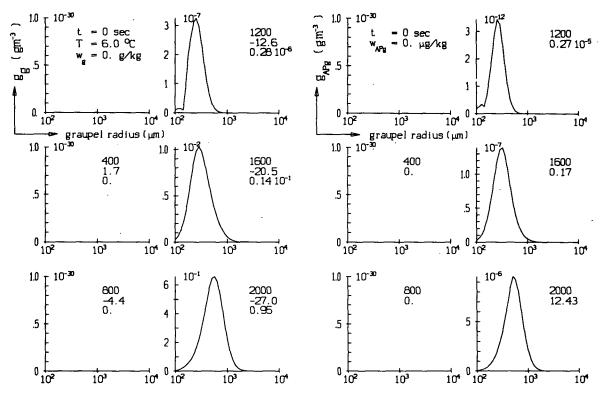


FIG. 3c. Evolution in time of the graupel mass distribution function g_g (left-hand side) and of the distribution function g_{APg} of the aerosol mass in the graupel (right-hand side) for the conditions in Fig. 3a.

that noticeable numbers of snow crystals develop after about 1200 model seconds. This is explicitly shown in Table 2 where we have listed the total number concentration of snow crystals and of graupel as a function of model time, and for the three different cases studied. With the onset of collision and coalescence at about 1200 model seconds a double maximum appears in the distribution of the snow crystals due to the fact that by that time the larger drops freeze more efficiently than the smaller ones. Subsequent growth of the snow crystals by vapor diffusion and riming shifts the dominant maximum to even larger sizes. From Fig. 3b (right-hand side) we notice that the main aerosol mass scavenged by the snow crystals is associated with the main snow crystal mass. This behavior is analogous to the behavior of all-water clouds where the main aerosol mass was found to be associated with the main water mass (Flossmann et al. 1985). Details on the behavior of the different scavenging mechanisms in mixed icewater clouds will be discussed later together with Figs. 6 to 8.

Figure 3c (left-hand side) illustrates the evolution in time of the mass density distribution function of the graupel. We notice from this figure and Table 2 (line 2) that very few graupel develop before 1200 model seconds. However, after this time the concentration of graupel increases rapidly and, after 1600 model seconds, is further increased due to drops larger than 100

 μ m which had formed by collision and coalescence and had frozen by that time. Figure 3c (right-hand side) shows that the main aerosol mass scavenger by graupel is associated with the main graupel mass, as was found for the snow crystal mass.

It is further noteworthy from Fig. 3a (left-hand side) that at 2000 model seconds the cloud air had become subsaturated with respect to water because of the high number concentration of ice particles ($\sim 100l^{-1}$) which had consumed substantial amounts of water vapor. This induced that no new drops become nucleated and that the drop-size spectrum shrank due to drop evaporation (Fig. 3a).

In Figs. 4a-c and Table 2 (case-2) we have illustrated the change in time of the same parameters as in case-

TABLE 2. Total number of snow crystals and of graupel per unit volume at selected model times, and for cases 1, 2 and 3 corresponding to Figs. 3, 4 and 5, respectively.

Case	800 sec	1200 sec	1600 sec	2000 sec
1: N _i	1.1 • 10-4	$3.4 \cdot 10^{-2}$	5	83
$1: N_g$	0	$2 \cdot 10^{-5}$	0.5	16
$2: N_i$	$0.95 \cdot 10^{-4}$	$2.8 \cdot 10^{-2}$	2.3	72
2: N _g	0	$8 \cdot 10^{-5}$	1.4	11
$3: N_i$	0.02	0.09	0.13	0.18
$3: N_g$	0	$4 \cdot 10^{-3}$	0.06	0.08

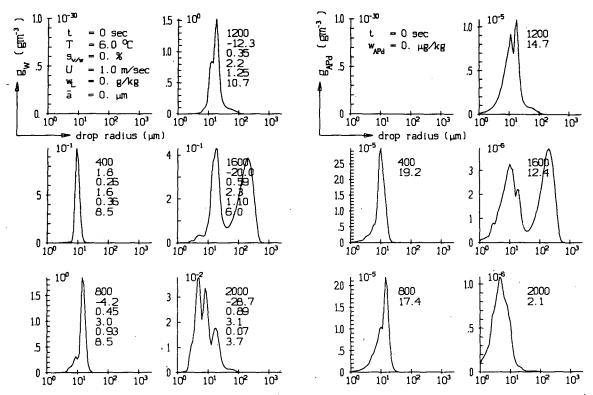


FIG. 4a. Evolution in time of the mass distribution function g_w of the cloud water mass (left-hand side) and of the distribution function g_{APd} of the aerosol mass in the cloud water (right-hand side) for the conditions in Fig. 3a, but for $\epsilon = 0.15$ (case-2).

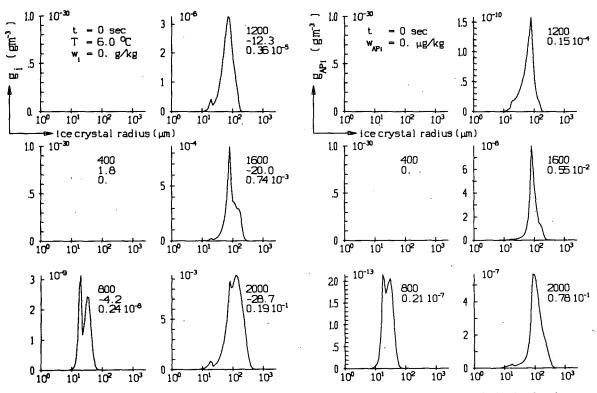


FIG. 4b. Evolution in time of the snow crystal mass distribution function g_i (left-hand side) and of the distribution function g_{APi} of the aerosol mass in the snow crystals (right-hand side) for the conditions in Fig. 4a.

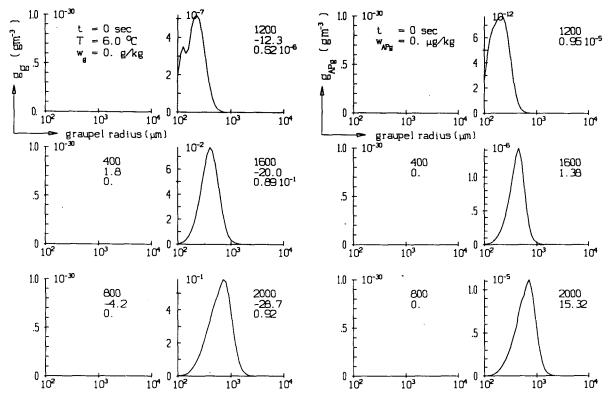


FIG. 4c. Evolution in time of the graupel mass distribution function g_g (left-hand side) and of the distribution function g_{APg} of the aerosol mass in the graupel (right-hand side) for the conditions in Fig. 4a.

1 with the exception that the aerosol particles in the size distribution according to Eq. (62) and Table 1 had a composition given by $\epsilon = 0.15$. The ice nucleation mechanism was the same as in case-1. We notice from these figures that a smaller number of drops became activated in case-2 than in case-1. This is an indirect result of the lower value of ϵ . Although the number of nucleated drops was less because of the smaller ϵ , their sizes were larger, allowing them to grow faster by collision and coalescence and to freeze more readily than in case-1. This in turn caused the graupel to develop earlier but in lower concentration than in case-1 (see Table 2). This resulted in a higher vertical velocity in the cloud due to the faster latent heat release and prevented any subsaturation to occur in the cloud due to the higher water vapor supply.

With regard to the scavenging of aerosol particles the main conclusion for *case-2* is the same as for *case-1*. Again, details on the relative behavior of the different scavenging mechanisms in mixed ice—water clouds will be discussed later together with Figs. 6 to 8.

In Figs. 5a to 5c and Table 2 (case-3) we have plotted the change in time of the same parameters as in case-l with the exception that we chose ice nucleation to proceed via Eq. (25) while the composition of the aerosol particles remained $\epsilon = 0.5$. Comparison with case-l shows that this mode of ice nucleation produces somewhat more snow crystals and graupel in the be-

ginning of the cloud life, but significantly fewer at later times. This results in a significantly less efficient removal of aerosol particle mass by riming of the liquid water and consequently allows a much more efficient growth of the drops by collision and coalescence. In addition, the fewer graupel grow individually to larger sizes than in *case-1*. With regard to the scavenging of aerosol particles, the main conclusion for *case-3* is the same as for *cases 1 and 2*. Again, details on the relative behavior of the different scavenging mechanisms will be discussed together with Figs. 6 to 8.

In Figs. 6, 7 and 8 we have plotted, for the conditions specified for cases 1, 2 and 3, respectively, the change in time of the total amount of aerosol particle mass scavenged by the various mechanisms in the cloud. We notice that over the whole model time in all three cases nucleation scavenging by drops (wAPd,nuc dominates impaction scavenging $(w_{APd,imp})$ by drops as well as all other scavenging modes. With regard to case-1 and case-2, in which ice nucleation takes place by drop freezing, aerosol particle mass is taken up and locked in the ice phase in five different ways: (i) by impaction of aerosol particles on the formed snow crystals $W_{AP/imp}$, (ii) by impaction of aerosol particles on the formed graupel, $w_{APg,imp}$, (iii) by freezing of drops containing previously captured aerosol mass $w_{APd, frd}$, (iv) by riming of snow crystals $w_{APi,rim}$, and (v) by riming of graupel $w_{APg,rim}$. We find from Fig. 6 (for case-1 with ϵ

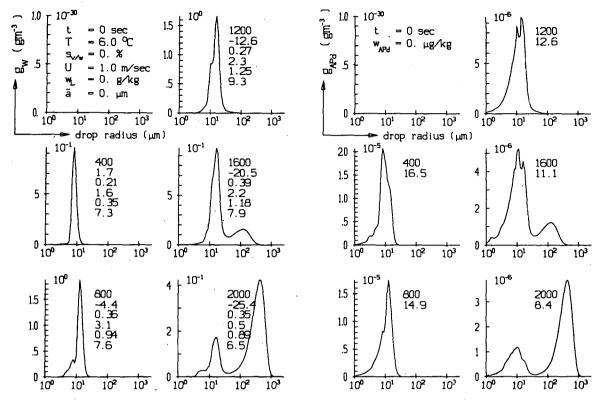


FIG. 5a. Evolution in time of the mass distribution function g_w of the cloud water mass (left-hand side) and of the distribution function g_{APd} of the aerosol mass in the cloud water (right-hand side) for the conditions in Fig. 3a, but for ice nucleation via Eq. (25), and for $\epsilon = 0.5$ (case-3).

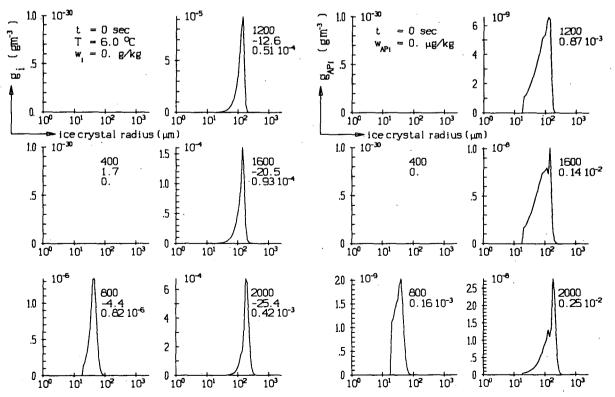


FIG. 5b. Evolution in time of the snow crystal mass distribution function g_i (left-hand side) and of the distribution function g_{APi} of the aerosol mass in the snow crystals (right-hand side) for the conditions in Fig. 5a.

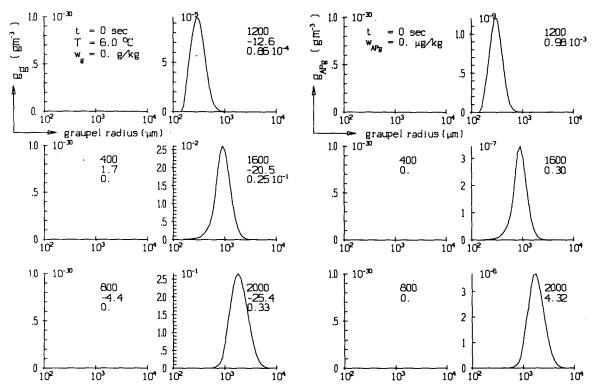


Fig. 5c. Evolution in time of the graupel mass distribution function g_g (left-hand side) and of the distribution function g_{APg} of the aerosol mass in the graupel (right-hand side) for the conditions in Fig. 5a.

= 0.5) that, inside the cloud, impaction scavenging by ice particles ($w_{APi,imp}$, $w_{APg,imp}$) starts fairly late and remains inefficient until the cloud air becomes subsa-

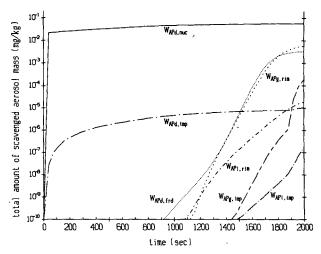


FIG. 6. Time evolution of the total amount of aerosol mass scavenged by drops via nucleation scavenging $(w_{APd,nuc})$, by graupel via riming $(w_{APg,rim})$, by ice crystals via riming $(w_{APd,rim})$, by ice particles via drop freezing $(w_{APd,fral})$, by drops via impaction scavenging $(w_{APd,imp})$, by graupel via impaction scavenging $(w_{APg,imp})$, and by ice crystals via impaction scavenging $(w_{APl,imp})$; for the aerosol particle distribution given by Eq. (62), for ice formation via drop freezing after Eq. (29) and for $\epsilon = 0.5$.

turated (\sim 1900 model seconds). Transfer of aerosol mass from the liquid phase into the ice phase by drop freezing ($w_{APd, frd}$), and scavenging of aerosol mass by riming of graupel ($w_{APg, rim}$) dominates the scavenging of aerosol mass due to the riming of snow crystals ($w_{APi, rim}$) as well as due to the impaction on ice particles ($w_{APi, imp}$, $w_{APg, imp}$). The efficient uptake of aerosol mass by riming of graupel is due to the fact that the graupel

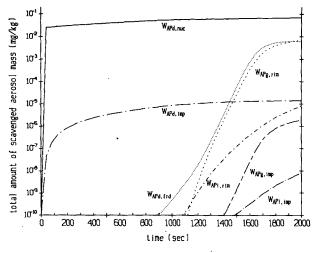


Fig. 7. As in Fig. 6, but for $\epsilon = 0.15$.

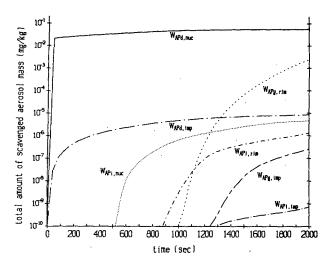


Fig. 8. As in Fig. 6, but for ice formation via Eq. (25) $(w_{APi,nuc})$ and $\epsilon = 0.5$.

acquire drops which themselves have previously, very efficiently scavenged aerosol particles by nucleation scavenging $(w_{APd,nuc})$.

In Fig. 7 we have plotted the same parameters as in Fig. 6 again as a function of model time, but for case-2 with $\epsilon = 0.15$. We notice that, in general, the relative behavior of the different scavenging mechanisms has not changed much, except that $w_{APi,rim}$, $w_{APi,imp}$ and $w_{APg,imp}$ are now somewhat lower than for case-1 because no subsaturation develops in the cloud so that aerosol particles may become activated to drops until the end of the model calculations.

In Fig. 8 (case-3 with $\epsilon = 0.5$) we have plotted the same parameters as in Figs. 6 and 7 as a function of model time, however, additionally, we plotted for the case that the ice phase is not initiated via drop freezing but via Eq. (25). We notice from this figure, that the relative importance of the various mechanisms by which aerosol becomes part of the ice phase in the cloud has now changed. The pronounced increase in the number of ice nuclei with decreasing temperature produces initially a significant increase in the amount of aerosol material scavenged by snow crystals via nucleation scavenging ($w_{APi,nuc}$). But after the time at which graupel have formed in the cloud, $w_{APi,nuc}$ remains much below $w_{APg,rim}$. This is in contrast to case-1 and case-2, where $w_{APd,frd}$ is of the same order-of-magnitude as $w_{APg,rim}$. We further notice that $w_{APi,rim}$ and impaction scavenging of ice particles $(w_{APi,imp}, w_{APg,imp})$ remain rather unimportant, as in cases 1 and 2. In fact, $w_{APg,imp}$ is even less than in cases 1 and 2.

4. Conclusion

From the results presented above the following conclusions may be drawn for aerosol particle scavenging inside convective mixed ice-water clouds: (i) Scavenging of aerosol particles by drop nucleation domi-

nates impaction scavenging by drops as well as all other scavenging mechanisms; (ii) scavenging of aerosol particles by nucleation of snow crystals via drop freezing or any other ice nucleation mechanism dominates impaction scavenging by snow crystals; (iii) impaction scavenging by snow crystals is, for the conditions studied, the least efficient scavenging mechanism; (iv) scavenging of aerosol particles by riming of graupel is an extremely efficient process due to the large efficiency of aerosol particle scavenging by nucleation of drops and the efficient uptake of drops by graupel; (v) the transfer of aerosol mass into the ice phase by riming of graupel and by freezing of drops dominates all other transfer mechanisms; (vi) inside mixed ice-water clouds the aerosol mass becomes redistributed in such a manner that the main aerosol mass is associated with the main, condensed water mass; i.e., it is associated with the graupel mass if riming is the dominant process of precipitation formation and is associated with the liquid water mass if collision and coalescence of drops is the dominant process of precipitation formation. This supports the conclusion of Flossmann et al. (1985) for the redistribution of aerosol mass inside a warm cloud.

Acknowledgments. This research was funded by the Ministry for Research and Technology of the Federal Republic of Germany under Project 325-4007-0704580. The authors of this paper are solely responsible for its content. The authors also acknowledge with gratitude the financial support of the German National Science Foundation under Project SFB 233.

APPENDIX

Definition of Terms

a, \bar{a}	drop radius, mean drop radius of distri-
	bution.
a_{g}	radius of graupel.
a_i .	radius of snow crystal (along a-axis).
b_i	thickness of a rimed snow crystal.
b'	parameter of the BIGG mechanism.
B'	parameter of the BIGG mechanism.
C_{pa}	specific heat of air.
C	rate of water mass flux from air to water,
$C_{ph,e}$, $C_{ph,m}$, $C_{ph,s}$	water to ice, air to ice.
C*	electrostatic capacitance.
d_i	width of an snow crystal = $2a_i$.
D_{v}^{*}	corrected diffusivity of water vapor in
	air.
$E_{\rm g}, E_{\rm g,AP}$	collection efficiency for graupel collect-
5	ing drops, aerosol particles.
$E_i, E_{i,AP}$	collection efficiency for snow crystals
., .,	collecting drops, aerosol particles.
f_d, f_g, f_i	cloud drop, graupel, snow crystal num-
	ber density distribution function.
f_{APa}, f'_{APa}	aerosol particle number density distri-
	bution function in parcel air, envi-
	cloud drop, graupel, snow crystal number density distribution function. aerosol particle number density distri-

ronment.

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$f_{v,L}$ •	ventilation coefficient for snow crystals	$u_{d,\infty}, u_{g,\infty},$	terminal fall velocity of drops, graupel,
	of characteristic length L^* .	$u_{i,\infty}$	snow crystals in air.
g	magnitude of acceleration of gravity.	$u_{\rm impact}$	impact velocity of drops on graupel.
g_g, g_i, g_w	graupel, snow crystal, liquid water mass	U	updraft of the rising air parcel.
087 017 011	density distribution function.	\overline{V}	total volume of aerosol particles per unit
$\alpha - \alpha' -$	aerosol particle mass density distribu-	•	volume.
g_{APa}, g_{APa}'		*	
	tion function for aerosol in parcel air,	w_L, w_L^*	liquid water content of drops, liquid
	environment.		water content of drops plus liquid
$g_{APd}, g_{APg},$	aerosol particle mass density distribu-		water attached to unactivated aerosol
g_{APi}	tion function for aerosol in cloud wa-		particles.
	ter, graupel, snow crystals.	w_g, w_i	solid water content of graupel, snow
h_i	thickness of an unrimed snow crystal.	87 •	crystals.
$\overset{\cdot \cdot \cdot_l}{H}$	vertical attenuation height = 3500 m.	w_v, w'_v	mixing ratio of moist air in the parcel,
		, , , , , , , , , , , , , , , , , , ,	environment.
k_a	thermal conductivity of air.		
K_d	collection kernel for drops colliding with	W_{APd} , W_{APg} ,	aerosol particle mass content of drops,
	drops.	w_{APi}	graupel, snow crystals.
$K_g, K_{g,AP}$	collection kernel for graupel colliding	$W_{APd,frd}$	cumulative amount of aerosol mass
0 0	with drops, aerosol particles.		scavenged by ice particles via drop
$K_i, K_{i,AP}$	collection kernel for snow crystals col-		freezing.
1,1,AF	liding with drops, aerosol particles.	Wini	cumulative amount of aerosol mass
ν		$W_{APd,imp}$,	scavenged by drops via impaction,
K_{AP}	collection kernel for drops colliding with	WAPd,nuc	
	aerosol particles.		nucleation scavenging.
L_e, L_m, L_s	latent heat of evaporation, melting, sub-	$W_{APg,imp}$,	cumulative amount of aerosol mass
	limation.	$w_{APg,rim}$	scavenged by graupel via impaction
L^*	characteristic length of ice particle		scavenging, riming.
	$=\Omega/P$.	$W_{APi,imp}$,	cumulative amount of aerosol mass
m, m'	mass of drop.		scavenged by snow crystals via im-
		$w_{APi,nuc}$,	
m_g, m_i	mass of graupel, snow crystal.	$w_{APi,rim}$	paction, nucleation scavenging, rim-
m_s	mass of water-soluble component in a	_	ing.
	dry aerosol particle.	\boldsymbol{Z}	height above ground level.
$m_{ m AP},m_{IN}$	mass of growing aerosol particle, as-	Δt	numerical time step = 2 sec .
	sumed $IN = 8.3776 \times 10^{-12}$ g.	ϵ	composition of dry aerosol particle
n_i	aerosol particle number concentration		$= m_s/m_{AP}$.
J	of component j in the "rural back-	μ	entrainment parameter.
	ground" aerosol particle distribution.	ν	kinematic viscosity of air.
N_i	total number of snow crystals per unit		density of graupel.
IVi		$ ho_g$	
3.7	volume.	$ ho_{g,\mathrm{accr}}, ho_{g,\mathrm{dep}}$	density of accreted rime, deposited water
N_{g}	total number of graupel per unit vol-		vapor on graupel surface.
	ume.	$ ho_{sat,i}$	saturation vapor density over ice.
N_{IN}	calculated Ice Nuclei concentration ac-	$ ho_w$	density of water.
	cording to FLETCHER.	σ_j	standard deviation of component j in
$N_{\rm Re}, N_{\rm Re,L^{\bullet}}$	Reynolds number.	,	the "rural background" aerosol par-
$N_{\rm Sc}$	Schmidt number.		ticle distribution.
P	perimeter of a body.	ν	$= N_{Sc}^{1/3} \cdot N_{Re,L}^{1/2} \cdot .$
-	•	X	
r_j	mean geometrical radius of component	ψ	$= -\bar{a}u_{\rm impact}(a_g,\bar{a})/T_g.$
	j in the "rural background" aerosol	Ω	total surface of ice particle.
	particle distribution.	$\Phi _{ m act}$	change due to activation of aerosol par-
r_{AP}	radius of aerosol particle.		ticles to drops or snow crystals.
R_v	specific gas constant for water vapor.	$\Phi _{\text{con/eva}}$	change due to condensation and/or
R_P	radius of rising air parcel.	, 55.1, 674	evaporation.
$S_{v/i}$	supersaturation of moist air with respect	$\Phi _{\mathrm{conv}}$	change due to conversion of densely
-0/1	to ice.	≖ I conv	rimed snow crystals to graupel.
٠.	supersaturation of moist air with respect	۸۱	
$S_{v/w}$		$\Phi _{ ext{dep}}$	change due to deposition of water vapor
	to water.	_ 1	on ice particles.
t m	time.	$\Phi _{d, \mathrm{coal}}$	change due to collision and coalescence
T, T'	absolute parcel temperature, environ-		of drops.
	mental temperature in K.	$\Phi _{\mathrm{frd}}$	change due to drop freezing.
T_{g}	surface temperature of graupel in °C.	$\Phi _{g/i, \text{rime}}$	change due to collision of graupel or
T_o^s	melting temperature = $273.15 K$.	. 01-,111110	snow crystals with drops.
-	- -		

- $\Phi|_{\mathsf{AP},\mathsf{coll},d/g/i}$ change due to aerosol particle collection of drops, graupel or snow crystals, respective.
- Φ refers to the time rate change of the number and mass density distribution functions.

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